



**Gesellschaft für Anlagen-
und Reaktorsicherheit
(GRS) mbH**

Forschungsvorhaben

Chemisch-toxische Stoffe in einem Endlager für hochradioaktive Abfälle

Kurztitel: CHEMOTOX

**AP V: Methoden der Stoffbewertung und
Identifizierung relevanter Stoffe, Anlagen**

Braunschweig, 31.08.2009

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Das diesem Bericht zugrunde liegende FE-Vorhaben wurde im Auftrag des Bundesministeriums für Wirtschaft und Technologie unter den Kennzeichen 02E10387, 02E10397 und 02E10407 durchgeführt. Die Verantwortung für den Inhalt dieser Veröffentlichung liegt bei den Autoren.

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Anlage 1: Verwendete thermodynamische Datenbasis


```
#####
Date:          January 2009
Editor:        Moog (GRS)
Serial No:     tdb-hmw-118.1.1.1
Status:        For internal use only
*****
### Source
This parameterfile was created from tdb-hmw-118#1.xls

### Purpose
Calculate reaction of BSK3+3SF with Opalinus, saturated NaCl- and IP21-
solution.

### Data
It contains the most recent values for
- Pitzer parameters
-- for the oceanic system (data0,hmw)
-- for radionuclide (INE)
-- Si, Al: Reardon, E, J, (1990): An ion interaction model for the determination of
   chemical equilibria in cement/water systems, Cement and Concrete Research (20),
   175-192,
-- Zn (KODA), I, Se (ISe-project), and Fe (FeS-project)
- Gibbs energies of formation
-- from the Chemical Thermodynamics series (NEA)
-- calculated from solubility constants (INE)
-- calculated from data in KODA
Some solids were added from a database for Geochemist's Workbench
Some additional elements ...
For detailed information please consult the references below.

There is a strong probability that data in this file are in part inconsistent. One
example is, that carbonate data originate in part from the
HMW-database including Pitzer parameters, others were taken from different sources.

###History of changes
118,1,1,1
First version issued

###References
1   CO2(g)
    Gibbs Free Energy of Formation directly entered
    deltaG_0^f [J mol-1] = -394373 +/- 133, Reference: NEA-9
    V0 [cm3 mol-1] = +/- unknown, Reference:
    Remark: dG^0_f value adopted from NEA-9 deviates from that given in data0,yppf
    (for 298,15 K) by 36 J only
2   H2
    Gibbs Free Energy of Formation directly entered
    deltaG_0^f [J mol-1] = 0 +/- , Reference: by convention
    V0 [cm3 mol-1] = +/- unknown, Reference:
    Remark:
```

3 H2O(g)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -228582 +/- 40, Reference: NEA-6
V0 [cm³ mol⁻¹] = +/- unknown, Reference:
Remark:

4 H2S(g)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -33443 +/- 500, Reference: NEA-6
V0 [cm³ mol⁻¹] = +/- unknown, Reference:
Remark:

5 O2(g)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 0 +/- unknown, Reference: by convention
V0 [cm³ mol⁻¹] = +/- unknown, Reference:
Remark:

6 Hg(g)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 31842 +/- 54, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

7 I2(g)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 19323 +/- 120, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

8 H2Se(g)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 15217 +/- 2003, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

##AQUEOUS

1 H2O
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -237140 +/- 41, Reference: NEA-9
V0 [cm³ mol⁻¹] = 18,06862038 +/- unknown, Reference: HbChemPhys
Remark:

2 H<+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 0 +/- unknown, Reference: by convention
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

3 OH<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -157246,238 +/- unknown, Reaction: 0 = +1OH<->+1H<+>-
1H2O, logK(298,15K) = -13,9967 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

4 O2(aq)
Gibbs Free Energy of Formation calculated from logK for reaction

- deltaG₀^f [J mol⁻¹] = 16543,62 +/- unknown, Reaction: 0 = +1O₂(aq)+1O₂(g),
logK(298,15K) = -2,8983 +/- , Reference:
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 5 (CO₂)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -385879,15 +/- unknown, Reaction: 0 = +1(CO₂)<0>+1H₂O-
1H<+>-1(HCO₃)<->, logK(298,15K) = 6,3374 +/- unknown, Reference: HMW
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 6 (CO₃)<2->
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -527827,834 +/- unknown, Reaction: 0 = +1(CO₃)<2->-
>+1H<+>-1(HCO₃)<->, logK(298,15K) = -10,3393 +/- unknown, Reference: HMW
V₀ [cm³ mol⁻¹] = -4,07 +/- unknown, Reference: KPP1996
Remark:
- 7 (HCO₃)<->
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -586845 +/- 251, Reference: NEA-9
V₀ [cm³ mol⁻¹] = 24,38 +/- unknown, Reference: KPP1996
Remark:
- 8 (HSO₄)<->
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -755297,933 +/- unknown, Reaction: 0 = +1(HSO₄)<->-
1H<+>-1(SO₄)<2->, logK(298,15K) = 1,9786 +/- unknown, Reference: HMW
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 9 (SO₄)<2->
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -744004 +/- 418, Reference: NEA-5
V₀ [cm³ mol⁻¹] = 14,18 +/- unknown, Reference: KPP1996
Remark:
- 10 Ca(CO₃)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -1098621,589 +/- unknown, Reaction: 0 =
+1Ca(CO₃)<0>+1H<+>-1Ca<2+>-1(HCO₃)<->, logK(298,15K) = -7,188 +/- unknown,
Reference: HMW
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 11 Ca<2+>
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -552806 +/- 1050, Reference: NEA-5
V₀ [cm³ mol⁻¹] = -18,03 +/- unknown, Reference: KPP1996
Remark:
- 12 Cl<->
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -131217 +/- 117, Reference: NEA-5
V₀ [cm³ mol⁻¹] = 17,82 +/- unknown, Reference: KPP1996
Remark:
- 13 K<+>
Gibbs Free Energy of Formation directly entered

- ΔG_0^f [J mol⁻¹] = -282510 +/- 116, Reference: NEA-5
V0 [cm³ mol⁻¹] = 9,02 +/- unknown, Reference: KPP1996
Remark:
- 14 Mg(CO₃)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -999918,837 +/- unknown, Reaction: 0 = +1Mg(CO₃)<0>+1H<+>-1(HCO₃)<->-1Mg<2+>, logK(298,15K) = -7,4108 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 15 Mg(OH)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -625108,153 +/- unknown, Reaction: 0 = +1Mg(OH)<+>+1H<+>-1H₂O-1Mg<2+>, logK(298,15K) = -11,8091 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 16 Mg<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -455375 +/- 1335, Reference: NEA-5
V0 [cm³ mol⁻¹] = -21,56 +/- unknown, Reference: KPP1996
Remark:
- 17 Na<+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -261953 +/- 96, Reference: NEA-5
V0 [cm³ mol⁻¹] = -1,2 +/- unknown, Reference: KPP1996
Remark:
- 18 H₂S<0>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -27648 +/- 2115, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 19 HS<->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 12243 +/- 2115, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 20 H₂SiO₄<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1178373,256 +/- unknown, Reaction: 0 = +1H₂SiO₄<2->+2H<+>-1SiO₂<0>-2H₂O, logK(298,15K) = -22,9119 +/- unknown, Reference: INTGRS
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: Grambow-Data0
- 21 H₃SiO₄<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1253139,483 +/- unknown, Reaction: 0 = +1H₃SiO₄<->+1H<+>-1SiO₂<0>-2H₂O, logK(298,15K) = -9,8135 +/- unknown, Reference: INTGRS
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: Grambow-Data0
- 22 SiO₂<0>
Gibbs Free Energy of Formation directly entered

- deltaG_0^f [J mol⁻¹] = -834875,36 +/- unknown, Reference: Solids_j_Mg_TJW_1.xls
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 23 Al(OH)₄<->
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol⁻¹] = -1313595,61 +/- unknown, Reaction: 0 = +1Al(OH)₄<->+4H<+>-1Al<3+>-4H₂O, logK(298,15K) = -22,1567 +/- unknown, Reference: REA90
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 24 Al<3+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol⁻¹] = -491507 +/- 3338, Reference: NEA-5
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 25 Pb<2+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol⁻¹] = -24238 +/- 399, Reference: NEA-5
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 26 Fe<2+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol⁻¹] = -90500 +/- 1000, Reference: PK1995
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: referenced value for 1 atm
- 27 Fe<3+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol⁻¹] = 72395,235 +/- 1100, Reaction: 0 = +1Fe<3+>+0,5H₂O-0,25O₂(g)-1Fe<2+>-1H<+>, logK(298,15K) = -7,7654 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: referenced value for 1 atm
- 28 Am|+II|<2+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol⁻¹] = -376780 +/- 15236, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 29 Am|+III|(CO₃)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol⁻¹] = -1172190,176 +/- 5289, Reaction: 0 = +1Am|+III|(CO₃)<+>-1Am|+III|<3+>-1(CO₃)<2->, logK(298,15K) = 8 +/- 0,4, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 30 Am|+III|(CO₃)₂<->
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol⁻¹] = -1727987,419 +/- 5911, Reaction: 0 = +1Am|+III|(CO₃)₂<->-1Am|+III|<3+>-2(CO₃)<2->, logK(298,15K) = 12,9 +/- 0,6, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 31 Am|+III|(CO₃)₃<3->
Gibbs Free Energy of Formation calculated from logK for reaction

- deltaG_0^f [J mol-1] = -2267802,143 +/- 7521, Reaction: 0 = +1Am|+III|(CO3)3<3->-1Am|+III|<3+>-3(CO3)<2->, logK(298,15K) = 15 +/- 1, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 32 Am|+III|(CO3)5<6->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -3210230 +/- 7919, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 33 Am|+III|(H2PO4)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1752974,128 +/- 5763, Reaction: 0 = +1Am|+III|(H2PO4)<2+>-1Am|+III|<3+>-1(H2PO4)<->, logK(298,15K) = 3 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 34 Am|+III|(HCO3)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1203237,932 +/- 5060, Reaction: 0 = +1Am|+III|(HCO3)<2+>-1Am|+III|<3+>-1(HCO3)<->, logK(298,15K) = 3,1 +/- 0,3, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 35 Am|+III|(OH)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -794740,092 +/- 5546, Reaction: 0 = +1Am|+III|(OH)<2+>+1H<+>-1Am|+III|<3+>-1H2O, logK(298,15K) = -7,2 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 36 Am|+III|(OH)2<+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -986786,555 +/- 6211, Reaction: 0 = +1Am|+III|(OH)2<+>+2H<+>-1Am|+III|<3+>-2H2O, logK(298,15K) = -15,1 +/- 0,7, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 37 Am|+III|(OH)3<0>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1160567,281 +/- 5547, Reaction: 0 = +1Am|+III|(OH)3<0>+3H<+>-1Am|+III|<3+>-3H2O, logK(298,15K) = -26,2 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 38 Am|+III|(SO4)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1361538,541 +/- 4849, Reaction: 0 = +1Am|+III|(SO4)<+>-1Am|+III|<3+>-1(SO4)<2->, logK(298,15K) = 3,3 +/- 0,15, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 39 Am|+III|(SO4)2<->

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2107825,758 +/- 4903, Reaction: 0 = +1Am|+III|(SO₄)₂<->-1Am|+III|<3+>-2(SO₄)<2->, logK(298,15K) = 3,7 +/- 0,15, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 40 Am|+III|<3+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -598698 +/- 4755, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 41 Am|+III|Cl<2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -731284,93 +/- 4759, Reaction: 0 = +1Am|+III|Cl<2+>-1Am|+III|<3+>-1Cl<->, logK(298,15K) = 0,24 +/- 0,03, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 42 Am|+III|Cl<2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -856908,048 +/- 4769, Reaction: 0 = +1Am|+III|Cl<2+>-1Am|+III|<3+>-2Cl<->, logK(298,15K) = -0,74 +/- 0,05, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 43 Am|+IV|<4+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -346358 +/- 8692, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 44 (Am|+V|O₂)(CO₃)<->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1296734,852 +/- 6844, Reaction: 0 = +1(Am|+V|O₂)(CO₃)<->-1(Am|+V|O₂)<+>-1(CO₃)<2->, logK(298,15K) = 5,1 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 45 (Am|+V|O₂)(CO₃)₂<3->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1833695,554 +/- 7746, Reaction: 0 = +1(Am|+V|O₂)(CO₃)₂<3->-1(Am|+V|O₂)<+>-2(CO₃)<2->, logK(298,15K) = 6,7 +/- 0,8, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 46 (Am|+V|O₂)(CO₃)₃<5->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2352390,52 +/- 8514, Reaction: 0 = +1(Am|+V|O₂)(CO₃)₃<5->-1(Am|+V|O₂)<+>-3(CO₃)<2->, logK(298,15K) = 5,1 +/- 1, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 47 (Am|+V|O₂)<+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -739796 +/- 6208, Reference: FZK-INE 002/04

- V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 48 (Am|+VI|O2)<2+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -585801 +/- 5715, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 49 Am(SiO)(OH)3<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1898263,848 +/- 5000, Reaction: 0 =
+1Am(SiO)(OH)3<2+>+1H<+>-1Am|+III|<3+>-1SiO2<0>-2H2O, logK(298,15K) = -1,68 +/-
0,18, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 50 (AsO4)<3->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -647500 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 51 (H2AsO4)<->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -748500 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 52 (HAsO4)<2->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -707100 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 53 H2AsO3<->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -587500 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 54 HAsO2<0>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -402700 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 55 B(OH)3<0>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -969268 +/- unknown, Reference: NTB 02-16
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 56 Ba<2+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -557656 +/- 2582, Reference: NEA-7
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 57 Cd<2+>

Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -77733 +/- 750, Reference: NEA 9
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

58 CdSeO4<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -530175,182 +/- 1652, Reaction: 0 = +1CdSeO4<0>-1Cd<2+>-1SeO4<2->, logK(298,15K) = 2,27 +/- 0,06, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

59 Cm(CO3)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1169292,176 +/- unknown, Reaction: 0 = +1Cm(CO3)<+>-1Cm<3+>-1(CO3)<2->, logK(298,15K) = 8 +/- 0,4, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

60 Cm(CO3)2<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1725089,419 +/- unknown, Reaction: 0 = +1Cm(CO3)2<->-1Cm<3+>-2(CO3)<2->, logK(298,15K) = 12,9 +/- 0,6, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

61 Cm(CO3)3<3->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2264904,143 +/- unknown, Reaction: 0 = +1Cm(CO3)3<3->-1Cm<3+>-3(CO3)<2->, logK(298,15K) = 15 +/- 1, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

62 Cm(CO3)4<5->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2781315,891 +/- unknown, Reaction: 0 = +1Cm(CO3)4<5->-1Cm<3+>-4(CO3)<2->, logK(298,15K) = 13 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

63 Cm(HCO3)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1200339,932 +/- unknown, Reaction: 0 = +1Cm(HCO3)<2+>-1Cm<3+>-1(HCO3)<->, logK(298,15K) = 3,1 +/- 0,3, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

64 Cm(OH)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -791842,092 +/- 6900, Reaction: 0 = +1Cm(OH)<2+>+1H<+>-1Cm<3+>-1H2O, logK(298,15K) = -7,2 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

65 Cm(OH)2<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -983888,555 +/- 7500, Reaction: 0 = +1Cm(OH)2<+>+2H<+>-1Cm<3+>-2H2O, logK(298,15K) = -15,1 +/- 0,7, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 66 Cm(OH)3<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1157669,281 +/- 6900, Reaction: 0 = +1Cm(OH)3<0>+3H<+>-
1Cm<3+>-3H2O, logK(298,15K) = -26,2 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 67 Cm(SO4)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1358640,541 +/- unknown, Reaction: 0 = +1Cm(SO4)<+>-
1Cm<3+>-1(SO4)<2->, logK(298,15K) = 3,3 +/- 0,15, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 68 Cm(SO4)2<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2104927,758 +/- unknown, Reaction: 0 = +1Cm(SO4)2<->-
1Cm<3+>-2(SO4)<2->, logK(298,15K) = 3,7 +/- 0,15, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 69 Cm<3+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -595800 +/- 6300, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 70 CmCl<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -728386,93 +/- unknown, Reaction: 0 = +1CmCl<2+>-
1Cm<3+>-1Cl<->, logK(298,15K) = 0,24 +/- 0,03, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 71 CmCl2<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -854010,048 +/- unknown, Reaction: 0 = +1CmCl2<+>-
1Cm<3+>-2Cl<->, logK(298,15K) = -0,74 +/- 0,05, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 72 Cm(SiO)(OH)3<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1895365,848 +/- unknown, Reaction: 0 =
+1Cm(SiO)(OH)3<2+>+1H<+>-1Cm<3+>-1SiO2<0>-2H2O, logK(298,15K) = -1,68 +/- 0,18,
Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 73 Co<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -53555,2 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 74 Co(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -634004,795 +/- unknown, Reaction: 0 = +1Co(OH)2-1Co<2+>-2H2O+2H<+>, logK(298,15K) = 18,6 +/- 0,4, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 75 Co(OH)3<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -945920,154 +/- unknown, Reaction: 0 = +1Co(OH)3<->-1Co<2+>-3H2O+3H<+>, logK(298,15K) = 31,7 +/- 0,5, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 76 Co(OH)4<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1267082,543 +/- unknown, Reaction: 0 = +1Co(OH)4<2->-1Co<2+>-4H2O+4H<+>, logK(298,15K) = 46,42 +/- 0,43, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 77 Co2(OH)<3+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -400360,46 +/- unknown, Reaction: 0 = +1Co2(OH)<3+>-2Co<2+>-1H2O+1H<+>, logK(298,15K) = 9,83 +/- 0,39, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 78 Co4(OH)4<4+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1333337,116 +/- unknown, Reaction: 0 = +1Co4(OH)4<4+>-4Co<2+>-4H2O+4H<+>, logK(298,15K) = 29,88 +/- 0,35, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 79 CoOH<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -343380,434 +/- unknown, Reaction: 0 = +1CoOH<+>-1Co<2+>+1H<+>-1H2O, logK(298,15K) = 9,23 +/- 0,72, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 80 Cr<3+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -215000 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 81 CrO4<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -727800 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 82 Cs<+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -291456 +/- 553, Reference: NEA 9
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 83 Cu<+>
Gibbs Free Energy of Formation directly entered

- deltaG₀^f [J mol⁻¹] = 50300 +/- unknown, Reference: /BPJ/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 84 Cu₂⁺
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = 65700 +/- 1557, Reference: /BPJ/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 85 Cu(CO₃)₂²⁻
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -931162,828 +/- unknown, Reaction: 0 = +1Cu(CO₃)₂²⁻-
1Cu₂⁺-2(CO₃)²⁻, logK(298,15K) = -10,3 +/- 0,1, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 86 Cu(HCO₃)⁻
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -510642,201 +/- unknown, Reaction: 0 = +1Cu(HCO₃)⁻-
1Cu₂⁺-1(HCO₃)⁻, logK(298,15K) = -1,84 +/- 0,1, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 87 Cu(OH)⁺
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -126061,06 +/- unknown, Reaction: 0 = +1Cu(OH)⁺+1H⁺-
1Cu₂⁺-1H₂O, logK(298,15K) = -7,95 +/- 0,16, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 88 Cu(OH)₂(aq)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -316109,708 +/- unknown, Reaction: 0 =
+1Cu(OH)₂(aq)+2H⁺-1Cu₂⁺-2H₂O, logK(298,15K) = -16,2 +/- 0,2, Reference:
Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 89 Cu(OH)₃⁻
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -493886,064 +/- unknown, Reaction: 0 = +1Cu(OH)₃⁻-
>+3H⁺-1Cu₂⁺-3H₂O, logK(298,15K) = -26,6 +/- 0,09, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 90 Cu(OH)₄²⁻
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -656022,382 +/- unknown, Reaction: 0 = +1Cu(OH)₄²⁻-
>+4H⁺-1Cu₂⁺-4H₂O, logK(298,15K) = -39,74 +/- 0,18, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 91 Cu₂(OH)₃⁺
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -69208,527 +/- unknown, Reaction: 0 =
+1Cu₂(OH)₃⁺+1H⁺-2Cu₂⁺-1H₂O, logK(298,15K) = -6,4 +/- 0,12, Reference:
Koda
V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 92 Cu₂(OH)₂<2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -283345,114 +/- unknown, Reaction: 0 = +1Cu₂(OH)₂<2+>+2H<+>-2Cu<2+>-2H₂O, logK(298,15K) = -10,43 +/- 0,07, Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 93 Cu₃(OH)₄<2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -631020,299 +/- unknown, Reaction: 0 = +1Cu₃(OH)₄<2+>+4H<+>-3Cu<2+>-4H₂O, logK(298,15K) = -21,1 +/- 0,2, Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 94 CuCO₃<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -500657,122 +/- unknown, Reaction: 0 = +1CuCO₃<0>-1Cu<2+>-1(CO₃)<2->, logK(298,15K) = 6,75 +/- 0,03, Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 95 Hg<2+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 164667 +/- 313, Reference: NEA-7
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 96 Hg₂<2+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 153567 +/- 559, Reference: NEA-7
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 97 Hg(OH)<+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -91880,345 +/- unknown, Reaction: 0 = +1Hg(OH)<+>+1H<+>-1Hg<2+>-1H₂O, logK(298,15K) = 3,4 +/- 0,08, Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 98 Hg(OH)₂
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -343747,095 +/- unknown, Reaction: 0 = +1Hg(OH)₂+2H<+>-1Hg<2+>-2H₂O, logK(298,15K) = 5,98 +/- 0,06, Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 99 Hg(OH)₃<->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -667249,782 +/- unknown, Reaction: 0 = +1Hg(OH)₃<->+3H<+>-1Hg<2+>-3H₂O, logK(298,15K) = 21,11 +/- , Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 100 Hg(SeO₃)₂<2->
 Gibbs Free Energy of Formation calculated from logK for reaction

deltaG_0^f [J mol-1] = -644878,058 +/- 6291, Reaction: 0 = +1Hg(SeO3)2<2->-
1HgSeO3-1SeO3<2->, logK(298,15K) = -1,35 +/- 0,15, Reference: NAE-7
V0 [cm3 mol-1] = dummy value 0,001
Remark:

101 HgSe2<2->
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = 65334,758 +/- 5641, Reaction: 0 = +1HgSe2<2->+1H<+>-
1HSe<->-1alpha-HgSe, logK(298,15K) = -12,8 +/- 0,6, Reference: NAE-7
V0 [cm3 mol-1] = dummy value 0,001
Remark:

102 I<->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -51724 +/- 112, Reference: NEA-6
V0 [cm3 mol-1] = dummy value 0,001
Remark:

103 IO3<->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -126338 +/- 779, Reference: NEA-6
V0 [cm3 mol-1] = dummy value 0,001
Remark:

104 Li<+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -292918 +/- 0,109, Reference: NEA 9
V0 [cm3 mol-1] = dummy value 0,001
Remark:

105 Mn<2+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -228100 +/- unknown, Reference: BPJ
V0 [cm3 mol-1] = dummy value 0,001
Remark:

106 MoO4<2->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -853535,058 +/- unknown, Reference: /DHH1976/
V0 [cm3 mol-1] = dummy value 0,001
Remark: DFG calculated from DFG=-204kcal/mol

107 (NH4)<+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -79398 +/- unknown, Reference: /NEA-9/
V0 [cm3 mol-1] = dummy value 0,001
Remark:

108 NO3<->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -110794 +/- unknown, Reference: /NEA-9/
V0 [cm3 mol-1] = dummy value 0,001
Remark:

109 Nd<3+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -672000 +/- unknown, Reference: BPJ
V0 [cm3 mol-1] = dummy value 0,001
Remark:

- 110 Ni<2+>
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -45773 +/- 771, Reference: NEA-6
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 111 HNi(P2O7)<->
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -2064270,34 +/- 4766, Reaction: 0 = +1HNi(P2O7)<->-
1Ni<2+>-1(HP2O7)<3->, logK(298,15K) = 5,14 +/- 0,25, Reference: NEA-6
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 112 Ni(HPO4)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1159167,53 +/- 1820, Reaction: 0 = +1Ni(HPO4)<0>-
1Ni<2+>-1(HPO4)<2->, logK(298,15K) = 3,05 +/- 0,09, Reference: NEA-6
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 113 Ni(P2O7)<2->
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -2031107,213 +/- 4843, Reaction: 0 = +1Ni(P2O7)<2->-
1Ni<2+>-1(P2O7)<4->, logK(298,15K) = 8,73 +/- 0,25, Reference:
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 114 Np|+III|(CO3)3<3->
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = -2185950 +/- 15451, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 115 Np|+III|(OH)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -711190,527 +/- 5922, Reaction: 0 =
+1Np|+III|(OH)<2+>+1H<+>-1H2O-1Np|+III|<3+>, logK(298,15K) = -6,8 +/- 0,3,
Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 116 Np|+III|<3+>
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -512865,218 +/- 5669, Reaction: 0 =
+1Np|+III|<3+>+1H<+>-0,5H2-1Np|+IV|<4+>, logK(298,15K) = 3,695 +/- 0,169,
Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 117 Np|+IV|(CO3)4<4->
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -2812456,343 +/- 8240, Reaction: 0 = +1Np|+IV|(CO3)4<4->-
4(CO3)<2->-1Np|+IV|<4+>, logK(298,15K) = 36,68 +/- 1,06, Reference: FZK-INE
002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 118 Np|+IV|(CO3)5<6->
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -3334176,571 +/- 8425, Reaction: 0 = +1Np|+IV|(CO₃)₅<6->-5(CO₃)<2->-1Np|+IV|<4+>, logK(298,15K) = 35,61 +/- 1,1, Reference: FZK-INE 002/04
- V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 119 Np|+IV|(OH)<3+>
- Gibbs Free Energy of Formation calculated from logK for reaction
- ΔG_0^f [J mol⁻¹] = -732053,423 +/- 5702, Reaction: 0 = +1Np|+IV|(OH)<3+>+1H<+>-1H₂O-1Np|+IV|<4+>, logK(298,15K) = 0,55 +/- 0,2, Reference: FZK-INE 002/04
- V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 120 Np|+IV|(OH)₂(CO₃)₂<2->
- Gibbs Free Energy of Formation calculated from logK for reaction
- ΔG_0^f [J mol⁻¹] = -2118498,665 +/- unknown, Reaction: 0 = +1Np|+IV|(OH)₂(CO₃)₂<2->+2OH<->-1Np|+IV|(OH)₄(am)-2(CO₃)<2->, logK(298,15K) = -11,75 +/- unknown, Reference: FZK-INE 002/04
- V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 121 Np|+IV|(OH)₂<2+>
- Gibbs Free Energy of Formation calculated from logK for reaction
- ΔG_0^f [J mol⁻¹] = -968051,815 +/- 5844, Reaction: 0 = +1Np|+IV|(OH)₂<2+>+2H<+>-2H₂O-1Np|+IV|<4+>, logK(298,15K) = 0,35 +/- 0,3, Reference: FZK-INE 002/04
- V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 122 Np|+IV|(OH)₃<+>
- Gibbs Free Energy of Formation calculated from logK for reaction
- ΔG_0^f [J mol⁻¹] = -1187211,48 +/- 8000, Reaction: 0 = +1Np|+IV|(OH)₃<+>+3H<+>-3H₂O-1Np|+IV|<4+>, logK(298,15K) = -2,8 +/- 1, Reference: FZK-INE 002/04
- V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 123 Np|+IV|(OH)₄(CO₃)<2->
- Gibbs Free Energy of Formation calculated from logK for reaction
- ΔG_0^f [J mol⁻¹] = -1935130,532 +/- unknown, Reaction: 0 = +1Np|+IV|(OH)₄(CO₃)<2->-1(CO₃)<2->-1Np|+IV|(OH)₄(am), logK(298,15K) = -6,5 +/- 0,4, Reference: FZK-INE 002/04
- V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 124 Np|+IV|(OH)₄(CO₃)₂<4->
- Gibbs Free Energy of Formation calculated from logK for reaction
- ΔG_0^f [J mol⁻¹] = -2460675,149 +/- unknown, Reaction: 0 = +1Np|+IV|(OH)₄(CO₃)₂<4->-2(CO₃)<2->-1Np|+IV|(OH)₄(am), logK(298,15K) = -6,9 +/- 0,4, Reference: FZK-INE 002/04
- V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 125 Np|+IV|(OH)₄<0>
- Gibbs Free Energy of Formation calculated from logK for reaction
- ΔG_0^f [J mol⁻¹] = -1392957,245 +/- 8409, Reaction: 0 = +1Np|+IV|(OH)₄<0>+4H<+>-4H₂O-1Np|+IV|<4+>, logK(298,15K) = -8,3 +/- 1,1, Reference: FZK-INE 002/04
- V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 126 $\text{Np|+IV|(SO}_4\text{)}<2+\text{>}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1274870,101 +/- 5809, Reaction: 0 = +1Np|+IV|(SO₄)<2+>+1H<+>-1Np|+IV|<4+>-1(HSO₄)<->, logK(298,15K) = 4,87 +/- 0,15, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 127 $\text{Np|+IV|(SO}_4\text{)}_2<0\text{>}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2042839,89 +/- 6360, Reaction: 0 = +1Np|+IV|(SO₄)₂<0>+2H<+>-1Np|+IV|<4+>-2(HSO₄)<->, logK(298,15K) = 7,09 +/- 0,25, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 128 $\text{Np|+IV|<4+\text{>}$
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -491774 +/- 5586, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 129 $\text{Np|+IV|Cl<3+\text{>}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -631553,064 +/- 5844, Reaction: 0 = +1Np|+IV|Cl<3+>-1Np|+IV|<4+>-1Cl<->, logK(298,15K) = 1,5 +/- 0,3, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 130 $(\text{Np|+V|O}_2)(\text{CO}_3)<->$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1463916,142 +/- 5652, Reaction: 0 = +1(Np|+V|O₂)(CO₃)<->-1(CO₃)<2->-1(Np|+V|O₂)<+>, logK(298,15K) = 4,962 +/- 0,061, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 131 $(\text{Np|+V|O}_2)(\text{CO}_3)_2(\text{OH})<4-\text{>}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2170298,338 +/- 8785, Reaction: 0 = +1(Np|+V|O₂)(CO₃)₂(OH)<4->+1(CO₃)<2->-1(Np|+V|O₂)(CO₃)₃<5->-1OH<->, logK(298,15K) = 3,195 +/- 1,164, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 132 $(\text{Np|+V|O}_2)(\text{CO}_3)_2<3-\text{>}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2000694,187 +/- 5685, Reaction: 0 = +1(Np|+V|O₂)(CO₃)₂<3->-2(CO₃)<2->-1(Np|+V|O₂)<+>, logK(298,15K) = 6,53 +/- 0,1, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 133 $(\text{Np|+V|O}_2)(\text{CO}_3)_3<5-\text{>}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2522642,737 +/- 5733, Reaction: 0 = +1(Np|+V|O₂)(CO₃)₃<5->-3(CO₃)<2->-1(Np|+V|O₂)<+>, logK(298,15K) = 5,5 +/- 0,15, Reference: FZK-INE 002/04

- V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 134 (Np|+V|O2)(HPO4)<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -2020588,726 +/- 5870, Reaction: 0 = +1(Np|+V|O2)(HPO4)<->-1(HPO4)<2->-1(Np|+V|O2)<+>, logK(298,15K) = 2,95 +/- 0,1, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 135 (Np|+V|O2)(HPO4)2<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -3042135,406 +/- 8598, Reaction: 0 = +1(Np|+V|O2)(HPO4)2<2->-2(HPO4)<2->-1(Np|+V|O2)<2+>, logK(298,15K) = 9,5 +/- 1, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 136 (Np|+V|O2)(OH)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -1080404,117 +/- 6902, Reaction: 0 = +1(Np|+V|O2)(OH)<0>+1H<+>-1H2O-1(Np|+V|O2)<+>, logK(298,15K) = -11,3 +/- 0,7, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 137 (Np|+V|O2)(OH)2<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -1247335,192 +/- 6311, Reaction: 0 = +1(Np|+V|O2)(OH)2<->+2H<+>-2H2O-1(Np|+V|O2)<+>, logK(298,15K) = -23,6 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 138 (Np|+V|O2)(PO4)<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -1973212,299 +/- unknown, Reaction: 0 = +1(Np|+V|O2)(PO4)<2->-1(PO4)<3->-1(Np|+V|O2)<+>, logK(298,15K) = 7 +/- 0,2, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 139 (Np|+V|O2)(SO4)<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -1654280,539 +/- 5850, Reaction: 0 = +1(Np|+V|O2)(SO4)<->-1(SO4)<2->-1(Np|+V|O2)<+>, logK(298,15K) = 0,44 +/- 0,27, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 140 (Np|+V|O2)<+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol-1] = -907765 +/- 5628, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 141 (Np|+VI|O2)(CO3)<0>
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -1376965,792 +/- 6617, Reaction: 0 = +1(Np|+VI|O₂)(CO₃)<0>-1(CO₃)<2->-1(Np|+VI|O₂)<2+>, logK(298,15K) = 9,32 +/- 0,61, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 142 (Np|+VI|O₂)(CO₃)₂<2->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1945868,701 +/- 7033, Reaction: 0 = +1(Np|+VI|O₂)(CO₃)₂<2->-2(CO₃)<2->-1(Np|+VI|O₂)<2+>, logK(298,15K) = 16,516 +/- 0,729, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 143 (Np|+VI|O₂)(CO₃)₃<4->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -248992,732 +/- 5759, Reaction: 0 = +1(Np|+VI|O₂)(CO₃)₃<4->-1(Np|+V|O₂)(CO₃)₃<5->, logK(298,15K) = -5,72 +/- 0,095, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 144 (Np|+VI|O₂)(H₂PO₄)<+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1952041,702 +/- 6491, Reaction: 0 = +1(Np|+VI|O₂)(H₂PO₄)<+>-1(H₂PO₄)<->-1(Np|+VI|O₂)<2+>, logK(298,15K) = 3,32 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 145 (Np|+VI|O₂)(HPO₄)<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1927313,865 +/- 7067, Reaction: 0 = +1(Np|+VI|O₂)(HPO₄)<0>-1(HPO₄)<2->-1(Np|+VI|O₂)<2+>, logK(298,15K) = 6,2 +/- 0,7, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 146 (Np|+VI|O₂)(OH)<+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1003967,982 +/- 6062, Reaction: 0 = +1(Np|+VI|O₂)(OH)<+>+1H<+>-1H₂O-1(Np|+VI|O₂)<2+>, logK(298,15K) = -5,1 +/- 0,4, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 147 (Np|+VI|O₂)(SO₄)<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1558665,38 +/- 5641, Reaction: 0 = +1(Np|+VI|O₂)(SO₄)<0>-1(SO₄)<2->-1(Np|+VI|O₂)<2+>, logK(298,15K) = 3,28 +/- 0,06, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 148 (Np|+VI|O₂)(SO₄)₂<2->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2310774,801 +/- 5705, Reaction: 0 = +1(Np|+VI|O₂)(SO₄)₂<2->-2(SO₄)<2->-1(Np|+VI|O₂)<2+>, logK(298,15K) = 4,7 +/- 0,1, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 149 (Np|+VI|O2)<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -795939 +/- 5615, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 150 (Np|+VI|O2)2(CO3)(OH)3<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2814482,093 +/- 14665, Reaction: 0 =
+1(Np|+VI|O2)2(CO3)(OH)3<->+5CO2(g)+2H2O-7H<+>-2(Np|+VI|O2)(CO3)3<4->,
logK(298,15K) = 49,166 +/- 1,586, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 151 (Np|+VI|O2)2(OH)2<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2030368,572 +/- 11294, Reaction: 0 =
+1(Np|+VI|O2)2(OH)2<2+>+2H<+>-2H2O-2(Np|+VI|O2)<2+>,
logK(298,15K) = -6,27 +/- 0,21, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 152 (Np|+VI|O2)3(CO3)6<6->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5839277,766 +/- 19185, Reaction: 0 =
+1(Np|+VI|O2)3(CO3)6<6->+3(CO3)<2->-3(Np|+VI|O2)(CO3)3<4->,
logK(298,15K) = -8,272 +/- 1,447, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 153 (Np|+VI|O2)3(OH)5<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3475795,309 +/- 16893, Reaction: 0 =
+1(Np|+VI|O2)3(OH)5<+>+5H<+>-5H2O-3(Np|+VI|O2)<2+>,
logK(298,15K) = -17,12 +/- 0,22, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 154 (Np|+VI|O2)Cl<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -929439,217 +/- 5699, Reaction: 0 = +1(Np|+VI|O2)Cl<+>-
1Cl<->-1(Np|+VI|O2)<2+>,
logK(298,15K) = 0,4 +/- 0,17, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 155 (H2P2O7)<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -2027117 +/- 4445, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 156 (H2PO4)<->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1137152 +/- 1567, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 157 (H3P2O7)<->

- Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -2039960 +/- 4362, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 158 (HP2O7)<3->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1989158 +/- 4482, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 159 (HPO4)<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1095985 +/- 1567, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 160 (P2O7)<4->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1935503 +/- 4563, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 161 (PO4)<3->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1025491 +/- 1576, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 162 H3PO4<0>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1149367 +/- 1576, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 163 H4P2O7<0>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -2045668 +/- 3299, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 164 Pu|+III|(OH)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -776738,505 +/- 3187, Reaction: 0 =
+1Pu|+III|(OH)<2+>+1H<+>-1H2O-1Pu|+III|<3+>, logK(298,15K) = -6,9 +/- 0,3,
Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 165 Pu|+III|(SO4)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1345298,456 +/- 4599, Reaction: 0 =
+1Pu|+III|(SO4)<+>+1H<+>-1(HSO4)<->-1Pu|+III|<3+>, logK(298,15K) = 1,93 +/-
0,61, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 166 Pu|+III|(SO4)2<->
Gibbs Free Energy of Formation calculated from logK for reaction

ΔG_0^f [J mol⁻¹] = -2099511,861 +/- 5766, Reaction: 0 = +1Pu|+III|(SO₄)₂<->+2H<+>-2(HSO₄)<->-1Pu|+III|<3+>, logK(298,15K) = 1,74 +/- 0,76, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

167 Pu|+III|<3+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -578984 +/- 2688, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: in forming reaction given by INE species PuCl₃*6H₂O(c) is not defined

168 Pu|+IV|(CO₃)₄<4->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2800496,916 +/- 7013, Reaction: 0 = +1Pu|+IV|(CO₃)₄<4->-4(CO₃)<2->-1Pu|+IV|<4+>, logK(298,15K) = 37 +/- 1,1, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

169 Pu|+IV|(CO₃)₅<6->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3320618,893 +/- 7261, Reaction: 0 = +1Pu|+IV|(CO₃)₅<6->-5(CO₃)<2->-1Pu|+IV|<4+>, logK(298,15K) = 35,65 +/- 1,13, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

170 Pu|+IV|(OH)<3+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -718552,826 +/- 2936, Reaction: 0 = +1Pu|+IV|(OH)<3+>+1H<+>-1H₂O-1Pu|+IV|<4+>, logK(298,15K) = 0,6 +/- 0,2, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

171 Pu|+IV|(OH)₂(CO₃)₂<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2112076,04 +/- unknown, Reaction: 0 = +1Pu|+IV|(OH)₂(CO₃)₂<2->+2OH<->-1Pu|+IV|(OH)₄(am)-2(CO₃)<2->, logK(298,15K) = -12,09 +/- unknown, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

172 Pu|+IV|(OH)₂<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -955692,826 +/- 3203, Reaction: 0 = +1Pu|+IV|(OH)₂<2+>+2H<+>-2H₂O-1Pu|+IV|<4+>, logK(298,15K) = 0,6 +/- 0,3, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

173 Pu|+IV|(OH)₃<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1176279,502 +/- 3542, Reaction: 0 = +1Pu|+IV|(OH)₃<+>+3H<+>-3H₂O-1Pu|+IV|<4+>, logK(298,15K) = -2,3 +/- 0,4, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

- 174 Pu|+IV|(OH)4(CO3)<2->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1925511,404 +/- unknown, Reaction: 0 = +1Pu|+IV|(OH)4(CO3)<2->-1Pu|+IV|(OH)4(am)-1(CO3)<2->, logK(298,15K) = -7,4 +/- 0,3, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 175 Pu|+IV|(OH)4(CO3)2<4->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2461330,497 +/- unknown, Reaction: 0 = +1Pu|+IV|(OH)4(CO3)2<4->-1Pu|+IV|(OH)4(am)-2(CO3)<2->, logK(298,15K) = -6 +/- 0,3, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 176 Pu|+IV|(OH)4<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1378029,637 +/- 3936, Reaction: 0 = +1Pu|+IV|(OH)4<0>+4H<+>-4H2O-1Pu|+IV|<4+>, logK(298,15K) = -8,5 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 177 Pu|+IV|(SO4)<2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1261312,423 +/- 3270, Reaction: 0 = +1Pu|+IV|(SO4)<2+>+1H<+>-1(HSO4)<->-1Pu|+IV|<4+>, logK(298,15K) = 4,91 +/- 0,22, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 178 Pu|+IV|(SO4)2<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2029567,613 +/- 4225, Reaction: 0 = +1Pu|+IV|(SO4)2<0>+2H<+>-2(HSO4)<->-1Pu|+IV|<4+>, logK(298,15K) = 7,18 +/- 0,32, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 179 Pu|+IV|<4+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -477988 +/- 2705, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 180 Pu|+IV|Cl<3+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -619479,477 +/- 3204, Reaction: 0 = +1Pu|+IV|Cl<3+>-1Cl<->-1Pu|+IV|<4+>, logK(298,15K) = 1,8 +/- 0,3, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 181 (Pu|+V|O2)(CO3)<->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1409701,54 +/- 3002, Reaction: 0 = +1(Pu|+V|O2)(CO3)<->-1(CO3)<2->-1(Pu|+V|O2)<+>, logK(298,15K) = 5,12 +/- 0,14, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 182 (Pu|+V|O2)(CO3)3<5->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2464814,944 +/- 6096, Reaction: 0 =
+1(Pu|+V|O2)(CO3)3<5->-3(CO3)<2->-1(Pu|+V|O2)<+>, logK(298,15K) = 5,025 +/-
0,92, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 183 (Pu|+V|O2)(OH)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1034249,272 +/- unknown, Reaction: 0 =
+1(Pu|+V|O2)(OH)<0>+1H<+>-1H2O-1(Pu|+V|O2)<+>, logK(298,15K) = -9,73 +/-
unknown, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 184 (Pu|+V|O2)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -852648,528 +/- 2868, Reaction: 0 =
+1(Pu|+V|O2)<+>+1H<+>-0,5H2-1(Pu|+VI|O2)<2+>, logK(298,15K) = 15,819 +/- 0,09,
Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 185 Pu|+V|(H3PO4)<4+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1641054,303 +/- 3569, Reaction: 0 =
+1Pu|+V|(H3PO4)<4+>-1H3PO4<0>-1Pu|+IV|<4+>, logK(298,15K) = 2,4 +/- 0,3,
Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 186 (Pu|+VI|O2)(CO3)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1344407,24 +/- 4032, Reaction: 0 =
+1(Pu|+VI|O2)(CO3)<0>-1(CO3)<2->-1(Pu|+VI|O2)<2+>, logK(298,15K) = 9,5 +/- 0,5,
Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 187 (Pu|+VI|O2)(CO3)2<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1901916,896 +/- 4088, Reaction: 0 =
+1(Pu|+VI|O2)(CO3)2<2->-2(CO3)<2->-1(Pu|+VI|O2)<2+>, logK(298,15K) = 14,7 +/-
0,5, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 188 (Pu|+VI|O2)(CO3)3<4->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2448581,271 +/- 4180, Reaction: 0 =
+1(Pu|+VI|O2)(CO3)3<4->-3(CO3)<2->-1(Pu|+VI|O2)<2+>, logK(298,15K) = 18 +/-
0,5, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 189 (Pu|+VI|O2)(OH)<+>
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -968098,765 +/- 4013, Reaction: 0 = +1(Pu|+VI|O₂)(OH)<+>+1H<+>-1H₂O-1(Pu|+VI|O₂)<2+>, logK(298,15K) = -5,5 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 190 (Pu|+VI|O₂)(OH)₂<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1161286,836 +/- 9015, Reaction: 0 = +1(Pu|+VI|O₂)(OH)₂<0>+2H<+>-2H₂O-1(Pu|+VI|O₂)<2+>, logK(298,15K) = -13,2 +/- 1,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 191 (Pu|+VI|O₂)(SO₄)<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1525650,184 +/- 3072, Reaction: 0 = +1(Pu|+VI|O₂)(SO₄)<0>-1(Pu|+VI|O₂)<2+>-1(SO₄)<2->, logK(298,15K) = 3,38 +/- 0,2, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 192 (Pu|+VI|O₂)(SO₄)₂<2->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2275476,388 +/- 3156, Reaction: 0 = +1(Pu|+VI|O₂)(SO₄)₂<2->-1(Pu|+VI|O₂)<2+>-2(SO₄)<2->, logK(298,15K) = 4,4 +/- 0,2, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 193 (Pu|+VI|O₂)<2+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -762353 +/- 2821, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 194 (Pu|+VI|O₂)₂(OH)₂<2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1956175,68 +/- 8026, Reaction: 0 = +1(Pu|+VI|O₂)₂(OH)₂<2+>+2H<+>-2H₂O-2(Pu|+VI|O₂)<2+>, logK(298,15K) = -7,5 +/- 1, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 195 (Pu|+VI|O₂)₃(OH)₅<+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3354602,516 +/- unknown, Reaction: 0 = +1(Pu|+VI|O₂)₃(OH)₅<+>+5H<+>-5H₂O-3(Pu|+VI|O₂)<2+>, logK(298,15K) = -20,7 +/- 0,6, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 196 (Pu|+VI|O₂)Cl<+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -894882,85 +/- 2829, Reaction: 0 = +1(Pu|+VI|O₂)Cl<+>-1Cl<->-1(Pu|+VI|O₂)<2+>, logK(298,15K) = 0,23 +/- 0,03, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

- 197 (Pu|+VI|O2)Cl2<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1018222,751 +/- 3308, Reaction: 0 = +1(Pu|+VI|O2)Cl2<0>-2Cl<->-1(Pu|+VI|O2)<2+>, logK(298,15K) = -1,15 +/- 0,3, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 198 Ra<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -561500 +/- unknown, Reference: NTB 02-16
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 199 RaCl<+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -692146 +/- unknown, Reference: NTB 02-16
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 200 RaCO3<0>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1103688 +/- unknown, Reference: NTB 02-16
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 201 RaOH<+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -721584 +/- unknown, Reference: NTB 02-16
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 202 RaSO4<0>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1321201 +/- unknown, Reference: NTB 02-16
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: 0
- 203 H2Se<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = 21495,847 +/- 2003, Reaction: 0 = +1H2Se<0>-1H2Se(g), logK(298,15K) = -1,1 +/- 0,01, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 204 H2SeO3<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -425178,782 +/- 849, Reaction: 0 = +1H2SeO3<0>+2H2-1Se(cr)-3H2O, logK(298,15K) = -50,147 +/- 0,147, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 205 HSe<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = 43471,811 +/- 2024, Reaction: 0 = +1HSe<->+1H<+>-1H2Se<0>, logK(298,15K) = -3,85 +/- 0,05, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

- 206 HSeO3<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -410109,549 +/- 1166, Reaction: 0 = +1HSeO3<->+1H<+>-
1H2SeO3<0>, logK(298,15K) = -2,64 +/- 0,14, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 207 HSeO4<->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -449474 +/- 1312, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 208 Se<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 128600 +/- 3000, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 209 Se2<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 112670 +/- 6294, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 210 Se3<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 100590 +/- 9198, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 211 Se4<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 97580 +/- 12149, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 212 SeO3<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -362390,312 +/- 1756, Reaction: 0 = +1SeO3<2->+1H<+>-
1HSeO3<->, logK(298,15K) = -8,36 +/- 0,23, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 213 SeO4<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -439484,925 +/- 1431, Reaction: 0 = +1SeO4<2->+1H<+>-
1HSeO4<->, logK(298,15K) = -1,75 +/- 0,1, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 214 ZnSeO4<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -599017,298 +/- 1493, Reaction: 0 = +1ZnSeO4<0>-1Zn<2+>-
1SeO4<2->, logK(298,15K) = 2,16 +/- 0,06, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 215 Sn(OH)6<2->

- Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1299000 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 216 Sn<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -27240 +/- unknown, Reference:
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 217 Sn<4+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 2720 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 218 SnO3<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -574965 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 219 Sr<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -563864 +/- 781, Reference: NEA-9
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 220 SrSO4<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1320939,418 +/- unknown, Reaction: 0 = +1SrSO4<0>-
1Sr<2+>-1(SO4)<2->, logK(298,15K) = 2,29 +/- unknown, Reference: /PSI/NAGRA/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 221 (TcO)(OH)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -345378,096 +/- 9009, Reaction: 0 = +1(TcO)(OH)<+>+1H2O-
1H<+>-1(TcO)(OH)2<0>, logK(298,15K) = 2,5 +/- 0,3, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 222 (TcO)(OH)2<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -568247,989 +/- 8845, Reaction: 0 =
+1(TcO)(OH)2<0>+0,6H2O-1TcO2*1,6H2O(s), logK(298,15K) = -8,4 +/- 0,5,
Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 223 (TcO)(OH)3<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -743170,324 +/- 9135, Reaction: 0 = +1(TcO)(OH)3<-
>+1H<+>-1H2O-1(TcO)(OH)2<0>, logK(298,15K) = -10,9 +/- 0,4, Reference: FZK-INE
002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 224 (TcO)<2+>

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -116800,16 +/- unknown, Reaction: 0 = +1(TcO)₂+2H₂O-
2H<+>-1(TcO)(OH)₂<0>, logK(298,15K) = 4 +/- unknown, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 225 (TcO₄)<->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -637406 +/- 7616, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 226 (TcO₄)<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -575759 +/- 8133, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 227 Tc(CO₃)(OH)₂<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -968899,836 +/- 9010, Reaction: 0 = +1Tc(CO₃)(OH)₂<0>-
1CO₂(g)-1(TcO)(OH)₂<0>, logK(298,15K) = 1,1 +/- 0,3, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 228 Tc(CO₃)(OH)₃<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1158663,082 +/- 9486, Reaction: 0 = +1Tc(CO₃)(OH)₃<-
>+1H<+>-1CO₂(g)-1H₂O-1(TcO)(OH)₂<0>, logK(298,15K) = -7,2 +/- 0,6, Reference:
FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 229 H₂TeO₄
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -550857 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 230 HTeO₂<+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -216540 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 231 HTeO₃<->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -436590 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 232 HTeO₄<->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -515753 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 233 TeO₃<2->
Gibbs Free Energy of Formation directly entered

- ΔG_0^f [J mol⁻¹] = -392420 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 234 TeO₄<2->
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -456424 +/- unknown, Reference: /BPJ 1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 235 Th(CO₃)₅<6->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3506989,191 +/- unknown, Reaction: 0 = +1Th(CO₃)₅<6->+4OH<->-1Th(OH)₄(am)-5(CO₃)<2->, logK(298,15K) = -18,4 +/- unknown, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 236 Th(HPO₄)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1874789,555 +/- unknown, Reaction: 0 = +1Th(HPO₄)<2+>-1Th<4+>-1(HPO₄)<2->, logK(298,15K) = 13 +/- 1, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 237 Th(OH)(CO₃)₄<5->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3172939,069 +/- unknown, Reaction: 0 = +1Th(OH)(CO₃)₄<5->+3OH<->-1Th(OH)₄(am)-4(CO₃)<2->, logK(298,15K) = -12 +/- 0,1, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 238 Th(OH)<3+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -929182,306 +/- 5600, Reaction: 0 = +1Th(OH)<3+>+1H<+>-1H₂O-1Th<4+>, logK(298,15K) = -2,2 +/- 0,2, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 239 Th(OH)₂(CO₃)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1717590,788 +/- unknown, Reaction: 0 = +1Th(OH)₂(CO₃)<0>+2OH<->-1Th(OH)₄(am)-1(CO₃)<2->, logK(298,15K) = -17,1 +/- 0,2, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 240 Th(OH)₂(CO₃)₂<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2281379,291 +/- unknown, Reaction: 0 = +1Th(OH)₂(CO₃)₂<2->+2OH<->-1Th(OH)₄(am)-2(CO₃)<2->, logK(298,15K) = -10,8 +/- 0,1, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 241 Th(OH)₂<2+>
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -1144631,744 +/- 6400, Reaction: 0 = +1Th(OH)₂<2+>+2H<+>-2H₂O-1Th<4+>, logK(298,15K) = -6 +/- 0,6, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 242 Th(OH)₃(CO₃)<->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1919359,76 +/- unknown, Reaction: 0 = +1Th(OH)₃(CO₃)<->+1OH<->-1Th(OH)₄(am)-1(CO₃)<2->, logK(298,15K) = -9,3 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 243 Th(OH)₃<+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1353231,53 +/- 7900, Reaction: 0 = +1Th(OH)₃<+>+3H<+>-3H₂O-1Th<4+>, logK(298,15K) = -11 +/- 1, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 244 Th(OH)₄(CO₃)<2->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2089163,692 +/- unknown, Reaction: 0 = +1Th(OH)₄(CO₃)<2->-1Th(OH)₄(am)-1(CO₃)<2->, logK(298,15K) = -7,1 +/- 0,2, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 245 Th(OH)₄<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1553269,252 +/- 7900, Reaction: 0 = +1Th(OH)₄<0>+4H<+>-4H₂O-1Th<4+>, logK(298,15K) = -17,5 +/- 1, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 246 Th(SO₄)<2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1491985,125 +/- unknown, Reaction: 0 = +1Th(SO₄)<2+>-1Th<4+>-1(SO₄)<2->, logK(298,15K) = 7,6 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 247 Th(SO₄)₂<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2258821,296 +/- unknown, Reaction: 0 = +1Th(SO₄)₂<0>-1Th<4+>-2(SO₄)<2->, logK(298,15K) = 11,6 +/- unknown, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 248 Th(SO₄)₃<2->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3007391,73 +/- unknown, Reaction: 0 = +1Th(SO₄)₃<2->-1Th<4+>-3(SO₄)<2->, logK(298,15K) = 12,4 +/- unknown, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 249 Th<4+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -704600 +/- 5400, Reference: FZK-INE 002/04

- V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 250 Th2(OH)2<6+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -1850944,157 +/- 10800, Reaction: 0 =
+1Th2(OH)2<6+>+2H<+>-2Th<4+>-2H2O, logK(298,15K) = -5,7 +/- unknown, Reference:
FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 251 Th4(OH)12<4+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -5511675,259 +/- 21600, Reaction: 0 =
+1Th4(OH)12<4+>+12H<+>-4Th<4+>-12H2O, logK(298,15K) = -26,7 +/- unknown,
Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 252 Th4(OH)8<8+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -4599075,929 +/- 21600, Reaction: 0 =
+1Th4(OH)8<8+>+8H<+>-4Th<4+>-8H2O, logK(298,15K) = -20,4 +/- unknown,
Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 253 Ti(OH)4<0>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol-1] = 0 +/- unknown, Reference: gesetzt
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 254 (UO2)(H2PO4)(H3PO4)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -3260703,27 +/- 3659, Reaction: 0 =
+1(UO2)(H2PO4)(H3PO4)<+>+1H<+>-1(U|+VI|O2)<2+>-2H3PO4<0>, logK(298,15K) = 1,65
+/- 0,11, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 255 (UO2)(H2PO4)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -2108311,008 +/- 2378, Reaction: 0 =
+1(UO2)(H2PO4)<+>+1H<+>-1(U|+VI|O2)<2+>-1H3PO4<0>, logK(298,15K) = 1,12 +/-
0,06, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 256 (UO2)(H2PO4)2<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -3254938,147 +/- 3659, Reaction: 0 =
+1(UO2)(H2PO4)2<0>+2H<+>-1(U|+VI|O2)<2+>-2H3PO4<0>, logK(298,15K) = 0,64 +/-
0,11, Reference: FZK-INE 002/04
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 257 (UO2)(H3PO4)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -2106256,112 +/- 2504, Reaction: 0 = +1(UO₂)(H₃PO₄)<2+>-1H₃PO₄<0>-1(U|+VI|O₂)<2+>, logK(298,15K) = 0,76 +/- 0,15, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 258 (UO₂)(HAsO₄)<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1707133,881 +/- 4717, Reaction: 0 = +1(UO₂)(HAsO₄)<0>-1(AsO₄)<3->-1(U|+VI|O₂)<2+>-1H<+>, logK(298,15K) = 18,76 +/- 0,31, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 259 (UO₂)(HPO₄)<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2089862,229 +/- 2777, Reaction: 0 = +1(UO₂)(HPO₄)<0>-1(HPO₄)<2->-1(U|+VI|O₂)<2+>, logK(298,15K) = 7,24 +/- 0,26, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 260 U|+III|<3+>
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -476473 +/- 1810, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 261 U|+IV|(CO₃)₄<4->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2841637,796 +/- 5958, Reaction: 0 = +1U|+IV|(CO₃)₄<4->+1(CO₃)<2->-1U|+IV|(CO₃)₅<6->, logK(298,15K) = 1,12 +/- 0,25, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 262 U|+IV|(CO₃)₅<6->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3363072,622 +/- 5772, Reaction: 0 = +1U|+IV|(CO₃)₅<6->-5(CO₃)<2->-1U|+IV|<4+>, logK(298,15K) = 34 +/- 0,9, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 263 U|+IV|(OH)<3+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -763917,657 +/- 1798, Reaction: 0 = +1U|+IV|(OH)<3+>+1H<+>-1H₂O-1U|+IV|<4+>, logK(298,15K) = -0,54 +/- 0,06, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 264 U|+IV|(OH)₂(CO₃)₂<2->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2141972,076 +/- unknown, Reaction: 0 = +1U|+IV|(OH)₂(CO₃)₂<2->+2OH<->-2(CO₃)<2->-1U|+IV|(OH)₄(am), logK(298,15K) = -12,11 +/- unknown, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 265 U|+IV|(OH)₂<2+>

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -997861,153 +/- 6000, Reaction: 0 = +1U|+IV|(OH)2<2+>+2H<+>-2H2O-1U|+IV|<4+>, logK(298,15K) = -1,1 +/- 1, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 266 U|+IV|(OH)3<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1214452,199 +/- 6000, Reaction: 0 = +1U|+IV|(OH)3<+>+3H<+>-3H2O-1U|+IV|<4+>, logK(298,15K) = -4,7 +/- 1, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 267 U|+IV|(OH)4<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1421414,919 +/- 8189, Reaction: 0 = +1U|+IV|(OH)4<0>-4OH<->-1U|+IV|<4+>, logK(298,15K) = 46 +/- 1,4, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 268 U|+IV|(SO4)<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1311422,921 +/- 2113, Reaction: 0 = +1U|+IV|(SO4)<2+>-1U|+IV|<4+>-1(SO4)<2->, logK(298,15K) = 6,58 +/- 0,19, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 269 U|+IV|(SO4)2<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2077859,529 +/- 2262, Reaction: 0 = +1U|+IV|(SO4)2<0>-2(SO4)<2->-1U|+IV|<4+>, logK(298,15K) = 10,51 +/- 0,2, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 270 U|+IV|<4+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -529860 +/- 1765, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 271 U|+IV|Cl<3+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -670894,833 +/- 1918, Reaction: 0 = +1U|+IV|Cl<3+>-1U|+IV|<4+>-1Cl<->, logK(298,15K) = 1,72 +/- 0,13, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 272 (U|+V|O2)(CO3)3<5->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2584175,399 +/- 2943, Reaction: 0 = +1(U|+V|O2)(CO3)3<5->-3(CO3)<2->-1(U|+V|O2)<+>, logK(298,15K) = 6,95 +/- 0,36, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 273 (U|+V|O2)<+>

Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -961021 +/- 1752, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

274 (U|+VI|O₂)(CO₃)<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1537116,779 +/- 1799, Reaction: 0 =
 +1(U|+VI|O₂)(CO₃)<0>-1(U|+VI|O₂)<2+>-1(CO₃)<2->, logK(298,15K) = 9,94 +/- 0,03,
 Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

275 (U|+VI|O₂)(CO₃)₂<2->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2103017,257 +/- 1982, Reaction: 0 =
 +1(U|+VI|O₂)(CO₃)₂<2->-1(U|+VI|O₂)<2+>-2(CO₃)<2->, logK(298,15K) = 16,61 +/-
 0,09, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

276 (U|+VI|O₂)(CO₃)₃<4->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2660698,155 +/- 2116, Reaction: 0 =
 +1(U|+VI|O₂)(CO₃)₃<4->-1(U|+VI|O₂)<2+>-3(CO₃)<2->, logK(298,15K) = 21,84 +/-
 0,04, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

277 (U|+VI|O₂)(H₂AsO₄)<+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1725399,618 +/- 4582, Reaction: 0 =
 +1(U|+VI|O₂)(H₂AsO₄)<+>-1(AsO₄)<3->-1(U|+VI|O₂)<2+>-2H<+>, logK(298,15K) =
 21,96 +/- 0,24, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

278 (U|+VI|O₂)(H₂AsO₄)₂<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2484606,014 +/- 8283, Reaction: 0 =
 +1(U|+VI|O₂)(H₂AsO₄)₂<0>-2(AsO₄)<3->-1(U|+VI|O₂)<2+>-4H<+>, logK(298,15K) =
 41,53 +/- 0,2, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

279 (U|+VI|O₂)(OH)<+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1159723,776 +/- 2221, Reaction: 0 =
 +1(U|+VI|O₂)(OH)<+>+1H<+>-1(U|+VI|O₂)<2+>-1H₂O, logK(298,15K) = -5,25 +/- 0,24,
 Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

280 (U|+VI|O₂)(OH)₂<0>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1357478,281 +/- 1794, Reaction: 0 =
 +1(U|+VI|O₂)(OH)₂<0>+2H<+>-1(U|+VI|O₂)<2+>-2H₂O, logK(298,15K) = -12,15 +/-
 0,07, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 281 (U|+VI|O2)(OH)3<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1548383,135 +/- 2969, Reaction: 0 = +1(U|+VI|O2)(OH)3<->+3H<+>-1(U|+VI|O2)<2+>-3H2O, logK(298,15K) = -20,25 +/- 0,42, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 282 (U|+VI|O2)(OH)4<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1718910,276 +/- 4260, Reaction: 0 = +1(U|+VI|O2)(OH)4<2->+4H<+>-1(U|+VI|O2)<2+>-4H2O, logK(298,15K) = -31,92 +/- 0,33, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 283 (U|+VI|O2)(PO4)<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2053559,405 +/- 2504, Reaction: 0 = +1(U|+VI|O2)(PO4)<->-1(PO4)<3->-1(U|+VI|O2)<2+>, logK(298,15K) = 13,23 +/- 0,15, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 284 (U|+VI|O2)(SO4)<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1714535,335 +/- 1800, Reaction: 0 = +1(U|+VI|O2)(SO4)<0>-1(U|+VI|O2)<2+>-1(SO4)<2->, logK(298,15K) = 3,15 +/- 0,02, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 285 (U|+VI|O2)(SO4)2<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2464190,297 +/- 1978, Reaction: 0 = +1(U|+VI|O2)(SO4)2<2->-1(U|+VI|O2)<2+>-2(SO4)<2->, logK(298,15K) = 4,14 +/- 0,07, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 286 (U|+VI|O2)(SO4)3<4->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3201801,289 +/- 3054, Reaction: 0 = +1(U|+VI|O2)(SO4)3<4->-1(U|+VI|O2)<2+>-3(SO4)<2->, logK(298,15K) = 3,02 +/- 0,38, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 287 (U|+VI|O2)<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -952551 +/- 1747, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
- Remark:
- 288 (U|+VI|O2)11(CO3)6(OH)12<2->
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -16698985,903 +/- 22383, Reaction: 0 =
 $+1(U|+VI|O_2)_{11}(CO_3)_6(OH)_{12} <-> +24H <+> -18H_2O - 11(U|+VI|O_2) <2+> - 6CO_2(g)$,
 $\log K(298,15K) = -72,5 +/- 2$, Reference: FZK-INE 002/04
 V_0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 289 (U|+VI|O₂)₂(CO₃)(OH)₃ <->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3139525,108 +/- 4517, Reaction: 0 =
 $+1(U|+VI|O_2)_2(CO_3)(OH)_3 <-> +5H <+> -4H_2O - 2(U|+VI|O_2) <2+> - 1CO_2(g)$, $\log K(298,15K) =$
 $-19,01 +/- 0,5$, Reference: FZK-INE 002/04
 V_0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 290 (U|+VI|O₂)₂(OH)₃ <3+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2126830,285 +/- 6693, Reaction: 0 =
 $+1(U|+VI|O_2)_2(OH)_3 <3+> + 1H <+> - 1H_2O - 2(U|+VI|O_2) <2+>$, $\log K(298,15K) = -2,7 +/- 1$,
 Reference: FZK-INE 002/04
 V_0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 291 (U|+VI|O₂)₂(OH)₂ <2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2347302,8 +/- 3503, Reaction: 0 =
 $+1(U|+VI|O_2)_2(OH)_2 <2+> + 2H <+> - 2H_2O - 2(U|+VI|O_2) <2+>$, $\log K(298,15K) = -5,62 +/-$
 $0,04$, Reference: FZK-INE 002/04
 V_0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 292 (U|+VI|O₂)₃(CO₃)₆ <6->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5716385,697 +/- 8096, Reaction: 0 =
 $+1(U|+VI|O_2)_3(CO_3)_6 <6-> - 6(CO_3) <2-> - 3(U|+VI|O_2) <2+>$, $\log K(298,15K) = -54 +/- 1$,
 Reference: FZK-INE 002/04
 V_0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 293 (U|+VI|O₂)₃(OH)₄ <2+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3738287,292 +/- 5517, Reaction: 0 =
 $+1(U|+VI|O_2)_3(OH)_4 <2+> + 4H <+> - 4H_2O - 3(U|+VI|O_2) <2+>$, $\log K(298,15K) = -11,9 +/-$
 $0,3$, Reference: FZK-INE 002/04
 V_0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 294 (U|+VI|O₂)₃(OH)₅ <+>
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3954592,936 +/- 5291, Reaction: 0 =
 $+1(U|+VI|O_2)_3(OH)_5 <+> + 5H <+> - 5H_2O - 3(U|+VI|O_2) <2+>$, $\log K(298,15K) = -15,55 +/-$
 $0,12$, Reference: FZK-INE 002/04
 V_0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 295 (U|+VI|O₂)₃(OH)₇ <->
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4333834,024 +/- 6958, Reaction: 0 =
 $+1(U|+VI|O_2)_3(OH)_7 <-> + 7H <+> - 7H_2O - 3(U|+VI|O_2) <2+>$, $\log K(298,15K) = -32,2 +/-$
 $0,8$, Reference: FZK-INE 002/04
 V_0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 296 (U|+VI|O2)3O(OH)2(HCO3)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4100695,252 +/- 5973, Reaction: 0 = +1(U|+VI|O2)3O(OH)2(HCO3)<+>+5H<+>-4H2O-3(U|+VI|O2)<2+>-1CO2(g), logK(298,15K) = -17,5 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 297 (U|+VI|O2)4(OH)7<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5345177,864 +/- 9029, Reaction: 0 = +1(U|+VI|O2)4(OH)7<+>+7H<+>-7H2O-4(U|+VI|O2)<2+>, logK(298,15K) = -21,9 +/- 1, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 298 (U|+VI|O2)Cl<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1084738,367 +/- 1755, Reaction: 0 = +1(U|+VI|O2)Cl<+>-1(U|+VI|O2)<2+>-1Cl<->, logK(298,15K) = 0,17 +/- 0,02, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 299 (U|+VI|O2)Cl2<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1208706,153 +/- 2885, Reaction: 0 = +1(U|+VI|O2)Cl2<0>-1(U|+VI|O2)<2+>-2Cl<->, logK(298,15K) = -1,1 +/- 0,4, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 300 (U|+VI|O2)SiO(OH)3<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2251203,561 +/- 2172, Reaction: 0 = +1(U|+VI|O2)SiO(OH)3<+>+1H<+>-1(U|+VI|O2)<2+>-1SiO2<0>-2H2O, logK(298,15K) = -1,84 +/- 0,1, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 301 V<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -218000 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 302 V<3+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -251300 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 303 VO<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -446400 +/- unknown, Reference: /BPJ 1985/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 304 VO2<+>

- Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 0 +/- unknown, Reference: gesetzt
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 305 Zn(OH)<+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -341532,68 +/- unknown, Reaction: 0 = +1Zn(OH)<+>+1H<+>-
1Zn<2+>-1H₂O, logK(298,15K) = -7,5 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 306 Zn(OH)₂<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -527871,099 +/- unknown, Reaction: 0 = +1Zn(OH)₂<0>+2H<+>-
1Zn<2+>-2H₂O, logK(298,15K) = -16,4 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 307 Zn(OH)₃<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -697656,195 +/- unknown, Reaction: 0 = +1Zn(OH)₃<->
>+3H<+>-1Zn<2+>-3H₂O, logK(298,15K) = -28,2 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 308 Zn(OH)₄<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -860020,836 +/- unknown, Reaction: 0 = +1Zn(OH)₄<2->
>+4H<+>-1Zn<2+>-4H₂O, logK(298,15K) = -41,3 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 309 Zn<2+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -147203 +/- 254, Reference: NEA 9
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 310 ZnCO₃<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -702429,439 +/- unknown, Reaction: 0 = +1ZnCO₃<0>+1H<+>-
1Zn<2+>-1(HCO₃)<->, logK(298,15K) = -5,5393 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 311 Zr(OH)<3+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -767476,275 +/- , Reaction: 0 = +1Zr(OH)<3+>+1H<+>-
1Zr<4+>-1H₂O, logK(298,15K) = 0,32 +/- 0,22, Reference: NEA-8
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 312 Zr(OH)₂<2+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1008383,583 +/- , Reaction: 0 = +1Zr(OH)₂<2+>+2H<+>-
1Zr<4+>-2H₂O, logK(298,15K) = 0,98 +/- 1,06, Reference: NEA-8
V0 [cm³ mol⁻¹] = dummy value 0,001

Remark:

313 Zr(OH)₄<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1464569,087 +/- , Reaction: 0 = +1Zr(OH)₄<0>+4H<+>-
1Zr<4+>-4H₂O, logK(298,15K) = -2,19 +/- 1,7, Reference: NEA-8
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

314 Zr<4+>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -528509,701 +/- , Reaction: 0 = +1Zr<4+>+2H₂O-4H<+>-
1ZrO₂(monoclinic), logK(298,15K) = -7 +/- 1,6, Reference: NEA-8
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

315 Ag<+>
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 77096 +/- 156, Reference: NEA-8
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

316 AgCl<0>
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -72337,429 +/- unknown, Reaction: 0 = +1AgCl<0>-
1AgCl(cr), logK(298,15K) = -6,50863830616573 +/- 0,0969100130080562, Reference:
/FRI1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

317 AgCl₂<->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -214437,259 +/- unknown, Reaction: 0 = +1AgCl₂<->-
1AgCl(cr)-1Cl<->, logK(298,15K) = -4,60205999132796 +/- 0,0969100130080562,
Reference: /FRI1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

318 AgCl₃<2->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -345101,093 +/- unknown, Reaction: 0 = +1AgCl₃<2->-
1AgCl(cr)-2Cl<->, logK(298,15K) = -4,69897000433602 +/- 0,0969100130080562,
Reference: /FRI1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

319 AgCl₄<3->
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -468275,829 +/- unknown, Reaction: 0 = +1AgCl₄<3->-
1AgCl(cr)-3Cl<->, logK(298,15K) = -6,10790539730952 +/- 0,0969100130080562,
Reference: /FRI1985/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

##MIXED-SOLID-PHASES
Automatic generation of references not yet implemented
##SOLIDS
1 Ca(CO₃)_Aragonite
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -1127551,091 +/- unknown, Reaction: 0 = +1Ca(CO₃)_Aragonite+1H<+>-1Ca<2+>-1(HCO₃)<->, logK(298,15K) = -2,1198 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 34,15 +/- unknown, Reference: YPF
 Remark:
- 2 Ca(CO₃)_Calcite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1128617,353 +/- unknown, Reaction: 0 = +1Ca(CO₃)_Calcite+1H<+>-1Ca<2+>-1(HCO₃)<->, logK(298,15K) = -1,933 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 36,934 +/- unknown, Reference: YPF
 Remark:
- 3 Ca(OH)₂_Portlandite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -896922,648 +/- unknown, Reaction: 0 = +1Ca(OH)₂_Portlandite+2H<+>-1Ca<2+>-2H₂O, logK(298,15K) = -22,8035 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 33,056 +/- unknown, Reference: YPF
 Remark:
- 4 Ca(SO₄):2H₂O_Gypsum
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1797235,69 +/- unknown, Reaction: 0 = +1Ca(SO₄):2H₂O_Gypsum-1Ca<2+>-1(SO₄)<2->-2H₂O, logK(298,15K) = 4,5805 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 74,69 +/- unknown, Reference: YPF
 Remark:
- 5 Ca(SO₄)_Anhydrite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1321709,053 +/- unknown, Reaction: 0 = +1Ca(SO₄)_Anhydrite-1Ca<2+>-1(SO₄)<2->, logK(298,15K) = 4,3621 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 45,94 +/- unknown, Reference: YPF
 Remark:
- 6 Ca₂Cl₂(OH)₂:H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1928024,206 +/- unknown, Reaction: 0 = +1Ca₂Cl₂(OH)₂:H₂O+2H<+>-2Ca<2+>-2Cl<->-3H₂O, logK(298,15K) = -26,5313 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 7 Ca₄Cl₂(OH)₆:13H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -6586979,679 +/- unknown, Reaction: 0 = +1Ca₄Cl₂(OH)₆:13H₂O+6H<+>-4Ca<2+>-2Cl<->-19H₂O, logK(298,15K) = -68,7343 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 354,662925 +/- unknown, Reference:
 Remark:
- 8 CaCl₂:4H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1731167,12 +/- unknown, Reaction: 0 = +1CaCl₂:4H₂O-1Ca<2+>-2Cl<->-4H₂O, logK(298,15K) = -5,717 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

- 9 CaCl₂·6H₂O_Antarcticite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2214428,154 +/- unknown, Reaction: 0 =
+1CaCl₂·6H₂O_Antarcticite-1Ca²⁺-2Cl⁻-6H₂O, logK(298,15K) = -4,1436 +/-
unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 128,11 +/- unknown, Reference:
Remark: V0 calculated from <http://webmineral.com/data/Antarcticite.shtml>: rho =
1,71 g/cm³
- 10 CaMg(CO₃)₂_Dolomite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2161344,878 +/- unknown, Reaction: 0 =
+1CaMg(CO₃)₂_Dolomite+2H⁺-1Ca²⁺-1Mg²⁺-2(HCO₃)⁻, logK(298,15K) = -3,596
+/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 64,365 +/- unknown, Reference: YPF
Remark:
- 11 CaNa₂(CO₃)₂·5H₂O_Gaylussite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3371843,138 +/- unknown, Reaction: 0 =
+1CaNa₂(CO₃)₂·5H₂O_Gaylussite+2H⁺-1Ca²⁺-2Na⁺-2(HCO₃)⁻-5H₂O,
logK(298,15K) = -11,2576 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 148,15 +/- unknown, Reference: YPF
Remark:
- 12 K(HCO₃)_Kalicinite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -867748,757 +/- unknown, Reaction: 0 =
+1K(HCO₃)_Kalicinite-1K⁺-1(HCO₃)⁻, logK(298,15K) = -0,2814 +/- unknown,
Reference: HMW
V0 [cm³ mol⁻¹] = 46,14 +/- unknown, Reference: YPF
Remark:
- 13 K(HSO₄)_Mercurite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1034513,822 +/- unknown, Reaction: 0 =
+1K(HSO₄)_Mercurite-1K⁺-1H⁺-1(SO₄)²⁻, logK(298,15K) = 1,4015 +/-
unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 58,59 +/- unknown, Reference: YPF
Remark:
- 14 K₂(CO₃):1,5H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1431247,053 +/- unknown, Reaction: 0 =
+1K₂(CO₃):1,5H₂O+1H⁺-2K⁺-1(HCO₃)⁻-1,5H₂O, logK(298,15K) = -13,372 +/-
unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 80,88 +/- unknown, Reference: YPF
Remark:
- 15 K₂(SO₄)_Arcanite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1319163,196 +/- unknown, Reaction: 0 =
+1K₂(SO₄)_Arcanite-2K⁺-1(SO₄)²⁻, logK(298,15K) = 1,7763 +/- unknown,
Reference: HMW
V0 [cm³ mol⁻¹] = 65,5 +/- unknown, Reference: YPF
Remark:
- 16 K₂Ca(SO₄)₂·H₂O_Syngenite
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -2885489,785 +/- unknown, Reaction: 0 =
 $+1K_2Ca(SO_4)_2 \cdot H_2O_{\text{Synge}} - 2K^{+} - 1Ca^{2+} - 2(SO_4)^{2-} - 1H_2O$, logK(298,15K) =
 7,4484 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 124,2 +/- unknown, Reference: YPF
 Remark:
- 17 $K_2Mg(SO_4)_2 \cdot 4H_2O_{\text{Leonite}}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3479675,302 +/- unknown, Reaction: 0 =
 $+1K_2Mg(SO_4)_2 \cdot 4H_2O_{\text{Leonite}} - 2K^{+} - 1Mg^{2+} - 2(SO_4)^{2-} - 4H_2O$, logK(298,15K) = 3,979
 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 166,3 +/- unknown, Reference: YPF
 Remark:
- 18 $K_2Mg(SO_4)_2 \cdot 6H_2O_{\text{Picromerite}}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3955945,696 +/- unknown, Reaction: 0 =
 $+1K_2Mg(SO_4)_2 \cdot 6H_2O_{\text{Picromerite}} - 2K^{+} - 1Mg^{2+} - 2(SO_4)^{2-} - 6H_2O$, logK(298,15K) =
 4,3277 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 197,5 +/- unknown, Reference:
 Remark:
- 19 $K_2MgCa_2(SO_4)_4 \cdot 2H_2O_{\text{Polyhalite}}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5654754,91 +/- unknown, Reaction: 0 =
 $+1K_2MgCa_2(SO_4)_4 \cdot 2H_2O_{\text{Polyhalite}} - 2K^{+} - 1Mg^{2+} - 2Ca^{2+} - 4(SO_4)^{2-} - 2H_2O$,
 logK(298,15K) = 13,7441 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 218,1 +/- unknown, Reference: YPF
 Remark:
- 20 $K_2Na(HCO_3)(CO_3) \cdot 2H_2O_{\text{Trona-K}}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2408868,41 +/- unknown, Reaction: 0 =
 $+1K_2Na(HCO_3)(CO_3) \cdot 2H_2O_{\text{Trona-K}} + 1H^{+} - 2K^{+} - 1Na^{+} - 2(HCO_3)^{-} - 2H_2O$,
 logK(298,15K) = -11,5757 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 21 $K_3(HSO_4)(SO_4)$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2355758,741 +/- unknown, Reaction: 0 =
 $+1K_3(HSO_4)(SO_4) - 3K^{+} - 1H^{+} - 2(SO_4)^{2-}$, logK(298,15K) = 3,5425 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 22 $K_8(HCO_3)_4(CO_3)_2 \cdot 3H_2O$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -6334529,138 +/- unknown, Reaction: 0 =
 $+1K_8(HCO_3)_4(CO_3)_2 \cdot 3H_2O + 2H^{+} - 8K^{+} - 6(HCO_3)^{-} - 3H_2O$, logK(298,15K) =
 -27,6874 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 23 $K_8(HSO_4)_6(SO_4)_{\text{Misenite}}$
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -7529789,68 +/- unknown, Reaction: 0 =
 $+1K_8(HSO_4)_6(SO_4)_{\text{Misenite}} - 8K^{+} - 6H^{+} - 7(SO_4)^{2-}$, logK(298,15K) = 10,8061 +/-
 unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

- 24 KCl_Sylvite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -408590,332 +/- unknown, Reaction: 0 = +1KCl_Sylvite-1K<+>-1Cl<->, logK(298,15K) = -0,8999 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 37,524 +/- unknown, Reference: YPF
Remark:
- 25 KMgCl(SO4):3H2O_Kainite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2325625,369 +/- unknown, Reaction: 0 = +1KMgCl(SO4):3H2O_Kainite-1K<+>-1Mg<2+>-1Cl<->-1(SO4)<2->-3H2O, logK(298,15K) = 0,1926 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 115,797553488372 +/- unknown, Reference:
Remark:
- 26 KMgCl3:6H2O_Carnallite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2529657,892 +/- unknown, Reaction: 0 = +1KMgCl3:6H2O_Carnallite-1K<+>-1Mg<2+>-3Cl<->-6H2O, logK(298,15K) = -4,3304 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 172,58 +/- unknown, Reference: YPF
Remark:
- 27 KNa(CO3):6H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2495792,967 +/- unknown, Reaction: 0 = +1KNa(CO3):6H2O+1H<+>-1(HCO3)<->-1K<+>-1Na<+>-6H2O, logK(298,15K) = -10,2233 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 28 Mg(CO3):3H2O_Nesquehonite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1724116,861 +/- unknown, Reaction: 0 = +1Mg(CO3):3H2O_Nesquehonite+1H<+>-1Mg<2+>-1(HCO3)<->-3H2O, logK(298,15K) = -5,1722 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 74,79 +/- unknown, Reference: YPF
Remark: poor G-fit from data0,y pf
- 29 Mg(CO3)_Magnesite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1027919,07 +/- unknown, Reaction: 0 = +1Mg(CO3)_Magnesite+1H<+>-1Mg<2+>-1(HCO3)<->, logK(298,15K) = -2,5054 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 28,018 +/- unknown, Reference: YPF
Remark:
- 30 Mg(OH)2_Brucite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -831996,097 +/- unknown, Reaction: 0 = +1Mg(OH)2_Brucite+2H<+>-1Mg<2+>-2H2O, logK(298,15K) = -17,109 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 24,63 +/- unknown, Reference: YPF
Remark:
- 31 Mg(SO4):6H2O_Hexahydrate
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2631552,221 +/- unknown, Reaction: 0 = +1Mg(SO4):6H2O_Hexahydrate-1Mg<2+>-1(SO4)<2->-6H2O, logK(298,15K) = 1,6351 +/- unknown, Reference: HMW

- V0 [cm³ mol⁻¹] = 132,58 +/- unknown, Reference: YPF
Remark:
- 32 Mg(SO₄):7H₂O_Epsomite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2870096,399 +/- unknown, Reaction: 0 =
 +1Mg(SO₄):7H₂O_Epsomite-1Mg<2+>-1(SO₄)<2->-7H₂O, logK(298,15K) = 1,8811 +/-
 unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 146,8 +/- unknown, Reference: YPF
Remark:
- 33 Mg(SO₄):H₂O_Kieserite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1437219,377 +/- unknown, Reaction: 0 =
 +1Mg(SO₄):H₂O_Kieserite-1Mg<2+>-1(SO₄)<2->-1H₂O, logK(298,15K) = 0,1227 +/-
 unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 56,6 +/- unknown, Reference: YPF
Remark:
- 34 Mg₂CaCl₆:12H₂O_Tachyhydrite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4997309,956 +/- unknown, Reaction: 0 =
 +1Mg₂CaCl₆:12H₂O_Tachyhydrite-2Mg<2+>-1Ca<2+>-6Cl<->-12H₂O, logK(298,15K) = -
 17,3839 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 203,78 +/- unknown, Reference: YPF
Remark:
- 35 Mg₂Cl(OH)₃:4H₂O_Oxychloride-Mg
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2313687,076 +/- unknown, Reaction: 0 =
 +1Mg₂Cl(OH)₃:4H₂O_Oxychloride-Mg-2Mg<2+>-1Cl<->-3OH<->-4H₂O, logK(298,15K) = -
 26,0297 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 86,3108916666667 +/- unknown, Reference:
Remark:
- 36 MgCl₂:6H₂O_Bischoffite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2115217,386 +/- unknown, Reaction: 0 =
 +1MgCl₂:6H₂O_Bischoffite-1Mg<2+>-2Cl<->-6H₂O, logK(298,15K) = -4,4554 +/-
 unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 129,57 +/- unknown, Reference: YPF
Remark:
- 37 Na(HCO₃)_Nahcolite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -851098,341 +/- unknown, Reaction: 0 =
 +1Na(HCO₃)_Nahcolite-1Na<+>-1(HCO₃)<->, logK(298,15K) = 0,403 +/- unknown,
 Reference: HMW
 V0 [cm³ mol⁻¹] = 38,62 +/- unknown, Reference: YPF
Remark:
- 38 Na₂(CO₃):10H₂O_Natron
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3427841,828 +/- unknown, Reaction: 0 =
 +1Na₂(CO₃):10H₂O_Natron+1H<+>-1(HCO₃)<->-2Na<+>-10H₂O, logK(298,15K) = -9,5145
 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 195,99 +/- unknown, Reference: YPF
Remark:
- 39 Na₂(CO₃):7H₂O

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2714340,675 +/- unknown, Reaction: 0 =
+1Na₂(CO₃):7H₂O+1H⁺-2Na⁺-1(HCO₃)⁻-7H₂O, logK(298,15K) = -9,8791 +/-
unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 153,71 +/- unknown, Reference: YPF
Remark:
- 40 Na₂(CO₃):H₂O_Thermonatrit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1286123,699 +/- unknown, Reaction: 0 =
+1Na₂(CO₃):H₂O_Thermonatrit+1H⁺-2Na⁺-1(HCO₃)⁻-1H₂O, logK(298,15K) = -
10,8211 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 54,92 +/- unknown, Reference: YPF
Remark:
- 41 Na₂(SO₄):10H₂O_Mirabilite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3646318,335 +/- unknown, Reaction: 0 =
+1Na₂(SO₄):10H₂O_Mirabilite-2Na⁺-1(SO₄)²⁻-10H₂O, logK(298,15K) = 1,2278 +/-
unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 219,8 +/- unknown, Reference: YPF
Remark:
- 42 Na₂(SO₄)_Thenardite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1269551,062 +/- unknown, Reaction: 0 =
+1Na₂(SO₄)_Thenardite-2Na⁺-1(SO₄)²⁻, logK(298,15K) = 0,2875 +/- unknown,
Reference: HMW
V0 [cm³ mol⁻¹] = 53,33 +/- unknown, Reference: YPF
Remark:
- 43 Na₂Ca(CO₃)₂:2H₂O_Pirssonite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2659408,248 +/- unknown, Reaction: 0 =
+1Na₂Ca(CO₃)₂:2H₂O_Pirssonite+2H⁺-2Na⁺-1Ca²⁺-2(HCO₃)⁻-2H₂O,
logK(298,15K) = -11,4354 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 102,3 +/- unknown, Reference: YPF
Remark:
- 44 Na₂Ca(SO₄)₂_Glauberite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2594655,83 +/- unknown, Reaction: 0 =
+1Na₂Ca(SO₄)₂_Glauberite-2Na⁺-1Ca²⁺-2(SO₄)²⁻, logK(298,15K) = 5,2445 +/-
unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 101,12 +/- unknown, Reference: YPF
Remark:
- 45 Na₂Mg(SO₄)₂:4H₂O_Bloedite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3429245,205 +/- unknown, Reaction: 0 =
+1Na₂Mg(SO₄)₂:4H₂O_Bloedite-2Na⁺-1Mg²⁺-2(SO₄)²⁻-4H₂O, logK(298,15K) =
2,3469 +/- unknown, Reference: HMW
V0 [cm³ mol⁻¹] = 149,98 +/- unknown, Reference: YPF
Remark:
- 46 Na₃(CO₃)(HCO₃):2H₂O_Trona
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2380773,885 +/- unknown, Reaction: 0 =
+1Na₃(CO₃)(HCO₃):2H₂O_Trona+1H⁺-3Na⁺-2(HCO₃)⁻-2H₂O, logK(298,15K) = -
9,2948 +/- unknown, Reference: HMW

- V0 [cm³ mol⁻¹] = 106,11588732394 +/- unknown, Reference:
 Remark: V0 calculated from density 2,11 - 2,17, Average = 2,13, according to
http://webmineral.com/data/Trona_shtml
- 47 Na₃(HSO₄)(SO₄)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2278515,059 +/- unknown, Reaction: 0 =
 +1Na₃(HSO₄)(SO₄)-3Na⁺-1H⁺-2(SO₄)²⁻, logK(298,15K) = 0,8143 +/- unknown,
 Reference: HMW
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 48 Na₄Ca(SO₄)₃:2H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4339287,731 +/- unknown, Reaction: 0 =
 +1Na₄Ca(SO₄)₃:2H₂O-4Na⁺-1Ca²⁺-3(SO₄)²⁻-2H₂O, logK(298,15K) = 5,6723 +/-
 unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 49 Na₆(CO₃)(SO₄)₂_Burkeite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3591961,585 +/- unknown, Reaction: 0 =
 +1Na₆(CO₃)(SO₄)₂_Burkeite+1H⁺-6Na⁺-1(HCO₃)⁻-2(SO₄)²⁻, logK(298,15K) = -
 9,5671 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 151,19 +/- unknown, Reference: YPF
 Remark:
- 50 NaCl_Halite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -384206,09 +/- unknown, Reaction: 0 = +1NaCl_Halite-
 1Na⁺-1Cl⁻, logK(298,15K) = -1,5704 +/- unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 27,015 +/- unknown, Reference: YPF
 Remark:
- 51 NaK₃(SO₄)₂_Glaserite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2619196,974 +/- unknown, Reaction: 0 =
 +1NaK₃(SO₄)₂_Glaserite-1Na⁺-3K⁺-2(SO₄)²⁻, logK(298,15K) = 3,8027 +/-
 unknown, Reference: HMW
 V0 [cm³ mol⁻¹] = 246,23 +/- unknown, Reference: YPF
 Remark: Value for V0 is twice as high as in hnv !
- 52 2Pb(CO₃):Pb(OH)₂_Hydrocerussite_(Schock)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1708697,11 +/- unknown, Reaction: 0 =
 +12Pb(CO₃):Pb(OH)₂_Hydrocerussite_(Schock)+4H⁺-3Pb²⁺-2(HCO₃)⁻-2H₂O,
 logK(298,15K) = -2,1 +/- unknown, Reference: ULT
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 53 2PbCl₂:KCl
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1047085,361 +/- unknown, Reaction: 0 = +12PbCl₂:KCl-
 2Pb²⁺-1K⁺-5Cl⁻, logK(298,15K) = 10,514 +/- unknown, Reference: ULT
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 54 3Pb(OH)₂:Pb(SO₄)
 Gibbs Free Energy of Formation calculated from logK for reaction

- deltaG_0^f [J mol-1] = -2177604,555 +/- unknown, Reaction: 0 =
+13Pb(OH)2:Pb(SO4)+6H<+>-4Pb<2+>-1(SO4)<2->-6H2O, logK(298,15K) = -15,1 +/-
unknown, Reference: HAG1998
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 55 3Pb(OH)2:PbCl2
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1688614,099 +/- unknown, Reaction: 0 =
+13Pb(OH)2:PbCl2+6H<+>-4Pb<2+>-2Cl<->-6H2O, logK(298,15K) = -16,4 +/- unknown,
Reference: HAG1998
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 56 3PbCl2:3KCl:H2O
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -2424134,59 +/- unknown, Reaction: 0 =
+13PbCl2:3KCl:H2O-3Pb<2+>-3K<+>-9Cl<->-1H2O, logK(298,15K) = 15,031 +/-
unknown, Reference: ULT
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 57 NaPb2(CO3)2(OH)_Auerbachs_Salz
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1700281,943 +/- unknown, Reaction: 0 =
+1NaPb2(CO3)2(OH)_Auerbachs_Salz+3H<+>-2Pb<2+>-1Na<+>-2(HCO3)<->-1H2O,
logK(298,15K) = -3,675 +/- unknown, Reference: HAG1998
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 58 Pb
Gibbs Free Energy of Formation directly entered
deltaG_0^f [J mol-1] = 0 +/- unknown, Reference:
V0 [cm3 mol-1] = 18,267 +/- unknown, Reference: COM
Remark:
- 59 Pb(CO3):PbCl2_Phosgenite
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -956547,84 +/- unknown, Reaction: 0 =
+1Pb(CO3):PbCl2_Phosgenite+1H<+>-2Pb<2+>-1(HCO3)<->-2Cl<->,
logK(298,15K) = 10,3 +/- unknown, Reference: ULT
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 60 Pb(CO3)_Cerussite
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -629708,343 +/- unknown, Reaction: 0 =
+1Pb(CO3)_Cerussite+1H<+>-1Pb<2+>-1(HCO3)<->,
logK(298,15K) = 3,263 +/-
unknown, Reference: ULT
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 61 Pb(OH)2:Pb(SO4)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -1261622,762 +/- unknown, Reaction: 0 =
+1Pb(OH)2:Pb(SO4)+2H<+>-2Pb<2+>-1(SO4)<2->-2H2O, logK(298,15K) = -0,9 +/-
unknown, Reference: HAG1998
V0 [cm3 mol-1] = dummy value 0,001
Remark:

- 62 Pb(OH)Cl_Laurionite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -390939,668 +/- unknown, Reaction: 0 =
 +1Pb(OH)Cl_Laurionite+1H<+>-1Pb<2+>-1Cl<->-1H2O, logK(298,15K) = -0,29 +/-
 unknown, Reference: ULT
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 63 Pb(SO4):K2(SO4)_Palmierite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2150249,034 +/- unknown, Reaction: 0 =
 +1Pb(SO4):K2(SO4)_Palmierite-1Pb<2+>-2K<+>-2(SO4)<2->, logK(298,15K) = 12,786
 +/- unknown, Reference: ULT
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 64 Pb(SO4)_Anglesite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -813012,462 +/- unknown, Reaction: 0 =
 +1Pb(SO4)_Anglesite-1Pb<2+>-1(SO4)<2->, logK(298,15K) = 7,8434 +/- unknown,
 Reference: ULT
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 65 Pb2(O)(SO4)_Lanarkite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1026708,898 +/- unknown, Reaction: 0 =
 +1Pb2(O)(SO4)_Lanarkite+2H<+>-2Pb<2+>-1(SO4)<2->-1H2O, logK(298,15K) = -0,51
 +/- unknown, Reference: ULT
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 66 Pb3(CO3)2(OH)2_Hydrocerussite_(RIC)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1701169,344 +/- unknown, Reaction: 0 =
 +1Pb3(CO3)2(OH)2_Hydrocerussite_(RIC)+4H<+>-3Pb<2+>-2(HCO3)<->-2H2O,
 logK(298,15K) = -3,4188 +/- unknown, Reference: HAG1998
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 67 Pb3O4_Minium
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -602116,505 +/- unknown, Reaction: 0 =
 +1Pb3O4_Minium+8H<+>-1Pb<4+>-2Pb<2+>-4H2O, logK(298,15K) = -16,2585 +/-
 unknown, Reference: COM
 V0 [cm3 mol⁻¹] = 76,81 +/- unknown, Reference: COM
 Remark:
- 68 PbCl2:3MgCl2:19H2O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -6909444,432 +/- unknown, Reaction: 0 =
 +1PbCl2:3MgCl2:19H2O-1Pb<2+>-3Mg<2+>-8Cl<->-19H2O, logK(298,15K) = -6,362 +/-
 unknown, Reference: ULT
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 69 PbCl2:Pb(SO4):2Na2(SO4):5w
 Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -4866792,316 +/- unknown, Reaction: 0 =
+1PbCl₂:Pb(SO₄):2Na₂(SO₄):5w-2Pb²⁺-4Na⁺-2Cl⁻-3(SO₄)²⁻-5H₂O,
logK(298,15K) = 15,83 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 70 PbCl₂_Cotunnite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -313891,373 +/- unknown, Reaction: 0 =
+1PbCl₂_Cotunnite-1Pb²⁺-2Cl⁻, logK(298,15K) = 4,7686 +/- unknown,
Reference: ULT
V0 [cm³ mol⁻¹] = 46,5059197324415 +/- unknown, Reference:
Remark: V0 calculated from rho=5,98 g/cc taken from
<http://www.matweb.com/search/SpecificMaterial.asp?bassnum=EINOR0401>
- 71 PbO_Litharge
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -189205,508 +/- unknown, Reaction: 0 =
+1PbO_Litharge+2H⁺-1H₂O-1Pb²⁺, logK(298,15K) = -12,644 +/- unknown,
Reference: COM
V0 [cm³ mol⁻¹] = 23,91 +/- unknown, Reference: COM
Remark:
- 72 PbO_Massicot
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -188205,459 +/- unknown, Reaction: 0 =
+1PbO_Massicot+2H⁺-1H₂O-1Pb²⁺, logK(298,15K) = -12,8192 +/- unknown,
Reference: COM
V0 [cm³ mol⁻¹] = 23,15 +/- unknown, Reference: COM
Remark:
- 73 Fe(OH)₂(cr)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -486975 +/- unknown, Reference: Barin
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 74 Fe(OH)₂_precipitated
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -483324 +/- unknown, Reference: calc, From emf (Moog)
V0 [cm³ mol⁻¹] = 26,43 +/- unknown, Reference: ymp
Remark:
- 75 Fe(SO₄):7H₂O_Melanterite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2507304,264 +/- unknown, Reaction: 0 =
+1Fe(SO₄):7H₂O_Melanterite-1Fe²⁺-1(SO₄)²⁻-7H₂O, logK(298,15K) = 2,246 +/-
0,009, Reference: MoHa
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 76 FeAl₂O₄_Hercynite
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1879669 +/- unknown, Reference: Barin
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 77 FeCl₂:4H₂O
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -1284285,964 +/- unknown, Reaction: 0 = +1FeCl₂:4H₂O-1Fe²⁺-2Cl⁻-4H₂O, logK(298,15K) = -3,0147 +/- 0,03, Reference: MoHa
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 78 FeK₂(SO₄)₂:6H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3591070,696 +/- unknown, Reaction: 0 = +1FeK₂(SO₄)₂:6H₂O-1Fe²⁺-2K⁺-2(SO₄)²⁻-6H₂O, logK(298,15K) = 4,3277 +/- unknown, Reference: MoHa
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark: logK: value for Picromerite in data0,hmw
- 79 FeNa₂(SO₄)₂:4H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3069987,49 +/- unknown, Reaction: 0 = +1FeNa₂(SO₄)₂:4H₂O-1Fe²⁺-2Na⁺-2(SO₄)²⁻-4H₂O, logK(298,15K) = 3,331 +/- 0,03, Reference: MoHa
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 80 FeS
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -101965 +/- unknown, Reference: Barin
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 81 FeS₂
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -160076 +/- unknown, Reference: Barin
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 82 Fe₃O₄_Magnetite
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1015227 +/- unknown, Reference: Barin
 V0 [cm³ mol⁻¹] = 44,524 +/- unknown, Reference: YMP
 Remark:
- 83 Fe(OH)₃(cr)
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -696486 +/- unknown, Reference: Barin
 V0 [cm³ mol⁻¹] = 34,36 +/- unknown, Reference: YPF
 Remark:
- 84 Fe₂(SO₄)₃
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -2262753 +/- unknown, Reference: Barin
 V0 [cm³ mol⁻¹] = 130,77 +/- unknown, Reference: YPF
 Remark:
- 85 KFe₃(SO₄)₂(OH)₆_Jarosite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3029660,08 +/- unknown, Reaction: 0 = +1KFe₃(SO₄)₂(OH)₆_Jarosite+6H⁺-1K⁺-2(SO₄)²⁻-3Fe³⁺-6H₂O, logK(298,15K) = 9,3706 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 156,75 +/- unknown, Reference: YPF
 Remark:
- 86 NaFe₃(SO₄)₂(OH)₆_Jarosite-Na

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2986713,854 +/- unknown, Reaction: 0 =
+1NaFe₃(SO₄)₂(OH)₆_Jarosite-Na+6H⁺-1Na⁺-2(SO₄)²⁻-3Fe³⁺-6H₂O,
logK(298,15K) = 5,4482 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = 168,42 +/- unknown, Reference: YPF
Remark:
- 87 Fe₂(SiO₄)_Fayalite
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1378985 +/- unknown, Reference: Barin
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 88 FeSiO₃(cr)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1117463 +/- unknown, Reference: Barin
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 89 Am^{III}(CO₃)(OH)(am,hyd)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1399074,535 +/- unknown, Reaction: 0 =
+1Am^{III}(CO₃)(OH)(am,hyd)-1Am^{III}+1(CO₃)²⁻-1OH⁻, logK(298,15K) =
20,2 +/- 1, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 90 Am^{III}(CO₃)(OH):0,5H₂O(c)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1530202,229 +/- 5560, Reaction: 0 =
+1Am^{III}(CO₃)(OH):0,5H₂O(c)-1Am^{III}+1(CO₃)²⁻-0,5H₂O-1OH⁻,
logK(298,15K) = 22,4 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 91 Am^{III}(OH)₃(am)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1213652,078 +/- unknown, Reaction: 0 =
+1Am^{III}(OH)₃(am)+3H⁺-3H₂O-1Am^{III}+1(CO₃)²⁻, logK(298,15K) = -16,9 +/- 0,8,
Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 92 Am^{III}(OH)₃(c)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1221072,534 +/- 5861, Reaction: 0 =
+1Am^{III}(OH)₃(c)+3H⁺-3H₂O-1Am^{III}+1(CO₃)²⁻, logK(298,15K) = -15,6 +/- 0,6,
Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 93 Am^{III}(PO₄)(am,hyd)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1765691,379 +/- unknown, Reaction: 0 =
+1Am^{III}(PO₄)(am,hyd)-1Am^{III}+1(CO₃)²⁻-1(PO₄)³⁻, logK(298,15K) = 24,79 +/-
0,6, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 94 Am^{III}₂(CO₃)₃(am,hyd)

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2876203,815 +/- unknown, Reaction: 0 = +1Am|+III|2(CO3)3(am,hyd)-3(CO3)<2->-2Am|+III|<3+>, logK(298,15K) = 16,7 +/- 1,1, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 95 NaAm|+III|(CO3)2:5H2O(c)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3221875,565 +/- 5605, Reaction: 0 = +1NaAm|+III|(CO3)2:5H2O(c)-1Am|+III|<3+>-2(CO3)<2->-5H2O-1Na<+>, logK(298,15K) = 21 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 96 (Am|+V|O2)(OH)(am)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -946683,374 +/- unknown, Reaction: 0 = +1(Am|+V|O2)(OH)(am)+1H<+>-1H2O-1(Am|+V|O2)<+>, logK(298,15K) = -5,3 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 97 Na(Am|+V|O2)(CO3)(s)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1591794,5 +/- 6627, Reaction: 0 = +1Na(Am|+V|O2)(CO3)(s)-1(Am|+V|O2)<+>-1(CO3)<2->-1Na<+>, logK(298,15K) = 10,9 +/- 0,4, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 98 K2HAsO4:3H2O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1991494,158 +/- unknown, Reaction: 0 = +1K2HAsO4:3H2O-2K<+>-1(HAsO4)<2->-3H2O, logK(298,15K) = 1,3935 +/- , Reference: Koda
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 99 K3AsO4:7H2O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3156410,183 +/- unknown, Reaction: 0 = +1K3AsO4:7H2O-3K<+>-1(AsO4)<3->-7H2O, logK(298,15K) = 0,2453 +/- , Reference: Koda
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 100 KH2AsO4
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1029378,641 +/- unknown, Reaction: 0 = +1KH2AsO4-1K<+>-1(H2AsO4)<->, logK(298,15K) = -0,2858 +/- , Reference: Koda
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 101 Na2HAsO4:7H2O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1937259,651 +/- unknown, Reaction: 0 = +1Na2HAsO4:7H2O-2Na<+>-1(HAsO4)<2->-3H2O, logK(298,15K) = -0,9051 +/- , Reference: Koda
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 102 Na3AsO4:12H2O

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4262084,972 +/- unknown, Reaction: 0 = +1Na₃AsO₄:12H₂O-3Na⁺-1(AsO₄)³⁻-12H₂O, logK(298,15K) = -2,9702 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 103 NaH₂AsO₄:H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1258950,863 +/- unknown, Reaction: 0 = +1NaH₂AsO₄:H₂O-1Na⁺-1(H₂AsO₄)⁻-1H₂O, logK(298,15K) = 1,9898 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 104 Na₂[B₄O₅(OH)₄]:8H₂O_Borax
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5517956,02 +/- unknown, Reaction: 0 = +1Na₂[B₄O₅(OH)₄]:8H₂O_Borax+2H⁺-2Na⁺-4B(OH)₃-5H₂O, logK(298,15K) = -12,0395 +/- unknown, Reference: /WEP1982/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 105 BaSeO₃
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -957148,59 +/- 3400, Reaction: 0 = +1BaSeO₃-1Ba²⁺-1SeO₃²⁻, logK(298,15K) = 6,5 +/- 0,25, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 106 BaSeO₄
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1040293,728 +/- 3007, Reaction: 0 = +1BaSeO₄-1Ba²⁺-1SeO₄²⁻, logK(298,15K) = 7,56 +/- 0,1, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 107 (CdCl₂)₂:5H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1889151,573 +/- unknown, Reaction: 0 = +1(CdCl₂)₂:5H₂O-2Cd²⁺-4Cl⁻-5H₂O, logK(298,15K) = 4,05 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 108 (CdSO₄)₃:8H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4395768,714 +/- unknown, Reaction: 0 = +1(CdSO₄)₃:8H₂O-3Cd²⁺-3(SO₄)²⁻-8H₂O, logK(298,15K) = 5,858 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 109 2CdSO₄:2K₂SO₄:3H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5025284,752 +/- unknown, Reaction: 0 = +12CdSO₄:2K₂SO₄:3H₂O-4K⁺-2Cd²⁺-4(SO₄)²⁻-3H₂O, logK(298,15K) = 9,17 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 110 3CdSO₄:K₂SO₄:5H₂O
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -5012448,993 +/- unknown, Reaction: 0 = +13CdSO₄:K₂SO₄:5H₂O-2K<+>-3Cd<2+>-4(SO₄)<2->-5H₂O, logK(298,15K) = 9,2 +/- unknown, Reference:
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 111 CdCl₂:H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -587216,162 +/- unknown, Reaction: 0 = +1CdCl₂:H₂O-1Cd<2+>-2Cl<->-1H₂O, logK(298,15K) = 1,736 +/- unknown, Reference: ULT
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 112 CdCO₃
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -669587,378 +/- unknown, Reaction: 0 = +1CdCO₃+1H<+>-1Cd<2+>-1(HCO₃)<->, logK(298,15K) = 0,8776 +/- unknown, Reference:
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 113 K₄CdCl₆
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2000897,204 +/- unknown, Reaction: 0 = +1K₄CdCl₆-4K<+>-1Cd<2+>-6Cl<->, logK(298,15K) = 1,02 +/- unknown, Reference: ULT
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 114 K₃CdCl₇:4H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2430878,26 +/- unknown, Reaction: 0 = +1K₃CdCl₇:4H₂O-1K<+>-3Cd<2+>-7Cl<->-4H₂O, logK(298,15K) = 8,425 +/- unknown, Reference: ULT
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 115 K₂CdCl₃:H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1008614,772 +/- unknown, Reaction: 0 = +1K₂CdCl₃:H₂O-1K<+>-1Cd<2+>-3Cl<->-1H₂O, logK(298,15K) = 3,08 +/- unknown, Reference: ULT
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 116 Mg₂CdCl₆:12H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4583352,399 +/- unknown, Reaction: 0 = +1Mg₂CdCl₆:12H₂O-2Mg<2+>-1Cd<2+>-6Cl<->-12H₂O, logK(298,15K) = -6,677 +/- unknown, Reference: ULT
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 117 MgCd₂Cl₆:12H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4251397,573 +/- unknown, Reaction: 0 = +1MgCd₂Cl₆:12H₂O-1Mg<2+>-2Cd<2+>-6Cl<->-12H₂O, logK(298,15K) = 1,327 +/- unknown, Reference: ULT
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 118 Na₂Cd(SO₄)₂:2H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -2582535,219 +/- unknown, Reaction: 0 = +1Na₂Cd(SO₄)₂·2H₂O-2Na⁺-1Cd²⁺-2(SO₄)²⁻-2H₂O, logK(298,15K) = 3,26 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 119 Na₂CdCl₄·3H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1845347,456 +/- unknown, Reaction: 0 = +1Na₂CdCl₄·3H₂O-2Na⁺-1Cd²⁺-4Cl⁻-3H₂O, logK(298,15K) = 1,3 +/- unknown, Reference:
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 120 Na₃Cd₄Cl₁₁·14H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5897240,278 +/- unknown, Reaction: 0 = +1Na₃Cd₄Cl₁₁·14H₂O-3Na⁺-4Cd²⁺-11Cl⁻-14H₂O, logK(298,15K) = 6,5 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 121 alpha-CdSe
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -140897 +/- 1918, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 122 CdSeO₃
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -493432 +/- 6467, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 123 Co(OH)₂(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -458197,079 +/- unknown, Reaction: 0 = +1Co(OH)₂(cr)-1Co²⁺-2H₂O+2H⁺, logK(298,15K) = -12,2 +/- 0,2, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 124 CoCl₂·6H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1753219,176 +/- unknown, Reaction: 0 = +1CoCl₂·6H₂O-1Co²⁺-2Cl⁻-6H₂O, logK(298,15K) = 2,521 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 125 CoCl₂·MgCl₂·8H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2981148,976 +/- unknown, Reaction: 0 = +1CoCl₂·MgCl₂·8H₂O-1Co²⁺-4Cl⁻-8H₂O-1Mg²⁺, logK(298,15K) = 8,8 +/- 0,09, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 126 CoSO₄·7H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2444165,256 +/- unknown, Reaction: 0 = +1CoSO₄·7H₂O-1Co²⁺-1(SO₄)²⁻-7H₂O, logK(298,15K) = -2,343 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 127 CoSO₄:K₂SO₄:6H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3499912,619 +/- unknown, Reaction: 0 = +1CoSO₄:K₂SO₄:6H₂O-1Co²⁺-2(SO₄)²⁻-6H₂O-2K⁺, logK(298,15K) = -5,17 +/- 0,04, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 128 CoSO₄:Na₂SO₄:4H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2995078,498 +/- unknown, Reaction: 0 = +1CoSO₄:Na₂SO₄:4H₂O-1Co²⁺-2(SO₄)²⁻-4H₂O-2Na⁺, logK(298,15K) = -3,32 +/- 0,03, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 129 K₂CrO₄
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1292510,03 +/- unknown, Reaction: 0 = +1K₂CrO₄-2K⁺-1CrO₄²⁻, logK(298,15K) = -0,054304 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 130 K₂CrO₄:MgCrO₄
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2488552,694 +/- unknown, Reaction: 0 = +1K₂CrO₄:MgCrO₄-2K⁺-1Mg²⁺-2CrO₄²⁻, logK(298,15K) = 2,2 +/- 0,2, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 131 MgCrO₄
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1181614,25 +/- unknown, Reaction: 0 = +1MgCrO₄-1Mg²⁺-1CrO₄²⁻, logK(298,15K) = -0,27343 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 132 MgCrO₄:5H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2369758,091 +/- unknown, Reaction: 0 = +1MgCrO₄:5H₂O-1Mg²⁺-1CrO₄²⁻-5H₂O, logK(298,15K) = 0,15471 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 133 Na₂CrO₄:4H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2194693,01 +/- unknown, Reaction: 0 = +1Na₂CrO₄:4H₂O-2Na⁺-1CrO₄²⁻-4H₂O, logK(298,15K) = -0,97634 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 134 Na₂CrO₄:6H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2670986,636 +/- unknown, Reaction: 0 = +1Na₂CrO₄:6H₂O-2Na⁺-1CrO₄²⁻-6H₂O, logK(298,15K) = -0,62357 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 135 Na₂CrO₄:MgCrO₄:2H₂O

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2949704,196 +/- unknown, Reaction: 0 = +1Na₂CrO₄:MgCrO₄:2H₂O-2K<+>-1Mg<2+>-2CrO₄<2->-2H₂O, logK(298,15K) = -0,1 +/- 0,2, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 136 Cs₂SeO₄
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1018766,61 +/- 1800, Reaction: 0 = +1Cs₂SeO₄-2Cs<+>-1SeO₄<2->, logK(298,15K) = -0,636 +/- 0,065, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 137 Cu(OH)₂(s)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -458068,73 +/- unknown, Reaction: 0 = +1Cu(OH)₂(s)+2H<+>-1Cu<2+>-2H₂O, logK(298,15K) = 8,67 +/- 0,05, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 138 Cu₂CO₃(OH)₂(s)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -521641,614 +/- unknown, Reaction: 0 = +1Cu₂CO₃(OH)₂(s)-2Cu<2+>-1(CO₃)<2->-2OH<->, logK(298,15K) = -33,16 +/- 0,08, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 139 Cu₃(CO₃)₂(OH)₂(s)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -916757,027 +/- unknown, Reaction: 0 = +1Cu₃(CO₃)₂(OH)₂(s)-3Cu<2+>-2(CO₃)<2->-2OH<->, logK(298,15K) = -44,9 +/- 0,2, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 140 CuO(s)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -215049,446 +/- unknown, Reaction: 0 = +1CuO(s)+2H<+>-1Cu<2+>-1H₂O, logK(298,15K) = 7,64 +/- 0,06, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 141 FeCO₃_Siderite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -678035,673 +/- unknown, Reaction: 0 = +1FeCO₃_Siderite+1H<+>-1Fe<2+>-1(HCO₃)<->, logK(298,15K) = 0,121 +/- unknown, Reference: PSI/NAGRA
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 142 Hg(liquid)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 0 +/- 0, Reference: Convention
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 143 Hg₂Cl₂
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -210725 +/- 471, Reference: NEA-7

- V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 144 Hg₂SO₄
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -625780 +/- 411, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 145 HgO(red)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -58523 +/- 154, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 146 HgO(s)(hx)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = 388164,778 +/- unknown, Reaction: 0 = +1HgO(s)(hx)+3H₂O-
1Hg(OH)₂, logK(298,15K) = -3,59 +/- 0,05, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 147 HgO(s)(O-rh,red)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -85943,981 +/- unknown, Reaction: 0 = +1HgO(s)(O-
rh,red)+1H₂O-1Hg(OH)₂, logK(298,15K) = -3,62 +/- 0,05, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 148 HgO(s)(O-rh,yellow)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = 151253,1 +/- unknown, Reaction: 0 = +1HgO(s)(O-
rh,yellow)+2H₂O-1Hg(OH)₂, logK(298,15K) = -3,63 +/- 0,05, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 149 alpha-HgSe
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -51200 +/- 4000, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 150 Hg₂SeO₃
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -295585,561 +/- 5998, Reaction: 0 = +1Hg₂SeO₃-1Hg₂<2+>-
1SeO₃<2->, logK(298,15K) = 15,2 +/- 1, Reference: NAE-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 151 HgSeO₃
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -290193,604 +/- 5,98, Reaction: 0 = +1HgSeO₃-1Hg<2+>-
1SeO₃<2->, logK(298,15K) = 16,2 +/- 1, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 152 I₂(cr)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 0 +/- 0, Reference: Convention

- V0 [cm³ mol⁻¹] = 51,3783279352227 +/- unknown, Reference:
Remark: V0 calculated from rho = 4,94 g/cm³
- 153 KI
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -324244,925 +/- unknown, Reaction: 0 = +1KI-1K<+>-1I<->,
logK(298,15K) = -1,75 +/- unknown, Reference: I/Se-AB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 154 MgI₂:8H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -2426609,368 +/- unknown, Reaction: 0 = +1MgI₂:8H₂O-
1Mg<2+>-2I<->-8H₂O, logK(298,15K) = -5,139 +/- unknown, Reference: I/Se-AB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 155 NaI:2H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -771002,401 +/- unknown, Reaction: 0 = +1NaI:2H₂O-
1Na<+>-1I<->-2H₂O, logK(298,15K) = -2,9703 +/- unknown, Reference: I/Se-AB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 156 beta-Ni(IO₃)₂
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -323735,629 +/- 1743, Reaction: 0 = +1beta-Ni(IO₃)₂-
1Ni<2+>-2IO₃<->, logK(298,15K) = 4,43 +/- 0,02, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 157 Ni(IO₃)₂:2H₂O(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -802068,34 +/- 1832, Reaction: 0 = +1Ni(IO₃)₂:2H₂O(cr)-
1Ni<2+>-2IO₃<->-2H₂O, logK(298,15K) = 5,14 +/- 0,1, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 158 NiI₂(cr)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -94360 +/- 898, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 159 Li₂SeO₄:H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -1252403,354 +/- 1500, Reaction: 0 = +1Li₂SeO₄:H₂O-
2Li<+>-1SeO₄<2->-1H₂O, logK(298,15K) = -1,762 +/- 0,065, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 160 Anatase_TiO₂
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = 425427,146 +/- unknown, Reaction: 0 =
+1Anatase_TiO₂+2H₂O-1Ti(OH)₄<0>, logK(298,15K) = 8,5586 +/- , Reference: Min-
GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 161 Bassanite_CaSO₄:0,5H₂O

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1436279,998 +/- unknown, Reaction: 0 = +1Bassanite_CaSO4:0,5H2O-0,5H2O-1Ca<2+>-1(SO4)<2->, logK(298,15K) = 3,6615 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 162 CaSO4:0,5H2O(beta)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1435320,476 +/- unknown, Reaction: 0 = +1CaSO4:0,5H2O(beta)-0,5H2O-1Ca<2+>-1(SO4)<2->, logK(298,15K) = 3,4934 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 163 Cassiterite_SnO2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -519615,416 +/- unknown, Reaction: 0 = +1Cassiterite_SnO2+2H<+>-0,5O2(aq)-1H2O-1Sn<2+>, logK(298,15K) = 46,1642 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 164 Chalcedony-SiO2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -856155,514 +/- unknown, Reaction: 0 = +1Chalcedony-SiO2-1SiO2<0>, logK(298,15K) = 3,7281 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 165 Coesite_SiO2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -853080,021 +/- unknown, Reaction: 0 = +1Coesite_SiO2-1SiO2<0>, logK(298,15K) = 3,1893 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 166 Dawsonite_NaAlCO3(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1793683,289 +/- unknown, Reaction: 0 = +1Dawsonite_NaAlCO3(OH)2+3H<+>-1Al<3+>-1(HCO3)<->-1Na<+>-2H2O, logK(298,15K) = -3,6618 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 167 Delafossite_CuFeO2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -388214,417 +/- unknown, Reaction: 0 = +1Delafossite_CuFeO2+4H<+>-1Cu<+>-1Fe<3+>-2H2O, logK(298,15K) = 6,4172 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 168 Dolomite-dis_CaMg(CO3)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2158708,333 +/- unknown, Reaction: 0 = +1Dolomite-dis_CaMg(CO3)2+2H<+>-1Ca<2+>-1Mg<2+>-2(HCO3)<->, logK(298,15K) = -4,0579 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 169 Dolomite-ord_CaMg(CO3)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2167523,835 +/- unknown, Reaction: 0 = +1Dolomite-
ord_CaMg(CO3)2+2H<+>-1Ca<2+>-1Mg<2+>-2(HCO3)<->, logK(298,15K) = -2,5135 +/- ,
Reference: Min-GWB
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 170 Eskolaite_Cr2O3
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1051056,585 +/- unknown, Reaction: 0 =
+1Eskolaite_Cr2O3+2H2O+1,5O2(aq)-2CrO4<2->-4H<+>, logK(298,15K) = 7,8698 +/- ,
Reference: Min-GWB
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 171 Gibbsite_Al(OH)3
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1163161,35 +/- unknown, Reaction: 0 =
+1Gibbsite_Al(OH)3+3H<+>-1Al<3+>-3H2O, logK(298,15K) = -6,9666 +/- , Reference:
Min-GWB
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 172 Goethite_FeOOH
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -398833,816 +/- unknown, Reaction: 0 =
+1Goethite_FeOOH+3H<+>-1Fe<3+>-2H2O, logK(298,15K) = -0,5345 +/- , Reference:
Min-GWB
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 173 Hematite_Fe2O3
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -566009,637 +/- unknown, Reaction: 0 =
+1Hematite_Fe2O3+6H<+>-2Fe<3+>-3H2O, logK(298,15K) = -0,1086 +/- , Reference:
Min-GWB
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 174 MnHPO4
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1397987,029 +/- unknown, Reaction: 0 = +1MnHPO4-
1(HPO4)<2->-1Mn<2+>, logK(298,15K) = 12,947 +/- , Reference: Min-GWB
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 175 Morenosite_NiSO4:7H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2461252,998 +/- unknown, Reaction: 0 =
+1Morenosite_NiSO4:7H2O-1Ni<2+>-1(SO4)<2->-7H2O, logK(298,15K) = 2,014 +/- ,
Reference: Min-GWB
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 176 Ni2SiO4
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -1318838,895 +/- unknown, Reaction: 0 = +1Ni₂SiO₄+4H<+>-1SiO₂<0>-2H₂O-2Ni<2+>, logK(298,15K) = -14,3416 +/- , Reference: Min-GWB
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 177 NiCO₃
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -612572,496 +/- unknown, Reaction: 0 = +1NiCO₃+1H<+>-1(HCO₃)<->-1Ni<2+>, logK(298,15K) = -3,5118 +/- , Reference: Min-GWB
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 178 NiSO₄·6H₂O(alpha)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2224074,183 +/- unknown, Reaction: 0 = +1NiSO₄·6H₂O(alpha)-1Ni<2+>-1(SO₄)<2->-6H₂O, logK(298,15K) = 2,0072 +/- , Reference: Min-GWB
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 179 Rhodochrosite_MnCO₃
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -813535,113 +/- unknown, Reaction: 0 = +1Rhodochrosite_MnCO₃+1H<+>-1(HCO₃)<->-1Mn<2+>, logK(298,15K) = -0,247 +/- , Reference: Min-GWB
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 180 Rutile-TiO₂
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = 419224,786 +/- unknown, Reaction: 0 = +1Rutile-TiO₂+2H₂O-1Ti(OH)₄<0>, logK(298,15K) = 9,6452 +/- , Reference: Min-GWB
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 181 SiO₂(am)-GWB
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -850364,705 +/- unknown, Reaction: 0 = +1SiO₂(am)-GWB-1SiO₂<0>, logK(298,15K) = 2,7136 +/- , Reference: Min-GWB
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 182 Trevorite_NiFe₂O₄
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -793674,491 +/- unknown, Reaction: 0 = +1Trevorite_NiFe₂O₄+8H<+>-1Ni<2+>-2Fe<3+>-4H₂O, logK(298,15K) = -9,7876 +/- , Reference: Min-GWB
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 183 Tridymite_SiO₂
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -856724,606 +/- unknown, Reaction: 0 = +1Tridymite_SiO₂-1SiO₂<0>, logK(298,15K) = 3,8278 +/- , Reference: Min-GWB
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 184 V₂O₄
 Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -1318151,229 +/- unknown, Reaction: 0 = +1V2O4+4H<+>-2H2O-2VO<2+>, logK(298,15K) = -8,5719 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 185 2MgCl₂:MnCl₂:12H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4692718,528 +/- unknown, Reaction: 0 = +12MgCl₂:MnCl₂:12H₂O-1Mn<2+>-6Cl<->-12H₂O-2Mg<2+>, logK(298,15K) = -13,86 +/- 0,07, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 186 CaCl₂:MnCl₂:8H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3160654,484 +/- unknown, Reaction: 0 = +1CaCl₂:MnCl₂:8H₂O -1Mn<2+>-4Cl<->-8H₂O-1Ca<2+>, logK(298,15K) = -7,4 +/- 0,1, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 187 MgCl₂:2MnCl₂:12H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4481197,726 +/- unknown, Reaction: 0 = +1MgCl₂:2MnCl₂:12H₂O-2Mn<2+>-6Cl<->-12H₂O-1Mg<2+>, logK(298,15K) = -11,1 +/- 0,1, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 188 MgCl₂:MnCl₂:8H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3057915,004 +/- unknown, Reaction: 0 = +1MgCl₂:MnCl₂:8H₂O-1Mn<2+>-4Cl<->-8H₂O-1Mg<2+>, logK(298,15K) = -8,33 +/- 0,06, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 189 Mn(OH)₂
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -683930,228 +/- unknown, Reaction: 0 = +1Mn(OH)₂-2Mn<2+>-2OH<->, logK(298,15K) = -15,2 +/- 0,1, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 190 MnCl₂:4H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1422580,632 +/- unknown, Reaction: 0 = +1MnCl₂:4H₂O-1Mn<2+>-2Cl<->-4H₂O, logK(298,15K) = -2,893 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 191 MnCl₂:KCl:2H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1356165,473 +/- unknown, Reaction: 0 = +1MnCl₂:KCl:2H₂O-1Mn<2+>-3Cl<->-2H₂O-1K<+>, logK(298,15K) = -3,92 +/- 0,05, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 192 MnSO₄:4H₂O
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -1929882,489 +/- unknown, Reaction: 0 = +1MnSO₄:4H₂O-1Mn²⁺-1(SO₄)²⁻-4H₂O, logK(298,15K) = 1,615 +/- , Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 193 MnSO₄:5H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2167416,344 +/- unknown, Reaction: 0 = +1MnSO₄:5H₂O-1Mn²⁺-1(SO₄)²⁻-5H₂O, logK(298,15K) = 1,684 +/- , Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 194 MnSO₄:H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1217292,34 +/- unknown, Reaction: 0 = +1MnSO₄:H₂O-1Mn²⁺-1(SO₄)²⁻-1H₂O, logK(298,15K) = 1,41 +/- , Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 195 MnSO₄:K₂SO₄:4H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3253661,779 +/- unknown, Reaction: 0 = +1MnSO₄:K₂SO₄:4H₂O-1Mn²⁺-2(SO₄)²⁻-4H₂O-2K⁺, logK(298,15K) = 4,2 +/- 0,3, Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 196 MnSO₄:Na₂SO₄:2H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2469293,887 +/- unknown, Reaction: 0 = +1MnSO₄:Na₂SO₄:2H₂O-1Mn²⁺-2(SO₄)²⁻-2H₂O-1Na⁺, logK(298,15K) = 2,97 +/- 0,02, Reference: Koda
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 197 CaMoO₄_Powellite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1451777,078 +/- unknown, Reaction: 0 = +1CaMoO₄_Powellite-1Ca²⁺-1MoO₄²⁻, logK(298,15K) = 7,96 +/- unknown, Reference: /GMR1992/
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark: logK value valid for 22C
- 198 MgMoO₄:5H₂O(s)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2505169,937 +/- unknown, Reaction: 0 = +1MgMoO₄:5H₂O(s)-1Mg²⁺-1MoO₄²⁻-5H₂O, logK(298,15K) = 1,85 +/- 0,03, Reference: /GMR1992/
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 199 Na_{0,5}Nd_{0,5}MoO₄(s)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1371541,46 +/- unknown, Reaction: 0 = +1Na_{0,5}Nd_{0,5}MoO₄(s)-0,5Na⁺-0,5Nd³⁺-1MoO₄²⁻, logK(298,15K) = 8,94 +/- 0,13, Reference: /GMR1992/
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark: logK-value valid for 22C
- 200 Nd(OH)₃

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1298712,075 +/- unknown, Reaction: 0 = +1Nd(OH)₃-
1Nd³⁺-3OH⁻, logK(298,15K) = 27,15 +/- 0,3, Reference: SM8097
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 201 Nd₂(CO₃)₃
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3124981,78 +/- unknown, Reaction: 0 = +1Nd₂(CO₃)₃-
2Nd³⁺-3(CO₃)²⁻, logK(298,15K) = 34,6 +/- 0,1, Reference: SM8097
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: logK: Alternatively 34,8
- 202 NdO(HCO₃)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1479796,991 +/- unknown, Reaction: 0 = +1NdO(HCO₃)-
1Nd³⁺-1OH⁻-1(CO₃)²⁻, logK(298,15K) = 21,5 +/- 0,3, Reference: SM8097
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: logK: Alternatively 21,65
- 203 alpha-NiS
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = 8733,305 +/- 1007, Reaction: 0 = +1alpha-NiS,
logK(298,15K) = -1,53 +/- 0,3, Reference:
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 204 alpha-NiSO₄·6H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2225871,075 +/- 1059, Reaction: 0 = +1alpha-NiSO₄·6H₂O-
1Ni²⁺-1(SO₄)²⁻-6H₂O, logK(298,15K) = 2,322 +/- , Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 205 beta-Ni(OH)₂
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = 62902,631 +/- 1400, Reaction: 0 = +1beta-Ni(OH)₂,
logK(298,15K) = -11,02 +/- 0,2, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 206 beta-NiS
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = 11986,89 +/- 1007, Reaction: 0 = +1beta-NiS,
logK(298,15K) = -2,1 +/- 0,3, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 207 beta-NiSO₄·6H₂O
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -2224915 +/- 1815, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 208 K₂Ni(SO₄)₂·6H₂O(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3560455,69 +/- unknown, Reaction: 0 =
+1K₂Ni(SO₄)₂·6H₂O(cr)-2K⁺-1Ni²⁺-2(SO₄)²⁻, logK(298,15K) = 6,8 +/- 0,1,
Reference: Koda

- V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 209 Na₂Ni(SO₄)₂·4H₂O(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -3026168,069 +/- unknown, Reaction: 0 = +1Na₂Ni(SO₄)₂·4H₂O(cr)-2Na⁺-1Ni²⁺-2(SO₄)²⁻, logK(298,15K) = 3,49 +/- 0,03, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 210 Ni(CO₃)(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -636416,414 +/- 1292, Reaction: 0 = +1Ni(CO₃)(cr)+2H⁺-1Ni²⁺-1H₂O-1CO₂(g), logK(298,15K) = -7,16 +/- 0,18, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 211 Ni(CO₃):5,5H₂O(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -1920879,506 +/- 1004, Reaction: 0 = +1Ni(CO₃):5,5H₂O(cr)+2H⁺-1Ni²⁺-6,5H₂O-1CO₂(g), logK(298,15K) = -10,63 +/- 0,1, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 212 Ni(cr)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = 0 +/- 0, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 213 Ni(OH)₂(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -460118,551 +/- unknown, Reaction: 0 = +1Ni(OH)₂(cr)-1Ni²⁺-2H₂O+2H⁺, logK(298,15K) = -10,5 +/- 1,3, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 214 Ni₃S₂(cr)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -211172 +/- 1624, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 215 Ni₉S₈(cr)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -746803 +/- 9255, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 216 NiCl₂(cr)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -258743 +/- 154, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 217 NiCl₂·2H₂O(cr)
Gibbs Free Energy of Formation calculated from logK for reaction

ΔG_0^f [J mol⁻¹] = -753430,642 +/- 1256, Reaction: 0 = +1NiCl₂:2H₂O(cr)+2H₂O(g)-1NiCl₂:4H₂O(cr), logK(298,15K) = -4,099 +/- 0,123, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

218 NiCl₂:4H₂O(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1233991,91 +/- 1040, Reaction: 0 = +1NiCl₂:4H₂O(cr)-1Ni²⁺+2Cl⁻-4H₂O, logK(298,15K) = -3,99 +/- unknown, Reference: Koda
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: log K (NEA 6): -3,777±0,105

219 NiCl₂:6H₂O(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1713421,706 +/- 846, Reaction: 0 = +1NiCl₂:6H₂O(cr)-1Ni²⁺+2Cl⁻-6H₂O, logK(298,15K) = -3,0878 +/- unknown, Reference: /BC1999/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark: log K (NEA 6): -3,045±0,014

220 NiO(cr)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -211660 +/- 422, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

221 NiS₂(cr)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -123832 +/- 7396, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

222 NiSO₄(cr)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -762688 +/- 1572, Reference: NEA-6
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

223 NiSO₄:7H₂O(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2463011,075 +/- 929, Reaction: 0 = +1NiSO₄:7H₂O(cr)-1Ni²⁺+1(SO₄)²⁻-7H₂O, logK(298,15K) = 2,322 +/- unknown, Reference: /BC1999/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

224 NiSe₂
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -112355 +/- 7017, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

225 NiSeO₃:2H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -915549,96 +/- 6022, Reaction: 0 = +1NiSeO₃:2H₂O-1Ni²⁺+1SeO₃²⁻-2H₂O, logK(298,15K) = 5,8 +/- 1, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:

226 NiSeO₄:6H₂O
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -1915980,732 +/- unknown, Reaction: 0 = +1NiSeO4:6H2O-1Ni<2+>-1SeO4<2->-6H2O, logK(298,15K) = 1,381 +/- 0,045, Reference: NEA-7
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 227 Ni2(SiO4)(oliv)
 Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1288441 +/- 5002, Reference: NEA-6
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 228 Np|+IV|(OH)4(am)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1444404,976 +/- unknown, Reaction: 0 = +1Np|+IV|(OH)4(am)-1Np|+IV|<4+>-4OH<->, logK(298,15K) = 56,7 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 229 (Np|+V|O2)(OH)(am,aged)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1118077,199 +/- 6310, Reaction: 0 = +1(Np|+V|O2)(OH)(am,aged)+1H<+>-1H2O-1(Np|+V|O2)<+>, logK(298,15K) = -4,7 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 230 (Np|+V|O2)(OH)(am,fresh)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1114652,374 +/- 5743, Reaction: 0 = +1(Np|+V|O2)(OH)(am,fresh)+1H<+>-1H2O-1(Np|+V|O2)<+>, logK(298,15K) = -5,3 +/- 0,2, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 231 K(Np|+V|O2)(CO3)(s)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1793163,596 +/- 5746, Reaction: 0 = +1K(Np|+V|O2)(CO3)(s)-1(Np|+V|O2)<+>-1(CO3)<2->-1K<+>, logK(298,15K) = 13,15 +/- 0,19, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 232 K3(Np|+V|O2)(CO3)2(s)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2899197,008 +/- 5765, Reaction: 0 = +1K3(Np|+V|O2)(CO3)2(s)-1(Np|+V|O2)<+>-2(CO3)<2->-3K<+>, logK(298,15K) = 15,46 +/- 0,16, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 233 Na(Np|+V|O2)(CO3):3,5H2O(c)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2590324,304 +/- 5808, Reaction: 0 = +1Na(Np|+V|O2)(CO3):3,5H2O(c)-1(Np|+V|O2)<+>-1(CO3)<2->-3,5H2O-1Na<+>, logK(298,15K) = 11 +/- 0,24, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 234 Na3(Np|+V|O2)(CO3)2(c)

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2830448,035 +/- 6365, Reaction: 0 =
+1Na3(Np|+V|O2)(CO3)2(c)-1(Np|+V|O2)<+>-2(CO3)<2->-3Na<+>, logK(298,15K) =
14,22 +/- 0,5, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 235 Np|+V|2O5(c)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -2031570 +/- 11226, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 236 Np|+V|O2,5(s,hyd)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1007517,296 +/- unknown, Reaction: 0 =
+1Np|+V|O2,5(s,hyd)+0,5H2O-1(Np|+V|O2)<+>-1OH<->, logK(298,15K) = 10,7 +/- 1,
Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = 24,212 +/- unknown, Reference: HbChemPhys
Remark: V0: chapter B-101, for NpO2 rho=11,11 and Mw=269,0g/mol
- 237 (Np|+VI|O2)(CO3)(s)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1407081,425 +/- 6233, Reaction: 0 =
+1(Np|+VI|O2)(CO3)(s)-1(CO3)<2->-1(Np|+VI|O2)<2+>, logK(298,15K) = 14,596 +/-
0,469, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 238 K4(Np|+VI|O2)(CO3)3(s)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3660177,397 +/- 7641, Reaction: 0 =
+1K4(Np|+VI|O2)(CO3)3(s)-4K<+>-1(Np|+VI|O2)(CO3)3<4->, logK(298,15K) = 7,033
+/- 0,876, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 239 Np|+VI|O3:H2O(c)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1238996,006 +/- 6062, Reaction: 0 =
+1Np|+VI|O3:H2O(c)+2H<+>-2H2O-1(Np|+VI|O2)<2+>, logK(298,15K) = -5,47 +/- 0,4,
Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 240 PbSe
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -97936 +/- 7694, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 241 PbSeO3
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -457978,846 +/- 5985, Reaction: 0 = +1PbSeO3-1Pb<2+>-
1SeO3<2->, logK(298,15K) = 12,5 +/- 1, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 242 PbSeO4
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -503108,42 +/- 2060, Reaction: 0 = +1PbSeO4-1Pb<2+>-1SeO4<2->, logK(298,15K) = 6,9 +/- 0,25, Reference: NEA-7
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 243 Pu|+III|(OH)3(c)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1200216,925 +/- 8975, Reaction: 0 = +1Pu|+III|(OH)3(c)+3H<+>-3H2O-1Pu|+III|<3+>, logK(298,15K) = -15,8 +/- 1,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 244 Pu|+III|(PO4)(s,hyd)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1744892,851 +/- 5528, Reaction: 0 = +1Pu|+III|(PO4)(s,hyd)-1(PO4)<3->-1Pu|+III|<3+>, logK(298,15K) = 24,6 +/- 0,8, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 245 Pu|+IV|(HPO4)2(am,hyd)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2843767,901 +/- unknown, Reaction: 0 = +1Pu|+IV|(HPO4)2(am,hyd)-2(HPO4)<2->-1Pu|+IV|<4+>, logK(298,15K) = 30,45 +/- 0,51, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 246 Pu|+IV|(OH)4(am)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1439923,086 +/- unknown, Reaction: 0 = +1Pu|+IV|(OH)4(am)-4OH<->-1Pu|+IV|<4+>, logK(298,15K) = 58,33 +/- 0,52, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark: logK: in INE-Report from 2000: 58,5
- 247 (Pu|+V|O2)(OH)(am)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1061248,314 +/- 4046, Reaction: 0 = +1(Pu|+V|O2)(OH)(am)+1H<+>-1H2O-1(Pu|+V|O2)<+>, logK(298,15K) = -5 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 248 (Pu|+VI|O2)(CO3)(s)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1373803,66 +/- 3912, Reaction: 0 = +1(Pu|+VI|O2)(CO3)(s)-1(CO3)<2->-1(Pu|+VI|O2)<2+>, logK(298,15K) = 14,65 +/- 0,47, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 249 (Pu|+VI|O2)(OH)2:H2O(c)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1442378,765 +/- 6368, Reaction: 0 = +1(Pu|+VI|O2)(OH)2:H2O(c)+2H<+>-3H2O-1(Pu|+VI|O2)<2+>, logK(298,15K) = -5,5 +/- 1, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:

- 250 RaCO3(cr)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1136794 +/- unknown, Reference: NTB 02-16
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 251 RaSO4(cr)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1364069 +/- unknown, Reference: NTB 02-16
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 252 3MgSeO4:2Na2SeO4:18H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -8902371,315 +/- unknown, Reaction: 0 = +13MgSeO4:2Na2SeO4:18H2O-4Na<+>-3Mg<2+>-5SeO4<2->-18H2O, logK(298,15K) = 3,94 +/- unknown, Reference: I/SE-AB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 253 CaSeO3
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -950586,177 +/- unknown, Reaction: 0 = +1CaSeO3-1Ca<2+>-1SeO3<2->, logK(298,15K) = 6,2 +/- 0,3, Reference: I/SE-AB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 254 CaSeO4:2H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1481868,48 +/- 2300, Reaction: 0 = +1CaSeO4:2H2O-1Ca<2+>-1SeO4<2->-2H2O, logK(298,15K) = 2,68 +/- 0,25, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 255 K2Mg(SeO4)2:6H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3338575,517 +/- unknown, Reaction: 0 = +1K2Mg(SeO4)2:6H2O-2K<+>-1Mg<2+>-2SeO4<2->-6H2O, logK(298,15K) = 2,868 +/- unknown, Reference: I/SE-AB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 256 K2SeO4
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -997158,674 +/- unknown, Reaction: 0 = +1K2SeO4-2K<+>-1SeO4<2->, logK(298,15K) = -1,287 +/- unknown, Reference: I/SE-AB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 257 MgSeO3
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -851442,764 +/- unknown, Reaction: 0 = +1MgSeO3-1Mg<2+>-1SeO3<2->, logK(298,15K) = 5,9 +/- 0,3, Reference: I/SE-AB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 258 MgSeO4:6H2O
Gibbs Free Energy of Formation calculated from logK for reaction

- deltaG₀^f [J mol⁻¹] = -2324349,795 +/- 2000, Reaction: 0 = +1MgSeO₄:6H₂O-1Mg<2+>-1SeO₄<2->-6H₂O, logK(298,15K) = 1,165 +/- unknown, Reference: I/SE-AB
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 259 Na₂SeO₃
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = 10731,12 +/- 967, Reaction: 0 = +1Na₂SeO₃, logK(298,15K) = -1,88 +/- unknown, Reference: I/SE-AB
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 260 Na₂SeO₃:5H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -2067771,79 +/- unknown, Reaction: 0 = +1Na₂SeO₃:5H₂O-2Na<+>-1SeO₄<2->-5H₂O, logK(298,15K) = -0,7401 +/- unknown, Reference: I/SE-AB
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 261 Na₂SeO₄(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -958583,612 +/- unknown, Reaction: 0 = +1Na₂SeO₄(cr)-2Na<+>-1SeO₄<2->, logK(298,15K) = -0,8422 +/- unknown, Reference: I/SE-AB
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 262 Na₂SeO₄:10H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -3338186,64 +/- 1600, Reaction: 0 = +1Na₂SeO₄:10H₂O-2Na<+>-1SeO₄<2->-10H₂O, logK(298,15K) = 0,5949 +/- unknown, Reference: I/SE-AB
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 263 Se(cr)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = 0 +/- 0, Reference: NEA-7
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 264 Se(mono)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = 1281 +/- 184, Reference: NEA-7
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 265 SeCl₄
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -85902 +/- 3114, Reference: NEA-7
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 266 SeO₂(cr)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -171797 +/- 620, Reference: NEA-7
V₀ [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 267 SeO₃(cr)
Gibbs Free Energy of Formation directly entered

- deltaG₀^f [J mol⁻¹] = -86154 +/- 2222, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 268 SrSeO3
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -962214,981 +/- 3400, Reaction: 0 = +1SrSeO3-1Sr<2+>-
1SeO3<2->, logK(298,15K) = 6,3 +/- 0,5, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 269 USE
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -273255 +/- 18006, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 270 alpha-ZnSe
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -172488 +/- 4013, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 271 ZnSeO4:6H2O
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -2018306,895 +/- 1524, Reaction: 0 = +1ZnSeO4:6H2O-
1Zn<2+>-1SeO4<2->-6H2O, logK(298,15K) = 1,538 +/- 0,068, Reference: NEA-7
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 272 SnO2(cr)_Cassiterite
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -519615,416 +/- , Reaction: 0 = +1SnO2(cr)_Cassiterite-
0,5O2(aq)-1H2O-1Sn<2+>+2H<+>, logK(298,15K) = 46,1642 +/- , Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 273 SnS_Herzenbergite
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -104171,468 +/- unknown, Reaction: 0 =
+1SnS_Herzenbergite-1HS<->-1Sn<2+>+1H<+>, logK(298,15K) = 15,6226 +/- ,
Reference: Min-GWB
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 274 SrCO3(s)
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG₀^f [J mol⁻¹] = -1144669,891 +/- unknown, Reaction: 0 =
+1SrCO3(s)+1H<+>-1Sr<2+>-1(HCO3)<->, logK(298,15K) = -1,058 +/- unknown,
Reference: /PSI/NAGRA/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 275 K(TcO4)(c)
Gibbs Free Energy of Formation directly entered
deltaG₀^f [J mol⁻¹] = -932921 +/- 7604, Reference: FZK-INE 002/04
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 276 TcO2:1,6H2O(s)

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -758479,548 +/- 8372, Reaction: 0 = +1TcO2:1,6H2O(s)+0,4H2O-1(TcO4)<->-1,5H2-1H<+>, logK(298,15K) = 37,829 +/- 0,609, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 277 Th(OH)4(am)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1601862,962 +/- unknown, Reaction: 0 = +1Th(OH)4(am)-1Th<4+>-4OH<->, logK(298,15K) = 47 +/- 0,8, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 278 UO2(OH)2:2H2O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1634859,982 +/- unknown, Reaction: 0 = +1UO2(OH)2:2H2O+2H<+>-1(U|+VI|O2)<2+>-3H2O, logK(298,15K) = -5,1 +/- unknown, Reference:
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 279 U|+IV|(OH)2(SO4)(c)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1766276,168 +/- 3385, Reaction: 0 = +1U|+IV|(OH)2(SO4)(c)-2OH<->-1(SO4)<2->-1U|+IV|<4+>, logK(298,15K) = 31,17 +/- 0,5, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 280 U|+IV|(OH)4(am)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1469933,282 +/- unknown, Reaction: 0 = +1U|+IV|(OH)4(am)-4OH<->-1U|+IV|<4+>, logK(298,15K) = 54,5 +/- 1, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 281 U|+IV|(SiO4)(c)_Coffinit
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1885022,184 +/- 4000, Reaction: 0 = +1U|+IV|(SiO4)(c)_Coffinit+4H<+>-1U|+IV|<4+>-1SiO2<0>-2H2O, logK(298,15K) = 8,06 +/- 0,77, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 282 (U|+VI|O2)(CO3)(c)_Rutherfordit
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1564629,544 +/- 1794, Reaction: 0 = +1(U|+VI|O2)(CO3)(c)_Rutherfordit-1(CO3)<2->-1(U|+VI|O2)<2+>, logK(298,15K) = 14,76 +/- 0,02, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 283 (U|+VI|O2)3(PO4)2:4H2O(c)
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -6138966,935 +/- 6355, Reaction: 0 = +1(U|+VI|O2)3(PO4)2:4H2O(c)+6H<+>-4H2O-2H3PO4<0>-3(U|+VI|O2)<2+>, logK(298,15K) = 5,96 +/- 0,3, Reference: FZK-INE 002/04
 V0 [cm3 mol⁻¹] = dummy value 0,001

- Remark:
- 284 Ca(U+VI|6019):11H2O(c)_Bequerelit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -10305456,27 +/- 13964, Reaction: 0 =
+1Ca(U+VI|6019):11H2O(c)_Bequerelit+14H<+>-18H2O-6(U+VI|O2)<2+>-1Ca<2+>,
logK(298,15K) = -40,5 +/- 1,6, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 285 K2(U+VI|6019):11H2O(c)_Compreignacit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -10337077,615 +/- 10956, Reaction: 0 =
+1K2(U+VI|6019):11H2O(c)_Compreignacit+14H<+>-18H2O-2K<+>-6(U+VI|O2)<2+>,
logK(298,15K) = -37,1 +/- 0,54, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 286 Na(U+VI|O2)(OH)3(c)_Clarkeit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1619202,388 +/- unknown, Reaction: 0 =
+1Na(U+VI|O2)(OH)3(c)_Clarkeit+1H2O-1Na<+>-1(U+VI|O2)<2+>-3OH<->,
logK(298,15K) = 29,8 +/- 0,2, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 287 Na2(U+VI|207)(c)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -3011450 +/- 4015, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 288 Na4(U+VI|O2)(CO3)3(c)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3738991,103 +/- 2342, Reaction: 0 =
+1Na4(U+VI|O2)(CO3)3(c)-1(U+VI|O2)(CO3)3<4->-4Na<+>,
logK(298,15K) = 5,34 +/- 0,16, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 289 U|+VI|O3:2H2O(c)_Metaschoepit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1633470,645 +/- unknown, Reaction: 0 =
+1U|+VI|O3:2H2O(c)_Metaschoepit-1(U+VI|O2)<2+>-2OH<->-1H2O,
logK(298,15K) = 22,65 +/- 0,13, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 290 U|+VI|O3:2H2O(c)_Schoepit
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1636510 +/- 1705, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 291 (U+VI|O2)2(SiO4):2H2O(c)_Soddyit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3653147,495 +/- unknown, Reaction: 0 =
+1(U+VI|O2)2(SiO4):2H2O(c)_Soddyit+4H<+>-2(U+VI|O2)<2+>-1SiO2<0>-2H2O-2H2O,
logK(298,15K) = -6,2 +/- 1, Reference: FZK-INE 002/04
V0 [cm3 mol⁻¹] = dummy value 0,001

- Remark:
- 292 Ca(U|+VI|O2)2(SiO3OH)2:5H2O(c)_Uranophan
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -6208148,958 +/- unknown, Reaction: 0 = +1Ca(U|+VI|O2)2(SiO3OH)2:5H2O(c)_Uranophan+6H<+>-5H2O-1Ca<2+>-2(U|+VI|O2)<2+>-2SiO2<0>, logK(298,15K) = -9,42 +/- 0,48, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 293 Na2(U|+VI|O2)2(Si2O5)3:4H2O(c)_Na-weeksite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -9089678,096 +/- unknown, Reaction: 0 = +1Na2(U|+VI|O2)2(Si2O5)3:4H2O(c)_Na-weeksite+5H2O+6H<+>-2Na<+>-2(U|+VI|O2)<2+>-6SiO2<0>, logK(298,15K) = -1,5 +/- 0,08, Reference: FZK-INE 002/04
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 294 Al2O3_Corundum
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1583736,215 +/- unknown, Reaction: 0 = +1Al2O3_Corundum+6H<+>-2Al<3+>-3H2O, logK(298,15K) = -19,3933 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 25,575 +/- unknown, Reference: YPF
 Remark:
- 295 AlOOH_Boehmite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -916689,841 +/- unknown, Reaction: 0 = +1AlOOH_Boehmite+3H<+>-1Al<3+>-2H2O, logK(298,15K) = -8,6014 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 19,535 +/- unknown, Reference: YPF
 Remark:
- 296 Beidellite-Ca_Ca,165Al2,33Si3,67O10(OH)2
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5377318,939 +/- unknown, Reaction: 0 = +1Beidellite-Ca_Ca,165Al2,33Si3,67O10(OH)2+7,32H<+>-0,165Ca<2+>-2,33Al<3+>-3,67SiO2<0>-4,66H2O, logK(298,15K) = -4,9352 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 129,766 +/- unknown, Reference: YPF
 Remark:
- 297 Beidellite-H_H,33Al2,33Si3,67O10(OH)2
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5291573,683 +/- unknown, Reaction: 0 = +1Beidellite-H_H,33Al2,33Si3,67O10(OH)2+6,99H<+>-2,33Al<3+>-3,67SiO2<0>-4,66H2O, logK(298,15K) = -3,9773 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 128,913 +/- unknown, Reference: YPF
 Remark:
- 298 Beidellite-K_K,33Al2,33Si3,67O10(OH)2
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5380949,625 +/- unknown, Reaction: 0 = +1Beidellite-K_K,33Al2,33Si3,67O10(OH)2+7,32H<+>-0,33K<+>-2,33Al<3+>-3,67SiO2<0>-4,66H2O, logK(298,15K) = -4,6522 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 134,152 +/- unknown, Reference: YPF
 Remark:
- 299 Beidellite-Mg_Mg,165Al2,33Si3,67O10(OH)2
 Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -5361460,3 +/- unknown, Reaction: 0 = +1Beidellite-Mg_Mg,165Al₂,33Si₃,67O₁₀(OH)₂+7,32H⁺-0,165Mg²⁺-2,33Al³⁺-3,67SiO₂<0>-4,66H₂O, logK(298,15K) = -4,8971 +/- unknown, Reference: YPF
V₀ [cm³ mol⁻¹] = 128,725 +/- unknown, Reference: YPF
Remark:
- 300 Beidellite-Na_Na,33Al₂,33Si₃,67O₁₀(OH)₂
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5372231,359 +/- unknown, Reaction: 0 = +1Beidellite-Na_Na,33Al₂,33Si₃,67O₁₀(OH)₂+7,32H⁺-0,33Na⁺-2,33Al³⁺-3,67SiO₂<0>-4,66H₂O, logK(298,15K) = -4,9911 +/- unknown, Reference: YPF
V₀ [cm³ mol⁻¹] = 130,727 +/- unknown, Reference: YPF
Remark:
- 301 Chabazite_K0,6Na0,2Ca1,55Al3,8Si8,2024:10,0H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -13879273,758 +/- unknown, Reaction: 0 = +1Chabazite_K0,6Na0,2Ca1,55Al₃,8Si₈,2024:10,0H₂O+15,2H⁺-0,2Na⁺-1,5Ca²⁺-3,8Al³⁺-8,2SiO₂<0>-0,6K⁺, logK(298,15K) = -10,3714 +/- unknown, Reference: YPF
V₀ [cm³ mol⁻¹] = 499,4 +/- unknown, Reference: YPF
Remark:
- 302 Chamosite-7A_Fe2Al2SiO5(OH)4
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3472687,847 +/- unknown, Reaction: 0 = +1Chamosite-7A_Fe2Al2SiO5(OH)4+10H⁺-1SiO₂<0>-2Al³⁺-2Fe²⁺-7H₂O, logK(298,15K) = -32,6174 +/- unknown, Reference: YPF
V₀ [cm³ mol⁻¹] = 106,2 +/- unknown, Reference: YPF
Remark:
- 303 Clinoptilolite-Ca_Ca1,7335Al3,45Fe,017Si14,533O36:10,922H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -19052582,422 +/- unknown, Reaction: 0 = +1Clinoptilolite-Ca_Ca1,7335Al₃,45Fe,017Si₁₄,533O₃₆:10,922H₂O+13,868H⁺-0,017Fe³⁺-1,7335Ca²⁺-3,45Al³⁺-14,533SiO₂<0>, logK(298,15K) = 5,6428 +/- unknown, Reference: YPF
V₀ [cm³ mol⁻¹] = 632,05 +/- unknown, Reference: YPF
Remark:
- 304 Clinoptilolite-Cs_Cs3,467Al3,45Fe,017Si14,533O36:10,922H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -19139295,699 +/- unknown, Reaction: 0 = +1Clinoptilolite-Cs_Cs3,467Al₃,45Fe,017Si₁₄,533O₃₆:10,922H₂O+13,868H⁺-0,017Fe³⁺-3,467Cs⁺-3,45Al³⁺-14,533SiO₂<0>, logK(298,15K) = 11,6912 +/- unknown, Reference: YPF
V₀ [cm³ mol⁻¹] = 632,05 +/- unknown, Reference: YPF
Remark:
- 305 Clinoptilolite-K_K3,467Al3,45Fe,017Si14,533O36:10,922H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -19096239,942 +/- unknown, Reaction: 0 = +1Clinoptilolite-K_K3,467Al₃,45Fe,017Si₁₄,533O₃₆:10,922H₂O+13,868H⁺-0,017Fe³⁺-3,467K⁺-3,45Al³⁺-14,533SiO₂<0>, logK(298,15K) = 9,5819 +/- unknown, Reference: YPF
V₀ [cm³ mol⁻¹] = 632,05 +/- unknown, Reference: YPF
Remark:
- 306 Clinoptilolite-Na_Na3,467Al3,45Fe,017Si14,533O36:10,922H2O
Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -19003208,052 +/- unknown, Reaction: 0 = +1Clinoptilolite-Na₃,467Al₃,45Fe,017Si₁₄,533O₃₆:10,922H₂O+13,868H<+>-0,017Fe<3+>-3,467Na<+>-3,45Al<3+>-14,533SiO₂<0>, logK(298,15K) = 5,7696 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 632,05 +/- unknown, Reference: YPF
 Remark:
- 307 Clinoptilolite-NH₄(NH₄)₃,467Al₃,45Fe,017Si₁₄,533O₃₆:10,922H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -18389152,665 +/- unknown, Reaction: 0 = +1Clinoptilolite-NH₄(NH₄)₃,467Al₃,45Fe,017Si₁₄,533O₃₆:10,922H₂O+13,868H<+>-0,017Fe<3+>-3,467(NH₄)<+>-3,45Al<3+>-14,533SiO₂<0>, logK(298,15K) = 9,0742 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 308 Cronstedtite-7A₂Fe₂SiO₅(OH)₄
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2431884,223 +/- unknown, Reaction: 0 = +1Cronstedtite-7A₂Fe₂SiO₅(OH)₄+10H<+>-1SiO₂<0>-2Fe<2+>-2Fe<3+>-7H₂O, logK(298,15K) = -17,3756 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 110,9 +/- unknown, Reference: YPF
 Remark:
- 309 Daphnite-7A₂Fe₅Al₃Si₃O₁₀(OH)₈
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -6471810,946 +/- unknown, Reaction: 0 = +1Daphnite-7A₂Fe₅Al₃Si₃O₁₀(OH)₈+16H<+>-2Al<3+>-3SiO₂<0>-5Fe<2+>-12H₂O, logK(298,15K) = -55,0117 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 221,2 +/- unknown, Reference: YPF
 Remark:
- 310 Erionite_K₁,5Na₀,9Ca₀,9Al₄,2Si₁₃,8O₃₆:13,0H₂O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -19845021,031 +/- unknown, Reaction: 0 = +1Erionite_K₁,5Na₀,9Ca₀,9Al₄,2Si₁₃,8O₃₆:13,0H₂O+16,8H<+>-1,5K<+>-4,2Al<3+>-0,9Ca<2+>-13,8SiO₂<0>-0,9Na<+>, logK(298,15K) = 4,8296 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 672 +/- unknown, Reference: YPF
 Remark:
- 311 Fe₃Si₂O₅(OH)₄_Greenalite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2994738,751 +/- unknown, Reaction: 0 = +1Fe₃Si₂O₅(OH)₄_Greenalite+6H<+>-2SiO₂<0>-3Fe<2+>-5H₂O, logK(298,15K) = -23,1624 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 115 +/- unknown, Reference: YPF
 Remark:
- 312 Ferroaluminoceladonite_KFeAlSi₄O₁₀(OH)₂
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5126466,999 +/- unknown, Reaction: 0 = +1Ferroaluminoceladonite_KFeAlSi₄O₁₀(OH)₂+6H<+>-1Al<3+>-1K<+>-1Fe<2+>-4H₂O, logK(298,15K) = -4,5745 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 153,678 +/- unknown, Reference: YPF
 Remark:
- 313 Ferroceladonite_KFeFeSi₄O₁₀(OH)₂
 Gibbs Free Energy of Formation calculated from logK for reaction

- ΔG_0^f [J mol⁻¹] = -4606065,186 +/- unknown, Reaction: 0 = +1Ferroceladonite_KFeFeSi4O10(OH)2+6H<+>-1Fe<3+>-1K<+>-1Fe<2+>-4H2O, logK(298,15K) = 3,0464 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = 156,72 +/- unknown, Reference: YPF
Remark:
- 314 Illite_K0,6Mg0,25Al1,8Al0,5Si3,5O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5473799,868 +/- unknown, Reaction: 0 = +1Illite_K0,6Mg0,25Al1,8Al0,5Si3,5O10(OH)2+8H<+>-0,25Mg<2+>-0,6K<+>-2,3Al<3+>-3,5SiO2<0>, logK(298,15K) = -8,3706 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = 139,346 +/- unknown, Reference: YPF
Remark:
- 315 K2Si4O9
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4058816,05 +/- unknown, Reaction: 0 = +1K2Si4O9+2H<+>-2K<+>-4SiO2<0>-1H2O, logK(298,15K) = -14,5138 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 316 K3AlCl6
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1945935,168 +/- unknown, Reaction: 0 = +1K3AlCl6-3K<+>-1Al<3+>-6Cl<->, logK(298,15K) = -31,6052 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 317 KAl(SO4)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2246653,812 +/- unknown, Reaction: 0 = +1KAl(SO4)2-1K<+>-1Al<3+>-2(SO4)<2->, logK(298,15K) = -2,6929 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 318 KAl(SO4)2:12H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5147168,695 +/- unknown, Reaction: 0 = +1KAl(SO4)2:12H2O-1K<+>-1Al<3+>-2(SO4)<2->-12H2O, logK(298,15K) = 6,9137 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 319 KAl(SO4)2:3H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2981412,857 +/- unknown, Reaction: 0 = +1KAl(SO4)2:3H2O-1K<+>-1Al<3+>-2(SO4)<2->-3H2O, logK(298,15K) = 1,3959 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 320 KAl3(OH)6(SO4)2_Alunite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4681588,577 +/- unknown, Reaction: 0 = +1KAl3(OH)6(SO4)2_Alunite+6H<+>-1K<+>-2(SO4)<2->-3Al<3+>-6H2O, logK(298,15K) = 2,4018 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = 293,6 +/- unknown, Reference: YPF
Remark:
- 321 KAlCl4

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1102052,851 +/- unknown, Reaction: 0 = +1KAlCl4-1K<+>-1Al<3+>-4Cl<->, logK(298,15K) = -34,4833 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = dummy value 0,001
 Remark:
- 322 Kaolinite_Al2Si2O5(OH)4
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3804479,604 +/- unknown, Reaction: 0 = +1Kaolinite_Al2Si2O5(OH)4+6H<+>-2Al<3+>-2SiO2<0>-5H2O, logK(298,15K) = -5,9539 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 99,52 +/- unknown, Reference: YPF
 Remark:
- 323 KMgAlSi4O10(OH)2_Celadonite
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5472718,368 +/- unknown, Reaction: 0 = +1KMgAlSi4O10(OH)2_Celadonite+6H<+>-1Al<3+>-1K<+>-1Mg<2+>-4H2O, logK(298,15K) = -7,8372 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 157,1 +/- unknown, Reference: YPF
 Remark:
- 324 Laumontite_K0,2Na0,2Ca1,8Al4Si8,0O24:8,0H2O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -13461785,055 +/- unknown, Reaction: 0 = +1Laumontite_K0,2Na0,2Ca1,8Al4Si8,0O24:8,0H2O+16H<+>-1,8Ca<2+>-4Al<3+>-0,2K<+>-8SiO2<0>-0,2Na<+>, logK(298,15K) = -14,2657 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 406,4 +/- unknown, Reference: YPF
 Remark:
- 325 Maximum-Microcline_KAlSi3O8
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3754009,321 +/- unknown, Reaction: 0 = +1Maximum-Microcline_KAlSi3O8+4H<+>-1Al<3+>-1K<+>-2H2O-3SiO2<0>, logK(298,15K) = 0,1903 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 108,741 +/- unknown, Reference: YPF
 Remark: together with K-Feldspar questionable
- 326 Mesolite_Na,676Ca,657Al1,99Si3,01O10:2,647H2O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5528504,225 +/- unknown, Reaction: 0 = +1Mesolite_Na,676Ca,657Al1,99Si3,01O10:2,647H2O+7,96H<+>-0,657Ca<2+>-0,676Na<+>-1,99Al<3+>-3,01SiO2<0>, logK(298,15K) = -13,029 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 171,7 +/- unknown, Reference: YPF
 Remark:
- 327 Mg2Al2SiO5(OH)4_Amesit-7A
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4165191,156 +/- unknown, Reaction: 0 = +1Mg2Al2SiO5(OH)4_Amesit-7A+10H<+>-1SiO2<0>-2Al<3+>-2Mg<2+>-7H2O, logK(298,15K) = -39,1427 +/- unknown, Reference: YPF
 V0 [cm³ mol⁻¹] = 103 +/- unknown, Reference: YPF
 Remark:
- 328 Mg4Al4Si2O10(OH)8_Amesit-14A
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -8368892,764 +/- unknown, Reaction: 0 = +1Mg4Al4Si2O10(OH)8_Amesit-14A+20H<+>-2SiO2<0>-4Al<3+>-4Mg<2+>-14H2O, logK(298,15K) = -71,5387 +/- unknown, Reference: YPF

- V0 [cm3 mol-1] = 205,4 +/- unknown, Reference: YPF
Remark:
- 329 Minnesotaite-Fe3Si4O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -4473939,658 +/- unknown, Reaction: 0 =
+1Minnesotaite-Fe3Si4O10(OH)2+6H<+>-3Fe<2+>-4H2O-4SiO2<0>, logK(298,15K) = -
15,0002 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 147,86 +/- unknown, Reference: YPF
Remark:
- 330 Montmorillonite-Ca_Ca0,165Mg0,33Al1,67Si4O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -5336651,868 +/- unknown, Reaction: 0 =
+1Montmorillonite-Ca_Ca0,165Mg0,33Al1,67Si4O10(OH)2+6H<+>-0,33Mg<2+>-4SiO2<0>-
1,67Al<3+>-4H2O, logK(298,15K) = -2,4024 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 133,07 +/- unknown, Reference: YPF
Remark:
- 331 Montmorillonite-H_H0,33Mg0,33Al1,67Si4O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -5250906,612 +/- unknown, Reaction: 0 =
+1Montmorillonite-H_H0,33Mg0,33Al1,67Si4O10(OH)2+5,67H<+>-0,33Mg<2+>-4SiO2<0>-
1,67Al<3+>-4H2O, logK(298,15K) = -1,4445 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 132,216 +/- unknown, Reference: YPF
Remark:
- 332 Montmorillonite-K_K0,33Mg0,33Al1,67Si4O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -5340282,554 +/- unknown, Reaction: 0 =
+1Montmorillonite-K_K0,33Mg0,33Al1,67Si4O10(OH)2+6H<+>-0,33Mg<2+>-4SiO2<0>-
1,67Al<3+>-4H2O, logK(298,15K) = -2,1194 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 137,455 +/- unknown, Reference: YPF
Remark:
- 333 Montmorillonite-Mg_Mg0,495Al1,67Si4O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -5320793,23 +/- unknown, Reaction: 0 =
+1Montmorillonite-Mg_Mg0,495Al1,67Si4O10(OH)2+6H<+>-0,495Mg<2+>-4SiO2<0>-
1,67Al<3+>-4H2O, logK(298,15K) = -2,3643 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 132,029 +/- unknown, Reference: YPF
Remark:
- 334 Montmorillonite-Na_Na0,33Mg0,33Al1,67Si4O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -5331564,289 +/- unknown, Reaction: 0 =
+1Montmorillonite-Na_Na0,33Mg0,33Al1,67Si4O10(OH)2+6H<+>-0,33Mg<2+>-4SiO2<0>-
1,67Al<3+>-4H2O, logK(298,15K) = -2,4583 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 134,03 +/- unknown, Reference: YPF
Remark:
- 335 Mordenite_Ca,2895Na,361Al,94Si5,06O12:3,468H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol-1] = -6235240,634 +/- unknown, Reaction: 0 =
+1Mordenite_Ca,2895Na,361Al,94Si5,06O12:3,468H2O+3,76H<+>-0,2895Ca<2+>-
0,361Na<+>-0,94Al<3+>-5,06SiO2<0>, logK(298,15K) = 4,5423 +/- unknown,
Reference: YPF
V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 336 Na,96Al,96Si2,04O6:H2O_Analcim

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3084635,342 +/- unknown, Reaction: 0 = +1Na,96Al,96Si2,0406:H2O_Analcim+3,84H<+>-0,96Al<3+>-0,96Na<+>-2,04SiO2<0>-2,92H2O, logK(298,15K) = -6,0057 +/- unknown, Reference: YPF
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 337 Na,96Al,96Si2,0406_Analcime-dehy
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2811176,208 +/- unknown, Reaction: 0 = +1Na,96Al,96Si2,0406_Analcime-dehy+3,84H<+>-0,96Al<3+>-0,96Na<+>-2,04SiO2<0>-1,92H2O, logK(298,15K) = -12,3685 +/- unknown, Reference: YPF
 V0 [cm3 mol⁻¹] = 89,1 +/- unknown, Reference: YPF
 Remark:
- 338 Nontronite-Ca_Ca,165Fe2Al,33Si3,67H2O12
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4341044,646 +/- unknown, Reaction: 0 = +1Nontronite-Ca_Ca,165Fe2Al,33Si3,67H2O12+7,32H<+>-0,165Ca<2+>-0,33Al<3+>-2Fe<3+>-3,67SiO2<0>, logK(298,15K) = 11,1001 +/- unknown, Reference: YPF
 V0 [cm3 mol⁻¹] = 135,85 +/- unknown, Reference: YPF
 Remark:
- 339 Nontronite-H_H,33Fe2Al,33Si3,67H2O12
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4259340,685 +/- unknown, Reaction: 0 = +1Nontronite-H_H,33Fe2Al,33Si3,67H2O12+6,99H<+>, logK(298,15K) = 12,766 +/- unknown, Reference: YPF
 V0 [cm3 mol⁻¹] = dummy value 0,001
 Remark:
- 340 Nontronite-K_K,33Fe2Al,33Si3,67H2O12
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4344675,332 +/- unknown, Reaction: 0 = +1Nontronite-K_K,33Fe2Al,33Si3,67H2O12+7,32H<+>-0,33K<+>-0,33Al<3+>-2Fe<3+>-3,67SiO2<0>, logK(298,15K) = 11,3831 +/- unknown, Reference: YPF
 V0 [cm3 mol⁻¹] = 140,235 +/- unknown, Reference: YPF
 Remark:
- 341 Nontronite-Mg_Mg,165Fe2Al,33Si3,67H2O12
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4325186,008 +/- unknown, Reaction: 0 = +1Nontronite-Mg_Mg,165Fe2Al,33Si3,67H2O12+7,32H<+>-0,165Mg<2+>-0,33Al<3+>-2Fe<3+>-3,67SiO2<0>, logK(298,15K) = 11,1382 +/- unknown, Reference: YPF
 V0 [cm3 mol⁻¹] = 134,809 +/- unknown, Reference: YPF
 Remark:
- 342 Nontronite-Na_Na,33Fe2Al,33Si3,67H2O12
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4335957,067 +/- unknown, Reaction: 0 = +1Nontronite-Na_Na,33Fe2Al,33Si3,67H2O12+7,32H<+>-0,33Na<+>-0,33Al<3+>-2Fe<3+>-3,67SiO2<0>, logK(298,15K) = 11,0442 +/- unknown, Reference: YPF
 V0 [cm3 mol⁻¹] = 136,81 +/- unknown, Reference: YPF
 Remark:
- 343 Phillipsite_K0,7Na0,7Ca1,1Al3,6Si12,4O32:12,6H2O
 Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -17845058,436 +/- unknown, Reaction: 0 = +1Phillipsite_K0,7Na0,7Ca1,1Al3,6Si12,4O32:12,6H2O+14,4H<+>-1,1Ca<2+>-

- 3,6Al<3+>-0,7K<+>-12,4SiO2<0>-0,7Na<+>, logK(298,15K) = 6,7617 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 609,2 +/- unknown, Reference: YPF
Remark:
- 344 Pyrophyllite_Al2Si4O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -5270523,472 +/- unknown, Reaction: 0 = +1Pyrophyllite_Al2Si4O10(OH)2+6H<+>-2Al<3+>-4H2O-4SiO2<0>, logK(298,15K) = -0,0967 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 126,6 +/- unknown, Reference: YPF
Remark:
- 345 Ripidolite-14A_Mg3Fe2Al2Si3O10(OH)8
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -7542655,67 +/- unknown, Reaction: 0 = +1Ripidolite-14A_Mg3Fe2Al2Si3O10(OH)8+16H<+>-2Al<3+>-2Fe<2+>-3Mg<2+>-3SiO2<0>, logK(298,15K) = -59,1778 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 209,634 +/- unknown, Reference: YPF
Remark: Phase should not be used simultaneously with ss-chlorite
- 346 Ripidolite-7A_Mg3Fe2Al2Si3O10(OH)8
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -7523400,729 +/- unknown, Reaction: 0 = +1Ripidolite-7A_Mg3Fe2Al2Si3O10(OH)8+16H<+>-2Al<3+>-2Fe<2+>-3Mg<2+>-3SiO2<0>, logK(298,15K) = -62,5511 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 215,38 +/- unknown, Reference: YPF
Remark: Phase should not be used simultaneously with ss-chlorite
- 347 Saponite-Ca_Ca,165Mg3Al,33Si3,67O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -5634464,852 +/- unknown, Reaction: 0 = +1Saponite-Ca_Ca,165Mg3Al,33Si3,67O10(OH)2+7,32H<+>-0,33Al<3+>-3Mg<2+>-3,67SiO2<0>-4,66H2O, logK(298,15K) = -27,0032 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 134,359 +/- unknown, Reference: YPF
Remark:
- 348 Saponite-H_H,33Mg3Al,33Si3,67O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -5548719,596 +/- unknown, Reaction: 0 = +1Saponite-H_H,33Mg3Al,33Si3,67O10(OH)2+6,99H<+>-0,33Al<3+>-3Mg<2+>-3,67SiO2<0>-4,66H2O, logK(298,15K) = -26,0453 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 133,506 +/- unknown, Reference: YPF
Remark:
- 349 Saponite-K_K,33Mg3Al,33Si3,67O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -5638095,538 +/- unknown, Reaction: 0 = +1Saponite-K_K,33Mg3Al,33Si3,67O10(OH)2+7,32H<+>-0,33Al<3+>-3Mg<2+>-3,67SiO2<0>-4,66H2O, logK(298,15K) = -26,7202 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 138,745 +/- unknown, Reference: YPF
Remark:
- 350 Saponite-Mg_Mg3,165Al,33Si3,67O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
deltaG_0^f [J mol-1] = -5618606,214 +/- unknown, Reaction: 0 = +1Saponite-Mg_Mg3,165Al,33Si3,67O10(OH)2+7,32H<+>-0,33Al<3+>-3,165Mg<2+>-3,67SiO2<0>-4,66H2O, logK(298,15K) = -26,9651 +/- unknown, Reference: YPF
V0 [cm3 mol-1] = 132,602 +/- unknown, Reference: YPF
Remark:

- 351 Saponite-Na_Na,33Mg3Al,33Si3,67O10(OH)2
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5629377,273 +/- unknown, Reaction: 0 = +1Saponite-Na_Na,33Mg3Al,33Si3,67O10(OH)2+7,32H<+>-0,33Al<3+>-3Mg<2+>-3,67SiO2<0>-4,66H2O, logK(298,15K) = -27,0591 +/- unknown, Reference: YPF
V0 [cm3 mol⁻¹] = 135,32 +/- unknown, Reference: YPF
Remark:
- 352 Scolecite_CaAl2Si3O10:3H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5613223,17 +/- unknown, Reaction: 0 = +1Scolecite_CaAl2Si3O10:3H2O+8H<+>-1Ca<2+>-2Al<3+>-3SiO2<0>-7H2O, logK(298,15K) = -15,2772 +/- unknown, Reference: YPF
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark: a Ca-zeolite (Natrolite = Na-zeolite)
- 353 Sepiolite_Mg4Si6O15(OH)2:6H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -9265517,078 +/- unknown, Reaction: 0 = +1Sepiolite_Mg4Si6O15(OH)2:6H2O+8H<+>-4Mg<2+>-6SiO2<0>-11H2O, logK(298,15K) = -30,4439 +/- unknown, Reference: YPF
V0 [cm3 mol⁻¹] = 285,6 +/- unknown, Reference: YPF
Remark:
- 354 Smectite-high-Fe-Mg_Ca,025Na,1K,2Fe++,5Fe+++,2Mg1,15Al1,25Si3,5H
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5273457,591 +/- unknown, Reaction: 0 = +1Smectite-high-Fe-Mg_Ca,025Na,1K,2Fe++,5Fe+++,2Mg1,15Al1,25Si3,5H+8H<+>-0,025Ca<2+>-0,1Na<+>-0,2Fe<3+>-0,2K<+>-1,15Mg<2+>-1,25Al<3+>-3,5SiO2<0>-5H2O, logK(298,15K) = -17,4595 +/- unknown, Reference: YPF
V0 [cm3 mol⁻¹] = 139,51 +/- unknown, Reference: YPF
Remark:
- 355 Smectite-low-Fe-Mg_Ca,02Na,15K,2Fe++,29Fe+++,16Mg,9Al1,25Si3,7
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5279978,603 +/- unknown, Reaction: 0 = +1Smectite-low-Fe-Mg_Ca,02Na,15K,2Fe++,29Fe+++,16Mg,9Al1,25Si3,7+7H<+>-0,02Ca<2+>-0,15Na<+>-0,16Fe<3+>-0,2K<+>-0,9Mg<2+>-1,25Al<3+>-3,75SiO2<0>-4,5H2O, logK(298,15K) = -11,1541 +/- unknown, Reference: YPF
V0 [cm3 mol⁻¹] = 139,602 +/- unknown, Reference: YPF
Remark:
- 356 Stellerite_Ca2,0Al4,0Si14,0O36:14,0H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -20027116,77 +/- unknown, Reaction: 0 = +1Stellerite_Ca2,0Al4,0Si14,0O36:14,0H2O+16H<+>-2Ca<2+>-4Al<3+>-22H2O-14SiO2<0>, logK(298,15K) = 8,7844 +/- unknown, Reference: YPF
V0 [cm3 mol⁻¹] = 665,5 +/- unknown, Reference: YPF
Remark:
- 357 Stilbite_Ca1,019Na,136K,006Al2,18Si6,82O18:7,33H2O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -10130643,987 +/- unknown, Reaction: 0 = +1Stilbite_Ca1,019Na,136K,006Al2,18Si6,82O18:7,33H2O+8,72H<+>-0,006K<+>-0,136Na<+>-1,019Ca<2+>-2,18Al<3+>-11,69H2O, logK(298,15K) = -1,3118 +/- unknown, Reference: YPF
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 358 Talc_Mg3Si4O10(OH)2

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5527669,956 +/- unknown, Reaction: 0 =
+1Talc_Mg3Si4O10(OH)2+6H<+>-3Mg<2+>-4H2O-4SiO2<0>, logK(298,15K) = -22,1646 +/-
unknown, Reference: YPF
V0 [cm3 mol⁻¹] = 136,25 +/- unknown, Reference: YPF
Remark:
- 359 C2AH8_Ca2Al2O5:8H2O_Dicalciumaluminathydrat
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4831760,378 +/- , Reaction: 0 =
+1C2AH8_Ca2Al2O5:8H2O_Dicalciumaluminathydrat+10H<+>-2Ca<2+>-2Al<3+>-13H2O,
logK(298,15K) = -59,51 +/- unknown, Reference: /RAR1997a/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 360 C2ASH8_Ca2Al2SiO7:8H2O_Gehlenithydrat
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5722802,878 +/- , Reaction: 0 =
+1C2ASH8_Ca2Al2SiO7:8H2O_Gehlenithydrat+10H<+>-2Ca<2+>-2Al<3+>-13H2O-1SiO2<0>,
logK(298,15K) = -49,67 +/- unknown, Reference: /RAR1997a/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 361 C3AH6_Ca3Al2O6:6H2O_Hydrogarnet
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5038117,36 +/- , Reaction: 0 =
+1C3AH6_Ca3Al2O6:6H2O_Hydrogarnet+12H<+>-3Ca<2+>-2Al<3+>-12H2O, logK(298,15K) =
-78,66 +/- unknown, Reference: /RAR1997a/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 362 C3AS(0,5)_Ca3Al2Si(0,5)O7-Tricalciumalumoheemisilikat
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4058629,554 +/- , Reaction: 0 =
+1C3AS(0,5)_Ca3Al2Si(0,5)O7-Tricalciumalumoheemisilikat+12H<+>-3Ca<2+>-2Al<3+>-
6H2O-0,5SiO2<0>, logK(298,15K) = -74,12 +/- unknown, Reference: /RAR1997a/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 363 C3ASH4_Ca3Al2SiO8:4H2O_Si-Hydrogarnet
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -5451740,437 +/- , Reaction: 0 =
+1C3ASH4_Ca3Al2SiO8:4H2O_Si-Hydrogarnet+12H<+>-3Ca<2+>-2Al<3+>-10H2O-1SiO2<0>,
logK(298,15K) = -69,37 +/- unknown, Reference: /RAR1997a/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 364 C9S6H16_Ca9Si6O21:16H2O_Jennit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -15073353,076 +/- , Reaction: 0 =
+1C9S6H16_Ca9Si6O21:16H2O_Jennit+18H<+>-9Ca<2+>, logK(298,15K) = -147,1 +/-
unknown, Reference: /RAR1997a/
V0 [cm3 mol⁻¹] = dummy value 0,001
Remark:
- 365 Ca4Al2Cl2O6:10H2O_Friedelsches_Salz_Monochlorid
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -6839704,603 +/- , Reaction: 0 =
+1Ca4Al2Cl2O6:10H2O_Friedelsches_Salz_Monochlorid+12H<+>-4Ca<2+>-2Al<3+>-16H2O-
2Cl<->, logK(298,15K) = -72,04 +/- unknown, Reference: /RAR1997a/

- V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 366 Ca6Al2(SO4)3(OH)12:26H2O_Ettringit
Gibbs Free Energy of Formation calculated from logK for reaction

$$\Delta G_0^f \text{ [J mol}^{-1}] = -15227966,757 \text{ +/- , Reaction: } 0 =$$

$$+1\text{Ca}_6\text{Al}_2(\text{SO}_4)_3(\text{OH})_{12}:26\text{H}_2\text{O_Ettringit} + 12\text{H}^+ - 6\text{Ca}^{2+} - 2\text{Al}^{3+} - 38\text{H}_2\text{O} - 3(\text{SO}_4)^{2-},$$

$$\log K(298,15\text{K}) = -55,223 \text{ +/- unknown, Reference: /RAR1997a/}$$
 V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 367 CSH(0,8)_Ca0,8SiO2,8_H2O
Gibbs Free Energy of Formation calculated from logK for reaction

$$\Delta G_0^f \text{ [J mol}^{-1}] = -1640727,047 \text{ +/- , Reaction: } 0 =$$

$$+1\text{CSH}(0,8)\text{Ca}_0,8\text{SiO}_2,8\text{H}_2\text{O} + 1,6\text{H}^+ - 0,8\text{Ca}^{2+}, \log K(298,15\text{K}) = -11,08 \text{ +/-}$$

$$\text{unknown, Reference: /RAR1997a/}$$
 V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 368 CSH(1,1)_Ca(1,1)SiO(3,1):3H2O
Gibbs Free Energy of Formation calculated from logK for reaction

$$\Delta G_0^f \text{ [J mol}^{-1}] = -2319797,486 \text{ +/- , Reaction: } 0 =$$

$$+1\text{CSH}(1,1)\text{Ca}(1,1)\text{SiO}(3,1):3\text{H}_2\text{O} + 2,2\text{H}^+ - 1,1\text{Ca}^{2+}, \log K(298,15\text{K}) = -16,72 \text{ +/-}$$

$$\text{unknown, Reference: /RAR1997a/}$$
 V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 369 CSH(1,8)_Ca(1,8)SiO(3,8):5H2O
Gibbs Free Energy of Formation calculated from logK for reaction

$$\Delta G_0^f \text{ [J mol}^{-1}] = -3256395,967 \text{ +/- , Reaction: } 0 =$$

$$+1\text{CSH}(1,8)\text{Ca}(1,8)\text{SiO}(3,8):5\text{H}_2\text{O} + 3,6\text{H}^+ - 1,8\text{Ca}^{2+}, \log K(298,15\text{K}) = -32,6 \text{ +/-}$$

$$\text{unknown, Reference: /RAR1997a/}$$
 V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 370 Mg4Al2O7:10H2O_Hydrotalcit
Gibbs Free Energy of Formation calculated from logK for reaction

$$\Delta G_0^f \text{ [J mol}^{-1}] = -6405279,257 \text{ +/- , Reaction: } 0 =$$

$$+1\text{Mg}_4\text{Al}_2\text{O}_7:10\text{H}_2\text{O_Hydrotalcit} + 14\text{H}^+ - 4\text{Mg}^{2+} - 2\text{Al}^{3+} - 17\text{H}_2\text{O}, \log K(298,15\text{K}) = -$$

$$75,44 \text{ +/- unknown, Reference: /RAR1997a/}$$
 V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 371 Ca2Zn2(OH)2(HSiO4)2_Clinohedrite
Gibbs Free Energy of Formation calculated from logK for reaction

$$\Delta G_0^f \text{ [J mol}^{-1}] = -4784175,542 \text{ +/- unknown, Reaction: } 0 =$$

$$+1\text{Ca}_2\text{Zn}_2(\text{OH})_2(\text{HSiO}_4)_2\text{Clinohedrite} + 8\text{H}^+ - 2\text{Zn}^{2+} - 2\text{SiO}_2 - 6\text{H}_2\text{O} - 2\text{Ca}^{2+},$$

$$\log K(298,15\text{K}) = 51,08 \text{ +/- unknown, Reference: ULT}$$
 V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 372 Ca2ZnSi2O7:2H2O_Hardystonite
Gibbs Free Energy of Formation calculated from logK for reaction

$$\Delta G_0^f \text{ [J mol}^{-1}] = -3880628,976 \text{ +/- unknown, Reaction: } 0 =$$

$$+1\text{Ca}_2\text{ZnSi}_2\text{O}_7:2\text{H}_2\text{O_Hardystonite} + 6\text{H}^+ - 1\text{Zn}^{2+} - 2\text{SiO}_2 - 5\text{H}_2\text{O} - 2\text{Ca}^{2+},$$

$$\log K(298,15\text{K}) = -39,88 \text{ +/- unknown, Reference: ULT}$$
 V0 [cm3 mol-1] = dummy value 0,001
Remark:
- 373 CaZn2(OH)6:2H2O_Calziumzinkat

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2493748,925 +/- unknown, Reaction: 0 = +1CaZn₂(OH)₆:2H₂O-Calziumzinkat+6H<+>-2Zn<2+>-1Ca<2+>-8H₂O, logK(298,15K) = -43,9 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 374 Goslarite_ZnSO₄:7H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2562168,703 +/- unknown, Reaction: 0 = +1Goslarite_ZnSO₄:7H₂O-7H₂O-1Zn<2+>-1(SO₄)<2->, logK(298,15K) = 1,9239 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 375 K₂Zn(SO₄)₂:6H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3654973,251 +/- unknown, Reaction: 0 = +1K₂Zn(SO₄)₂:6H₂O-2K<+>-1Zn<2+>-2(SO₄)<2->-6H₂O, logK(298,15K) = 5,589 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 376 K₂ZnCl₄
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1226582,493 +/- unknown, Reaction: 0 = +1K₂ZnCl₄-2K<+>-1Zn<2+>-4Cl<->, logK(298,15K) = -1,841 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 377 KZnCl₃:2H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1292190,536 +/- unknown, Reaction: 0 = +1KZnCl₃:2H₂O-1K<+>-1Zn<2+>-3Cl<->-2H₂O, logK(298,15K) = -0,9554 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 378 MgZnCl₄:5H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2280301,922 +/- unknown, Reaction: 0 = +1MgZnCl₄:5H₂O-1Mg<2+>-1Zn<2+>-4Cl<->-5H₂O, logK(298,15K) = -5,754 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 379 Na₂Zn(SO₄)₂:4H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3127187,09 +/- unknown, Reaction: 0 = +1Na₂Zn(SO₄)₂:4H₂O-2Na<+>-1Zn<2+>-2(SO₄)<2->-4H₂O, logK(298,15K) = 3,418 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 380 Na₂ZnCl₄:3H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1892027,524 +/- unknown, Reaction: 0 = +1Na₂ZnCl₄:3H₂O-2Na<+>-1Zn<2+>-4Cl<->-3H₂O, logK(298,15K) = -2,6926 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 381 Zn(OH)₂

- Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -555726,348 +/- unknown, Reaction: 0 = +1Zn(OH)₂+2H⁺-
1Zn²⁺-2H₂O, logK(298,15K) = -11,52 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 382 Zn₂(PO₄)OH_Tarbuttit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1628958,338 +/- unknown, Reaction: 0 =
+1Zn₂(PO₄)OH_Tarbuttit+1H⁺-2Zn²⁺-1(PO₄)³⁻-1H₂O, logK(298,15K) = 12,6 +/-
unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 383 Zn₂Ca(PO₄)₂:2H₂O_Scholzit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3567118,257 +/- unknown, Reaction: 0 =
+1Zn₂Ca(PO₄)₂:2H₂O_Scholzit-1Ca²⁺-2Zn²⁺-2(PO₄)³⁻-2H₂O, logK(298,15K) =
34,1 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 384 Zn₂SiO₄_Willemit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1524162,486 +/- unknown, Reaction: 0 =
+1Zn₂SiO₄_Willemit+4H⁺-2Zn²⁺-1SiO₂<0>-2H₂O, logK(298,15K) = -13,91 +/-
unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 385 Zn₃(PO₄)₂_Zn-Orthophosphat
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2675613,682 +/- unknown, Reaction: 0 = +1Zn₃(PO₄)₂_Zn-
Orthophosphat, logK(298,15K) = 32,064 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 386 Zn₄(PO₄)(OH)₂:3H₂O_Spencerit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3967053,459 +/- unknown, Reaction: 0 =
+1Zn₄(PO₄)(OH)₂:3H₂O_Spencerit+2H⁺-4Zn²⁺-2(PO₄)³⁻-5H₂O, logK(298,15K) =
24,8 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 387 Zn₅(OH)₆(CO₃)₂
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -3140765,039 +/- unknown, Reaction: 0 =
+1Zn₅(OH)₆(CO₃)₂+8H⁺-5Zn²⁺-2(HCO₃)⁻-6H₂O, logK(298,15K) = -33,5982 +/-
unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
- 388 Zn₅(PO₄)₃OH
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -4329892,897 +/- unknown, Reaction: 0 =
+1Zn₅(PO₄)₃OH+1H⁺-5Zn²⁺-3(PO₄)³⁻-1H₂O, logK(298,15K) = 49,1 +/- unknown,
Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001

- Remark:
- 389 ZnAl₆(PO₄)₄(OH)₈:4H₂O_Faustit
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -10419078,648 +/- unknown, Reaction: 0 = +1ZnAl₆(PO₄)₄(OH)₈:4H₂O_Faustit+8H<+>-1Zn<2+>-4(PO₄)<3->-12H₂O-6Al<3+>,
logK(298,15K) = 65,73 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 390 ZnCl₂:4Zn(OH)₂
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2661539,249 +/- unknown, Reaction: 0 = +1ZnCl₂:4Zn(OH)₂+8H<+>-5Zn<2+>-8H₂O-2Cl<->,
logK(298,15K) = -41 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 391 ZnCO₃
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -731403,464 +/- unknown, Reaction: 0 = +1ZnCO₃+1H<+>-1Zn<2+>-1(HCO₃)<->,
logK(298,15K) = -0,4633 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 392 ZnCO₃:H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -970495,044 +/- unknown, Reaction: 0 = +1ZnCO₃:H₂O+1H<+>-1Zn<2+>-1(HCO₃)<->-1H₂O,
logK(298,15K) = -0,1214 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 393 ZnO-aktiv
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -319670,876 +/- unknown, Reaction: 0 = +1ZnO-aktiv-1H₂O-1Zn<2+>+2H<+>,
logK(298,15K) = -11,33 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 394 ZnSO₄:2H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1369351,345 +/- unknown, Reaction: 0 = +1ZnSO₄:2H₂O-2H₂O-1Zn<2+>-1(SO₄)<2->,
logK(298,15K) = 0,677 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 395 ZnSO₄:6H₂O
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -2324149,094 +/- unknown, Reaction: 0 = +1ZnSO₄:6H₂O-6H₂O-1Zn<2+>-1(SO₄)<2->,
logK(298,15K) = 1,7698 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001
- Remark:
- 396 ZnSO₄:H₂O_Gunningite
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -1132363,179 +/- unknown, Reaction: 0 = +1ZnSO₄:H₂O_Gunningite-1H₂O-1Zn<2+>-1(SO₄)<2->,
logK(298,15K) = 0,7036 +/- unknown, Reference: ULT
V0 [cm³ mol⁻¹] = dummy value 0,001

Remark:
397 ZrO2(monoclinic)
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = -1042746 +/- 1313, Reference: NEA-8
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
398 BSK3+3SF
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 0 +/- , Reference: gesetzt
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
399 HAW-Kokille
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 0 +/- , Reference: gesetzt
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
400 SpentFuel
Gibbs Free Energy of Formation directly entered
 ΔG_0^f [J mol⁻¹] = 0 +/- unknown, Reference: gesetzt
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
401 NaOH
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -379766,367 +/- unknown, Reaction: 0 = +1NaOH+1H<+>-
1Na<+>-1H2O, logK(298,15K) = -20,905 +/- unknown, Reference: YPF
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
402 AgCl(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -109489,014 +/- unknown, Reaction: 0 = +1AgCl(cr)-
1Ag<+>-1Cl<->, logK(298,15K) = 9,7 +/- 0,1, Reference: /BR1973a/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
403 AgI(cr)
Gibbs Free Energy of Formation calculated from logK for reaction
 ΔG_0^f [J mol⁻¹] = -65956,683 +/- unknown, Reaction: 0 = +1AgI(cr)-1Ag<+>-
1I<->, logK(298,15K) = 16 +/- 0,1, Reference: /BR1973a/
V0 [cm³ mol⁻¹] = dummy value 0,001
Remark:
##Pitzer-Parameters
1 beta0 (Pb<2+> Cl<->)
Value in this parameterfile (chemapp) = -4,7625519008 = R * -0,5728; Reference:
HAG1998
Remark: Value was actually copied from a data file for EQ36, Though
calculations with the given Pitzer parameters for Pb compare very well with
experiments, all Pitzer parameters for Pb are still preliminary until a final
release by Hagemann,
2 beta1 (Pb<2+> Cl<->)
Value in this parameterfile (chemapp) = -4,61347271857 = R * -0,55487, alpha1 =
2; Reference: HAG1998

- Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 3 beta2 (Pb<2+> Cl<->)
Value in this parameterfile (chemapp) = -210,7104950175 = R * -25,3425, alpha2 = 14; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 4 cphi (Pb<2+> Cl<->)
Value in this parameterfile (chemapp) = 2,01668464305 = R * 0,24255; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 5 beta0 (Pb<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,73891059257 = R * 0,08887; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 6 beta1 (Pb<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 11,28827900426 = R * 1,35766, alpha1 = 1,4; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 7 beta2 (Pb<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -1213,918606 = R * -146, alpha2 = 12; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 8 cphi (Pb<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -8,3369601797 = R * -1,0027; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 9 psi (Pb<2+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -1,57335491653 = R * -0,18923; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 10 psi (Pb<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -1,32333757076 = R * -0,15916; Reference: HAG1998

- Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 11 psi (Pb<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -3,04618739507 = R * -0,36637; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 12 psi (Pb<2+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -2,89943627592 = R * -0,34872; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 13 theta (Pb<2+> Na<+>)
Value in this parameterfile (chemapp) = 1,47324820409 = R * 0,17719; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 14 theta (Pb<2+> K<+>)
Value in this parameterfile (chemapp) = 0,11856492686 = R * 0,01426; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 15 theta (Pb<2+> Mg<2+>)
Value in this parameterfile (chemapp) = -0,3250973801 = R * -0,0391; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 16 psi (Pb<2+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 3,0688860101 = R * 0,3691; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 17 psi (Pb<2+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,4082424901 = R * -0,0491; Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 18 psi (Pb<2+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 8,39150337186 = R * 1,00926; Reference: HAG1998

- Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 19 psi (Pb<2+> Cl<-> (SO4)<2->)
Value in this parameterfile (chemapp) = -1,25657204743 = R * -0,15113;
Reference: HAG1998
Remark: Value was actually copied from a data file for EQ36, Though calculations with the given Pitzer parameters for Pb compare very well with experiments, all Pitzer parameters for Pb are still preliminary until a final release by Hagemann,
- 20 beta0 (Fe<2+> Cl<->)
Value in this parameterfile (chemapp) = 3,10330808564 = R * 0,37324; Reference: MoHa
Remark:
- 21 beta1 (Fe<2+> Cl<->)
Value in this parameterfile (chemapp) = 9,43688683989 = R * 1,13499, alpha1 = 2; Reference: MoHa
Remark:
- 22 cphi (Fe<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,17901142183 = R * -0,02153;
Reference: MoHa
Remark:
- 23 beta0 (Fe<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 2,39981730993 = R * 0,28863; Reference: MoHa
Remark:
- 24 beta1 (Fe<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 22,50413861771 = R * 2,70661, alpha1 = 1,4; Reference: MoHa
Remark:
- 25 beta2 (Fe<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -349,209462 = R * -42, alpha2 = 12; Reference: MoHa
Remark:
- 26 cphi (Fe<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,06219254228 = R * 0,00748; Reference: MoHa
Remark:
- 27 theta (Fe<2+> Na<+>)
Value in this parameterfile (chemapp) = 1,07722804516 = R * 0,12956; Reference: MoHa
Remark:
- 28 psi (Fe<2+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,17618448809 = R * -0,02119;
Reference: MoHa
Remark:
- 29 psi (Fe<2+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,5446004705 = R * -0,0655; Reference: MoHa
Remark:
- 30 theta (Fe<2+> K<+>)

- Value in this parameterfile (chemapp) = -0,310685678407767 = R * -0,03737;
Reference: INTGRS
Remark:
- 31 psi (Fe<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,106525336812285 = R * -0,01281;
Reference: INTGRS
Remark:
- 32 psi (Fe<2+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,139617438206258 = R * -0,01679;
Reference: INTGRS
Remark:
- 33 theta (Fe<2+> Mg<2+>)
Value in this parameterfile (chemapp) = 0,90910863274 = R * 0,10934; Reference:
MoHa
Remark:
- 34 psi (Fe<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,25043307132 = R * -0,03012;
Reference: MoHa
Remark:
- 35 psi (Fe<2+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,20927624187 = R * 0,02517; Reference:
MoHa
Remark:
- 36 theta (Fe<2+> Ca<2+>)
Value in this parameterfile (chemapp) = -0,06352286404 = R * -0,00764;
Reference: MoHa
Remark:
- 37 psi (Fe<2+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,07557890499 = R * -0,00909;
Reference: MoHa
Remark:
- 38 psi (Fe<2+> Cl<-> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,10734033701 = R * 0,01291; Reference:
MoHa
Remark:
- 39 beta0 (Fe<3+> Cl<->)
Value in this parameterfile (chemapp) = 4,52342656444 = R * 0,54404; Reference:
MoHa
Remark:
- 40 beta1 (Fe<3+> Cl<->)
Value in this parameterfile (chemapp) = 50,94018196326 = R * 6,12666, alpha1 =
2; Reference: MoHa
Remark:
- 41 beta2 (Fe<3+> Cl<->)
Value in this parameterfile (chemapp) = 757,73838700395 = R * 91,13445, alpha2
= 12; Reference: MoHa
Remark:
- 42 cphi (Fe<3+> Cl<->)
Value in this parameterfile (chemapp) = -0,52248387124 = R * -0,06284;
Reference: MoHa
Remark:
- 43 beta0 (Fe<3+> (SO4)<2->)

- Value in this parameterfile (chemapp) = 4,95752718375 = R * 0,59625; Reference: MoHa
Remark:
- 44 beta1 (Fe<3+> (SO4)<2->)
Value in this parameterfile (chemapp) = 163,58018828466 = R * 19,67406, alpha1 = 1,559; Reference: MoHa
Remark:
- 45 beta2 (Fe<3+> (SO4)<2->)
Value in this parameterfile (chemapp) = 615,82556494996 = R * 74,06636, alpha2 = 5,268; Reference: MoHa
Remark:
- 46 cphi (Fe<3+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,26830926997 = R * -0,03227; Reference: MoHa
Remark:
- 47 theta (Fe<3+> Na<+>)
Value in this parameterfile (chemapp) = 2,04029785429 = R * 0,24539; Reference: MoHa
Remark:
- 48 psi (Fe<3+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,22790074651 = R * -0,02741; Reference: MoHa
Remark:
- 49 theta (Fe<3+> K<+>)
Value in this parameterfile (chemapp) = 1,24085762164 = R * 0,14924; Reference: MoHa
Remark:
- 50 psi (Fe<3+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,29757634869 = R * -0,03579; Reference: MoHa
Remark:
- 51 theta (Fe<3+> Mg<2+>)
Value in this parameterfile (chemapp) = 1,2787717918 = R * 0,1538; Reference: MoHa
Remark:
- 52 psi (Fe<3+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,64146452365 = R * -0,07715; Reference: MoHa
Remark:
- 53 theta (Fe<3+> Ca<2+>)
Value in this parameterfile (chemapp) = 1,35451698701 = R * 0,16291; Reference: MoHa
Remark:
- 54 psi (Fe<3+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,4082424901 = R * -0,0491; Reference: MoHa
Remark:
- 55 beta0 ((TcO)(OH)<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE 002/04
Remark:
- 56 beta1 ((TcO)(OH)<+> (SO4)<2->)

- Value in this parameterfile (chemapp) = 8,314511 = R * 1, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 57 beta0 ((TcO)(OH)<+> Cl<->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE
002/04
Remark:
- 58 beta1 ((TcO)(OH)<+> Cl<->)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 59 beta0 ((TcO)<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE
002/04
Remark:
- 60 beta1 ((TcO)<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 24,943533 = R * 3, alpha1 = 1,4;
Reference: FZK-INE 002/04
Remark:
- 61 beta2 ((TcO)<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -332,58044 = R * -40, alpha2 = 12;
Reference: FZK-INE 002/04
Remark:
- 62 beta0 ((TcO)<2+> Cl<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3; Reference: FZK-INE
002/04
Remark:
- 63 beta1 ((TcO)<2+> Cl<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 64 psi ((TcO4)<-> (SO4)<2-> Na<+>)
Value in this parameterfile (chemapp) = -0,024943533 = R * -0,003; Reference:
FZK-INE 002/04
Remark:
- 65 psi ((TcO4)<-> (SO4)<2-> K<+>)
Value in this parameterfile (chemapp) = 0,016629022 = R * 0,002; Reference:
FZK-INE 002/04
Remark:
- 66 psi ((TcO4)<-> (SO4)<2-> Ca<2+>)
Value in this parameterfile (chemapp) = -0,24943533 = R * -0,03; Reference:
FZK-INE 002/04
Remark:
- 67 theta ((TcO4)<-> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,488297469 = R * 0,179; Reference:
FZK-INE 002/04
Remark:
- 68 psi ((TcO4)<-> Cl<-> Na<+>)
Value in this parameterfile (chemapp) = -0,0706733435 = R * -0,0085; Reference:
FZK-INE 002/04
Remark:
- 69 psi ((TcO4)<-> Cl<-> K<+>)

- Value in this parameterfile (chemapp) = -0,091459621 = R * -0,011; Reference: FZK-INE 002/04
Remark:
- 70 psi ((TcO4)<-> Cl<-> Mg<2+>)
Value in this parameterfile (chemapp) = -0,274378863 = R * -0,033; Reference: FZK-INE 002/04
Remark:
- 71 psi ((TcO4)<-> Cl<-> Ca<2+>)
Value in this parameterfile (chemapp) = -0,0956168765 = R * -0,0115; Reference: FZK-INE 002/04
Remark:
- 72 theta ((TcO4)<-> Cl<->)
Value in this parameterfile (chemapp) = 0,557072237 = R * 0,067; Reference: FZK-INE 002/04
Remark:
- 73 beta0 (Ca<2+> (TcO)(OH)3<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3; Reference: FZK-INE 002/04
Remark:
- 74 beta1 (Ca<2+> (TcO)(OH)3<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 75 beta0 (Ca<2+> (TcO4)<->)
Value in this parameterfile (chemapp) = 2,4644210604 = R * 0,2964; Reference: FZK-INE 002/04
Remark:
- 76 beta1 (Ca<2+> (TcO4)<->)
Value in this parameterfile (chemapp) = 13,810402771 = R * 1,661, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 77 beta0 (Ca<2+> Tc(CO3)(OH)3<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3; Reference: FZK-INE 002/04
Remark:
- 78 beta1 (Ca<2+> Tc(CO3)(OH)3<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 79 beta0 (Cs<+> (TcO4)<->)
Value in this parameterfile (chemapp) = -1,5664538724 = R * -0,1884; Reference: FZK-INE 002/04
Remark:
- 80 beta1 (Cs<+> (TcO4)<->)
Value in this parameterfile (chemapp) = -1,3203443468 = R * -0,1588, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 81 beta1 (K<+> (TcO)(OH)3<->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 82 beta0 (K<+> (TcO4)<->)

- Value in this parameterfile (chemapp) = -0,4805787358 = R * -0,0578; Reference: FZK-INE 002/04
Remark:
- 83 beta1 (K<+> (TcO4)<->)
Value in this parameterfile (chemapp) = 0,049887066 = R * 0,006, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 84 beta1 (K<+> Tc(CO3)(OH)3<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 85 beta0 (Mg<2+> (TcO)(OH)3<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3; Reference: FZK-INE 002/04
Remark:
- 86 beta1 (Mg<2+> (TcO)(OH)3<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 87 beta0 (Mg<2+> (TcO4)<->)
Value in this parameterfile (chemapp) = 2,6090935518 = R * 0,3138; Reference: FZK-INE 002/04
Remark:
- 88 beta1 (Mg<2+> (TcO4)<->)
Value in this parameterfile (chemapp) = 15,29870024 = R * 1,84, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 89 cphi (Mg<2+> (TcO4)<->)
Value in this parameterfile (chemapp) = 0,0947854254 = R * 0,0114; Reference: FZK-INE 002/04
Remark:
- 90 beta0 (Mg<2+> Tc(CO3)(OH)3<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3; Reference: FZK-INE 002/04
Remark:
- 91 beta1 (Mg<2+> Tc(CO3)(OH)3<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 92 beta1 (Na<+> (TcO)(OH)3<->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 93 beta0 (Na<+> (TcO4)<->)
Value in this parameterfile (chemapp) = 0,09237421721 = R * 0,01111; Reference: FZK-INE 002/04
Remark:
- 94 beta1 (Na<+> (TcO4)<->)
Value in this parameterfile (chemapp) = 1,3261645045 = R * 0,1595, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 95 cphi (Na<+> (TcO4)<->)

- Value in this parameterfile (chemapp) = 0,01962224596 = R * 0,00236; Reference: FZK-INE 002/04
Remark:
- 96 psi (Na<+> Mg<2+> (TcO4)<->)
Value in this parameterfile (chemapp) = -0,16629022 = R * -0,02; Reference: FZK-INE 002/04
Remark:
- 97 beta1 (Na<+> Tc(CO3)(OH)3<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 98 beta0 (K<+> Th(CO3)5<6->)
Value in this parameterfile (chemapp) = 10,89200941 = R * 1,31; Reference: FZK-INE 002/04
Remark:
- 99 beta1 (K<+> Th(CO3)5<6->)
Value in this parameterfile (chemapp) = 249,43533 = R * 30, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 100 beta1 (K<+> Th(OH)(CO3)4<5->)
Value in this parameterfile (chemapp) = 191,233753 = R * 23, alpha1 = 2; Reference: FZK-INE 002/04
Remark: Schätzwert [siehe Kapitel 3,6]
- 101 beta1 (K<+> Th(OH)2(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 102 beta1 (K<+> Th(OH)3(CO3)<->)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 103 beta1 (K<+> Th(OH)4(CO3)<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 104 beta0 (Na<+> Th(CO3)5<6->)
Value in this parameterfile (chemapp) = 10,89200941 = R * 1,31; Reference: FZK-INE 002/04
Remark:
- 105 beta1 (Na<+> Th(CO3)5<6->)
Value in this parameterfile (chemapp) = 249,43533 = R * 30, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 106 beta1 (Na<+> Th(OH)(CO3)4<5->)
Value in this parameterfile (chemapp) = 191,233753 = R * 23, alpha1 = 2; Reference: FZK-INE 002/04
Remark: Schätzwert [siehe Kapitel 3,6]
- 107 beta1 (Na<+> Th(OH)2(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 108 beta1 (Na<+> Th(OH)3(CO3)<->)

- Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 109 beta1 (Na<+> Th(OH)4(CO3)<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 110 psi (Th(CO3)5<6-> Cl<-> Na<+>)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3; Reference: FZK-INE
002/04
Remark: Keine gcc'-Parameter und ycc'a-Parameter für Th4+ und kationische Th-
Komplexe bekannt, alle Werte gleich Null gesetzt,
- 111 theta (Th(CO3)5<6-> Cl<->)
Value in this parameterfile (chemapp) = 14,9661198 = R * 1,8; Reference: FZK-
INE 002/04
Remark:
- 112 beta0 (Th(OH)<3+> Cl<->)
Value in this parameterfile (chemapp) = 4,9887066 = R * 0,6; Reference: FZK-INE
002/04
Remark:
- 113 beta1 (Th(OH)<3+> Cl<->)
Value in this parameterfile (chemapp) = 49,0556149 = R * 5,9, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 114 beta0 (Th(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 1,91233753 = R * 0,23; Reference: FZK-
INE 002/04
Remark:
- 115 beta1 (Th(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 16,04700623 = R * 1,93, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 116 beta0 (Th(OH)3<+> Cl<->)
Value in this parameterfile (chemapp) = 0,66516088 = R * 0,08; Reference: FZK-
INE 002/04
Remark:
- 117 beta1 (Th(OH)3<+> Cl<->)
Value in this parameterfile (chemapp) = 3,24265929 = R * 0,39, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 118 beta0 (Th<4+> Cl<->)
Value in this parameterfile (chemapp) = 8,430914154 = R * 1,014; Reference:
FZK-INE 002/04
Remark:
- 119 beta1 (Th<4+> Cl<->)
Value in this parameterfile (chemapp) = 110,83243163 = R * 13,33, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 120 cphi (Th<4+> Cl<->)
Value in this parameterfile (chemapp) = -0,8597204374 = R * -0,1034; Reference:
FZK-INE 002/04
Remark:

- 121 beta0 ((U|+VI|O2)(OH)<+> Cl<->)
Value in this parameterfile (chemapp) = 1,24717665 = R * 0,15; Reference: FZK-INE 002/04
Remark:
- 122 beta1 ((U|+VI|O2)(OH)<+> Cl<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 123 beta0 ((U|+VI|O2)<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 2,677272542 = R * 0,322; Reference: FZK-INE 002/04
Remark:
- 124 beta1 ((U|+VI|O2)<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 15,190611597 = R * 1,827, alpha1 = 1,4;
Reference: FZK-INE 002/04
Remark:
- 125 cphi ((U|+VI|O2)<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,1463353936 = R * -0,0176; Reference: FZK-INE 002/04
Remark:
- 126 beta0 ((U|+VI|O2)<2+> Cl<->)
Value in this parameterfile (chemapp) = 3,55320627585 = R * 0,42735; Reference: FZK-INE 002/04
Remark:
- 127 beta1 ((U|+VI|O2)<2+> Cl<->)
Value in this parameterfile (chemapp) = 13,669056084 = R * 1,644, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 128 cphi ((U|+VI|O2)<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,30647287546 = R * -0,03686;
Reference: FZK-INE 002/04
Remark:
- 129 theta ((U|+VI|O2)<2+> Na<+>)
Value in this parameterfile (chemapp) = 0,1920652041 = R * 0,0231; Reference: FZK-INE 002/04
Remark: nur gültig in Kombination mit den binären Parametern für UO22+/ClO4- aus Kim und Frederick, 1988,
- 130 beta0 ((U|+VI|O2)2(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 4,1572555 = R * 0,5; Reference: FZK-INE 002/04
Remark:
- 131 beta1 ((U|+VI|O2)2(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 13,3032176 = R * 1,6, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 132 beta0 ((U|+VI|O2)3(OH)4<2+> Cl<->)
Value in this parameterfile (chemapp) = 0,58201577 = R * 0,07; Reference: FZK-INE 002/04
Remark:
- 133 beta1 ((U|+VI|O2)3(OH)4<2+> Cl<->)
Value in this parameterfile (chemapp) = 13,3032176 = R * 1,6, alpha1 = 2;
Reference: FZK-INE 002/04

- Remark:
- 134 theta ((U|+VI|O2)3(OH)4<2+> Na<+>)
Value in this parameterfile (chemapp) = 0,41572555 = R * 0,05; Reference: FZK-INE 002/04
Remark:
- 135 beta0 ((U|+VI|O2)3(OH)5<+> Cl<->)
Value in this parameterfile (chemapp) = 2,57749841 = R * 0,31; Reference: FZK-INE 002/04
Remark:
- 136 beta1 ((U|+VI|O2)3(OH)5<+> Cl<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 137 psi ((U|+VI|O2)3(OH)5<+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,66516088 = R * -0,08; Reference: FZK-INE 002/04
Remark:
- 138 beta0 ((U|+VI|O2)4(OH)7<+> Cl<->)
Value in this parameterfile (chemapp) = 1,91233753 = R * 0,23; Reference: FZK-INE 002/04
Remark:
- 139 beta1 ((U|+VI|O2)4(OH)7<+> Cl<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 140 beta0 (K<+> U|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE 002/04
Remark:
- 141 beta1 (K<+> U|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 142 beta0 (K<+> U|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 12,4717665 = R * 1,5; Reference: FZK-INE 002/04
Remark:
- 143 beta1 (K<+> U|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 260,2441943 = R * 31,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 144 beta1 (K<+> U|+IV|(OH)2(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 145 beta0 (Mg<2+> (U|+VI|O2)(OH)3<->)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE 002/04
Remark:
- 146 beta1 (Mg<2+> (U|+VI|O2)(OH)3<->)
Value in this parameterfile (chemapp) = 13,3032176 = R * 1,6, alpha1 = 2;
Reference: FZK-INE 002/04

- Remark:
- 147 beta1 (Mg<2+> (U|+VI|O2)(OH)4<2->)
Value in this parameterfile (chemapp) = 24,943533 = R * 3, alpha1 = 1,4;
Reference: FZK-INE 002/04
- Remark:
- 148 beta2 (Mg<2+> (U|+VI|O2)(OH)4<2->)
Value in this parameterfile (chemapp) = -332,58044 = R * -40, alpha2 = 12;
Reference: FZK-INE 002/04
- Remark:
- 149 beta0 (Mg<2+> (U|+VI|O2)3(OH)7<->)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE
002/04
- Remark:
- 150 beta1 (Mg<2+> (U|+VI|O2)3(OH)7<->)
Value in this parameterfile (chemapp) = 13,3032176 = R * 1,6, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 151 beta0 (Na<+> (U|+VI|O2)(CO3)2<2->)
Value in this parameterfile (chemapp) = -0,99774132 = R * -0,12; Reference:
FZK-INE 002/04
- Remark:
- 152 beta1 (Na<+> (U|+VI|O2)(CO3)2<2->)
Value in this parameterfile (chemapp) = 20,7862775 = R * 2,5, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 153 beta0 (Na<+> (U|+VI|O2)(CO3)3<4->)
Value in this parameterfile (chemapp) = 7,31676968 = R * 0,88; Reference: FZK-
INE 002/04
- Remark:
- 154 beta1 (Na<+> (U|+VI|O2)(CO3)3<4->)
Value in this parameterfile (chemapp) = 98,1112298 = R * 11,8, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 155 beta0 (Na<+> (U|+VI|O2)(OH)3<->)
Value in this parameterfile (chemapp) = -1,99548264 = R * -0,24; Reference:
FZK-INE 002/04
- Remark:
- 156 beta1 (Na<+> (U|+VI|O2)(OH)3<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 157 beta0 (Na<+> (U|+VI|O2)(OH)4<2->)
Value in this parameterfile (chemapp) = 1,33032176 = R * 0,16; Reference: FZK-
INE 002/04
- Remark:
- 158 beta1 (Na<+> (U|+VI|O2)(OH)4<2->)
Value in this parameterfile (chemapp) = 13,3032176 = R * 1,6, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 159 beta0 (Na<+> (U|+VI|O2)3(OH)7<->)
Value in this parameterfile (chemapp) = -1,99548264 = R * -0,24; Reference:
FZK-INE 002/04

- Remark:
- 160 $\text{beta1 (Na}^{+}> \text{U}^{+}\text{VI}|\text{O}_2\text{)3(OH)7}^{<->}$
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 161 $\text{beta0 (Na}^{+}> \text{U}^{+}\text{IV}|\text{CO}_3\text{)4}^{<->}$
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE
002/04
Remark:
- 162 $\text{beta1 (Na}^{+}> \text{U}^{+}\text{IV}|\text{CO}_3\text{)4}^{<->}$
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 163 $\text{beta0 (Na}^{+}> \text{U}^{+}\text{IV}|\text{CO}_3\text{)5}^{<->}$
Value in this parameterfile (chemapp) = 12,4717665 = R * 1,5; Reference: FZK-
INE 002/04
Remark:
- 164 $\text{beta1 (Na}^{+}> \text{U}^{+}\text{IV}|\text{CO}_3\text{)5}^{<->}$
Value in this parameterfile (chemapp) = 260,2441943 = R * 31,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 165 $\text{beta1 (Na}^{+}> \text{U}^{+}\text{IV}|\text{OH)2(CO}_3\text{)2}^{<->}$
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 166 $\text{beta0 (U}^{+}\text{IV}|\text{OH)3}^{+> \text{Cl}^{<->}}$
Value in this parameterfile (chemapp) = 4,9887066 = R * 0,6; Reference: FZK-INE
002/04
Remark:
- 167 $\text{beta1 (U}^{+}\text{IV}|\text{OH)3}^{+> \text{Cl}^{<->}}$
Value in this parameterfile (chemapp) = 49,0556149 = R * 5,9, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 168 $\text{beta0 (U}^{+}\text{IV}|\text{OH)2}^{<2+> \text{Cl}^{<->}}$
Value in this parameterfile (chemapp) = 1,91233753 = R * 0,23; Reference: FZK-
INE 002/04
Remark:
- 169 $\text{beta1 (U}^{+}\text{IV}|\text{OH)2}^{<2+> \text{Cl}^{<->}}$
Value in this parameterfile (chemapp) = 16,04700623 = R * 1,93, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 170 $\text{beta0 (U}^{+}\text{IV}|\text{OH)3}^{<+> \text{Cl}^{<->}}$
Value in this parameterfile (chemapp) = 0,66516088 = R * 0,08; Reference: FZK-
INE 002/04
Remark:
- 171 $\text{beta1 (U}^{+}\text{IV}|\text{OH)3}^{<+> \text{Cl}^{<->}}$
Value in this parameterfile (chemapp) = 3,24265929 = R * 0,39, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 172 $\text{beta0 (U}^{+}\text{IV}|\text{4}^{+> \text{Cl}^{<->}}$
Value in this parameterfile (chemapp) = 10,55942897 = R * 1,27; Reference: FZK-
INE 002/04

- Remark:
- 173 β_{11} (U|+IV|<4+> Cl<->)
Value in this parameterfile (chemapp) = 112,2458985 = R * 13,5, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 174 θ ((Np|+V|O2)(CO3)<-> Cl<->)
Value in this parameterfile (chemapp) = -2,07862775 = R * -0,25; Reference:
FZK-INE 002/04
- Remark:
- 175 θ ((Np|+V|O2)(CO3)2<3-> Cl<->)
Value in this parameterfile (chemapp) = -2,07862775 = R * -0,25; Reference:
FZK-INE 002/04
- Remark:
- 176 θ ((Np|+V|O2)(CO3)3<5-> (CO3)<2->)
Value in this parameterfile (chemapp) = -6,90104413 = R * -0,83; Reference:
FZK-INE 002/04
- Remark:
- 177 θ ((Np|+V|O2)(CO3)3<5-> Cl<->)
Value in this parameterfile (chemapp) = -2,07862775 = R * -0,25; Reference:
FZK-INE 002/04
- Remark:
- 178 λ ((Np|+V|O2)(OH)<0> Cl<->)
Value in this parameterfile (chemapp) = -1,57975709 = R * -0,19; Reference:
FZK-INE 002/04
- Remark:
- 179 θ ((Np|+V|O2)(OH)2<-> Cl<->)
Value in this parameterfile (chemapp) = -1,99548264 = R * -0,24; Reference:
FZK-INE 002/04
- Remark:
- 180 θ ((Np|+V|O2)<+> Ca<2+>)
Value in this parameterfile (chemapp) = 0,41572555 = R * 0,05; Reference: FZK-
INE 002/04
- Remark:
- 181 β_{10} ((Np|+V|O2)<+> Cl<->)
Value in this parameterfile (chemapp) = 1,1765033065 = R * 0,1415; Reference:
FZK-INE 002/04
- Remark:
- 182 β_{11} ((Np|+V|O2)<+> Cl<->)
Value in this parameterfile (chemapp) = 2,336377591 = R * 0,281, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 183 θ ((Np|+V|O2)<+> Mg<2+>)
Value in this parameterfile (chemapp) = 0,41572555 = R * 0,05; Reference: FZK-
INE 002/04
- Remark:
- 184 λ ((Np|+VI|O2)(CO3)<0> Cl<->)
Value in this parameterfile (chemapp) = -2,07862775 = R * -0,25; Reference:
FZK-INE 002/04
- Remark:
- 185 β_{10} ((Np|+VI|O2)(OH)<+> Cl<->)
Value in this parameterfile (chemapp) = 1,24717665 = R * 0,15; Reference: FZK-
INE 002/04

- Remark:
- 186 $\beta_{10}((\text{Np}|+\text{VI}|\text{O}_2)(\text{OH})\langle+\rangle \text{Cl}\langle-\rangle)$
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 187 $\beta_{00}((\text{Np}|+\text{VI}|\text{O}_2)\langle 2+\rangle \text{Cl}\langle-\rangle)$
Value in this parameterfile (chemapp) = 3,5536220014 = R * 0,4274; Reference:
FZK-INE 002/04
Remark:
- 188 $\beta_{10}((\text{Np}|+\text{VI}|\text{O}_2)\langle 2+\rangle \text{Cl}\langle-\rangle)$
Value in this parameterfile (chemapp) = 13,669056084 = R * 1,644, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 189 $\text{cphi}((\text{Np}|+\text{VI}|\text{O}_2)\langle 2+\rangle \text{Cl}\langle-\rangle)$
Value in this parameterfile (chemapp) = -0,3059740048 = R * -0,0368; Reference:
FZK-INE 002/04
Remark:
- 190 $\beta_{00}((\text{Np}|+\text{VI}|\text{O}_2)_2(\text{OH})_2\langle 2+\rangle \text{Cl}\langle-\rangle)$
Value in this parameterfile (chemapp) = 4,1572555 = R * 0,5; Reference: FZK-INE
002/04
Remark:
- 191 $\beta_{10}((\text{Np}|+\text{VI}|\text{O}_2)_2(\text{OH})_2\langle 2+\rangle \text{Cl}\langle-\rangle)$
Value in this parameterfile (chemapp) = 13,3032176 = R * 1,6, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 192 $\beta_{00}((\text{Np}|+\text{VI}|\text{O}_2)_3(\text{OH})_5\langle+\rangle \text{Cl}\langle-\rangle)$
Value in this parameterfile (chemapp) = 2,57749841 = R * 0,31; Reference: FZK-
INE 002/04
Remark:
- 193 $\beta_{10}((\text{Np}|+\text{VI}|\text{O}_2)_3(\text{OH})_5\langle+\rangle \text{Cl}\langle-\rangle)$
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 194 $\beta_{00}(\text{K}\langle+\rangle (\text{Np}|+\text{V}|\text{O}_2)(\text{CO}_3)\langle-\rangle)$
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE
002/04
Remark:
- 195 $\beta_{10}(\text{K}\langle+\rangle (\text{Np}|+\text{V}|\text{O}_2)(\text{CO}_3)\langle-\rangle)$
Value in this parameterfile (chemapp) = 2,82693374 = R * 0,34, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 196 $\beta_{00}(\text{K}\langle+\rangle (\text{Np}|+\text{V}|\text{O}_2)(\text{CO}_3)_2\langle 3-\rangle)$
Value in this parameterfile (chemapp) = 3,99096528 = R * 0,48; Reference: FZK-
INE 002/04
Remark:
- 197 $\beta_{10}(\text{K}\langle+\rangle (\text{Np}|+\text{V}|\text{O}_2)(\text{CO}_3)_2\langle 3-\rangle)$
Value in this parameterfile (chemapp) = 36,5838484 = R * 4,4, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 198 $\beta_{00}(\text{K}\langle+\rangle (\text{Np}|+\text{V}|\text{O}_2)(\text{CO}_3)_3\langle 5-\rangle)$
Value in this parameterfile (chemapp) = 14,9661198 = R * 1,8; Reference: FZK-
INE 002/04

- Remark:
- 199 beta1 (K<+> (Np|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 188,7393997 = R * 22,7, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 200 beta2 (K<+> (Np|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = -790,7099961 = R * -95,1, alpha2 = 12;
Reference: FZK-INE 002/04
- Remark:
- 201 cphi (K<+> (Np|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = -1,820877909 = R * -0,219; Reference:
FZK-INE 002/04
- Remark:
- 202 beta0 (K<+> Np|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE
002/04
- Remark:
- 203 beta1 (K<+> Np|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 204 beta0 (K<+> Np|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 12,4717665 = R * 1,5; Reference: FZK-
INE 002/04
- Remark:
- 205 beta1 (K<+> Np|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 260,2441943 = R * 31,3, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 206 beta1 (K<+> Np|+IV|(OH)2(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 207 beta1 (K<+> Np|+IV|(OH)4(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 208 beta0 (K<+> Np|+IV|(OH)4(CO3)2<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE
002/04
- Remark:
- 209 beta1 (K<+> Np|+IV|(OH)4(CO3)2<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 210 beta0 (Mg<2+> (Np|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 3,3258044 = R * 0,4; Reference: FZK-INE
002/04
- Remark:
- 211 beta1 (Mg<2+> (Np|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 2;
Reference: FZK-INE 002/04

- Remark:
- 212 β_0 (Mg²⁺ (Np|+V|O₂)(OH)₂<->)
Value in this parameterfile (chemapp) = 3,3258044 = R * 0,4; Reference: FZK-INE 002/04
- Remark:
- 213 β_1 (Mg²⁺ (Np|+V|O₂)(OH)₂<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 1,4;
Reference: FZK-INE 002/04
- Remark:
- 214 β_0 (Na⁺ (Np|+V|O₂)(CO₃)<->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE 002/04
- Remark:
- 215 β_1 (Na⁺ (Np|+V|O₂)(CO₃)<->)
Value in this parameterfile (chemapp) = 2,82693374 = R * 0,34, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 216 β_0 (Na⁺ (Np|+V|O₂)(CO₃)₂<3->)
Value in this parameterfile (chemapp) = 3,99096528 = R * 0,48; Reference: FZK-INE 002/04
- Remark:
- 217 β_1 (Na⁺ (Np|+V|O₂)(CO₃)₂<3->)
Value in this parameterfile (chemapp) = 36,5838484 = R * 4,4, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 218 β_0 (Na⁺ (Np|+V|O₂)(CO₃)₃<5->)
Value in this parameterfile (chemapp) = 14,9661198 = R * 1,8; Reference: FZK-INE 002/04
- Remark:
- 219 β_1 (Na⁺ (Np|+V|O₂)(CO₃)₃<5->)
Value in this parameterfile (chemapp) = 188,7393997 = R * 22,7, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 220 λ (Na⁺ (Np|+VI|O₂)(CO₃)<0>)
Value in this parameterfile (chemapp) = 0,41572555 = R * 0,05; Reference: FZK-INE 002/04
- Remark:
- 221 β_0 (Na⁺ (Np|+VI|O₂)(CO₃)₂<2->)
Value in this parameterfile (chemapp) = 1,762676332 = R * 0,212; Reference: FZK-INE 002/04
- Remark:
- 222 β_1 (Na⁺ (Np|+VI|O₂)(CO₃)₂<2->)
Value in this parameterfile (chemapp) = 20,7862775 = R * 2,5, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 223 β_0 (Na⁺ (Np|+VI|O₂)(CO₃)₃<4->)
Value in this parameterfile (chemapp) = 10,39313875 = R * 1,25; Reference: FZK-INE 002/04
- Remark:
- 224 β_1 (Na⁺ (Np|+VI|O₂)(CO₃)₃<4->)
Value in this parameterfile (chemapp) = 96,4483276 = R * 11,6, alpha1 = 2;
Reference: FZK-INE 002/04

- Remark:
- 225 beta0 (Na<+> Np|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE 002/04
- Remark:
- 226 beta1 (Na<+> Np|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 227 beta0 (Na<+> Np|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 12,4717665 = R * 1,5; Reference: FZK-INE 002/04
- Remark:
- 228 beta1 (Na<+> Np|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 260,2441943 = R * 31,3, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 229 beta1 (Na<+> Np|+IV|(OH)2(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 230 beta1 (Na<+> Np|+IV|(OH)4(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 231 beta0 (Na<+> Np|+IV|(OH)4(CO3)2<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE 002/04
- Remark:
- 232 beta1 (Na<+> Np|+IV|(OH)4(CO3)2<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 233 beta0 (Np|+IV|(OH)3<+> Cl<->)
Value in this parameterfile (chemapp) = 4,9887066 = R * 0,6; Reference: FZK-INE 002/04
- Remark:
- 234 beta1 (Np|+IV|(OH)3<+> Cl<->)
Value in this parameterfile (chemapp) = 49,0556149 = R * 5,9, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 235 beta0 (Np|+IV|(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 1,91233753 = R * 0,23; Reference: FZK-INE 002/04
- Remark:
- 236 beta1 (Np|+IV|(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 15,7975709 = R * 1,9, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 237 beta0 (Np|+IV|(OH)3<+> Cl<->)
Value in this parameterfile (chemapp) = 0,66516088 = R * 0,08; Reference: FZK-INE 002/04

- Remark:
- 238 $\beta_{1\text{ (Np|+IV|OH)3<+> Cl<->}}$
Value in this parameterfile (chemapp) = 3,24265929 = R * 0,39, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 239 $\beta_{0\text{ (Np|+IV|<4+> Cl<->)}}$
Value in this parameterfile (chemapp) = 10,97515452 = R * 1,32; Reference: FZK-
INE 002/04
Remark:
- 240 $\beta_{1\text{ (Np|+IV|<4+> Cl<->)}}$
Value in this parameterfile (chemapp) = 112,2458985 = R * 13,5, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 241 $\theta_{\text{ ((Pu|+V|O2)(CO3)<-> Cl<->)}}$
Value in this parameterfile (chemapp) = -1,74604731 = R * -0,21; Reference:
FZK-INE 002/04
Remark:
- 242 $\theta_{\text{ ((Pu|+V|O2)(CO3)3<5-> (CO3)<2->)}}$
Value in this parameterfile (chemapp) = -6,90104413 = R * -0,83; Reference:
FZK-INE 002/04
Remark:
- 243 $\theta_{\text{ ((Pu|+V|O2)(CO3)3<5-> Cl<->)}}$
Value in this parameterfile (chemapp) = -2,16177286 = R * -0,26; Reference:
FZK-INE 002/04
Remark:
- 244 $\theta_{\text{ ((Pu|+V|O2)<+> Ca<2+>)}}$
Value in this parameterfile (chemapp) = 0,41572555 = R * 0,05; Reference: FZK-
INE 002/04
Remark:
- 245 $\beta_{0\text{ ((Pu|+V|O2)<+> Cl<->)}}$
Value in this parameterfile (chemapp) = 1,1765033065 = R * 0,1415; Reference:
FZK-INE 002/04
Remark:
- 246 $\beta_{1\text{ ((Pu|+V|O2)<+> Cl<->)}}$
Value in this parameterfile (chemapp) = 2,336377591 = R * 0,281, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 247 $\theta_{\text{ ((Pu|+V|O2)<+> Mg<2+>)}}$
Value in this parameterfile (chemapp) = 0,41572555 = R * 0,05; Reference: FZK-
INE 002/04
Remark:
- 248 $\beta_{0\text{ ((Pu|+VI|O2)(OH)<+> Cl<->)}}$
Value in this parameterfile (chemapp) = 1,24717665 = R * 0,15; Reference: FZK-
INE 002/04
Remark:
- 249 $\beta_{1\text{ ((Pu|+VI|O2)(OH)<+> Cl<->)}}$
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 250 $\beta_{0\text{ ((Pu|+VI|O2)<2+> Cl<->)}}$
Value in this parameterfile (chemapp) = 3,55320627585 = R * 0,42735; Reference:
FZK-INE 002/04

- Remark:
- 251 beta1 ((Pu|+VI|O2)<2+> Cl<->)
Value in this parameterfile (chemapp) = 13,669056084 = R * 1,644, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 252 cphi ((Pu|+VI|O2)<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,30647287546 = R * -0,03686;
Reference: FZK-INE 002/04
- Remark:
- 253 beta0 ((Pu|+VI|O2)2(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 4,1572555 = R * 0,5; Reference: FZK-INE
002/04
- Remark:
- 254 beta1 ((Pu|+VI|O2)2(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 13,3032176 = R * 1,6, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 255 beta0 ((Pu|+VI|O2)3(OH)5<+> Cl<->)
Value in this parameterfile (chemapp) = 2,57749841 = R * 0,31; Reference: FZK-
INE 002/04
- Remark:
- 256 beta1 ((Pu|+VI|O2)3(OH)5<+> Cl<->)
Value in this parameterfile (chemapp) = 2,4943533 = R * 0,3, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 257 beta0 (K<+> (Pu|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE
002/04
- Remark:
- 258 beta1 (K<+> (Pu|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 2,82693374 = R * 0,34, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 259 beta0 (K<+> (Pu|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 14,9661198 = R * 1,8; Reference: FZK-
INE 002/04
- Remark:
- 260 beta1 (K<+> (Pu|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 188,7393997 = R * 22,7, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 261 beta2 (K<+> (Pu|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = -790,7099961 = R * -95,1, alpha2 = 12;
Reference: FZK-INE 002/04
- Remark:
- 262 cphi (K<+> (Pu|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = -1,820877909 = R * -0,219; Reference:
FZK-INE 002/04
- Remark:
- 263 beta0 (K<+> Pu|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE
002/04

- Remark:
- 264 beta1 (K<+> Pu|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 265 beta0 (K<+> Pu|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 12,4717665 = R * 1,5; Reference: FZK-
INE 002/04
Remark:
- 266 beta1 (K<+> Pu|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 260,2441943 = R * 31,3, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 267 beta1 (K<+> Pu|+IV|(OH)2(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 268 beta1 (K<+> Pu|+IV|(OH)4(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 269 beta0 (K<+> Pu|+IV|(OH)4(CO3)2<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE
002/04
Remark:
- 270 beta1 (K<+> Pu|+IV|(OH)4(CO3)2<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 271 beta0 (Mg<2+> (Pu|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 3,3258044 = R * 0,4; Reference: FZK-INE
002/04
Remark:
- 272 beta1 (Mg<2+> (Pu|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 2;
Reference: FZK-INE 002/04
Remark: INE-FZK 002/04)
- 273 beta0 (Na<+> (Pu|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE
002/04
Remark:
- 274 beta1 (Na<+> (Pu|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 2,82693374 = R * 0,34, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 275 beta0 (Na<+> (Pu|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 14,9661198 = R * 1,8; Reference: FZK-
INE 002/04
Remark:
- 276 beta1 (Na<+> (Pu|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 188,7393997 = R * 22,7, alpha1 = 2;
Reference: FZK-INE 002/04

- Remark:
- 277 beta0 (Na<+> (Pu|+VI|O2)(CO3)2<2->)
Value in this parameterfile (chemapp) = 1,762676332 = R * 0,212; Reference: FZK-INE 002/04
- Remark:
- 278 beta1 (Na<+> (Pu|+VI|O2)(CO3)2<2->)
Value in this parameterfile (chemapp) = 20,7862775 = R * 2,5, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
- 279 beta0 (Na<+> (Pu|+VI|O2)(CO3)3<4->)
Value in this parameterfile (chemapp) = 10,39313875 = R * 1,25; Reference: FZK-INE 002/04
- Remark:
- 280 beta1 (Na<+> (Pu|+VI|O2)(CO3)3<4->)
Value in this parameterfile (chemapp) = 96,4483276 = R * 11,6, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
- 281 beta0 (Na<+> Pu|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE 002/04
- Remark:
- 282 beta1 (Na<+> Pu|+IV|(CO3)4<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
- 283 beta0 (Na<+> Pu|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 12,4717665 = R * 1,5; Reference: FZK-INE 002/04
- Remark:
- 284 beta1 (Na<+> Pu|+IV|(CO3)5<6->)
Value in this parameterfile (chemapp) = 260,2441943 = R * 31,3, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
- 285 beta1 (Na<+> Pu|+IV|(OH)2(CO3)2<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
- 286 beta1 (Na<+> Pu|+IV|(OH)4(CO3)<2->)
Value in this parameterfile (chemapp) = 16,629022 = R * 2, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
- 287 beta0 (Na<+> Pu|+IV|(OH)4(CO3)2<4->)
Value in this parameterfile (chemapp) = 8,314511 = R * 1; Reference: FZK-INE 002/04
- Remark:
- 288 beta1 (Na<+> Pu|+IV|(OH)4(CO3)2<4->)
Value in this parameterfile (chemapp) = 108,088643 = R * 13, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
- 289 beta0 (Pu|+III|(OH)<2+> Cl<->)
Value in this parameterfile (chemapp) = 0,457298105 = R * 0,055; Reference: FZK-INE 002/04

- Remark: INE-FZK 002/04)
- 290 beta1 (Pu|+III|(OH)<2+> Cl<->)
Value in this parameterfile (chemapp) = 15,04926491 = R * 1,81, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 291 theta (Pu|+III|<3+> Ca<2+>)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE
002/04
Remark:
- 292 beta0 (Pu|+III|<3+> Cl<->)
Value in this parameterfile (chemapp) = 4,8689776416 = R * 0,5856; Reference:
FZK-INE 002/04
Remark:
- 293 beta1 (Pu|+III|<3+> Cl<->)
Value in this parameterfile (chemapp) = 46,5612616 = R * 5,6, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 294 cphi (Pu|+III|<3+> Cl<->)
Value in this parameterfile (chemapp) = -0,157975709 = R * -0,019; Reference:
FZK-INE 002/04
Remark: Unter Vernachlässigung von Chloridkomplexen,
- 295 theta (Pu|+III|<3+> K<+>)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE
002/04
Remark: a) analog zu Na+,
- 296 theta (Pu|+III|<3+> Mg<2+>)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE
002/04
Remark: b) analog zu Ca2+,
- 297 theta (Pu|+III|<3+> Na<+>)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE
002/04
Remark:
- 298 beta0 (Pu|+IV|(OH)<3+> Cl<->)
Value in this parameterfile (chemapp) = 4,9887066 = R * 0,6; Reference: FZK-INE
002/04
Remark:
- 299 beta1 (Pu|+IV|(OH)<3+> Cl<->)
Value in this parameterfile (chemapp) = 49,0556149 = R * 5,9, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 300 beta0 (Pu|+IV|(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 1,91233753 = R * 0,23; Reference: FZK-
INE 002/04
Remark:
- 301 beta1 (Pu|+IV|(OH)2<2+> Cl<->)
Value in this parameterfile (chemapp) = 15,7975709 = R * 1,9, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 302 beta0 (Pu|+IV|(OH)3<+> Cl<->)
Value in this parameterfile (chemapp) = 0,66516088 = R * 0,08; Reference: FZK-
INE 002/04

- Remark:
- 303 beta1 (Pu|+IV|(OH)3<+> Cl<->)
Value in this parameterfile (chemapp) = 3,24265929 = R * 0,39, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 304 beta0 (Pu|+IV|<4+> Cl<->)
Value in this parameterfile (chemapp) = 10,97515452 = R * 1,32; Reference: FZK-
INE 002/04
- Remark:
- 305 beta1 (Pu|+IV|<4+> Cl<->)
Value in this parameterfile (chemapp) = 112,2458985 = R * 13,5, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 306 theta ((Am|+V|O2)(CO3)<-> Cl<->)
Value in this parameterfile (chemapp) = -1,74604731 = R * -0,21; Reference:
FZK-INE 002/04
- Remark:
- 307 theta ((Am|+V|O2)(CO3)2<3-> Cl<->)
Value in this parameterfile (chemapp) = -2,16177286 = R * -0,26; Reference:
FZK-INE 002/04
- Remark:
- 308 theta ((Am|+V|O2)(CO3)3<5-> (CO3)<2->)
Value in this parameterfile (chemapp) = -6,90104413 = R * -0,83; Reference:
FZK-INE 002/04
- Remark:
- 309 theta ((Am|+V|O2)(CO3)3<5-> Cl<->)
Value in this parameterfile (chemapp) = -2,16177286 = R * -0,26; Reference:
FZK-INE 002/04
- Remark:
- 310 theta ((Am|+V|O2)<+> Ca<2+>)
Value in this parameterfile (chemapp) = 0,41572555 = R * 0,05; Reference: FZK-
INE 002/04
- Remark:
- 311 beta0 ((Am|+V|O2)<+> Cl<->)
Value in this parameterfile (chemapp) = 1,1765033065 = R * 0,1415; Reference:
FZK-INE 002/04
- Remark:
- 312 beta1 ((Am|+V|O2)<+> Cl<->)
Value in this parameterfile (chemapp) = 2,336377591 = R * 0,281, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 313 theta ((Am|+V|O2)<+> Mg<2+>)
Value in this parameterfile (chemapp) = 0,41572555 = R * 0,05; Reference: FZK-
INE 002/04
- Remark:
- 314 beta0 (Am|+III|(CO3)<+> Cl<->)
Value in this parameterfile (chemapp) = -0,598644792 = R * -0,072; Reference:
FZK-INE 002/04
- Remark:
- 315 beta1 (Am|+III|(CO3)<+> Cl<->)
Value in this parameterfile (chemapp) = 3,350747933 = R * 0,403, alpha1 = 2;
Reference: FZK-INE 002/04

- Remark:
- 316 $\text{cphi}(\text{Am}|\text{+III}|(\text{CO}_3)\langle\text{+}\rangle\text{Cl}\langle\text{-}\rangle)$
Value in this parameterfile (chemapp) = 0,3226030268 = R * 0,0388; Reference:
FZK-INE 002/04
Remark:
- 317 $\text{beta0}(\text{Am}|\text{+III}|(\text{OH})\langle\text{2+}\rangle\text{Cl}\langle\text{-}\rangle)$
Value in this parameterfile (chemapp) = 0,457298105 = R * 0,055; Reference:
FZK-INE 002/04
Remark:
- 318 $\text{beta1}(\text{Am}|\text{+III}|(\text{OH})\langle\text{2+}\rangle\text{Cl}\langle\text{-}\rangle)$
Value in this parameterfile (chemapp) = 15,04926491 = R * 1,81, $\text{alpha1} = 2$;
Reference: FZK-INE 002/04
Remark:
- 319 $\text{beta0}(\text{Am}|\text{+III}|(\text{OH})\langle\text{2+}\rangle\text{Cl}\langle\text{-}\rangle)$
Value in this parameterfile (chemapp) = -3,442207554 = R * -0,414; Reference:
FZK-INE 002/04
Remark:
- 320 $\text{lambda}(\text{Am}|\text{+III}|(\text{OH})\langle\text{3+}\rangle\text{Cl}\langle\text{-}\rangle)$
Value in this parameterfile (chemapp) = -1,6629022 = R * -0,2; Reference: FZK-
INE 002/04
Remark: $l = -0,2$
- 321 $\text{beta0}(\text{Am}|\text{+III}|(\text{SO}_4)\langle\text{2-}\rangle)$
Value in this parameterfile (chemapp) = 14,899603712 = R * 1,792; Reference:
FZK-INE 002/04
Remark:
- 322 $\text{beta1}(\text{Am}|\text{+III}|(\text{SO}_4)\langle\text{2-}\rangle)$
Value in this parameterfile (chemapp) = 125,083503484 = R * 15,044, $\text{alpha1} = 2$;
Reference: FZK-INE 002/04
Remark:
- 323 $\text{cphi}(\text{Am}|\text{+III}|(\text{SO}_4)\langle\text{2-}\rangle)$
Value in this parameterfile (chemapp) = -4,9887066 = R * -0,6; Reference: FZK-
INE 002/04
Remark:
- 324 $\text{theta}(\text{Am}|\text{+III}|(\text{Ca})\langle\text{2+}\rangle)$
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE
002/04
Remark:
- 325 $\text{beta0}(\text{Am}|\text{+III}|(\text{Cl})\langle\text{-}\rangle)$
Value in this parameterfile (chemapp) = 4,8689776416 = R * 0,5856; Reference:
FZK-INE 002/04
Remark:
- 326 $\text{beta1}(\text{Am}|\text{+III}|(\text{Cl})\langle\text{-}\rangle)$
Value in this parameterfile (chemapp) = 46,5612616 = R * 5,6, $\text{alpha1} = 2$;
Reference: FZK-INE 002/04
Remark:
- 327 $\text{cphi}(\text{Am}|\text{+III}|(\text{Cl})\langle\text{-}\rangle)$
Value in this parameterfile (chemapp) = -0,133032176 = R * -0,016; Reference:
FZK-INE 002/04
Remark: b) Unter Einbeziehung von Chloridkomplexen,
- 328 $\text{theta}(\text{Am}|\text{+III}|(\text{K})\langle\text{+}\rangle)$
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE
002/04

- Remark: a) analog zu Na+,
329 theta (Am|+III|<3+> Mg<2+>)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE 002/04
- Remark: b) analog zu Ca2+,
330 theta (Am|+III|<3+> Na<+>)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE 002/04
- Remark:
331 theta (Am|+III|Cl<2+> Ca<2+>)
Value in this parameterfile (chemapp) = -0,116403154 = R * -0,014; Reference: FZK-INE 002/04
- Remark:
332 beta0 (Am|+III|Cl<2+> Cl<->)
Value in this parameterfile (chemapp) = 4,930505023 = R * 0,593; Reference: FZK-INE 002/04
- Remark:
333 beta1 (Am|+III|Cl<2+> Cl<->)
Value in this parameterfile (chemapp) = 26,19070965 = R * 3,15, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
334 cphi (Am|+III|Cl<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: FZK-INE 002/04
- Remark:
335 theta (Am|+III|Cl<2+> Mg<2+>)
Value in this parameterfile (chemapp) = -0,116403154 = R * -0,014; Reference: FZK-INE 002/04
- Remark: b) analog zu Ca2+,
336 theta (Am|+III|Cl2<+> Ca<2+>)
Value in this parameterfile (chemapp) = -1,629644156 = R * -0,196; Reference: FZK-INE 002/04
- Remark:
337 beta0 (Am|+III|Cl2<+> Cl<->)
Value in this parameterfile (chemapp) = 4,290287676 = R * 0,516; Reference: FZK-INE 002/04
- Remark:
338 beta1 (Am|+III|Cl2<+> Cl<->)
Value in this parameterfile (chemapp) = 14,55039425 = R * 1,75, alpha1 = 2; Reference: FZK-INE 002/04
- Remark:
339 cphi (Am|+III|Cl2<+> Cl<->)
Value in this parameterfile (chemapp) = 0,08314511 = R * 0,01; Reference: FZK-INE 002/04
- Remark:
340 theta (Am|+III|Cl2<+> Mg<2+>)
Value in this parameterfile (chemapp) = -1,629644156 = R * -0,196; Reference: FZK-INE 002/04
- Remark: b) analog zu Ca2+,
341 beta0 (K<+> (Am|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE 002/04

- Remark:
- 342 beta1 (K<+> (Am|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 2,82693374 = R * 0,34, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 343 beta0 (K<+> (Am|+V|O2)(CO3)2<3->)
Value in this parameterfile (chemapp) = 3,99096528 = R * 0,48; Reference: FZK-
INE 002/04
Remark:
- 344 beta1 (K<+> (Am|+V|O2)(CO3)2<3->)
Value in this parameterfile (chemapp) = 36,5838484 = R * 4,4, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 345 beta0 (K<+> (Am|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 14,9661198 = R * 1,8; Reference: FZK-
INE 002/04
Remark:
- 346 beta1 (K<+> (Am|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 188,7393997 = R * 22,7, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 347 beta2 (K<+> (Am|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = -790,7099961 = R * -95,1, alpha2 = 12;
Reference: FZK-INE 002/04
Remark:
- 348 cphi (K<+> (Am|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = -1,820877909 = R * -0,219; Reference:
FZK-INE 002/04
Remark:
- 349 beta0 (Mg<2+> (Am|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 3,3258044 = R * 0,4; Reference: FZK-INE
002/04
Remark:
- 350 beta1 (Mg<2+> (Am|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 14,1346687 = R * 1,7, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 351 beta0 (Na<+> (Am|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE
002/04
Remark:
- 352 beta1 (Na<+> (Am|+V|O2)(CO3)<->)
Value in this parameterfile (chemapp) = 2,82693374 = R * 0,34, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:
- 353 beta0 (Na<+> (Am|+V|O2)(CO3)2<3->)
Value in this parameterfile (chemapp) = 3,99096528 = R * 0,48; Reference: FZK-
INE 002/04
Remark:
- 354 beta1 (Na<+> (Am|+V|O2)(CO3)2<3->)
Value in this parameterfile (chemapp) = 36,5838484 = R * 4,4, alpha1 = 2;
Reference: FZK-INE 002/04

- Remark:
- 355 beta0 (Na<+> (Am|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 14,9661198 = R * 1,8; Reference: FZK-INE 002/04
- Remark:
- 356 beta1 (Na<+> (Am|+V|O2)(CO3)3<5->)
Value in this parameterfile (chemapp) = 188,7393997 = R * 22,7, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 357 beta0 (Na<+> Am|+III|(CO3)2<->)
Value in this parameterfile (chemapp) = -1,99548264 = R * -0,24; Reference:
FZK-INE 002/04
- Remark:
- 358 beta1 (Na<+> Am|+III|(CO3)2<->)
Value in this parameterfile (chemapp) = 1,862450464 = R * 0,224, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 359 cphi (Na<+> Am|+III|(CO3)2<->)
Value in this parameterfile (chemapp) = 0,2361321124 = R * 0,0284; Reference:
FZK-INE 002/04
- Remark:
- 360 beta0 (Na<+> Am|+III|(CO3)3<3->)
Value in this parameterfile (chemapp) = 1,039313875 = R * 0,125; Reference:
FZK-INE 002/04
- Remark:
- 361 beta1 (Na<+> Am|+III|(CO3)3<3->)
Value in this parameterfile (chemapp) = 39,32763703 = R * 4,73, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 362 cphi (Na<+> Am|+III|(CO3)3<3->)
Value in this parameterfile (chemapp) = 0,0058201577 = R * 0,0007; Reference:
FZK-INE 002/04
- Remark:
- 363 beta0 (Cm(CO3)<+> Cl<->)
Value in this parameterfile (chemapp) = -0,598644792 = R * -0,072; Reference:
FZK-INE 002/04
- Remark:
- 364 beta1 (Cm(CO3)<+> Cl<->)
Value in this parameterfile (chemapp) = 3,350747933 = R * 0,403, alpha1 = 2;
Reference: FZK-INE 002/04
- Remark:
- 365 cphi (Cm(CO3)<+> Cl<->)
Value in this parameterfile (chemapp) = 0,3226030268 = R * 0,0388; Reference:
FZK-INE 002/04
- Remark:
- 366 beta0 (Cm(OH)<2+> Cl<->)
Value in this parameterfile (chemapp) = 0,457298105 = R * 0,055; Reference:
FZK-INE 002/04
- Remark:
- 367 beta1 (Cm(OH)<2+> Cl<->)
Value in this parameterfile (chemapp) = 15,04926491 = R * 1,81, alpha1 = 2;
Reference: FZK-INE 002/04

- Remark:
- 368 beta0 (Cm(OH)₂<+> Cl<->)
Value in this parameterfile (chemapp) = -3,442207554 = R * -0,414; Reference: FZK-INE 002/04
Remark:
- 369 beta0 (Cm(SO₄)<+> Cl<->)
Value in this parameterfile (chemapp) = -0,756620501 = R * -0,091; Reference: FZK-INE 002/04
Remark:
- 370 beta1 (Cm(SO₄)<+> Cl<->)
Value in this parameterfile (chemapp) = -3,24265929 = R * -0,39, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 371 cphi (Cm(SO₄)<+> Cl<->)
Value in this parameterfile (chemapp) = 0,399096528 = R * 0,048; Reference: FZK-INE 002/04
Remark:
- 372 beta0 (Cm<3+> (SO₄)<2->)
Value in this parameterfile (chemapp) = 14,899603712 = R * 1,792; Reference: FZK-INE 002/04
Remark:
- 373 beta1 (Cm<3+> (SO₄)<2->)
Value in this parameterfile (chemapp) = 125,083503484 = R * 15,044, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 374 cphi (Cm<3+> (SO₄)<2->)
Value in this parameterfile (chemapp) = -4,9887066 = R * -0,6; Reference: FZK-INE 002/04
Remark:
- 375 theta (Cm<3+> Ca<2+>)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE 002/04
Remark:
- 376 beta0 (Cm<3+> Cl<->)
Value in this parameterfile (chemapp) = 4,8689776416 = R * 0,5856; Reference: FZK-INE 002/04
Remark:
- 377 beta1 (Cm<3+> Cl<->)
Value in this parameterfile (chemapp) = 46,5612616 = R * 5,6, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 378 cphi (Cm<3+> Cl<->)
Value in this parameterfile (chemapp) = -0,133032176 = R * -0,016; Reference: FZK-INE 002/04
Remark: b) Unter Einbeziehung von Chloridkomplexen,
- 379 theta (Cm<3+> K<+>)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE 002/04
Remark: a) analog zu Na+,
- 380 theta (Cm<3+> Mg<2+>)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: FZK-INE 002/04

Remark: b) analog zu Ca²⁺,

381 theta (Cm³⁺ Na⁺)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: FZK-INE 002/04
Remark:

382 theta (CmCl²⁺ Ca²⁺)
Value in this parameterfile (chemapp) = -0,116403154 = R * -0,014; Reference: FZK-INE 002/04
Remark:

383 beta0 (CmCl²⁺ Cl⁻)
Value in this parameterfile (chemapp) = 4,930505023 = R * 0,593; Reference: FZK-INE 002/04
Remark:

384 beta1 (CmCl²⁺ Cl⁻)
Value in this parameterfile (chemapp) = 26,19070965 = R * 3,15, alpha1 = 2; Reference: FZK-INE 002/04
Remark:

385 cphi (CmCl²⁺ Cl⁻)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: FZK-INE 002/04
Remark:

386 theta (CmCl²⁺ Mg²⁺)
Value in this parameterfile (chemapp) = -0,116403154 = R * -0,014; Reference: FZK-INE 002/04
Remark: b) analog zu Ca²⁺,

387 theta (CmCl²⁺ Ca²⁺)
Value in this parameterfile (chemapp) = -1,629644156 = R * -0,196; Reference: FZK-INE 002/04
Remark:

388 beta0 (CmCl²⁺ Cl⁻)
Value in this parameterfile (chemapp) = 4,290287676 = R * 0,516; Reference: FZK-INE 002/04
Remark:

389 beta1 (CmCl²⁺ Cl⁻)
Value in this parameterfile (chemapp) = 14,55039425 = R * 1,75, alpha1 = 2; Reference: FZK-INE 002/04
Remark:

390 cphi (CmCl²⁺ Cl⁻)
Value in this parameterfile (chemapp) = 0,08314511 = R * 0,01; Reference: FZK-INE 002/04
Remark:

391 theta (CmCl²⁺ Mg²⁺)
Value in this parameterfile (chemapp) = -1,629644156 = R * -0,196; Reference: FZK-INE 002/04
Remark: b) analog zu Ca²⁺,

392 beta0 (Na⁺ Cm(CO₃)²⁻)
Value in this parameterfile (chemapp) = -1,99548264 = R * -0,24; Reference: FZK-INE 002/04
Remark:

393 beta1 (Na⁺ Cm(CO₃)²⁻)
Value in this parameterfile (chemapp) = 1,862450464 = R * 0,224, alpha1 = 2; Reference: FZK-INE 002/04

Remark:

394 cphi (Na<+> Cm(CO3)2<->)
Value in this parameterfile (chemapp) = 0,2361321124 = R * 0,0284; Reference:
FZK-INE 002/04
Remark:

395 beta0 (Na<+> Cm(CO3)3<3->)
Value in this parameterfile (chemapp) = 1,039313875 = R * 0,125; Reference:
FZK-INE 002/04
Remark:

396 beta1 (Na<+> Cm(CO3)3<3->)
Value in this parameterfile (chemapp) = 39,32763703 = R * 4,73, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:

397 cphi (Na<+> Cm(CO3)3<3->)
Value in this parameterfile (chemapp) = 0,0058201577 = R * 0,0007; Reference:
FZK-INE 002/04
Remark:

398 beta0 (Na<+> Cm(CO3)4<5->)
Value in this parameterfile (chemapp) = 16,811941242 = R * 2,022; Reference:
FZK-INE 002/04
Remark:

399 beta1 (Na<+> Cm(CO3)4<5->)
Value in this parameterfile (chemapp) = 159,80490142 = R * 19,22, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:

400 cphi (Na<+> Cm(CO3)4<5->)
Value in this parameterfile (chemapp) = -2,535925855 = R * -0,305; Reference:
FZK-INE 002/04
Remark:

401 beta0 (Na<+> Cm(SO4)2<->)
Value in this parameterfile (chemapp) = -2,943336894 = R * -0,354; Reference:
FZK-INE 002/04
Remark:

402 beta1 (Na<+> Cm(SO4)2<->)
Value in this parameterfile (chemapp) = -3,3258044 = R * -0,4, alpha1 = 2;
Reference: FZK-INE 002/04
Remark:

403 cphi (Na<+> Cm(SO4)2<->)
Value in this parameterfile (chemapp) = 0,424040061 = R * 0,051; Reference:
FZK-INE 002/04
Remark:

404 beta1 (Ra<2+> (HSO4)<->)
Value in this parameterfile (chemapp) = 26,007790408 = R * 3,128, alpha1 = 2;
Reference: SM4759
Remark: valid with logK of RaSO4, higher temp, Available

405 beta1 (Ra<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 43,73432786 = R * 5,26, alpha1 = 2;
Reference: SM4759
Remark: approximated using parameters for BaSO4, higher temp, Available

406 beta2 (Ra<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -1272,120183 = R * -153, alpha2 = 12;
Reference: SM4759

Remark: approximated using parameters for BaSO4, higher temp, Available

407 theta (Ra<2+> H<+>)
Value in this parameterfile (chemapp) = 0,407411039 = R * 0,049; Reference: SM4759
Remark: valid with logK of RaSO4

408 lambda ((CO2)<0> Ca<2+>)
Value in this parameterfile (chemapp) = 1,521555513 = R * 0,183; Reference: HMW
Remark:

409 lambda ((CO2)<0> Cl<->)
Value in this parameterfile (chemapp) = -0,041572555 = R * -0,005; Reference: HMW
Remark:

410 beta0 (Cl<-> Ca<2+>)
Value in this parameterfile (chemapp) = 2,6265540249 = R * 0,3159; Reference: HMW
Remark:

411 beta1 (Cl<-> Ca<2+>)
Value in this parameterfile (chemapp) = 13,419620754 = R * 1,614, alpha1 = 2;
Reference: HMW
Remark:

412 cphi (Cl<-> Ca<2+>)
Value in this parameterfile (chemapp) = -0,00282693374 = R * -0,00034;
Reference: HMW
Remark:

413 beta0 ((HCO3)<-> Ca<2+>)
Value in this parameterfile (chemapp) = 3,3258044 = R * 0,4; Reference: HMW
Remark:

414 beta1 ((HCO3)<-> Ca<2+>)
Value in this parameterfile (chemapp) = 24,752299247 = R * 2,977, alpha1 = 2;
Reference: HMW
Remark:

415 beta0 ((HSO4)<-> Ca<2+>)
Value in this parameterfile (chemapp) = 1,7834626095 = R * 0,2145; Reference: HMW
Remark:

416 beta1 ((HSO4)<-> Ca<2+>)
Value in this parameterfile (chemapp) = 21,03571283 = R * 2,53, alpha1 = 2;
Reference: HMW
Remark:

417 beta0 (OH<-> Ca<2+>)
Value in this parameterfile (chemapp) = -1,4525450717 = R * -0,1747; Reference: HMW
Remark:

418 beta1 (OH<-> Ca<2+>)
Value in this parameterfile (chemapp) = -1,9148318833 = R * -0,2303, alpha1 = 2;
Reference: HMW
Remark:

419 beta2 (OH<-> Ca<2+>)
Value in this parameterfile (chemapp) = -47,55900292 = R * -5,72, alpha2 = 12;
Reference: HMW
Remark:

420 beta0 ((SO4)<2-> Ca<2+>)

Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: HMW
Remark:

421 beta1 ((SO4)<2-> Ca<2+>)
Value in this parameterfile (chemapp) = 26,5839860203 = R * 3,1973, alpha1 = 1,4; Reference: HMW
Remark:

422 beta2 ((SO4)<2-> Ca<2+>)
Value in this parameterfile (chemapp) = -450,97907664 = R * -54,24, alpha2 = 12; Reference: HMW
Remark:

423 lambda ((CO2)<0> (HSO4)<->)
Value in this parameterfile (chemapp) = -0,024943533 = R * -0,003; Reference: HMW
Remark:

424 lambda ((CO2)<0> K<+>)
Value in this parameterfile (chemapp) = 0,424040061 = R * 0,051; Reference: HMW
Remark:

425 lambda ((CO2)<0> Mg<2+>)
Value in this parameterfile (chemapp) = 1,521555513 = R * 0,183; Reference: HMW
Remark:

426 lambda ((CO2)<0> Na<+>)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: HMW
Remark:

427 lambda ((CO2)<0> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,806507567 = R * 0,097; Reference: HMW
Remark:

428 beta0 (H<+> Cl<->)
Value in this parameterfile (chemapp) = 1,4758257025 = R * 0,1775; Reference: HMW
Remark:

429 beta1 (H<+> Cl<->)
Value in this parameterfile (chemapp) = 2,4486234895 = R * 0,2945, alpha1 = 2; Reference: HMW
Remark:

430 cphi (H<+> Cl<->)
Value in this parameterfile (chemapp) = 0,0066516088 = R * 0,0008; Reference: HMW
Remark:

431 beta0 (H<+> (HSO4)<->)
Value in this parameterfile (chemapp) = 1,7169465215 = R * 0,2065; Reference: HMW
Remark:

432 beta1 (H<+> (HSO4)<->)
Value in this parameterfile (chemapp) = 4,6195423116 = R * 0,5556, alpha1 = 2; Reference: HMW
Remark:

433 beta0 (H<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,2477724278 = R * 0,0298; Reference: HMW
Remark:

434 cphi (H<+> (SO4)<2->)

- Value in this parameterfile (chemapp) = 0,3641755818 = R * 0,0438; Reference:
HMW
Remark:
- 435 beta0 ((CO3)<2-> K<+>)
Value in this parameterfile (chemapp) = 1,2371992368 = R * 0,1488; Reference:
HMW
Remark:
- 436 beta1 ((CO3)<2-> K<+>)
Value in this parameterfile (chemapp) = 11,88975073 = R * 1,43, alpha1 = 2;
Reference: HMW
Remark:
- 437 cphi ((CO3)<2-> K<+>)
Value in this parameterfile (chemapp) = -0,0124717665 = R * -0,0015; Reference:
HMW
Remark:
- 438 beta0 (Cl<-> K<+>)
Value in this parameterfile (chemapp) = 0,40200660685 = R * 0,04835; Reference:
HMW
Remark:
- 439 beta1 (Cl<-> K<+>)
Value in this parameterfile (chemapp) = 1,7643392342 = R * 0,2122, alpha1 = 2;
Reference: HMW
Remark:
- 440 cphi (Cl<-> K<+>)
Value in this parameterfile (chemapp) = -0,00698418924 = R * -0,00084;
Reference: HMW
Remark:
- 441 beta0 ((HCO3)<-> K<+>)
Value in this parameterfile (chemapp) = 0,2461095256 = R * 0,0296; Reference:
HMW
Remark:
- 442 beta1 ((HCO3)<-> K<+>)
Value in this parameterfile (chemapp) = -0,108088643 = R * -0,013, alpha1 = 2;
Reference: HMW
Remark:
- 443 cphi ((HCO3)<-> K<+>)
Value in this parameterfile (chemapp) = -0,066516088 = R * -0,008; Reference:
HMW
Remark:
- 444 beta0 (K<+> (HSO4)<->)
Value in this parameterfile (chemapp) = -0,0024943533 = R * -0,0003; Reference:
HMW
Remark:
- 445 beta1 (K<+> (HSO4)<->)
Value in this parameterfile (chemapp) = 1,4425676585 = R * 0,1735, alpha1 = 2;
Reference: HMW
Remark:
- 446 beta0 (OH<-> K<+>)
Value in this parameterfile (chemapp) = 1,0792235278 = R * 0,1298; Reference:
HMW
Remark:
- 447 beta1 (OH<-> K<+>)

Value in this parameterfile (chemapp) = 2,66064352 = R * 0,32, alpha1 = 2;
Reference: HMW
Remark:

448 cphi (OH<-> K<+>)
Value in this parameterfile (chemapp) = 0,0340894951 = R * 0,0041; Reference:
HMW
Remark:

449 beta0 (K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,41530982445 = R * 0,04995; Reference:
HMW
Remark:

450 beta1 (K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 6,4794984223 = R * 0,7793, alpha1 = 2;
Reference: HMW
Remark:

451 beta0 (Cl<-> Mg<2+>)
Value in this parameterfile (chemapp) = 2,92961795085 = R * 0,35235; Reference:
HMW
Remark:

452 beta1 (Cl<-> Mg<2+>)
Value in this parameterfile (chemapp) = 13,9808502465 = R * 1,6815, alpha1 = 2;
Reference: HMW
Remark:

453 cphi (Cl<-> Mg<2+>)
Value in this parameterfile (chemapp) = 0,04315231209 = R * 0,00519; Reference:
HMW
Remark:

454 beta0 ((HCO3)<-> Mg<2+>)
Value in this parameterfile (chemapp) = 2,735474119 = R * 0,329; Reference: HMW
Remark:

455 beta1 ((HCO3)<-> Mg<2+>)
Value in this parameterfile (chemapp) = 5,0485710792 = R * 0,6072, alpha1 = 2;
Reference: HMW
Remark:

456 beta0 (Mg<2+> (HSO4)<->)
Value in this parameterfile (chemapp) = 3,9460669206 = R * 0,4746; Reference:
HMW
Remark:

457 beta1 (Mg<2+> (HSO4)<->)
Value in this parameterfile (chemapp) = 14,375789519 = R * 1,729, alpha1 = 2;
Reference: HMW
Remark:

458 beta0 (Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,837506931 = R * 0,221; Reference: HMW
Remark:

459 beta1 (Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 27,795410273 = R * 3,343, alpha1 = 1,4;
Reference: HMW
Remark:

460 beta2 (Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -309,54924453 = R * -37,23, alpha2 =
12; Reference: HMW

- Remark:
- 461 cphi (Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,207862775 = R * 0,025; Reference: HMW
Remark:
- 462 beta0 (Na<+> (CO3)<2->)
Value in this parameterfile (chemapp) = 0,3317489889 = R * 0,0399; Reference: SM 4191
Remark: This value instead of that from hmw was used to estimate Pitzer-parameters in SM 5325 (Np(V) with carbonate), It was therefore adopted for this database, (hmw=0,0399, SM4191=0,0362
- 463 beta1 (Na<+> (CO3)<2->)
Value in this parameterfile (chemapp) = 11,548855779 = R * 1,389, alpha1 = 2; Reference: SM 4191
Remark: This value instead of that from hmw was used to estimate Pitzer-parameters in SM 5325 (Np(V) with carbonate), It was therefore adopted for this database, (hmw=1,389, SM4191=1,51
- 464 cphi (Na<+> (CO3)<2->)
Value in this parameterfile (chemapp) = 0,0365838484 = R * 0,0044; Reference: SM 4191
Remark: This value instead of that from hmw was used to estimate Pitzer-parameters in SM 5325 (Np(V) with carbonate), It was therefore adopted for this database, Hmw=0,0044, SM4191=0,0052
- 465 beta0 (Na<+> Cl<->)
Value in this parameterfile (chemapp) = 0,6360600915 = R * 0,0765; Reference: HMW
Remark:
- 466 beta1 (Na<+> Cl<->)
Value in this parameterfile (chemapp) = 2,1983567084 = R * 0,2644, alpha1 = 2; Reference: HMW
Remark:
- 467 cphi (Na<+> Cl<->)
Value in this parameterfile (chemapp) = 0,01055942897 = R * 0,00127; Reference: HMW
Remark:
- 468 beta0 (Na<+> (HCO3)<->)
Value in this parameterfile (chemapp) = 0,2303119547 = R * 0,0277; Reference: SM 4191
Remark: This value instead of that from hmw was used to estimate Pitzer-parameters in SM 5325 (Np(V) with carbonate), It was therefore adopted for this database, Hmw=0,0277, SM4191=0,028
- 469 beta1 (Na<+> (HCO3)<->)
Value in this parameterfile (chemapp) = 0,3417264021 = R * 0,0411, alpha1 = 2; Reference: SM 4191
Remark: This value instead of that from hmw was used to estimate Pitzer-parameters in SM 5325 (Np(V) with carbonate), It was therefore adopted for this database, Hmw=0,0411, SM4191=0,044
- 470 beta0 (Na<+> (HSO4)<->)
Value in this parameterfile (chemapp) = 0,3774787994 = R * 0,0454; Reference: HMW
Remark:
- 471 beta1 (Na<+> (HSO4)<->)
Value in this parameterfile (chemapp) = 3,309175378 = R * 0,398, alpha1 = 2; Reference: HMW
Remark:

- 472 beta0 (Na<+> OH<->)
Value in this parameterfile (chemapp) = 0,7183737504 = R * 0,0864; Reference: HMW
Remark:
- 473 betal (Na<+> OH<->)
Value in this parameterfile (chemapp) = 2,103571283 = R * 0,253, alpha1 = 2;
Reference: HMW
Remark:
- 474 cphi (Na<+> OH<->)
Value in this parameterfile (chemapp) = 0,0365838484 = R * 0,0044; Reference: HMW
Remark:
- 475 beta0 (Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,16279812538 = R * 0,01958; Reference: HMW
Remark:
- 476 betal (Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 9,254050743 = R * 1,113, alpha1 = 2;
Reference: HMW
Remark:
- 477 cphi (Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,04132311967 = R * 0,00497; Reference: HMW
Remark:
- 478 theta (H<+> Ca<2+>)
Value in this parameterfile (chemapp) = 0,764935012 = R * 0,092; Reference: HMW
Remark:
- 479 psi (H<+> Cl<-> Ca<2+>)
Value in this parameterfile (chemapp) = -0,124717665 = R * -0,015; Reference: HMW
Remark:
- 480 theta (Mg<2+> Ca<2+>)
Value in this parameterfile (chemapp) = 0,058201577 = R * 0,007; Reference: HMW
Remark:
- 481 psi (Mg<2+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,099774132 = R * -0,012; Reference: HMW
Remark:
- 482 psi (Mg<2+> Ca<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,199548264 = R * 0,024; Reference: HMW
Remark:
- 483 theta ((CO3)<2-> Cl<->)
Value in this parameterfile (chemapp) = -0,16629022 = R * -0,02; Reference: HMW
Remark:
- 484 psi ((CO3)<2-> Cl<-> K<+>)
Value in this parameterfile (chemapp) = 0,033258044 = R * 0,004; Reference: HMW
Remark:
- 485 psi ((CO3)<2-> Cl<-> Na<+>)
Value in this parameterfile (chemapp) = 0,0706733435 = R * 0,0085; Reference: HMW
Remark:

- 486 theta ((HCO3)<-> Cl<->)
Value in this parameterfile (chemapp) = 0,24943533 = R * 0,03; Reference: HMW
Remark:
- 487 psi ((HCO3)<-> Cl<-> Mg<2+>)
Value in this parameterfile (chemapp) = -0,798193056 = R * -0,096; Reference: HMW
Remark:
- 488 psi ((HCO3)<-> Cl<-> Na<+>)
Value in this parameterfile (chemapp) = -0,124717665 = R * -0,015; Reference: HMW
Remark:
- 489 theta (Cl<-> (HSO4)<->)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: HMW
Remark:
- 490 psi (Cl<-> (HSO4)<-> H<+>)
Value in this parameterfile (chemapp) = 0,108088643 = R * 0,013; Reference: HMW
Remark:
- 491 psi (Cl<-> (HSO4)<-> Na<+>)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: HMW
Remark:
- 492 theta (OH<-> Cl<->)
Value in this parameterfile (chemapp) = -0,41572555 = R * -0,05; Reference: HMW
Remark:
- 493 psi (OH<-> Cl<-> Ca<2+>)
Value in this parameterfile (chemapp) = -0,207862775 = R * -0,025; Reference: HMW
Remark:
- 494 psi (OH<-> Cl<-> K<+>)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: HMW
Remark:
- 495 psi (OH<-> Cl<-> Na<+>)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: HMW
Remark:
- 496 theta (Cl<-> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,16629022 = R * 0,02; Reference: HMW
Remark:
- 497 psi (Cl<-> (SO4)<2-> Ca<2+>)
Value in this parameterfile (chemapp) = -0,149661198 = R * -0,018; Reference: HMW
Remark:
- 498 psi (Cl<-> (SO4)<2-> Mg<2+>)
Value in this parameterfile (chemapp) = -0,033258044 = R * -0,004; Reference: HMW
Remark:
- 499 psi (Cl<-> (SO4)<2-> Na<+>)
Value in this parameterfile (chemapp) = 0,0116403154 = R * 0,0014; Reference: HMW
Remark:

500 theta ((HCO3)<-> (CO3)<2->)
Value in this parameterfile (chemapp) = -0,33258044 = R * -0,04; Reference: HMW
Remark:

501 psi ((HCO3)<-> (CO3)<2-> K<+>)
Value in this parameterfile (chemapp) = 0,099774132 = R * 0,012; Reference: HMW
Remark:

502 psi ((HCO3)<-> (CO3)<2-> Na<+>)
Value in this parameterfile (chemapp) = 0,016629022 = R * 0,002; Reference: HMW
Remark:

503 theta (K<+> Ca<2+>)
Value in this parameterfile (chemapp) = 0,266064352 = R * 0,032; Reference: HMW
Remark:

504 psi (K<+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,207862775 = R * -0,025; Reference:
HMW
Remark:

505 theta (H<+> K<+>)
Value in this parameterfile (chemapp) = 0,041572555 = R * 0,005; Reference: HMW
Remark:

506 psi (H<+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,091459621 = R * -0,011; Reference:
HMW
Remark:

507 psi (H<+> K<+> (HSO4)<->)
Value in this parameterfile (chemapp) = -0,2203345415 = R * -0,0265; Reference:
HMW
Remark:

508 psi (H<+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,637958667 = R * 0,197; Reference: HMW
Remark:

509 psi (K<+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,182919242 = R * -0,022; Reference:
HMW
Remark:

510 psi (K<+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,399096528 = R * -0,048; Reference:
HMW
Remark:

511 theta (H<+> Mg<2+>)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: HMW
Remark:

512 psi (H<+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,091459621 = R * -0,011; Reference:
HMW
Remark:

513 psi (H<+> Mg<2+> (HSO4)<->)
Value in this parameterfile (chemapp) = -0,1479982958 = R * -0,0178; Reference:
HMW
Remark:

514 theta (Na<+> Ca<2+>)
Value in this parameterfile (chemapp) = 0,58201577 = R * 0,07; Reference: HMW

Remark:
515 psi (Na<+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,058201577 = R * -0,007; Reference:
HMW
Remark:
516 psi (Na<+> Ca<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,457298105 = R * -0,055; Reference:
HMW
Remark:
517 theta (H<+> Na<+>)
Value in this parameterfile (chemapp) = 0,299322396 = R * 0,036; Reference: HMW
Remark:
518 psi (H<+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,033258044 = R * -0,004; Reference:
HMW
Remark:
519 psi (H<+> Na<+> (HSO4)<->)
Value in this parameterfile (chemapp) = -0,1072571919 = R * -0,0129; Reference:
HMW
Remark:
520 theta (Na<+> K<+>)
Value in this parameterfile (chemapp) = -0,099774132 = R * -0,012; Reference:
HMW
Remark:
521 psi (Na<+> K<+> (CO3)<2->)
Value in this parameterfile (chemapp) = 0,024943533 = R * 0,003; Reference: HMW
Remark:
522 psi (Na<+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,0149661198 = R * -0,0018; Reference:
HMW
Remark:
523 psi (Na<+> K<+> (HCO3)<->)
Value in this parameterfile (chemapp) = -0,024943533 = R * -0,003; Reference:
HMW
Remark:
524 psi (Na<+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,08314511 = R * -0,01; Reference: HMW
Remark:
525 theta (Na<+> Mg<2+>)
Value in this parameterfile (chemapp) = 0,58201577 = R * 0,07; Reference: HMW
Remark:
526 psi (Na<+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,099774132 = R * -0,012; Reference:
HMW
Remark:
527 psi (Na<+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,124717665 = R * -0,015; Reference:
HMW
Remark:
528 theta (OH<-> (CO3)<2->)
Value in this parameterfile (chemapp) = 0,8314511 = R * 0,1; Reference: HMW

Remark:
529 $\psi(\text{OH} \leftrightarrow (\text{CO}_3)^{2-} \rightarrow \text{K}^{+})$
Value in this parameterfile (chemapp) = -0,08314511 = R * -0,01; Reference: HMW
Remark:
530 $\psi(\text{OH} \leftrightarrow (\text{CO}_3)^{2-} \rightarrow \text{Na}^{+})$
Value in this parameterfile (chemapp) = -0,141346687 = R * -0,017; Reference: HMW
Remark:
531 $\theta((\text{CO}_3)^{2-} \rightarrow (\text{SO}_4)^{2-})$
Value in this parameterfile (chemapp) = 0,16629022 = R * 0,02; Reference: HMW
Remark:
532 $\psi((\text{CO}_3)^{2-} \rightarrow (\text{SO}_4)^{2-} \rightarrow \text{K}^{+})$
Value in this parameterfile (chemapp) = -0,074830599 = R * -0,009; Reference: HMW
Remark:
533 $\psi((\text{CO}_3)^{2-} \rightarrow (\text{SO}_4)^{2-} \rightarrow \text{Na}^{+})$
Value in this parameterfile (chemapp) = -0,041572555 = R * -0,005; Reference: HMW
Remark:
534 $\theta((\text{HCO}_3)^{-} \rightarrow (\text{SO}_4)^{2-})$
Value in this parameterfile (chemapp) = 0,08314511 = R * 0,01; Reference: HMW
Remark:
535 $\psi((\text{HCO}_3)^{-} \rightarrow (\text{SO}_4)^{2-} \rightarrow \text{Mg}^{2+})$
Value in this parameterfile (chemapp) = -1,338636271 = R * -0,161; Reference: HMW
Remark:
536 $\psi((\text{HCO}_3)^{-} \rightarrow (\text{SO}_4)^{2-} \rightarrow \text{Na}^{+})$
Value in this parameterfile (chemapp) = -0,041572555 = R * -0,005; Reference: HMW
Remark:
537 $\psi((\text{SO}_4)^{2-} \rightarrow (\text{HSO}_4)^{-} \rightarrow \text{K}^{+})$
Value in this parameterfile (chemapp) = -0,5628923947 = R * -0,0677; Reference: HMW
Remark:
538 $\psi((\text{SO}_4)^{2-} \rightarrow (\text{HSO}_4)^{-} \rightarrow \text{Mg}^{2+})$
Value in this parameterfile (chemapp) = -0,3533667175 = R * -0,0425; Reference: HMW
Remark:
539 $\psi((\text{SO}_4)^{2-} \rightarrow (\text{HSO}_4)^{-} \rightarrow \text{Na}^{+})$
Value in this parameterfile (chemapp) = -0,0781564034 = R * -0,0094; Reference: HMW
Remark:
540 $\theta(\text{OH} \leftrightarrow (\text{SO}_4)^{2-})$
Value in this parameterfile (chemapp) = -0,108088643 = R * -0,013; Reference: HMW
Remark:
541 $\psi(\text{OH} \leftrightarrow (\text{SO}_4)^{2-} \rightarrow \text{K}^{+})$
Value in this parameterfile (chemapp) = -0,41572555 = R * -0,05; Reference: HMW
Remark:
542 $\psi(\text{OH} \leftrightarrow (\text{SO}_4)^{2-} \rightarrow \text{Na}^{+})$
Value in this parameterfile (chemapp) = -0,074830599 = R * -0,009; Reference: HMW

Remark:
543 beta0 (Mg(OH)<+> Cl<->)
Value in this parameterfile (chemapp) = -0,8314511 = R * -0,1; Reference: HMW
Remark:
544 beta1 (Mg(OH)<+> Cl<->)
Value in this parameterfile (chemapp) = 13,785459238 = R * 1,658, alpha1 = 2;
Reference: HMW
Remark:
545 psi (Mg<2+> Mg(OH)<+> Cl<->)
Value in this parameterfile (chemapp) = 0,232806308 = R * 0,028; Reference: HMW
Remark:
546 theta ((SO4)<2-> I<->)
Value in this parameterfile (chemapp) = 0,86811809351 = R * 0,10441; Reference:
I/Se-AB
Remark:
547 theta ((SO4)<2-> SeO3<2->)
Value in this parameterfile (chemapp) = 0,47326196612 = R * 0,05692; Reference:
I/Se-AB
Remark:
548 theta ((SO4)<2-> SeO4<2->)
Value in this parameterfile (chemapp) = -0,49038985878 = R * -0,05898;
Reference: I/Se-AB
Remark:
549 beta0 (Ca<2+> I<->)
Value in this parameterfile (chemapp) = 3,68191490613 = R * 0,44283; Reference:
I/Se-AB
Remark:
550 beta1 (Ca<2+> I<->)
Value in this parameterfile (chemapp) = 15,03246959778 = R * 1,80798, alpha1 =
2; Reference: I/Se-AB
Remark:
551 cphi (Ca<2+> I<->)
Value in this parameterfile (chemapp) = -0,0223411390193305 = R * -0,00269;
Reference: I/Se-AB
Remark: calculated from cphi = cgamma*2*WURZEL(2) where cgamma = -0,00095
552 theta (Cl<-> I<->)
Value in this parameterfile (chemapp) = 0,14317587942 = R * 0,01722; Reference:
I/Se-AB
Remark:
553 theta (Cl<-> SeO3<2->)
Value in this parameterfile (chemapp) = -0,06443746025 = R * -0,00775;
Reference: I/Se-AB
Remark:
554 theta (Cl<-> SeO4<2->)
Value in this parameterfile (chemapp) = 0,02635699987 = R * 0,00317; Reference:
I/Se-AB
Remark:
555 psi (K<+> (SO4)<2-> SeO3<2->)
Value in this parameterfile (chemapp) = -0,85448229547 = R * -0,10277;
Reference: I/Se-AB
Remark:
556 psi (K<+> (SO4)<2-> SeO4<2->)

- Value in this parameterfile (chemapp) = 0,06052964008 = R * 0,00728; Reference: I/Se-AB
Remark:
- 557 psi (K<+> (SO4)<2-> I<->)
Value in this parameterfile (chemapp) = -0,05753641612 = R * -0,00692; Reference: I/Se-AB
Remark:
- 558 psi (K<+> Cl<-> SeO3<2->)
Value in this parameterfile (chemapp) = 0,02111885794 = R * 0,00254; Reference: I/Se-AB
Remark:
- 559 psi (K<+> Cl<-> I<->)
Value in this parameterfile (chemapp) = -0,02261546992 = R * -0,00272; Reference: I/Se-AB
Remark:
- 560 beta0 (K<+> HSeO3<->)
Value in this parameterfile (chemapp) = 0,22241316925 = R * 0,02675; Reference: I/Se-AB
Remark:
- 561 betal (K<+> HSeO3<->)
Value in this parameterfile (chemapp) = 11,84102769554 = R * 1,42414, alpha1 = 2; Reference: I/Se-AB
Remark:
- 562 beta0 (K<+> I<->)
Value in this parameterfile (chemapp) = 0,55399586793 = R * 0,06663; Reference: I/Se-AB
Remark:
- 563 betal (K<+> I<->)
Value in this parameterfile (chemapp) = 2,67777141266 = R * 0,32206, alpha1 = 2; Reference: I/Se-AB
Remark:
- 564 cphi (K<+> I<->)
Value in this parameterfile (chemapp) = -0,01962224596 = R * -0,00236; Reference: I/Se-AB
Remark: calculated from cgamma=-0,00118
- 565 psi (K<+> Mg<2+> SeO4<2->)
Value in this parameterfile (chemapp) = -0,37980686248 = R * -0,04568; Reference: I/Se-AB
Remark:
- 566 psi (K<+> Mg<2+> I<->)
Value in this parameterfile (chemapp) = -0,13353104666 = R * -0,01606; Reference: I/Se-AB
Remark:
- 567 beta0 (K<+> SeO3<2->)
Value in this parameterfile (chemapp) = 1,98592095235 = R * 0,23885; Reference: I/Se-AB
Remark:
- 568 betal (K<+> SeO3<2->)
Value in this parameterfile (chemapp) = 19,43508631739 = R * 2,33749, alpha1 = 2; Reference: I/Se-AB
Remark:
- 569 cphi (K<+> SeO3<2->)

- Value in this parameterfile (chemapp) = $-0,0475043166516292 = R * -0,00571$;
Reference: I/Se-AB
Remark: calculated from $cphi = cgamma * 2 * WURZEL(2)$ where $cgamma = 0,00202$
- 570 beta0 (K<+> SeO4<2->)
Value in this parameterfile (chemapp) = $0,78829878791 = R * 0,09481$; Reference:
I/Se-AB
Remark:
- 571 beta1 (K<+> SeO4<2->)
Value in this parameterfile (chemapp) = $13,49736143185 = R * 1,62335$, $alpha1 = 2$;
Reference: I/Se-AB
Remark:
- 572 cphi (K<+> SeO4<2->)
Value in this parameterfile (chemapp) = $0,00493856757269412 = R * 0,00059$;
Reference: I/Se-AB
Remark: calculated from $cphi = cgamma * 2 * WURZEL(2)$ with $cgamma = 0,00021$
- 573 psi (Mg<2+> (SO4)<2-> SeO4<2->)
Value in this parameterfile (chemapp) = $0,5030279155 = R * 0,0605$; Reference:
I/Se-AB
Remark:
- 574 psi (Mg<2+> (SO4)<2-> I<->)
Value in this parameterfile (chemapp) = $-0,47259680524 = R * -0,05684$;
Reference: I/Se-AB
Remark:
- 575 psi (Mg<2+> Cl<-> SeO4<2->)
Value in this parameterfile (chemapp) = $0,07807325829 = R * 0,00939$; Reference:
I/Se-AB
Remark:
- 576 psi (Mg<2+> Cl<-> I<->)
Value in this parameterfile (chemapp) = $-0,09436969985 = R * -0,01135$;
Reference: I/Se-AB
Remark:
- 577 beta0 (Mg<2+> I<->)
Value in this parameterfile (chemapp) = $4,1140200428 = R * 0,4948$; Reference:
I/Se-AB
Remark:
- 578 beta1 (Mg<2+> I<->)
Value in this parameterfile (chemapp) = $15,22337077034 = R * 1,83094$, $alpha1 = 2$;
Reference: I/Se-AB
Remark:
- 579 cphi (Mg<2+> I<->)
Value in this parameterfile (chemapp) = $0,0592628108723294 = R * 0,00713$;
Reference: I/Se-AB
Remark: calculated from $cphi = cgamma * 2 * WURZEL(2)$ where $cgamma = 0,00252$
- 580 beta0 (Mg<2+> SeO4<2->)
Value in this parameterfile (chemapp) = $2,72391694871 = R * 0,32761$; Reference:
I/Se-AB
Remark:
- 581 beta1 (Mg<2+> SeO4<2->)
Value in this parameterfile (chemapp) = $32,46010037933 = R * 3,90403$, $alpha1 = 1,4$;
Reference: I/Se-AB
Remark:
- 582 cphi (Mg<2+> SeO4<2->)

- Value in this parameterfile (chemapp) = 0,07449801856 = R * 0,00896; Reference: I/Se-AB
Remark: calculated from $cphi = cgamma * 2 * WURZEL(4)$ with $cgamma = 0,00224$
- 583 $\psi (Na^{+} (SO_4)^{2-} SeO_3^{2-})$
Value in this parameterfile (chemapp) = -0,16446102758 = R * -0,01978; Reference: I/Se-AB
Remark:
- 584 $\psi (Na^{+} (SO_4)^{2-} SeO_4^{2-})$
Value in this parameterfile (chemapp) = 0,21601099578 = R * 0,02598; Reference: I/Se-AB
Remark:
- 585 $\psi (Na^{+} (SO_4)^{2-} I^{-})$
Value in this parameterfile (chemapp) = -0,15664538724 = R * -0,01884; Reference: I/Se-AB
Remark:
- 586 $\psi (Na^{+} Cl^{-} I^{-})$
Value in this parameterfile (chemapp) = -0,03334118911 = R * -0,00401; Reference: I/Se-AB
Remark:
- 587 $\beta_0 (Na^{+} HSeO_3^{-})$
Value in this parameterfile (chemapp) = -0,93795998591 = R * -0,11281; Reference: I/Se-AB
Remark:
- 588 $\beta_1 (Na^{+} HSeO_3^{-})$
Value in this parameterfile (chemapp) = 14,72840793051 = R * 1,77141, $\alpha_1 = 2$; Reference: I/Se-AB
Remark:
- 589 $\beta_0 (Na^{+} I^{-})$
Value in this parameterfile (chemapp) = 1,04064419676 = R * 0,12516; Reference: I/Se-AB
Remark:
- 590 $\beta_1 (Na^{+} I^{-})$
Value in this parameterfile (chemapp) = 2,62123273786 = R * 0,31526, $\alpha_1 = 2$; Reference: I/Se-AB
Remark:
- 591 $cphi (Na^{+} I^{-})$
Value in this parameterfile (chemapp) = 0,0016629022 = R * 0,0002; Reference: I/Se-AB
Remark: calculated from $cgamma=0,0001$
- 592 $\psi (Na^{+} K^{+} SeO_3^{2-})$
Value in this parameterfile (chemapp) = 0,00706733435 = R * 0,00085; Reference: I/Se-AB
Remark:
- 593 $\psi (Na^{+} K^{+} SeO_4^{2-})$
Value in this parameterfile (chemapp) = 0,16204981939 = R * 0,01949; Reference: I/Se-AB
Remark:
- 594 $\psi (Na^{+} K^{+} I^{-})$
Value in this parameterfile (chemapp) = -0,03051425537 = R * -0,00367; Reference: I/Se-AB
Remark:
- 595 $\psi (Na^{+} Mg^{2+} SeO_4^{2-})$

Value in this parameterfile (chemapp) = -0,12945693627 = R * -0,01557;
Reference: I/Se-AB
Remark:

596 psi (Na<+> Mg<2+> I<->)
Value in this parameterfile (chemapp) = -0,15157353553 = R * -0,01823;
Reference: I/Se-AB
Remark:

597 beta0 (Na<+> SeO3<2->)
Value in this parameterfile (chemapp) = 0,76460243156 = R * 0,09196; Reference:
I/Se-AB
Remark:

598 beta1 (Na<+> SeO3<2->)
Value in this parameterfile (chemapp) = 13,30554566308 = R * 1,60028, alpha1 =
2; Reference: I/Se-AB
Remark:

599 cphi (Na<+> SeO3<2->)
Value in this parameterfile (chemapp) = 0,0277500463608527 = R * 0,00334;
Reference: I/Se-AB
Remark: calculated from cphi =cgamma*2*WURZEL(2) where cgamma = 0,00118

600 beta0 (Na<+> SeO4<2->)
Value in this parameterfile (chemapp) = 0,81241086981 = R * 0,09771; Reference:
I/Se-AB
Remark:

601 beta1 (Na<+> SeO4<2->)
Value in this parameterfile (chemapp) = 0,650735203415 = R * 0,07827, alpha1 =
2; Reference: I/Se-AB
Remark:

602 beta0 (Cs<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,5936560854 = R * 0,0714; Reference:
SM5322
Remark:

603 beta1 (Cs<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 9,985727711 = R * 1,201, alpha1 = 2;
Reference: SM5322
Remark:

604 cphi (Cs<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,02419522701 = R * 0,00291; Reference:
SM5322
Remark:

605 beta0 (Cs<+> Cl<->)
Value in this parameterfile (chemapp) = 0,28917869258 = R * 0,03478; Reference:
SM5322
Remark:

606 beta1 (Cs<+> Cl<->)
Value in this parameterfile (chemapp) = 0,33041866714 = R * 0,03974, alpha1 =
2; Reference: SM5322
Remark:

607 cphi (Cs<+> Cl<->)
Value in this parameterfile (chemapp) = -0,004123997456 = R * -0,0005;
Reference: SM5322
Remark:

608 psi (Cs<+> K<+> Cl<->)

- Value in this parameterfile (chemapp) = -0,0108088643 = R * -0,0013; Reference: SM5322
Remark:
- 609 psi (Cs<+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,0349209462 = R * -0,0042; Reference: SM5322
Remark:
- 610 psi (Cs<+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,365838484 = R * -0,044; Reference: SM5322
Remark:
- 611 psi (Cs<+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,66516088 = R * -0,08; Reference: SM5322
Remark:
- 612 theta (Cs<+> Mg<2+>)
Value in this parameterfile (chemapp) = -0,648531858 = R * -0,078; Reference: SM5322
Remark:
- 613 psi (Cs<+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,01122458985 = R * -0,00135; Reference: SM5322
Remark:
- 614 psi (Cs<+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,0191233753 = R * 0,0023; Reference: SM5322
Remark:
- 615 theta (Cs<+> Na<+>)
Value in this parameterfile (chemapp) = -0,28152934246 = R * -0,03386; Reference: SM5322
Remark:
- 616 beta0 (Li<+> Cl<->)
Value in this parameterfile (chemapp) = 1,74371924692 = R * 0,20972; Reference: SM250
Remark:
- 617 beta1 (Li<+> Cl<->)
Value in this parameterfile (chemapp) = -2,8585288818 = R * -0,3438, alpha1 = 2; Reference: SM250
Remark:
- 618 cphi (Li<+> Cl<->)
Value in this parameterfile (chemapp) = -0,03600183263 = R * -0,00433; Reference: SM250
Remark:
- 619 psi (Li<+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,0182919242 = R * -0,0022; Reference: SM251
Remark:
- 620 theta (Li<+> Na<+>)
Value in this parameterfile (chemapp) = 0,099774132 = R * 0,012; Reference: SM251
Remark:
- 621 beta0 (Li<+> OH<->)

- Value in this parameterfile (chemapp) = 0,42279288435 = R * 0,05085; Reference: SM250
Remark:
- 622 beta1 (Li<+> OH<->)
Value in this parameterfile (chemapp) = -0,60255261217 = R * -0,07247, alpha1 = 2; Reference: SM250
Remark:
- 623 cphi (Li<+> OH<->)
Value in this parameterfile (chemapp) = -0,02801990207 = R * -0,00337; Reference: SM250
Remark:
- 624 psi (Cd<2+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,48415397553 = R * -0,05823; Reference: ULT
Remark:
- 625 psi (Cd<2+> Ca<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,60238632195 = R * 0,07245; Reference: ULT
Remark:
- 626 theta (Cd<2+> Ca<2+>)
Value in this parameterfile (chemapp) = -1,67994694755 = R * -0,20205; Reference: ULT
Remark:
- 627 psi (Cd<2+> Cl<-> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,0440669083 = R * -0,0053; Reference: ULT
Remark:
- 628 psi (Cd<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,46212052138 = R * -0,05558; Reference: ULT
Remark:
- 629 psi (Cd<2+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,29391796385 = R * -0,03535; Reference: ULT
Remark:
- 630 theta (Cd<2+> K<+>)
Value in this parameterfile (chemapp) = -0,45912729742 = R * -0,05522; Reference: ULT
Remark:
- 631 psi (Cd<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,77383153877 = R * -0,09307; Reference: ULT
Remark:
- 632 psi (Cd<2+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,16595763956 = R * 0,01996; Reference: ULT
Remark:
- 633 theta (Cd<2+> Mg<2+>)
Value in this parameterfile (chemapp) = -0,55017119287 = R * -0,06617; Reference: ULT
Remark:
- 634 psi (Cd<2+> Na<+> (SO4)<2->)

- Value in this parameterfile (chemapp) = -0,1180660562 = R * -0,0142; Reference: ULT
Remark:
- 635 psi (Cd<2+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,48091131624 = R * -0,05784; Reference: ULT
Remark:
- 636 theta (Cd<2+> Na<+>)
Value in this parameterfile (chemapp) = 0,224491797 = R * 0,027; Reference: ULT
Remark:
- 637 psi (Cl<-> (SO4)<2-> Zn<2+>)
Value in this parameterfile (chemapp) = 0,226187957244 = R * 0,0272; Reference: ULT
Remark:
- 638 beta0 (Zn<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,51382301777 = R * 0,18207; Reference: ULT
Remark:
- 639 beta1 (Zn<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 24,469605873 = R * 2,943, alpha1 = 1,4; Reference: ULT
Remark:
- 640 beta2 (Zn<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -1645,35858179 = R * -197,89, alpha2 = 20; Reference: ULT
Remark:
- 641 cphi (Zn<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,27803724784 = R * 0,03344; Reference: ULT
Remark:
- 642 psi (Zn<2+> Ca<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,291007885 = R * 0,035; Reference: ULT
Remark:
- 643 psi (Zn<2+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,2386264657 = R * -0,0287; Reference: ULT
Remark:
- 644 beta0 (Zn<2+> Cl<->)
Value in this parameterfile (chemapp) = 0,54227240742 = R * 0,06522; Reference: ULT
Remark:
- 645 beta1 (Zn<2+> Cl<->)
Value in this parameterfile (chemapp) = 45,8852918557 = R * 5,5187, alpha1 = 2; Reference: ULT
Remark:
- 646 beta2 (Zn<2+> Cl<->)
Value in this parameterfile (chemapp) = -36,2329760358 = R * -4,3578, alpha2 = 2,5; Reference: ULT
Remark:
- 647 cphi (Zn<2+> Cl<->)
Value in this parameterfile (chemapp) = 0,03101312603 = R * 0,00373; Reference: ULT

Remark:
648 beta0 (Zn<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 1,837506931 = R * 0,221; Reference: ULT
Remark:
649 beta1 (Zn<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 27,795410273 = R * 3,343, alpha1 = 1,4;
Reference: ULT
Remark:
650 beta2 (Zn<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = -309,54924453 = R * -37,23, alpha2 =
12; Reference: ULT
Remark:
651 cphi (Zn<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 0,207862775 = R * 0,025; Reference: ULT
Remark:
652 beta0 (Zn<2+> H3SiO4<->)
Value in this parameterfile (chemapp) = 3,9460669206 = R * 0,4746; Reference:
ULT
Remark:
653 beta1 (Zn<2+> H3SiO4<->)
Value in this parameterfile (chemapp) = 14,375789519 = R * 1,729, alpha1 = 2;
Reference: ULT
Remark:
654 psi (Zn<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,2145143838 = R * -0,0258; Reference:
ULT
Remark:
655 psi (Zn<2+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,1779305354 = R * -0,0214; Reference:
ULT
Remark:
656 theta (Zn<2+> K<+>)
Value in this parameterfile (chemapp) = -0,814822078 = R * -0,098; Reference:
ULT
Remark:
657 psi (Zn<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,19347867097 = R * -0,02327;
Reference: ULT
Remark:
658 psi (Zn<2+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,21476381913 = R * -0,02583;
Reference: ULT
Remark:
659 psi (Zn<2+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,4373432786 = R * 0,0526; Reference:
ULT
Remark:
660 theta (Zn<2+> Na<+>)
Value in this parameterfile (chemapp) = -1,06558772976 = R * -0,12816;
Reference: ULT
Remark:
661 lambda (Zn<2+> SiO2<0>)

- Value in this parameterfile (chemapp) = 2,4319944675 = R * 0,2925; Reference: ULT
Remark:
- 662 beta0 (Cd<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,86968408857 = R * 0,22487; Reference: ULT
Remark:
- 663 beta1 (Cd<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 19,4434839735 = R * 2,3385, alpha1 = 1,4; Reference: ULT
Remark:
- 664 beta2 (Cd<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -497,407306064 = R * -59,824, alpha2 = 12; Reference: ULT
Remark:
- 665 cphi (Cd<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,04331860231 = R * 0,00521; Reference: ULT
Remark:
- 666 beta0 (Cd<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,33324560088 = R * -0,04008; Reference: ULT
Remark:
- 667 beta1 (Cd<2+> Cl<->)
Value in this parameterfile (chemapp) = -26,40788467732 = R * -3,17612, alpha1 = 2,5; Reference: ULT
Remark:
- 668 beta2 (Cd<2+> Cl<->)
Value in this parameterfile (chemapp) = -373,5718106811 = R * -44,9301, alpha2 = 12; Reference: ULT
Remark:
- 669 cphi (Cd<2+> Cl<->)
Value in this parameterfile (chemapp) = 0,04431634363 = R * 0,00533; Reference: ULT
Remark:
- 670 beta0 (Ni<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,41712525484 = R * 0,17044; Reference: Koda
Remark:
- 671 beta1 (Ni<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 24,6425477018 = R * 2,9638, alpha1 = 1,4; Reference: Koda
Remark:
- 672 beta2 (Ni<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -618,566360356 = R * -74,396, alpha2 = 12; Reference: Koda
Remark:
- 673 cphi (Ni<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,0731984604907 = R * 0,0088; Reference: Koda
Remark:
- 674 beta0 (Ni<2+> Cl<->)

- Value in this parameterfile (chemapp) = 3,35947816955 = R * 0,40405; Reference: Koda
Remark:
- 675 beta1 (Ni<2+> Cl<->)
Value in this parameterfile (chemapp) = 15,3760251923 = R * 1,8493, alpha1 = 2,5; Reference: Koda
Remark:
- 676 cphi (Ni<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,0533733404623 = R * -0,00642; Reference: Koda
Remark:
- 677 psi (Ni<2+> Cl<-> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,0385344326806 = R * 0,00463; Reference: Koda
Remark:
- 678 psi (Ni<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = 0,124784181088 = R * 0,01501; Reference: Koda
Remark:
- 679 psi (Ni<2+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,86346196735 = R * -0,10385; Reference: Koda
Remark:
- 680 theta (Ni<2+> K<+>)
Value in this parameterfile (chemapp) = -2,03256535906 = R * -0,24446; Reference: Koda
Remark:
- 681 psi (Ni<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = 0,185205732525 = R * 0,02228; Reference: Koda
Remark:
- 682 psi (Ni<2+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,413971188179 = R * 0,04979; Reference: Koda
Remark:
- 683 theta (Ni<2+> Mg<2+>)
Value in this parameterfile (chemapp) = -1,24534745758 = R * -0,14978; Reference: Koda
Remark:
- 684 psi (Ni<2+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,0308202293748 = R * -0,00371; Reference: Koda
Remark:
- 685 psi (Ni<2+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,15440046927 = R * -0,01857; Reference: Koda
Remark:
- 686 theta (Ni<2+> Na<+>)
Value in this parameterfile (chemapp) = 0,373920188692 = R * 0,04497; Reference: Koda
Remark:
- 687 psi (Ni<2+> Ca<2+> (SO4)<2->)

- Value in this parameterfile (chemapp) = 0,171387015243 = R * 0,02061;
Reference: Koda
Remark:
- 688 theta (Ni<2+> Ca<2+>)
Value in this parameterfile (chemapp) = -0,509879072564 = R * -0,06132;
Reference: Koda
Remark:
- 689 psi ((CO3)<2-> H2SiO4<2-> K<+>)
Value in this parameterfile (chemapp) = -0,074830599 = R * -0,009; Reference:
R1990
Remark:
- 690 psi ((CO3)<2-> H2SiO4<2-> Na<+>)
Value in this parameterfile (chemapp) = -0,041572555 = R * -0,005; Reference:
R1990
Remark:
- 691 psi ((HCO3)<-> H2SiO4<2-> Mg<2+>)
Value in this parameterfile (chemapp) = -1,338636271 = R * -0,161; Reference:
R1990
Remark:
- 692 psi ((HCO3)<-> H2SiO4<2-> Na<+>)
Value in this parameterfile (chemapp) = -0,041572555 = R * -0,005; Reference:
R1990
Remark:
- 693 psi ((SO4)<2-> Al(OH)4<-> K<+>)
Value in this parameterfile (chemapp) = -0,5628923947 = R * -0,0677; Reference:
R1990
Remark:
- 694 psi ((SO4)<2-> Al(OH)4<-> Mg<2+>)
Value in this parameterfile (chemapp) = -0,3533667175 = R * -0,0425; Reference:
R1990
Remark:
- 695 psi ((SO4)<2-> Al(OH)4<-> Na<+>)
Value in this parameterfile (chemapp) = -0,0781564034 = R * -0,0094; Reference:
R1990
Remark:
- 696 psi ((SO4)<2-> H3SiO4<-> K<+>)
Value in this parameterfile (chemapp) = -0,5628923947 = R * -0,0677; Reference:
R1990
Remark:
- 697 psi ((SO4)<2-> H3SiO4<-> Mg<2+>)
Value in this parameterfile (chemapp) = -0,3533667175 = R * -0,0425; Reference:
R1990
Remark:
- 698 psi ((SO4)<2-> H3SiO4<-> Na<+>)
Value in this parameterfile (chemapp) = -0,0781564034 = R * -0,0094; Reference:
R1990
Remark:
- 699 beta0 (Al<3+> Cl<->)
Value in this parameterfile (chemapp) = 5,81458697763 = R * 0,69933; Reference:
R1990
Remark:
- 700 beta1 (Al<3+> Cl<->)

- Value in this parameterfile (chemapp) = 48,59557300637 = R * 5,84467, alpha1 = 2; Reference: R1990
Remark:
- 701 cphi (Al<3+> Cl<->)
Value in this parameterfile (chemapp) = 0,02269861503 = R * 0,00273; Reference: R1990
Remark:
- 702 beta0 (Ca<2+> Al(OH)4<->)
Value in this parameterfile (chemapp) = 1,7834626095 = R * 0,2145; Reference: R1990
Remark:
- 703 beta1 (Ca<2+> Al(OH)4<->)
Value in this parameterfile (chemapp) = 21,03571283 = R * 2,53, alpha1 = 2; Reference: R1990
Remark:
- 704 beta0 (Ca<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 1,6629022 = R * 0,2; Reference: R1990
Remark:
- 705 beta1 (Ca<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 26,5839860203 = R * 3,1973, alpha1 = 1,4; Reference: R1990
Remark:
- 706 beta2 (Ca<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = -450,97907664 = R * -54,24, alpha2 = 12; Reference: R1990
Remark:
- 707 beta0 (Ca<2+> H3SiO4<->)
Value in this parameterfile (chemapp) = 1,7834626095 = R * 0,2145; Reference: R1990
Remark:
- 708 beta1 (Ca<2+> H3SiO4<->)
Value in this parameterfile (chemapp) = 21,03571283 = R * 2,53, alpha1 = 2; Reference: R1990
Remark:
- 709 psi (Ca<2+> Mg<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 0,199548264 = R * 0,024; Reference: R1990
Remark:
- 710 psi (Cl<-> Al(OH)4<-> H<+>)
Value in this parameterfile (chemapp) = 0,108088643 = R * 0,013; Reference: R1990
Remark:
- 711 psi (Cl<-> Al(OH)4<-> Na<+>)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: R1990
Remark:
- 712 theta (Cl<-> Al(OH)4<->)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: R1990
Remark:
- 713 psi (Cl<-> H2SiO4<2-> Ca<2+>)

- Value in this parameterfile (chemapp) = -0,149661198 = R * -0,018; Reference: R1990
Remark:
- 714 psi (Cl<-> H2SiO4<2-> Mg<2+>)
Value in this parameterfile (chemapp) = -0,033258044 = R * -0,004; Reference: R1990
Remark:
- 715 psi (Cl<-> H2SiO4<2-> Na<+>)
Value in this parameterfile (chemapp) = 0,0116403154 = R * 0,0014; Reference: R1990
Remark:
- 716 psi (Cl<-> H3SiO4<-> H<+>)
Value in this parameterfile (chemapp) = 0,108088643 = R * 0,013; Reference: R1990
Remark:
- 717 theta (Cl<-> H3SiO4<->)
Value in this parameterfile (chemapp) = -0,049887066 = R * -0,006; Reference: R1990
Remark:
- 718 beta0 (H<+> Al(OH)4<->)
Value in this parameterfile (chemapp) = 1,7510360166 = R * 0,2106; Reference: R1990
Remark:
- 719 beta1 (H<+> Al(OH)4<->)
Value in this parameterfile (chemapp) = 4,423319852 = R * 0,532, alpha1 = 2; Reference: R1990
Remark:
- 720 beta0 (H<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 0,1804248887 = R * 0,0217; Reference: R1990
Remark:
- 721 cphi (H<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 0,3417264021 = R * 0,0411; Reference: R1990
Remark:
- 722 beta0 (H<+> H3SiO4<->)
Value in this parameterfile (chemapp) = 1,7510360166 = R * 0,2106; Reference: R1990
Remark:
- 723 beta1 (H<+> H3SiO4<->)
Value in this parameterfile (chemapp) = 4,423319852 = R * 0,532, alpha1 = 2; Reference: R1990
Remark:
- 724 psi (H2SiO4<2-> Al(OH)4<-> K<+>)
Value in this parameterfile (chemapp) = -0,5628923947 = R * -0,0677; Reference: R1990
Remark:
- 725 psi (H2SiO4<2-> Al(OH)4<-> Mg<2+>)
Value in this parameterfile (chemapp) = -0,3533667175 = R * -0,0425; Reference: R1990
Remark:
- 726 psi (H2SiO4<2-> Al(OH)4<-> Na<+>)

- Value in this parameterfile (chemapp) = -0,0781564034 = R * -0,0094; Reference: R1990
Remark:
- 727 psi (H3SiO4<-> H2SiO4<2-> K<+>)
Value in this parameterfile (chemapp) = -0,5628923947 = R * -0,0677; Reference: R1990
Remark:
- 728 psi (H3SiO4<-> H2SiO4<2-> Mg<2+>)
Value in this parameterfile (chemapp) = -0,3533667175 = R * -0,0425; Reference: R1990
Remark:
- 729 psi (H3SiO4<-> H2SiO4<2-> Na<+>)
Value in this parameterfile (chemapp) = -0,0781564034 = R * -0,0094; Reference: R1990
Remark:
- 730 beta0 (K<+> Al(OH)4<->)
Value in this parameterfile (chemapp) = -0,0024943533 = R * -0,0003; Reference: R1990
Remark:
- 731 beta1 (K<+> Al(OH)4<->)
Value in this parameterfile (chemapp) = 1,4425676585 = R * 0,1735, alpha1 = 2; Reference: R1990
Remark:
- 732 psi (K<+> H<+> H3SiO4<->)
Value in this parameterfile (chemapp) = -0,2203345415 = R * -0,0265; Reference: R1990
Remark:
- 733 psi (K<+> H<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 1,637958667 = R * 0,197; Reference: R1990
Remark:
- 734 psi (K<+> H<+> Al(OH)4<->)
Value in this parameterfile (chemapp) = -0,2203345415 = R * -0,0265; Reference: R1990
Remark:
- 735 beta0 (K<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 0,4148940989 = R * 0,0499; Reference: R1990
Remark:
- 736 beta1 (K<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 6,4794984223 = R * 0,7793, alpha1 = 2; Reference: R1990
Remark:
- 737 beta0 (K<+> H3SiO4<->)
Value in this parameterfile (chemapp) = -0,0024943533 = R * -0,0003; Reference: R1990
Remark:
- 738 beta1 (K<+> H3SiO4<->)
Value in this parameterfile (chemapp) = 1,4425676585 = R * 0,1735, alpha1 = 2; Reference: R1990
Remark:
- 739 psi (K<+> Mg<2+> H2SiO4<2->)

- Value in this parameterfile (chemapp) = -0,698418924 = R * -0,084; Reference: R1990
Remark:
- 740 beta0 (Mg<2+> Al(OH)4<->)
Value in this parameterfile (chemapp) = 3,9460669206 = R * 0,4746; Reference: R1990
Remark:
- 741 beta1 (Mg<2+> Al(OH)4<->)
Value in this parameterfile (chemapp) = 14,375789519 = R * 1,729, alpha1 = 2; Reference: R1990
Remark:
- 742 psi (Mg<2+> H<+> H3SiO4<->)
Value in this parameterfile (chemapp) = -0,1479982958 = R * -0,0178; Reference: R1990
Remark:
- 743 psi (Mg<2+> H<+> Al(OH)4<->)
Value in this parameterfile (chemapp) = -0,1479982958 = R * -0,0178; Reference: R1990
Remark:
- 744 beta0 (Mg<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 1,837506931 = R * 0,221; Reference: R1990
Remark:
- 745 beta1 (Mg<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 27,795410273 = R * 3,343, alpha1 = 1,4; Reference: R1990
Remark:
- 746 beta2 (Mg<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = -309,54924453 = R * -37,23, alpha2 = 12; Reference: R1990
Remark:
- 747 cphi (Mg<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 0,207862775 = R * 0,025; Reference: R1990
Remark:
- 748 beta0 (Mg<2+> H3SiO4<->)
Value in this parameterfile (chemapp) = 3,9460669206 = R * 0,4746; Reference: R1990
Remark:
- 749 beta1 (Mg<2+> H3SiO4<->)
Value in this parameterfile (chemapp) = 14,375789519 = R * 1,729, alpha1 = 2; Reference: R1990
Remark:
- 750 psi (Na<+> Ca<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = -0,457298105 = R * -0,055; Reference: R1990
Remark:
- 751 psi (Na<+> H<+> H3SiO4<->)
Value in this parameterfile (chemapp) = -0,1072571919 = R * -0,0129; Reference: R1990
Remark:
- 752 psi (Na<+> H<+> Al(OH)4<->)

- Value in this parameterfile (chemapp) = -0,1072571919 = R * -0,0129; Reference: R1990
Remark:
- 753 beta0 (Na<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 0,1629644156 = R * 0,0196; Reference: R1990
Remark:
- 754 beta1 (Na<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 9,254050743 = R * 1,113, alpha1 = 2; Reference: R1990
Remark:
- 755 cphi (Na<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = 0,041572555 = R * 0,005; Reference: R1990
Remark:
- 756 beta0 (Na<+> H3SiO4<->)
Value in this parameterfile (chemapp) = 0,3774787994 = R * 0,0454; Reference: R1990
Remark:
- 757 beta1 (Na<+> H3SiO4<->)
Value in this parameterfile (chemapp) = 3,309175378 = R * 0,398, alpha1 = 2; Reference: R1990
Remark:
- 758 psi (Na<+> K<+> H2SiO4<2->)
Value in this parameterfile (chemapp) = -0,08314511 = R * -0,01; Reference: R1990
Remark:
- 759 psi (Na<+> Mg<2+> H2SiO4<2->)
Value in this parameterfile (chemapp) = -0,124717665 = R * -0,015; Reference: R1990
Remark:
- 760 psi (OH<-> H2SiO4<2-> K<+>)
Value in this parameterfile (chemapp) = -0,41572555 = R * -0,05; Reference: R1990
Remark:
- 761 psi (OH<-> H2SiO4<2-> Na<+>)
Value in this parameterfile (chemapp) = -0,074830599 = R * -0,009; Reference: R1990
Remark:
- 762 beta0 (Sr<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,82919242 = R * 0,22; Reference: FZK-INE 002/04
Remark:
- 763 beta1 (Sr<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 23,94579168 = R * 2,88, alpha1 = 1,4; Reference: FZK-INE 002/04
Remark:
- 764 beta2 (Sr<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -347,5465598 = R * -41,8, alpha2 = 12; Reference: FZK-INE 002/04
Remark:
- 765 cphi (Sr<2+> (SO4)<2->)

- Value in this parameterfile (chemapp) = 1,57975709 = R * 0,19; Reference: FZK-INE 002/04
Remark:
- 766 beta0 (Sr<2+> Cl<->)
Value in this parameterfile (chemapp) = 2,37587151825 = R * 0,28575; Reference: FZK-INE 002/04
Remark:
- 767 beta1 (Sr<2+> Cl<->)
Value in this parameterfile (chemapp) = 13,86236846475 = R * 1,66725, alpha1 = 2; Reference: FZK-INE 002/04
Remark:
- 768 cphi (Sr<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,0108088643 = R * -0,0013; Reference: FZK-INE 002/04
Remark:
- 769 psi (Sr<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,207862775 = R * -0,025; Reference: FZK-INE 002/04
Remark:
- 770 theta (Sr<2+> K<+>)
Value in this parameterfile (chemapp) = 0,266064352 = R * 0,032; Reference: FZK-INE 002/04
Remark:
- 771 psi (Sr<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,099774132 = R * -0,012; Reference: FZK-INE 002/04
Remark:
- 772 psi (Sr<2+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,199548264 = R * 0,024; Reference: FZK-INE 002/04
Remark:
- 773 theta (Sr<2+> Mg<2+>)
Value in this parameterfile (chemapp) = 0,058201577 = R * 0,007; Reference: FZK-INE 002/04
Remark:
- 774 psi (Sr<2+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,058201577 = R * -0,007; Reference: FZK-INE 002/04
Remark:
- 775 psi (Sr<2+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,457298105 = R * -0,055; Reference: FZK-INE 002/04
Remark:
- 776 theta (Sr<2+> Na<+>)
Value in this parameterfile (chemapp) = 0,58201577 = R * 0,07; Reference: FZK-INE 002/04
Remark:
- 777 beta0 (Ag<+> Cl<->)
Value in this parameterfile (chemapp) = 0,4980392089 = R * 0,0599; Reference: /FRI1985/
Remark:
- 778 beta1 (Ag<+> Cl<->)

Value in this parameterfile (chemapp) = 1,49661198 = R * 0,18, alpha1 = 2;
Reference: /FRI1985/
Remark:

779 cphi (Ag<+> Cl<->)
Value in this parameterfile (chemapp) = 0,54593079226 = R * 0,06566; Reference:
/FRI1985/
Remark: calculated from cgamma = 0,03283

780 beta0 (H<+> AgCl2<->)
Value in this parameterfile (chemapp) = 1,8483157953 = R * 0,2223; Reference:
/FRI1985/
Remark:

781 beta1 (H<+> AgCl2<->)
Value in this parameterfile (chemapp) = 3,059740048 = R * 0,368, alpha1 = 2;
Reference: /FRI1985/
Remark:

782 cphi (H<+> AgCl2<->)
Value in this parameterfile (chemapp) = 0,54593079226 = R * 0,06566; Reference:
/FRI1985/
Remark: calculated from cgamma = -0,01011

783 beta0 (H<+> AgCl3<2->)
Value in this parameterfile (chemapp) = 2,9624602693 = R * 0,3563; Reference:
/FRI1985/
Remark:

784 beta1 (H<+> AgCl3<2->)
Value in this parameterfile (chemapp) = 13,6565843175 = R * 1,6425, alpha1 = 2;
Reference: /FRI1985/
Remark:

785 cphi (H<+> AgCl3<2->)
Value in this parameterfile (chemapp) = -0,00376271815062409 = R * -0,00045;
Reference: /FRI1985/
Remark: calculated from cgamma = -0,00016

786 beta0 (H<+> AgCl4<3->)
Value in this parameterfile (chemapp) = 5,1782774508 = R * 0,6228; Reference:
/FRI1985/
Remark:

787 beta1 (H<+> AgCl4<3->)
Value in this parameterfile (chemapp) = 42,90287676 = R * 5,16, alpha1 = 2;
Reference: /FRI1985/
Remark:

788 cphi (H<+> AgCl4<3->)
Value in this parameterfile (chemapp) = 0,0198735945790846 = R * 0,00239;
Reference: /FRI1985/
Remark: calculated from cgamma = 0,00069

789 beta0 (Na<+> AgCl2<->)
Value in this parameterfile (chemapp) = 1,230547628 = R * 0,148; Reference:
/FRI1985/
Remark:

790 beta1 (Na<+> AgCl2<->)
Value in this parameterfile (chemapp) = 2,436151723 = R * 0,293, alpha1 = 2;
Reference: /FRI1985/
Remark:

791 cphi (Na<+> AgCl2<->)

- Value in this parameterfile (chemapp) = 0,54593079226 = R * 0,06566; Reference: /FRI1985/
Remark: calculated from cgamma = -0,0152
- 792 beta0 (Na<+> AgCl3<2->)
Value in this parameterfile (chemapp) = 1,0709090168 = R * 0,1288; Reference: /FRI1985/
Remark:
- 793 beta1 (Na<+> AgCl3<2->)
Value in this parameterfile (chemapp) = 9,9774132 = R * 1,2, alpha1 = 2;
Reference: /FRI1985/
Remark:
- 794 cphi (Na<+> AgCl3<2->)
Value in this parameterfile (chemapp) = -0,25304279562947 = R * -0,03043;
Reference: /FRI1985/
Remark: calculated from cgamma = -0,01076
- 795 beta0 (Na<+> AgCl4<3->)
Value in this parameterfile (chemapp) = 2,8568659796 = R * 0,3436; Reference: /FRI1985/
Remark:
- 796 beta1 (Na<+> AgCl4<3->)
Value in this parameterfile (chemapp) = 33,258044 = R * 4, alpha1 = 2;
Reference: /FRI1985/
Remark:
- 797 cphi (Na<+> AgCl4<3->)
Value in this parameterfile (chemapp) = -0,561933087301364 = R * -0,06758;
Reference: /FRI1985/
Remark: calculated from cgamma = -0,01951
- 798 beta0 (K<+> AgCl2<->)
Value in this parameterfile (chemapp) = 1,8000916315 = R * 0,2165; Reference: /FRI1985/
Remark:
- 799 beta1 (K<+> AgCl2<->)
Value in this parameterfile (chemapp) = 3,009852982 = R * 0,362, alpha1 = 2;
Reference: /FRI1985/
Remark:
- 800 cphi (K<+> AgCl2<->)
Value in this parameterfile (chemapp) = 0,54593079226 = R * 0,06566; Reference: /FRI1985/
Remark: calculated from cgamma = -0,012
- 801 beta0 (K<+> AgCl3<2->)
Value in this parameterfile (chemapp) = 0,16629022 = R * 0,02; Reference: /FRI1985/
Remark:
- 802 beta1 (K<+> AgCl3<2->)
Value in this parameterfile (chemapp) = 6,992503751 = R * 0,841, alpha1 = 2;
Reference: /FRI1985/
Remark:
- 803 cphi (K<+> AgCl3<2->)
Value in this parameterfile (chemapp) = -0,572638668548104 = R * -0,06887;
Reference: /FRI1985/
Remark: calculated from cgamma = -0,02435
- 804 beta0 (K<+> AgCl4<3->)

- Value in this parameterfile (chemapp) = 1,8632819151 = R * 0,2241; Reference: /FRI1985/
Remark:
- 805 beta1 (K<+> AgCl4<3->)
Value in this parameterfile (chemapp) = 29,208877143 = R * 3,513, alpha1 = 2; Reference: /FRI1985/
Remark:
- 806 cphi (K<+> AgCl4<3->)
Value in this parameterfile (chemapp) = -12,6154122110711 = R * -1,51728; Reference: /FRI1985/
Remark: calculated from cgamma = -0,438
- 807 beta0 (Na<+> MoO4<2->)
Value in this parameterfile (chemapp) = 0,964483276 = R * 0,116; Reference: /GMR1992/
Remark:
- 808 beta1 (Na<+> MoO4<2->)
Value in this parameterfile (chemapp) = 19,015286657 = R * 2,287, alpha1 = 2; Reference: /GMR1992/
Remark:
- 809 cphi (Na<+> MoO4<2->)
Value in this parameterfile (chemapp) = 0,141101930648403 = R * 0,01697; Reference: /GMR1992/
Remark: calculated from cgamma = 0,006
- 810 theta (Cl<-> MoO4<2->)
Value in this parameterfile (chemapp) = 0,291007885 = R * 0,035; Reference: /GMR1992/
Remark:
- 811 beta0 (Mg<2+> MoO4<2->)
Value in this parameterfile (chemapp) = 4,115682945 = R * 0,495; Reference: /GMR1992/
Remark:
- 812 beta1 (Mg<2+> MoO4<2->)
Value in this parameterfile (chemapp) = 54,29375683 = R * 6,53, alpha1 = 2; Reference: /GMR1992/
Remark:
- 813 cphi (Mg<2+> MoO4<2->)
Value in this parameterfile (chemapp) = -2,760417652 = R * -0,332; Reference: /GMR1992/
Remark: calculated from cgamma = -0,083
- 814 psi (Mg<2+> Na<+> MoO4<2->)
Value in this parameterfile (chemapp) = 0,49887066 = R * 0,06; Reference: /GMR1992/
Remark:
- 815 beta0 (Na<+> (AsO4)<3->)
Value in this parameterfile (chemapp) = 1,98517264636 = R * 0,23876; Reference: Koda
Remark:
- 816 beta1 (Na<+> (AsO4)<3->)
Value in this parameterfile (chemapp) = 32,6361185772 = R * 3,9252, alpha1 = 2; Reference: Koda
Remark:
- 817 cphi (Na<+> (AsO4)<3->)

- Value in this parameterfile (chemapp) = -0,114781824355 = R * -0,01381;
Reference: Koda
Remark:
- 818 beta0 (K<+> (AsO4)<3->)
Value in this parameterfile (chemapp) = 2,07180985098 = R * 0,24918; Reference:
Koda
Remark:
- 819 beta1 (K<+> (AsO4)<3->)
Value in this parameterfile (chemapp) = 61,2929121898 = R * 7,3718, alpha1 = 2;
Reference: Koda
Remark:
- 820 cphi (K<+> (AsO4)<3->)
Value in this parameterfile (chemapp) = -0,0360957866043 = R * -0,00434;
Reference: Koda
Remark:
- 821 beta0 (Na<+> (HAsO4)<2->)
Value in this parameterfile (chemapp) = 0,16163409384 = R * 0,01944; Reference:
Koda
Remark:
- 822 beta1 (Na<+> (HAsO4)<2->)
Value in this parameterfile (chemapp) = 14,4830467109 = R * 1,7419, alpha1 = 2;
Reference: Koda
Remark:
- 823 cphi (Na<+> (HAsO4)<2->)
Value in this parameterfile (chemapp) = 0,0131003435316 = R * 0,00158;
Reference: Koda
Remark:
- 824 beta0 (K<+> (HAsO4)<2->)
Value in this parameterfile (chemapp) = 0,589698378164 = R * 0,07092;
Reference: Koda
Remark:
- 825 beta1 (K<+> (HAsO4)<2->)
Value in this parameterfile (chemapp) = 17,5619101342 = R * 2,1122, alpha1 = 2;
Reference: Koda
Remark:
- 826 cphi (K<+> (HAsO4)<2->)
Value in this parameterfile (chemapp) = 0,0026589806178 = R * 0,00032;
Reference: Koda
Remark:
- 827 beta0 (Na<+> (H2AsO4)<->)
Value in this parameterfile (chemapp) = -0,766065785496 = R * -0,09214;
Reference: Koda
Remark:
- 828 beta1 (Na<+> (H2AsO4)<->)
Value in this parameterfile (chemapp) = 3,32755044731 = R * 0,40021, alpha1 =
2; Reference: Koda
Remark:
- 829 cphi (Na<+> (H2AsO4)<->)
Value in this parameterfile (chemapp) = 0,145138104016 = R * 0,01746;
Reference: Koda
Remark:
- 830 beta0 (K<+> (H2AsO4)<->)

Value in this parameterfile (chemapp) = -2,0777962989 = R * -0,2499; Reference:
Koda
Remark:

831 beta1 (K<+> (H2AsO4)<->)
Value in this parameterfile (chemapp) = 4,66626986342 = R * 0,56122, alpha1 =
2; Reference: Koda
Remark:

832 cphi (K<+> (H2AsO4)<->)
Value in this parameterfile (chemapp) = 0,509962217674 = R * 0,06133;
Reference: Koda
Remark:

833 theta ((AsO4)<3-> Cl<->)
Value in this parameterfile (chemapp) = 0,813117603245 = R * 0,0978; Reference:
Koda
Remark:

834 theta ((AsO4)<3-> (SO4)<2->)
Value in this parameterfile (chemapp) = -2,16085826379 = R * -0,25989;
Reference: Koda
Remark:

835 theta ((HAsO4)<2-> Cl<->)
Value in this parameterfile (chemapp) = -0,113734195969 = R * -0,01368;
Reference: Koda
Remark:

836 theta ((H2AsO4)<-> Cl<->)
Value in this parameterfile (chemapp) = 3,69546755906 = R * 0,44446; Reference:
Koda
Remark:

837 psi (Na<+> (AsO4)<3-> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,62159284236 = R * 0,07476; Reference:
Koda
Remark:

838 psi (Na<+> (HAsO4)<2-> Cl<->)
Value in this parameterfile (chemapp) = 0,207264130208 = R * 0,02493;
Reference: Koda
Remark:

839 psi (K<+> (HAsO4)<2-> Cl<->)
Value in this parameterfile (chemapp) = 0,103923072989 = R * 0,0125; Reference:
Koda
Remark:

840 psi (Na<+> (H2AsO4)<-> Cl<->)
Value in this parameterfile (chemapp) = -0,696656247668 = R * -0,08379;
Reference: Koda
Remark:

841 psi (K<+> (H2AsO4)<-> Cl<->)
Value in this parameterfile (chemapp) = -1,48139642487 = R * -0,17817;
Reference: Koda
Remark:

842 beta0 (Cu<2+> Cl<->)
Value in this parameterfile (chemapp) = 1,24476544181 = R * 0,14971; Reference:
Koda
Remark:

843 beta1 (Cu<2+> Cl<->)

Value in this parameterfile (chemapp) = 7,5138235907 = R * 0,9037, alpha1 = 1;
Reference: Koda
Remark: 1 kein Tippfehler!

844 cphi (Cu<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,02269861503 = R * -0,00273;
Reference: Koda
Remark:

845 beta0 (Cu<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,84191362183 = R * 0,22153; Reference:
Koda
Remark:

846 beta1 (Cu<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 20,64725887608 = R * 2,48328, alpha1 =
1,4; Reference: Koda
Remark:

847 beta2 (Cu<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -325,3651073542 = R * -39,1322, alpha2
= 12; Reference: Koda
Remark:

848 cphi (Cu<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,03167828691 = R * 0,00381; Reference:
Koda
Remark:

849 theta (Cu<2+> Na<+>)
Value in this parameterfile (chemapp) = -0,8505744753 = R * -0,1023; Reference:
Koda
Remark:

850 theta (Cu<2+> K<+>)
Value in this parameterfile (chemapp) = -1,3660741573 = R * -0,1643; Reference:
Koda
Remark:

851 theta (Cu<2+> Mg<2+>)
Value in this parameterfile (chemapp) = 0,9137647589 = R * 0,1099; Reference:
Koda
Remark:

852 theta (Cu<2+> Ca<2+>)
Value in this parameterfile (chemapp) = -1,0667517613 = R * -0,1283; Reference:
Koda
Remark:

853 psi (Cu<2+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,025666895457 = R * -0,00309;
Reference: Koda
Remark:

854 psi (Cu<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,011947952307 = R * -0,00144;
Reference: Koda
Remark:

855 psi (Cu<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,51816032552 = R * -0,06232;
Reference: Koda
Remark:

856 psi (Cu<2+> Ca<2+> Cl<->)

Value in this parameterfile (chemapp) = -0,18774165838 = R * -0,02258;
Reference: Koda
Remark:

857 psi (Cu<2+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,46602834155 = R * 0,05605; Reference:
Koda
Remark:

858 psi (Cu<2+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,15331958284 = R * 0,01844; Reference:
Koda
Remark:

859 psi (Cu<2+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,52206814569 = R * -0,06279;
Reference: Koda
Remark:

860 psi (Cu<2+> Ca<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,7333398702 = R * 0,0882; Reference:
Koda
Remark:

861 psi (Cu<2+> Cl<-> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,15722740301 = R * 0,01891; Reference:
Koda
Remark:

862 beta0 (Co<2+> Cl<->)
Value in this parameterfile (chemapp) = 3,15884901912 = R * 0,37992; Reference:
Koda
Remark:

863 beta1 (Co<2+> Cl<->)
Value in this parameterfile (chemapp) = 10,55518856939 = R * 1,26949, alpha1 =
2; Reference: Koda
Remark:

864 cphi (Co<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,05612294925 = R * -0,00675;
Reference: Koda
Remark:

865 beta0 (Co<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,58025596066 = R * 0,19006; Reference:
Koda
Remark:

866 beta1 (Co<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 23,5076169503 = R * 2,8273, alpha1 =
1,4; Reference: Koda
Remark:

867 beta2 (Co<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -481,4243215687 = R * -57,9017, alpha2 =
12; Reference: Koda
Remark:

868 cphi (Co<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,05786899656 = R * 0,00696; Reference:
Koda
Remark:

869 theta (Co<2+> Na<+>)

- Value in this parameterfile (chemapp) = 0,63747355837 = R * 0,07667; Reference:
Koda
Remark:
- 870 theta (Co<2+> K<+>)
Value in this parameterfile (chemapp) = 0,36642049977 = R * 0,04407; Reference:
Koda
Remark:
- 871 theta (Co<2+> Mg<2+>)
Value in this parameterfile (chemapp) = 4,1165143961 = R * 0,4951; Reference:
Koda
Remark:
- 872 theta (Co<2+> Ca<2+>)
Value in this parameterfile (chemapp) = -0,43792529437 = R * -0,05267;
Reference: Koda
Remark:
- 873 psi (Co<2+> Na<+> Cl<->)
Value in this parameterfile (chemapp) = -0,10326622662 = R * -0,01242;
Reference: Koda
Remark:
- 874 psi (Co<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = -0,24103767389 = R * -0,02899;
Reference: Koda
Remark:
- 875 psi (Co<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,75262953572 = R * -0,09052;
Reference: Koda
Remark:
- 876 psi (Co<2+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = 0,0057877311071 = R * 0,0007;
Reference: Koda
Remark:
- 877 psi (Co<2+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,17161150704 = R * -0,02064;
Reference: Koda
Remark:
- 878 psi (Co<2+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,35103865442 = R * -0,04222;
Reference: Koda
Remark:
- 879 psi (Co<2+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -2,1276833649 = R * -0,2559; Reference:
Koda
Remark:
- 880 psi (Co<2+> Ca<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,09362139386 = R * -0,01126;
Reference: Koda
Remark:
- 881 psi (Co<2+> Cl<-> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,133032176 = R * 0,016; Reference:
Koda
Remark:
- 882 beta0 (Na<+> CrO4<2->)

Value in this parameterfile (chemapp) = 0,507318203176 = R * 0,06102;
Reference: Koda
Remark:

883 beta1 (Na<+> CrO4<2->)
Value in this parameterfile (chemapp) = 13,7962681023 = R * 1,6593, alpha1 = 2;
Reference: Koda
Remark:

884 cphi (Na<+> CrO4<2->)
Value in this parameterfile (chemapp) = 0,0289029031382 = R * 0,00348;
Reference: Koda
Remark:

885 beta0 (K<+> CrO4<2->)
Value in this parameterfile (chemapp) = 0,604065853172 = R * 0,07265;
Reference: Koda
Remark:

886 beta1 (K<+> CrO4<2->)
Value in this parameterfile (chemapp) = 10,592687014 = R * 1,274, alpha1 = 2;
Reference: Koda
Remark:

887 beta0 (Mg<2+> CrO4<2->)
Value in this parameterfile (chemapp) = 0,87177647835 = R * 0,10485; Reference:
Koda
Remark:

888 beta1 (Mg<2+> CrO4<2->)
Value in this parameterfile (chemapp) = 76,5691632501 = R * 9,2091, alpha1 =
1,4; Reference: Koda
Remark:

889 cphi (Mg<2+> CrO4<2->)
Value in this parameterfile (chemapp) = 0,122730496871 = R * 0,01476;
Reference: Koda
Remark:

890 theta (CrO4<2-> Cl<->)
Value in this parameterfile (chemapp) = 0,669368022566 = R * 0,08051;
Reference: Koda
Remark:

891 psi (Na<+> CrO4<2-> Cl<->)
Value in this parameterfile (chemapp) = -0,0700281374464 = R * -0,00842;
Reference: Koda
Remark:

892 psi (K<+> CrO4<2-> Cl<->)
Value in this parameterfile (chemapp) = 0,195083371593 = R * 0,02346;
Reference: Koda
Remark:

893 psi (Na<+> Ca<2+> CrO4<2->)
Value in this parameterfile (chemapp) = -12,7702574449 = R * -1,5359;
Reference: Koda
Remark:

894 psi (K<+> Mg<2+> CrO4<2->)
Value in this parameterfile (chemapp) = -0,0157343806164 = R * -0,00189;
Reference: Koda
Remark:

895 psi (Mg<2+> CrO4<2-> (SO4)<2->)

- Value in this parameterfile (chemapp) = -1,06625289064 = R * -0,12824;
Reference: Koda
Remark:
- 896 psi (Na<+> Mg<2+> CrO4<2->)
Value in this parameterfile (chemapp) = -0,197378176629 = R * -0,02374;
Reference: Koda
Remark:
- 897 beta0 (Mn<2+> Cl<->)
Value in this parameterfile (chemapp) = 2,58772525853 = R * 0,31123; Reference:
Koda
Remark:
- 898 beta1 (Mn<2+> Cl<->)
Value in this parameterfile (chemapp) = 16,6814034193 = R * 2,0063, alpha1 = 2;
Reference: Koda
Remark: a-Wert geraten von MOO
- 899 cphi (Mn<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,05288028996 = R * -0,00636;
Reference: Koda
Remark:
- 900 beta0 (Mn<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 1,81439259042 = R * 0,21822; Reference:
Koda
Remark:
- 901 beta1 (Mn<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 23,06104456449 = R * 2,77359, alpha1 =
1,4; Reference: Koda
Remark: a-Wert geraten von MOO
- 902 beta2 (Mn<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -489,1368619723 = R * -58,8293, alpha2
= 12; Reference: Koda
Remark: a-Wert geraten von MOO
- 903 cphi (Mn<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,03009852982 = R * 0,00362; Reference:
Koda
Remark:
- 904 theta (Mn<2+> Na<+>)
Value in this parameterfile (chemapp) = 0,35128808975 = R * 0,04225; Reference:
Koda
Remark:
- 905 theta (Mn<2+> K<+>)
Value in this parameterfile (chemapp) = -2,0736390434 = R * -0,2494; Reference:
Koda
Remark:
- 906 theta (Mn<2+> Mg<2+>)
Value in this parameterfile (chemapp) = 2,6257225738 = R * 0,3158; Reference:
Koda
Remark:
- 907 theta (Mn<2+> Ca<2+>)
Value in this parameterfile (chemapp) = -0,5288028996 = R * -0,0636; Reference:
Koda
Remark:
- 908 psi (Mn<2+> Na<+> Cl<->)

Value in this parameterfile (chemapp) = -0,10900323921 = R * -0,01311;
Reference: Koda
Remark:

909 psi (Mn<2+> K<+> Cl<->)
Value in this parameterfile (chemapp) = 0,066707321753 = R * 0,00802;
Reference: Koda
Remark:

910 psi (Mn<2+> Mg<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,51566597222 = R * -0,06202;
Reference: Koda
Remark:

911 psi (Mn<2+> Ca<2+> Cl<->)
Value in this parameterfile (chemapp) = -0,040333692861 = R * -0,00485;
Reference: Koda
Remark:

912 psi (Mn<2+> Na<+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,13278274067 = R * -0,01597;
Reference: Koda
Remark:

913 psi (Mn<2+> K<+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,60180430618 = R * 0,07238; Reference:
Koda
Remark:

914 psi (Mn<2+> Mg<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = -0,8788438127 = R * -0,1057; Reference:
Koda
Remark:

915 psi (Mn<2+> Ca<2+> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,13461193309 = R * 0,01619; Reference:
Koda
Remark:

916 psi (Mn<2+> Cl<-> (SO4)<2->)
Value in this parameterfile (chemapp) = 0,15539821059 = R * 0,01869; Reference:
Koda
Remark:

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/YPF/ Parameterfile for EQ3/6 called "data0,y pf"

Anlage 2: Output-Datei der geochemischen Modellierung mit OPW

This is NATHAN 2.7.1

This program contains ChemApp
Copyright GTT-Technologies, Kaiserstrasse 100,
D-52134 Herzogenrath, Germany
<http://www.gtt-technologies.de>

No warranty is given for the correctness of
calculations done with NATHAN. Propositions
and error messages are welcome.

Helge C. Moog
Gesellschaft fuer Anlagen- und Reaktorsicherheit (GRS)
Theodor-Heuss-Str. 4, 38122 Braunschweig,
Germany
<http://www.grs.de/geochemie>
helge.moog@grs.de

Used input file:
Opa+3BSK3+3SF.ni

Version of prepare.f: 236
Active Chemapp-Version: 554
Number of phases = 406
Number of system components = 48
Name of paramaterfile used:
tdb-hmw-118#1-1-1.dat

> 3*(BSK3+3SF) mit Opalinustonloesung

> Es sollen ca. 1.5 m3 Lösung übrig bleiben

> p = 40 bar, Gasphase aktiv aber unbestimmtes Volumen

Step 1:
Charge balance before calculation = 0.000000D+00
Relative error of charge balance prior of calculation = 0.000000D+00

Active CHEMAPP-stream:

SYSTEM UNITS:
Pressure : Pa

Braunschweig

Volume : m3
Temperature : K
Energy : J
Amount : mol

TARGET LIMITS:

Pressure/bar : 2.00000E+01 6.00000E+01
Volume/dm3 : 1.00000E-07 1.00000E+50
Temperature/K: 298.15 6000.00

T = 298.15 K

P = 4.00000E+06 Pa

STREAM CONSTITUENT

OH<->/AQUEOUS/

AMOUNT/mol = 2.59438350000000E+00

Na<+>/AQUEOUS/

AMOUNT/mol = 2.59438350000000E+00

BSK3+3SF

AMOUNT/mol = 3.00000000000000E+00

H

AMOUNT/mol = 7.28928089466300E+05

O

AMOUNT/mol = 3.64775198961700E+05

NA

AMOUNT/mol = 8.43292552100000E+02

K

AMOUNT/mol = 5.28434525000000E+00

MG

AMOUNT/mol = 3.24120756000000E+01

CA

AMOUNT/mol = 4.55539549000000E+01

CL

AMOUNT/mol = 8.54050679100000E+02

S

AMOUNT/mol = 7.53460921000000E+01

C

AMOUNT/mol = 3.90794535000000E+00

SR

AMOUNT/mol = 2.07092515000000E+00

ELIMINATED PHASES:

K2HAsO4: 3H2O

K3AsO4: 7H2O

KH2AsO4

Na2HAsO4: 7H2O

Na3AsO4: 12H2O

NaH2AsO4: H2O

T = 298.15 K
 P = 4.00000E+06 Pa
 V = 1.51472E+02 m3

| STREAM CONSTITUENTS | AMOUNT/mol |
|---------------------|------------|
| OH<->/AQUEOUS/ | 2.5944E+00 |
| Na<+>/AQUEOUS/ | 2.5944E+00 |
| BSK3+3SF | 3.0000E+00 |
| H | 7.2893E+05 |
| O | 3.6478E+05 |
| NA | 8.4329E+02 |
| K | 5.2843E+00 |
| MG | 3.2412E+01 |
| CA | 4.5554E+01 |
| CL | 8.5405E+02 |
| S | 7.5346E+01 |
| C | 3.9079E+00 |
| SR | 2.0709E+00 |

| | EQUIL AMOUNT | MOLE FRACTION | FUGACITY |
|----------------|--------------|---------------|-------------|
| PHASE: GAS | mol | | Pa |
| H2 | 2.3834E+05 | 9.9921E-01 | 3.9968E+06 |
| H2O(g) | 1.8931E+02 | 7.9363E-04 | 3.1745E+03 |
| CO2(g) | 9.4309E-02 | 3.9537E-07 | 1.5815E+00 |
| H2S(g) | 5.4748E-08 | 2.2952E-13 | 9.1808E-07 |
| O2(g) | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| TOTAL: | 2.3853E+05 | 1.0000E+00 | 1.0000E+00 |
| PHASE: AQUEOUS | mol | MOLALITY | ACTIVITY |
| H2O | 8.2484E+04 | 5.5508E+01 | 9.7414E-01 |
| H<+> | 9.0851E-05 | 6.1139E-08 | 4.4975E-08 |
| OH<-> | 5.8668E-04 | 3.9482E-07 | 2.2454E-07 |
| O2(aq) | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (CO2)<0> | 6.9683E-04 | 4.6894E-07 | 5.1406E-07 |
| (CO3)<2-> | 8.2280E-05 | 5.5372E-08 | 5.3966E-09 |
| (HCO3)<-> | 1.3142E-02 | 8.8441E-06 | 5.0694E-06 |
| (HSO4)<-> | 3.6046E-47 | 2.4258E-50 | 1.5493E-50 |
| (SO4)<2-> | 5.1332E-41 | 3.4545E-44 | 3.5390E-45 |
| Ca(CO3)<0> | 8.5964E-06 | 5.7851E-09 | 5.7851E-09 |
| Ca<2+> | 6.1909E+00 | 4.1662E-03 | 7.8341E-04 |
| Cl<-> | 8.5405E+02 | 5.7474E-01 | 3.8455E-01 |
| K<+> | 5.2843E+00 | 3.5562E-03 | 2.0748E-03 |
| Mg(CO3)<0> | 2.8957E-05 | 1.9487E-08 | 1.9487E-08 |
| Mg(OH)<+> | 2.4649E-04 | 1.6588E-07 | 1.4818E-07 |
| Mg<2+> | 3.2412E+01 | 2.1812E-02 | 4.4324E-03 |
| Na<+> | 6.5199E+02 | 4.3877E-01 | 2.7762E-01 |
| H2S<0> | 1.3172E-09 | 8.8643E-13 | 8.8643E-13 |
| HS<-> | 5.8105E-09 | 3.9103E-12 | 2.0235E-12 |
| H2SiO4<2-> | 2.7283E-14 | 1.8360E-17 | 1.7347E-18 |
| H3SiO4<-> | 2.2733E-09 | 1.5299E-12 | 9.7858E-13 |

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|---|------------|------------|-------------|
| SiO ₂ <0> | 4.2378E-07 | 2.8519E-10 | 2.8519E-10 |
| Al(OH) ₄ <-> | 7.0998E-06 | 4.7779E-09 | 2.5949E-09 |
| Al<3+> | 9.3246E-12 | 6.2751E-15 | 1.5095E-16 |
| Fe<2+> | 5.5272E+01 | 3.7196E-02 | 6.4619E-03 |
| Fe<3+> | 9.6498E-36 | 6.4940E-39 | 1.3324E-39 |
| (AsO ₄)<3-> | 6.3678E-17 | 4.2853E-20 | 1.8321E-22 |
| (H ₂ AsO ₄)<-> | 3.0798E-16 | 2.0726E-19 | 1.8333E-19 |
| (HAsO ₄)<2-> | 3.1333E-15 | 2.1086E-18 | 2.2769E-19 |
| H ₂ AsO ₃ <-> | 4.5831E-01 | 3.0843E-04 | 1.5961E-04 |
| HAsO ₂ <0> | 1.5725E+01 | 1.0582E-02 | 1.0582E-02 |
| B(OH) ₃ <0> | 6.1072E+02 | 4.1099E-01 | 4.1099E-01 |
| Cr<3+> | 1.0828E-08 | 7.2870E-12 | 7.4630E-15 |
| CrO ₄ <2-> | 7.5213E-53 | 5.0616E-56 | 7.3779E-57 |
| Cu<+> | 8.7266E-10 | 5.8727E-13 | 3.0188E-13 |
| Cu<2+> | 5.0152E-20 | 3.3750E-23 | 4.3054E-24 |
| Cu(CO ₃) ₂ <2-> | 1.4782E-46 | 9.9475E-50 | 6.2041E-51 |
| Cu(HCO ₃)<-> | 1.8599E-11 | 1.2517E-14 | 6.4772E-15 |
| Cu(OH)<+> | 3.1118E-21 | 2.0941E-24 | 1.0765E-24 |
| Cu(OH) ₂ (aq) | 2.0045E-22 | 1.3490E-25 | 1.3490E-25 |
| Cu(OH) ₃ <-> | 3.4363E-25 | 2.3125E-28 | 1.1967E-28 |
| Cu(OH) ₄ <2-> | 4.6028E-30 | 3.0975E-33 | 1.9319E-34 |
| Cu ₂ (OH)<3+> | 2.3859E-40 | 1.6056E-43 | 1.6444E-46 |
| Cu ₂ (OH) ₂ <2+> | 7.8932E-39 | 5.3118E-42 | 3.4199E-43 |
| Cu ₃ (OH) ₄ <2+> | 3.6079E-58 | 2.4280E-61 | 1.5632E-62 |
| CuCO ₃ <0> | 1.9291E-22 | 1.2982E-25 | 1.2982E-25 |
| Mn<2+> | 2.3707E-08 | 1.5954E-11 | 3.4376E-12 |
| MoO ₄ <2-> | 1.3036E-01 | 8.7730E-05 | 1.4399E-05 |
| (NH ₄)<+> | 8.6564E+01 | 5.8255E-02 | 2.9946E-02 |
| NO ₃ <-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| Ni<2+> | 4.1372E-05 | 2.7842E-08 | 5.2877E-09 |
| HNi(P ₂ O ₇)<-> | 2.1451E-07 | 1.4436E-10 | 7.4703E-11 |
| Ni(HPO ₄)<0> | 1.3102E-05 | 8.8169E-09 | 8.8169E-09 |
| Ni(P ₂ O ₇)<2-> | 6.1307E-05 | 4.1257E-08 | 2.5732E-09 |
| (H ₂ P ₂ O ₇)<2-> | 4.8997E-04 | 3.2973E-07 | 2.0565E-08 |
| (H ₂ PO ₄)<-> | 3.1277E+00 | 2.1048E-03 | 1.0892E-03 |
| (H ₃ P ₂ O ₇)<-> | 4.7227E-10 | 3.1782E-13 | 1.6447E-13 |
| (HP ₂ O ₇)<3-> | 1.6688E-01 | 1.1231E-04 | 1.0235E-07 |
| (HPO ₄)<2-> | 3.5407E+01 | 2.3827E-02 | 1.4861E-03 |
| (P ₂ O ₇)<4-> | 1.6096E+00 | 1.0832E-03 | 9.0615E-10 |
| (PO ₄)<3-> | 2.4070E-02 | 1.6198E-05 | 1.4761E-08 |
| H ₃ PO ₄ <0> | 1.0048E-05 | 6.7618E-09 | 6.7618E-09 |
| H ₄ P ₂ O ₇ <0> | 1.0992E-16 | 7.3970E-20 | 7.3970E-20 |
| Sn(OH) ₆ <2-> | 2.7345E-11 | 1.8402E-14 | 1.1477E-15 |
| Sn<2+> | 1.2555E-12 | 8.4489E-16 | 5.4396E-17 |
| Sn<4+> | 2.1513E-29 | 1.4478E-32 | 1.5526E-38 |
| SnO ₃ <2-> | 1.6747E-13 | 1.1270E-16 | 7.0291E-18 |
| Sr<2+> | 2.0709E+00 | 1.3937E-03 | 2.5874E-04 |
| SrSO ₄ <0> | 2.7130E-43 | 1.8257E-46 | 1.8257E-46 |
| Ti(OH) ₄ <0> | 3.3787E-07 | 2.2737E-10 | 2.2737E-10 |

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| (UO2)(H2PO4)(H3PO4)<+> | 1.8591E-28 | 1.2511E-31 | 6.4313E-32 |
| (UO2)(H2PO4)<+> | 8.1140E-21 | 5.4604E-24 | 2.8069E-24 |
| (UO2)(H2PO4)2<0> | 2.0765E-22 | 1.3974E-25 | 1.3974E-25 |
| (UO2)(H3PO4)<2+> | 1.2719E-27 | 8.5594E-31 | 5.5107E-32 |
| (UO2)(HASO4)<0> | 9.9790E-32 | 6.7155E-35 | 6.7155E-35 |
| (UO2)(HPO4)<0> | 5.4350E-17 | 3.6576E-20 | 3.6576E-20 |
| U +III <3+> | 1.1257E-23 | 7.5752E-27 | 7.7582E-30 |
| U +IV (CO3)4<4-> | 1.7739E-19 | 1.1938E-22 | 1.3557E-26 |
| U +IV (CO3)5<6-> | 1.0400E-20 | 6.9988E-24 | 5.5145E-36 |
| U +IV (OH)<3+> | 5.4171E-17 | 3.6455E-20 | 7.9950E-22 |
| U +IV (OH)2(CO3)2<2-> | 5.6770E-12 | 3.8204E-15 | 4.4268E-16 |
| U +IV (OH)2<2+> | 3.9987E-11 | 2.6910E-14 | 4.9069E-15 |
| U +IV (OH)3<+> | 6.1903E-08 | 4.1658E-11 | 2.7467E-11 |
| U +IV (OH)4<0> | 4.6990E-06 | 3.1623E-09 | 3.1623E-09 |
| U +IV (SO4)<2+> | 3.9503E-62 | 2.6584E-65 | 1.7115E-66 |
| U +IV (SO4)2<0> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| U +IV <4+> | 1.7499E-22 | 1.1776E-25 | 1.2440E-28 |
| U +IV Cl<3+> | 3.7462E-21 | 2.5211E-24 | 2.5820E-27 |
| (U +V O2)(CO3)3<5-> | 4.9293E-19 | 3.3172E-22 | 8.3351E-33 |
| (U +V O2)<+> | 1.7535E-11 | 1.1800E-14 | 6.0660E-15 |
| (U +VI O2)(CO3)<0> | 9.8286E-20 | 6.6143E-23 | 6.6143E-23 |
| (U +VI O2)(CO3)2<2-> | 2.0372E-20 | 1.3710E-23 | 1.6589E-24 |
| (U +VI O2)(CO3)3<4-> | 3.3030E-20 | 2.2228E-23 | 1.5106E-27 |
| (U +VI O2)(H2AsO4)<+> | 1.3837E-35 | 9.3121E-39 | 4.7869E-39 |
| (U +VI O2)(H2AsO4)2<0> | 9.7941E-53 | 6.5911E-56 | 6.5911E-56 |
| (U +VI O2)(OH)<+> | 3.8267E-19 | 2.5753E-22 | 1.7748E-22 |
| (U +VI O2)(OH)2<0> | 7.3985E-19 | 4.9789E-22 | 4.9789E-22 |
| (U +VI O2)(OH)3<-> | 2.7559E-19 | 1.8546E-22 | 8.8131E-23 |
| (U +VI O2)(OH)4<2-> | 5.1696E-23 | 3.4790E-26 | 4.1988E-27 |
| (U +VI O2)(PO4)<-> | 1.0195E-15 | 6.8608E-19 | 3.5504E-19 |
| (U +VI O2)(SO4)<0> | 1.0758E-62 | 7.2395E-66 | 7.2395E-66 |
| (U +VI O2)(SO4)2<2-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)(SO4)3<4-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)<2+> | 1.0210E-20 | 6.8708E-24 | 1.4163E-24 |
| (U +VI O2)11(CO3)6(OH)12 | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)2(CO3)(OH)3<-> | 4.8805E-32 | 3.2844E-35 | 1.6997E-35 |
| (U +VI O2)2(OH)<3+> | 1.2940E-37 | 8.7080E-41 | 8.9184E-44 |
| (U +VI O2)2(OH)2<2+> | 1.6303E-35 | 1.0972E-38 | 2.3894E-39 |
| (U +VI O2)3(CO3)6<6-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)3(OH)4<2+> | 9.4411E-51 | 6.3535E-54 | 8.8189E-55 |
| (U +VI O2)3(OH)5<+> | 7.9006E-48 | 5.3168E-51 | 4.3996E-51 |
| (U +VI O2)3(OH)7<-> | 1.5295E-49 | 1.0293E-52 | 4.8910E-53 |
| (U +VI O2)30(OH)2(HCO3)< | 2.2517E-54 | 1.5153E-57 | 7.7895E-58 |
| (U +VI O2)4(OH)7<+> | 2.7183E-63 | 1.8293E-66 | 1.3821E-66 |
| (U +VI O2)Cl<+> | 2.3948E-21 | 1.6116E-24 | 8.2847E-25 |
| (U +VI O2)Cl2<0> | 2.6146E-23 | 1.7596E-26 | 1.7596E-26 |
| (U +VI O2)SiO(OH)3<+> | 3.7692E-25 | 2.5365E-28 | 1.3039E-28 |
| V<2+> | 1.2193E-03 | 8.2051E-07 | 5.2826E-08 |
| V<3+> | 3.7195E-04 | 2.5031E-07 | 2.5635E-10 |

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| VO<2+> | 8.9986E-07 | 6.0557E-10 | 3.8988E-11 |
| VO2<+> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| Zr(OH)<3+> | 2.7517E-23 | 1.8518E-26 | 1.8965E-29 |
| Zr(OH)2<2+> | 4.4585E-17 | 3.0004E-20 | 1.9317E-21 |
| Zr(OH)4<0> | 9.6371E-07 | 6.4854E-10 | 6.4854E-10 |
| Zr<4+> | 5.6441E-28 | 3.7983E-31 | 4.0734E-37 |
| TOTAL: | 8.4845E+04 | | 1.0000E+00 |
| PHASE: BASRSO4-IDMIX | mol | MOLE FRACTION | ACTIVITY |
| SrSO4 | 0.0000E+00 | 1.0000E+00 | 4.0127E-42 |
| TOTAL: | 0.0000E+00 | 1.0000E+00 | 4.0127E-42 |
| | mol | | ACTIVITY |
| Fe3O4_Magnetite | 4.7818E+04 | | 1.0000E+00 |
| U +IV (OH)4(am) | 1.9453E+04 | | 1.0000E+00 |
| ZrO2(monoclinic) | 1.5274E+04 | | 1.0000E+00 |
| Mn(OH)2 | 2.3728E+03 | | 1.0000E+00 |
| Eskolaite_Cr2O3 | 2.0938E+03 | | 1.0000E+00 |
| Ni(cr) | 1.9371E+03 | | 1.0000E+00 |
| U +IV (SiO4)(c)_Coffinit | 1.0690E+03 | | 1.0000E+00 |
| FeCO3_Siderite | 1.0251E+03 | | 1.0000E+00 |
| Rutile-TiO2 | 1.8953E+02 | | 1.0000E+00 |
| Cassiterite_SnO2 | 1.8768E+02 | | 1.0000E+00 |
| FeAl2O4_Hercynite | 1.2706E+02 | | 1.0000E+00 |
| Na2[B4O5(OH)4]:8H2O_Bora | 9.6947E+01 | | 1.0000E+00 |
| Cu(OH)2(s) | 8.3477E+01 | | 1.0000E+00 |
| Ni3S2(cr) | 4.8564E+01 | | 1.0000E+00 |
| CaMoO4_Powellite | 3.9363E+01 | | 1.0000E+00 |
| V2O4 | 1.4875E+01 | | 1.0000E+00 |
| SnO2(cr)_Cassiterite | 0.0000E+00 | | 1.0000E+00 |
| Gibbsite_Al(OH)3 | 0.0000E+00 | | 1.8040E-01 |
| CuO(s) | 0.0000E+00 | | 9.3118E-02 |
| Anatase_TiO2 | 0.0000E+00 | | 8.1922E-02 |
| Fe(OH)2(cr) | 0.0000E+00 | | 7.5090E-02 |
| MnHPO4 | 0.0000E+00 | | 4.5217E-02 |
| Fe(OH)2_precipitated | 0.0000E+00 | | 1.6516E-02 |
| FeS | 0.0000E+00 | | 4.1394E-03 |
| AlOOH_Boehmite | 0.0000E+00 | | 4.0468E-03 |
| NaCl_Halite | 0.0000E+00 | | 2.8242E-03 |
| SrCO3(s) | 0.0000E+00 | | 2.6516E-03 |
| Mg(CO3)_Magnesite | 0.0000E+00 | | 1.4997E-03 |
| Ca(CO3)_Calcite | 0.0000E+00 | | 9.8191E-04 |
| Ca(CO3)_Aragonite | 0.0000E+00 | | 6.4147E-04 |
| Dawsonite_NaAlCO3(OH)2 | 0.0000E+00 | | 5.3003E-04 |
| Dolomite-ord_CaMg(CO3)2 | 0.0000E+00 | | 1.3720E-04 |
| KCl_Sylvite | 0.0000E+00 | | 9.8796E-05 |
| Fe2(SiO4)_Fayalite | 0.0000E+00 | | 9.7681E-05 |
| Ni(OH)2(cr) | 0.0000E+00 | | 8.3034E-05 |
| Chamosite-7A_Fe2Al2SiO5(| 0.0000E+00 | | 1.6620E-05 |
| Mg(OH)2_Brucite | 0.0000E+00 | | 1.5925E-05 |
| Fe3Si2O5(OH)4_Greenalite | 0.0000E+00 | | 1.5395E-05 |

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| FeSiO3(cr) | 0.0000E+00 | 1.1688E-05 |
| SnS_Herzenbergite | 0.0000E+00 | 1.0264E-05 |
| CaMg(CO3)2_Dolomite | 0.0000E+00 | 1.0254E-05 |
| MgMoO4:5H2O(s) | 0.0000E+00 | 4.4165E-06 |
| Dolomite-dis_CaMg(CO3)2 | 0.0000E+00 | 3.9170E-06 |
| Na(HCO3)_Nahcolite | 0.0000E+00 | 3.4742E-06 |
| Mg(CO3):3H2O_Nesquehonit | 0.0000E+00 | 3.0210E-06 |
| Ni(CO3)(cr) | 0.0000E+00 | 2.8665E-06 |
| Tridymite_SiO2 | 0.0000E+00 | 1.9184E-06 |
| Chalcedony-SiO2 | 0.0000E+00 | 1.5249E-06 |
| FeCl2:4H2O | 0.0000E+00 | 9.8581E-07 |
| NiO(cr) | 0.0000E+00 | 8.6174E-07 |
| Coesite_SiO2 | 0.0000E+00 | 4.4100E-07 |
| Fe(OH)3(cr) | 0.0000E+00 | 1.6288E-07 |
| SiO2(am)-GWB | 0.0000E+00 | 1.4748E-07 |
| Al2O3_Corundum | 0.0000E+00 | 1.0764E-07 |
| MgCl2:6H2O_Bischoffite | 0.0000E+00 | 1.9411E-08 |
| CaCl2:6H2O_Antarcticite | 0.0000E+00 | 7.0894E-09 |
| K(HCO3)_Kalicinite | 0.0000E+00 | 5.3930E-09 |
| Na2(CO3):10H2O_Natron | 0.0000E+00 | 2.0662E-09 |
| Daphnite-7A_Fe5AlAlSi3O1 | 0.0000E+00 | 1.5001E-09 |
| Ni(CO3):5.5H2O(cr) | 0.0000E+00 | 9.8329E-10 |
| Na2(CO3):7H2O | 0.0000E+00 | 9.4745E-10 |
| CaCl2:4H2O | 0.0000E+00 | 2.3057E-10 |
| Rhodochrosite_MnCO3 | 0.0000E+00 | 2.2798E-10 |
| NiCO3 | 0.0000E+00 | 1.9059E-10 |
| Na2(CO3):H2O_Thermonatri | 0.0000E+00 | 1.2481E-10 |
| KMgCl3:6H2O_Carnallite | 0.0000E+00 | 2.0135E-11 |
| FeS2 | 0.0000E+00 | 1.9940E-11 |
| Ni2SiO4 | 0.0000E+00 | 8.9149E-12 |
| Ca(OH)2_Portlandite | 0.0000E+00 | 5.6438E-12 |
| CSH(0.8)_Ca0.8SiO2.8_H2O | 0.0000E+00 | 4.3397E-12 |
| KNa(CO3):6H2O | 0.0000E+00 | 4.1393E-12 |
| CaNa2(CO3)2:5H2O_Gayluss | 0.0000E+00 | 3.5500E-12 |
| Na2Ca(CO3)2:2H2O_Pirsson | 0.0000E+00 | 2.5168E-12 |
| Mg2Al2SiO5(OH)4_Amesit-7 | 0.0000E+00 | 2.1908E-12 |
| NiCl2:6H2O(cr) | 0.0000E+00 | 6.8476E-13 |
| Kaolinite_Al2Si2O5(OH)4 | 0.0000E+00 | 2.1531E-13 |
| Na.96Al.96Si2.04O6:H2O_A | 0.0000E+00 | 1.0347E-13 |
| NiCl2:4H2O(cr) | 0.0000E+00 | 8.5389E-14 |
| Ripidolite-14A_Mg3Fe2Al2 | 0.0000E+00 | 3.0378E-14 |
| CSH(1.1)_Ca(1.1)SiO(3.1) | 0.0000E+00 | 2.9676E-14 |
| Mg4Al2O7:10H2O_Hydrotalc | 0.0000E+00 | 2.0901E-14 |
| U +VI O3:2H2O(c)_Schoepi | 0.0000E+00 | 1.0894E-14 |
| NaOH | 0.0000E+00 | 7.6846E-15 |
| NiCl2:2H2O(cr) | 0.0000E+00 | 6.7459E-15 |
| Na3(CO3)(HCO3):2H2O_Tron | 0.0000E+00 | 5.6595E-15 |
| MnCl2:4H2O | 0.0000E+00 | 6.9404E-16 |
| NiS2(cr) | 0.0000E+00 | 4.9978E-16 |

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| Delafossite_CuFeO2 | 0.0000E+00 | 2.5806E-16 |
| Minnesotaites_Fe3Si4O10(O | 0.0000E+00 | 1.7236E-16 |
| Ni2(SiO4)(oliv) | 0.0000E+00 | 4.2137E-17 |
| Mg4Al4Si2O10(OH)8_Amesit | 0.0000E+00 | 2.6812E-17 |
| K2(CO3):1.5H2O | 0.0000E+00 | 1.9459E-17 |
| Ripidolite-7A_Mg3Fe2Al2S | 0.0000E+00 | 1.2745E-17 |
| Ni9S8(cr) | 0.0000E+00 | 7.1021E-18 |
| Mesolite_Na.676Ca.657Al1 | 0.0000E+00 | 5.0439E-18 |
| (U +VI O2)(CO3)(c)_Ruthe | 0.0000E+00 | 4.3700E-18 |
| Goethite_FeOOH | 0.0000E+00 | 4.2968E-18 |
| Maximum-Microcline_KAlSi | 0.0000E+00 | 2.3629E-18 |
| NiCl2(cr) | 0.0000E+00 | 1.7859E-18 |
| Ca2Cl2(OH)2:H2O | 0.0000E+00 | 1.3282E-19 |
| MnCl2:KCl:2H2O | 0.0000E+00 | 5.4041E-20 |
| Na.96Al.96Si2.04O6_Analc | 0.0000E+00 | 3.8918E-20 |
| Ferroaluminoceladonite_K | 0.0000E+00 | 3.4628E-20 |
| Scolecite_CaAl2Si3O10:3H | 0.0000E+00 | 1.2899E-20 |
| alpha-NiS | 0.0000E+00 | 9.3808E-21 |
| beta-NiS | 0.0000E+00 | 2.5249E-21 |
| K2Na(HCO3)(CO3):2H2O_Tro | 0.0000E+00 | 2.0203E-21 |
| Saponite-Na_Na.33Mg3Al.3 | 0.0000E+00 | 1.2427E-21 |
| Saponite-Mg_Mg3.165Al.33 | 0.0000E+00 | 9.6252E-22 |
| Saponite-Ca_Ca.165Mg3Al. | 0.0000E+00 | 6.6118E-22 |
| Saponite-K_K.33Mg3Al.33S | 0.0000E+00 | 5.3887E-22 |
| CSH(1.8)_Ca(1.8)SiO(3.8) | 0.0000E+00 | 4.9891E-22 |
| Illite_K0.6Mg0.25Al1.8Al | 0.0000E+00 | 2.0962E-22 |
| Saponite-H_H.33Mg3Al.33S | 0.0000E+00 | 7.3928E-23 |
| KMgAlSi4O10(OH)2_Celadon | 0.0000E+00 | 1.2472E-23 |
| Beidellite-Na_Na.33Al2.3 | 0.0000E+00 | 4.2408E-24 |
| Talc_Mg3Si4O10(OH)2 | 0.0000E+00 | 3.5044E-24 |
| Beidellite-Mg_Mg.165Al2. | 0.0000E+00 | 3.2811E-24 |
| CaCl2:MnCl2:8H2O | 0.0000E+00 | 2.5953E-24 |
| C2ASH8_Ca2Al2SiO7:8H2O_G | 0.0000E+00 | 2.4491E-24 |
| Beidellite-Ca_Ca.165Al2. | 0.0000E+00 | 2.2564E-24 |
| Na4(U +VI O2)(CO3)3(c) | 0.0000E+00 | 1.9484E-24 |
| Beidellite-K_K.33Al2.33S | 0.0000E+00 | 1.8390E-24 |
| MgCl2:MnCl2:8H2O | 0.0000E+00 | 1.7156E-24 |
| C2AH8_Ca2Al2O5:8H2O_Dica | 0.0000E+00 | 1.2413E-24 |
| Montmorillonite-Na_Na0.3 | 0.0000E+00 | 9.3319E-25 |
| Montmorillonite-Mg_Mg0.4 | 0.0000E+00 | 7.2199E-25 |
| Smectite-high-Fe-Mg_Ca.0 | 0.0000E+00 | 6.0326E-25 |
| Montmorillonite-Ca_Ca0.1 | 0.0000E+00 | 4.9652E-25 |
| Montmorillonite-K_K0.33M | 0.0000E+00 | 4.0467E-25 |
| Beidellite-H_H.33Al2.33S | 0.0000E+00 | 2.5229E-25 |
| Smectite-low-Fe-Mg_Ca.02 | 0.0000E+00 | 8.8450E-26 |
| Montmorillonite-H_H0.33M | 0.0000E+00 | 5.5516E-26 |
| Pyrophyllite_Al2Si4O10(O | 0.0000E+00 | 1.2052E-26 |
| Na2(U +VI 2O7)(c) | 0.0000E+00 | 4.7509E-28 |
| Mg2CaCl6:12H2O_Tachyhydr | 0.0000E+00 | 1.6473E-28 |

| | | |
|--------------------------|------------|------------|
| Ca4Al2Cl2O6:10H2O_Friede | 0.0000E+00 | 1.6536E-29 |
| C3AH6_Ca3Al2O6:6H2O_Hydr | 0.0000E+00 | 3.3009E-32 |
| C3AS(0.5)_Ca3Al2Si(0.5)O | 0.0000E+00 | 1.9072E-32 |
| C3ASH4_Ca3Al2SiO8:4H2O_S | 0.0000E+00 | 1.8274E-32 |
| Mordenite_Ca.2895Na.361A | 0.0000E+00 | 7.9329E-33 |
| 2MgCl2:MnCl2:12H2O | 0.0000E+00 | 3.4235E-33 |
| Hematite_Fe2O3 | 0.0000E+00 | 1.6816E-34 |
| (U +VI O2)2(SiO4):2H2O(c | 0.0000E+00 | 8.9004E-35 |
| Ferrocaldonite_KFeFeSi4 | 0.0000E+00 | 1.2707E-35 |
| Ca(U +VI O2)2(SiO3OH)2:5 | 0.0000E+00 | 5.8225E-36 |
| Cronstedtite-7A_Fe2Fe2Si | 0.0000E+00 | 2.2428E-36 |
| Trevorite_NiFe2O4 | 0.0000E+00 | 9.2260E-38 |
| (U +VI O2)3(PO4)2:4H2O(c | 0.0000E+00 | 1.4441E-38 |
| Ca4Cl2(OH)6:13H2O | 0.0000E+00 | 6.9964E-39 |
| Sepiolite_Mg4Si6O15(OH)2 | 0.0000E+00 | 2.5485E-39 |
| MgCl2:2MnCl2:12H2O | 0.0000E+00 | 1.5806E-39 |
| Stilbite_Ca1.019Na.136K. | 0.0000E+00 | 8.1294E-41 |
| Laumontite_K0.2Na0.2Ca1. | 0.0000E+00 | 1.3218E-43 |
| Ca(SO4):2H2O_Gypsum | 0.0000E+00 | 9.3679E-44 |
| Ca(SO4)_Anhydrite | 0.0000E+00 | 5.9013E-44 |
| K2Si4O9 | 0.0000E+00 | 4.4469E-44 |
| Chabazite_K0.6Na0.2Ca1.5 | 0.0000E+00 | 1.5699E-44 |
| Bassanite_CaSO4:0.5H2O | 0.0000E+00 | 1.2654E-44 |
| CaSO4:0.5H2O(beta) | 0.0000E+00 | 8.5925E-45 |
| Fe(SO4):7H2O_Melanterite | 0.0000E+00 | 4.1851E-45 |
| Na2(SO4):10H2O_Mirabilit | 0.0000E+00 | 3.3973E-45 |
| Mg(SO4):7H2O_Epsomite | 0.0000E+00 | 9.5072E-46 |
| Mg(SO4):6H2O_Hexahydrate | 0.0000E+00 | 5.5055E-46 |
| Na2(SO4)_Thenardite | 0.0000E+00 | 4.9532E-46 |
| Mg(SO4):H2O_Kieserite | 0.0000E+00 | 1.8857E-47 |
| K2(SO4)_Arcanite | 0.0000E+00 | 8.6379E-49 |
| KMgCl(SO4):3H2O_Kainite | 0.0000E+00 | 1.6871E-50 |
| NiSO4:7H2O(cr) | 0.0000E+00 | 4.0796E-51 |
| alpha-NiSO4:6H2O | 0.0000E+00 | 4.0705E-51 |
| beta-NiSO4:6H2O | 0.0000E+00 | 2.7679E-51 |
| Morenosite_NiSO4:7H2O | 0.0000E+00 | 2.0073E-51 |
| NiSO4:6H2O(alpha) | 0.0000E+00 | 1.9717E-51 |
| Mg2Cl(OH)3:4H2O_Oxychlor | 0.0000E+00 | 6.7613E-52 |
| K2CrO4:MgCrO4 | 0.0000E+00 | 1.1593E-52 |
| K(HSO4)_Mercurite | 0.0000E+00 | 7.8732E-54 |
| Nontronite-Na_Na.33Fe2Al | 0.0000E+00 | 3.5495E-54 |
| Nontronite-Mg_Mg.165Fe2A | 0.0000E+00 | 2.7462E-54 |
| Nontronite-Ca_Ca.165Fe2A | 0.0000E+00 | 1.8886E-54 |
| Nontronite-K_K.33Fe2Al.3 | 0.0000E+00 | 1.5392E-54 |
| Nontronite-H_H.33Fe2Al.3 | 0.0000E+00 | 1.3331E-54 |
| MnSO4:5H2O | 0.0000E+00 | 6.0765E-55 |
| MnSO4:4H2O | 0.0000E+00 | 5.1724E-55 |
| U +IV (OH)2(SO4)(c) | 0.0000E+00 | 3.3572E-55 |
| MnSO4:H2O | 0.0000E+00 | 3.2047E-55 |

| | | |
|--------------------------|------------|-------------|
| KAlCl4 | 0.0000E+00 | 2.5539E-55 |
| UO2(OH)2:2H2O | 0.0000E+00 | 1.6002E-56 |
| U +VI O3:2H2O(c)_Metasch | 0.0000E+00 | 9.1364E-57 |
| Na(U +VI O2)(OH)3(c)_Cla | 0.0000E+00 | 7.9940E-57 |
| NiSO4(cr) | 0.0000E+00 | 3.4361E-58 |
| Na2CrO4:6H2O | 0.0000E+00 | 1.3660E-58 |
| K3AlCl6 | 0.0000E+00 | 1.3362E-58 |
| Na2CrO4:4H2O | 0.0000E+00 | 6.0357E-59 |
| MgCrO4:5H2O | 0.0000E+00 | 4.5646E-59 |
| MgCrO4 | 0.0000E+00 | 1.6843E-59 |
| K2CrO4 | 0.0000E+00 | 2.8835E-62 |
| Na2(U +VI O2)2(Si2O5)3:4 | 0.0000E+00 | 3.2159E-64 |
| K8(HCO3)4(CO3)2:3H2O | 0.0000E+00 | 8.4024E-67 |
| Phillipsite_K0.7Na0.7Ca1 | 0.0000E+00 | 1.4296E-69 |
| Na2CrO4:MgCrO4:2H2O | 0.0000E+00 | 4.5625E-71 |
| K2Ca(SO4)2:H2O_Syngenite | 0.0000E+00 | <1.0000E-75 |
| K2Mg(SO4)2:4H2O_Leonite | 0.0000E+00 | <1.0000E-75 |
| K2Mg(SO4)2:6H2O_Picromer | 0.0000E+00 | <1.0000E-75 |
| K2MgCa2(SO4)4:2H2O_Polyh | 0.0000E+00 | <1.0000E-75 |
| K3(HSO4)(SO4) | 0.0000E+00 | <1.0000E-75 |
| K8(HSO4)6(SO4)_Misenite | 0.0000E+00 | <1.0000E-75 |
| Na2Ca(SO4)2_Glauberite | 0.0000E+00 | <1.0000E-75 |
| Na2Mg(SO4)2:4H2O_Bloedit | 0.0000E+00 | <1.0000E-75 |
| Na3(HSO4)(SO4) | 0.0000E+00 | <1.0000E-75 |
| Na4Ca(SO4)3:2H2O | 0.0000E+00 | <1.0000E-75 |
| Na6(CO3)(SO4)2_Burkeite | 0.0000E+00 | <1.0000E-75 |
| NaK3(SO4)2_Glaserite | 0.0000E+00 | <1.0000E-75 |
| FeK2(SO4)2:6H2O | 0.0000E+00 | <1.0000E-75 |
| FeNa2(SO4)2:4H2O | 0.0000E+00 | <1.0000E-75 |
| Fe2(SO4)3 | 0.0000E+00 | <1.0000E-75 |
| KFe3(SO4)2(OH)6_Jarosite | 0.0000E+00 | <1.0000E-75 |
| NaFe3(SO4)2(OH)6_Jarosit | 0.0000E+00 | <1.0000E-75 |
| Cu2CO3(OH)2(s) | 0.0000E+00 | <1.0000E-75 |
| Cu3(CO3)2(OH)2(s) | 0.0000E+00 | <1.0000E-75 |
| MnSO4:K2SO4:4H2O | 0.0000E+00 | <1.0000E-75 |
| MnSO4:Na2SO4:2H2O | 0.0000E+00 | <1.0000E-75 |
| beta-Ni(OH)2 | 0.0000E+00 | <1.0000E-75 |
| K2Ni(SO4)2:6H2O(cr) | 0.0000E+00 | <1.0000E-75 |
| Na2Ni(SO4)2:4H2O(cr) | 0.0000E+00 | <1.0000E-75 |
| Ca(U +VI 6O19):11H2O(c)_ | 0.0000E+00 | <1.0000E-75 |
| K2(U +VI 6O19):11H2O(c)_ | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-Ca_Ca1.73 | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-K_K3.467A | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-Na_Na3.46 | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-NH4_(NH4) | 0.0000E+00 | <1.0000E-75 |
| Erionite_K1.5Na0.9Ca0.9A | 0.0000E+00 | <1.0000E-75 |
| KAl(SO4)2 | 0.0000E+00 | <1.0000E-75 |
| KAl(SO4)2:12H2O | 0.0000E+00 | <1.0000E-75 |
| KAl(SO4)2:3H2O | 0.0000E+00 | <1.0000E-75 |

| | | |
|---|------------|-------------|
| KAl ₃ (OH) ₆ (SO ₄) ₂ _Alunite | 0.0000E+00 | <1.0000E-75 |
| Stellerite_Ca ₂ .0Al ₄ .0Si ₁ | 0.0000E+00 | <1.0000E-75 |
| C ₉ S ₆ H ₁₆ _Ca ₉ Si ₆ O ₂₁ :16H ₂ O_ | 0.0000E+00 | <1.0000E-75 |
| Ca ₆ Al ₂ (SO ₄) ₃ (OH) ₁₂ :26H ₂ O | 0.0000E+00 | <1.0000E-75 |
| BSK3+3SF | 0.0000E+00 | <1.0000E-75 |
| HAW-Kokille | 0.0000E+00 | <1.0000E-75 |

| Cp_EQUIL | H_EQUIL | S_EQUIL | G_EQUIL | V_EQUIL |
|----------|---------|---------|---------|---------|
| J.K-1 | J | J.K-1 | J | m3 |

| | | | | |
|-------------|--------------|--------------|--------------|-------------|
| 4.28888E+04 | -1.21067E+11 | -7.25259E+06 | -1.18904E+11 | 1.51472E+02 |
|-------------|--------------|--------------|--------------|-------------|

Properties for AQUEOUS:

pH = 7.3470

Eh/V = -0.4820

Total solute molality = 1.5890

Ionic strength = 0.7251

Osmotic coefficient = 0.9151

Debye-Hueckel slope = 0.3905

Mole fraction of system components:

| | GAS | AQUEOUS |
|----|------------|------------|
| H | 9.9960E-01 | 6.5807E-01 |
| O | 3.9705E-04 | 3.3263E-01 |
| NA | 0.0000E+00 | 2.5661E-03 |
| K | 0.0000E+00 | 2.0798E-05 |
| MG | 0.0000E+00 | 1.2756E-04 |
| CA | 0.0000E+00 | 2.4366E-05 |
| CL | 0.0000E+00 | 3.3613E-03 |
| S | 1.1471E-13 | 2.8053E-14 |
| C | 1.9761E-07 | 5.4937E-08 |
| FE | 0.0000E+00 | 2.1754E-04 |
| SI | 0.0000E+00 | 1.6768E-12 |
| AL | 0.0000E+00 | 2.7943E-11 |
| U | 0.0000E+00 | 1.8738E-11 |
| P | 0.0000E+00 | 1.6574E-04 |
| AS | 0.0000E+00 | 6.3693E-05 |
| N | 0.0000E+00 | 3.4069E-04 |
| NI | 0.0000E+00 | 4.5652E-10 |
| B | 0.0000E+00 | 2.4036E-03 |
| ZR | 0.0000E+00 | 3.7929E-12 |
| MN | 0.0000E+00 | 9.3303E-14 |
| SR | 0.0000E+00 | 8.1506E-06 |
| MO | 0.0000E+00 | 5.1307E-07 |
| CU | 0.0000E+00 | 3.5077E-15 |
| CR | 0.0000E+00 | 4.2617E-14 |
| SN | 0.0000E+00 | 1.1322E-16 |
| Ti | 0.0000E+00 | 1.3297E-12 |
| V | 0.0000E+00 | 6.2661E-09 |

System density/g.cm-3 including gaseous phase = 0.146689

System density/g.cm-3 excluding gaseous phase = 5.97786

| | Cp/J.K-1 | H/J | S/J.K-1 | G/J | V/m3 |
|---------------|-------------|--------------|--------------|--------------|-------------|
| AQUEOUS | 2.43061E+04 | -2.05131E+10 | 5.06503E+04 | -2.05282E+10 | 1.50690E+00 |
| FeAl2O4_Hercy | 0.00000E+00 | -2.38834E+08 | 7.29037E-08 | -2.38834E+08 | 1.27062E-07 |
| Fe3O4_Magneti | 0.00000E+00 | -4.85375E+10 | 5.56182E-05 | -4.85375E+10 | 2.12903E+00 |
| Na2[B4O5(OH)4 | 0.00000E+00 | -5.34949E+08 | -5.23076E-07 | -5.34949E+08 | 9.69470E-08 |
| Cu(OH)2(s) | 0.00000E+00 | -3.82380E+07 | -5.95742E-09 | -3.82380E+07 | 8.34765E-08 |
| FeCO3_Siderit | 0.00000E+00 | -6.95074E+08 | 1.72967E-07 | -6.95074E+08 | 1.02513E-06 |
| Cassiterite_S | 0.00000E+00 | -9.75238E+07 | -8.78174E-09 | -9.75238E+07 | 1.87685E-07 |
| Eskolaite_Cr2 | 0.00000E+00 | -2.20070E+09 | -1.88218E-06 | -2.20070E+09 | 2.09379E-06 |
| Rutile-TiO2 | 0.00000E+00 | 7.94539E+07 | 3.74870E-08 | 7.94539E+07 | 1.89526E-07 |
| V2O4 | 0.00000E+00 | -1.96076E+07 | -1.62260E-08 | -1.96076E+07 | 1.48751E-08 |
| Mn(OH)2 | 0.00000E+00 | -1.62285E+09 | -4.96806E-07 | -1.62285E+09 | 2.37283E-06 |
| CaMoO4_Powell | 0.00000E+00 | -5.71464E+07 | -3.92172E-08 | -5.71464E+07 | 3.93631E-08 |
| Ni(cr) | 0.00000E+00 | 7.55453E+00 | 4.31089E-15 | 7.55453E+00 | 1.93706E-06 |
| Ni3S2(cr) | 0.00000E+00 | -1.02553E+07 | -1.01508E-08 | -1.02553E+07 | 4.85639E-08 |
| U +IV (OH)4(a | 0.00000E+00 | -2.85945E+10 | -1.27428E-05 | -2.85945E+10 | 1.94529E-05 |
| U +IV (SiO4)(| 0.00000E+00 | -2.01500E+09 | 7.44687E-07 | -2.01500E+09 | 1.06896E-06 |
| ZrO2(monoclin | 0.00000E+00 | -1.59274E+10 | -6.87257E-06 | -1.59274E+10 | 1.52744E-05 |

| | Dens/g.cm-3 | Thermal exp | Bulk mod/bar | Cv/J.K-1 | Grueneisen |
|---------------|-------------|-------------|--------------|-------------|-------------|
| AQUEOUS | 1.04908E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| FeAl2O4_Hercy | 1.73806E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| Fe3O4_Magneti | 5.20018E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Na2[B4O5(OH)4 | 3.81372E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Cu(OH)2(s) | 9.75607E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| FeCO3_Siderit | 1.15854E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Cassiterite_S | 1.50709E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Eskolaite_Cr2 | 1.51990E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Rutile-TiO2 | 7.98788E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| V2O4 | 1.65881E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| Mn(OH)2 | 8.89527E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| CaMoO4_Powell | 2.00016E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| Ni(cr) | 5.86934E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Ni3S2(cr) | 2.40212E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| U +IV (OH)4(a | 3.06058E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| U +IV (SiO4)(| 3.30112E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| ZrO2(monoclin | 1.23223E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |

System Volume : 1.51472129055100254D+02 m3

System composition in [mol]:

| | | | | | | |
|-----|--|------|--|--------------------------|----|----|
| 500 | | 1 IA | | -1.66777633992811499E-10 | << | EA |
| 500 | | 2 IA | | 7.28930683849802241E+05 | << | H |
| 500 | | 3 IA | | 4.06009593591355486E+05 | << | O |
| 500 | | 4 IA | | 8.45886935600006609E+02 | << | NA |
| 500 | | 5 IA | | 5.28434524999914679E+00 | << | K |

500 | 6 | IA | 3.24120756000014651E+01<< MG
500 | 7 | IA | 4.55539549000069499E+01<< CA
500 | 8 | IA | 8.54050679099954436E+02<< CL
500 | 9 | IA | 9.71278323151877885E+01<< S
500 | 10 | IA | 1.02523726004436094E+03<< C
500 | 11 | IA | 1.44660489927476883E+05<< FE
500 | 12 | IA | 1.06895515479519167E+03<< SI
500 | 13 | IA | 2.54123094442883769E+02<< AL
500 | 16 | IA | 2.05218938456352153E+04<< U
500 | 23 | IA | 4.21126758835344646E+01<< P
500 | 27 | IA | 1.61833169606606688E+01<< AS
500 | 29 | IA | 8.65642870911732558E+01<< N
500 | 31 | IA | 2.08275138260861968E+03<< NI
500 | 32 | IA | 9.98510776061101183E+02<< B
500 | 33 | IA | 1.52744485003947011E+04<< ZR
500 | 37 | IA | 2.37282560746382023E+03<< MN
500 | 38 | IA | 2.07092514996680954E+00<< SR
500 | 39 | IA | 3.94934333958771902E+01<< MO
500 | 41 | IA | 8.34765366820890051E+01<< CU
500 | 44 | IA | 4.18758829989180049E+03<< CR
500 | 45 | IA | 1.87684609552692024E+02<< SN
500 | 46 | IA | 1.89525769319153994E+02<< Ti
500 | 47 | IA | 2.97517740938135979E+01<< V

Aq. phase composition in [mol]:

500 | 1 | IA | -1.66777633992811499E-10<< EA
500 | 2 | IA | 1.67204112196569680E+05<< H
500 | 3 | IA | 8.45158310510683950E+04<< O
500 | 4 | IA | 6.51992993951642120E+02<< NA
500 | 5 | IA | 5.28434524999914679E+00<< K
500 | 6 | IA | 3.24120756000014651E+01<< MG
500 | 7 | IA | 6.19088489366154260E+00<< CA
500 | 8 | IA | 8.54050679099954436E+02<< CL
500 | 9 | IA | 7.12771849521058518E-09<< S
500 | 10 | IA | 1.39586128549676006E-02<< C
500 | 11 | IA | 5.52722374065751012E+01<< FE
500 | 12 | IA | 4.26056899013073178E-07<< SI
500 | 13 | IA | 7.09985479598024753E-06<< AL
500 | 16 | IA | 4.76099495212858821E-06<< U
500 | 23 | IA | 4.21126758835344646E+01<< P
500 | 27 | IA | 1.61833169606606688E+01<< AS
500 | 29 | IA | 8.65642870911732558E+01<< N
500 | 31 | IA | 1.15994827469500988E-04<< NI
500 | 32 | IA | 6.10722892764372205E+02<< B
500 | 33 | IA | 9.63706675097301375E-07<< ZR
500 | 37 | IA | 2.37067573824971334E-08<< MN
500 | 38 | IA | 2.07092514996680954E+00<< SR
500 | 39 | IA | 1.30363389531777840E-01<< MO
500 | 41 | IA | 8.91255339980191924E-10<< CU

| | | |
|-----|--------------------------|-------------------------|
| 79 | FeAl2O4_Hercynite | 1.27061543671514485D+02 |
| 85 | Fe3O4_Magnetite | 4.78176757180016793D+04 |
| 107 | Na2[B4O5(OH)4]:8H2O_Bora | 9.69469708241822445D+01 |
| 140 | Cu(OH)2(s) | 8.34765366811977430D+01 |
| 144 | FeCO3_Siderite | 1.02512899239376793D+03 |
| 166 | Cassiterite_SnO2 | 1.87684609552663261D+02 |
| 173 | Eskolaite_Cr2O3 | 2.09379414994048602D+03 |
| 183 | Rutile-TiO2 | 1.89525768981288138D+02 |
| 187 | V2O4 | 1.48750909970968479D+01 |
| 192 | Mn(OH)2 | 2.37282560744011334D+03 |
| 200 | CaMoO4_Powellite | 3.93630700063454100D+01 |
| 215 | Ni(cr) | 1.93705951823382384D+03 |
| 217 | Ni3S2(cr) | 4.85639161266561956D+01 |
| 283 | U +IV (OH)4(am) | 1.94529386865050874D+04 |
| 284 | U +IV (SiO4)(c)_Coffinit | 1.06895515436913479D+03 |
| 400 | ZrO2(monoclinic) | 1.52744484994309951D+04 |

Other information: *****

Total volume of solid phases = 2.12907820631774758D+00 m3
Charge balance after calculation = -1.66777633992811499D-10
Relative error of charge balance = -8.91156296945682992D-14
kw = 13.984 (Should be at current T: 13.9960)
Eh = -0.482
pH = 7.347
pHCl= 3.881
phi = 0.0006
T = 298.1500 K

CPU-Time: 3.530 s

End of calculation

* Nathan: *
* Kaum war der Vater tot, so kommt ein jeder mit *
* seinem Ring, und jeder will der Fuerst des Hauses *
* sein. Man untersucht, man zankt, man klagt. *
* Umsonst; der rechte Ring war nicht erweislich; - *
* * *
* Fast so unerweislich, als uns ist - *
* der rechte Glaube. *
* * *
* Aus: "Nathan, der Weise" von G. E. Lessing *

Anlage 3: Output-Datei der geochemischen Modellierung mit NaCl-Lösung

This is NATHAN 2.7.1

This program contains ChemApp
Copyright GTT-Technologies, Kaiserstrasse 100,
D-52134 Herzogenrath, Germany
<http://www.gtt-technologies.de>

No warranty is given for the correctness of
calculations done with NATHAN. Propositions
and error messages are welcome.

Helge C. Moog
Gesellschaft fuer Anlagen- und Reaktorsicherheit (GRS)
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Germany
<http://www.grs.de/geochemie>
helge.moog@grs.de

Used input file:
NaCl+3BSK3+3SF.ni

Version of prepare.f: 236
Active Chemapp-Version: 554
Number of phases = 406
Number of system components = 48
Name of paramaterfile used:
tdb-hmw-118#1-1-1.dat

> n*3.0684 m3 ges. NaCl-Loesung

> reagiert mit 3BSK3+3SF

>

Step 1:
Charge balance before calculation = (no charged constituents set for aq. phase)

Active CHEMAPP-stream:

SYSTEM UNITS:
Pressure : Pa
Volume : m3

Temperature : K
 Energy : J
 Amount : mol

TARGET LIMITS:

Pressure/bar : 2.00000E+01 6.00000E+01
 Volume/dm3 : 1.00000E-07 1.00000E+50
 Temperature/K: 298.15 6000.00

T = 298.15 K
 P = 4.00000E+06 Pa

STREAM CONSTITUENT

BSK3+3SF
 AMOUNT/mol = 3.000000000000000E+00
 H
 AMOUNT/mol = 7.12210450028760E+05
 O
 AMOUNT/mol = 3.56105225014380E+05
 NA
 AMOUNT/mol = 3.92619034774110E+04
 CL
 AMOUNT/mol = 3.92619034774098E+04

ELIMINATED PHASES:

K2HAsO4:3H2O
 K3AsO4:7H2O
 KH2AsO4
 Na2HAsO4:7H2O
 Na3AsO4:12H2O
 NaH2AsO4:H2O

T = 298.15 K
 P = 4.00000E+06 Pa
 V = 1.52377E+02 m3

| STREAM CONSTITUENTS | AMOUNT/mol |
|---------------------|------------|
| BSK3+3SF | 3.0000E+00 |
| H | 7.1221E+05 |
| O | 3.5611E+05 |
| NA | 3.9262E+04 |
| CL | 3.9262E+04 |

| PHASE: GAS | EQUIL AMOUNT mol | MOLE FRACTION | FUGACITY Pa |
|------------|---------------------|---------------|----------------|
| H2 | 2.3852E+05 | 9.9939E-01 | 3.9976E+06 |
| H2O(g) | 1.4517E+02 | 6.0826E-04 | 2.4330E+03 |
| CO2(g) | 8.6349E-02 | 3.6180E-07 | 1.4472E+00 |
| H2S(g) | 5.4788E-08 | 2.2956E-13 | 9.1825E-07 |

| | | | |
|----------------|------------|------------|-------------|
| O2 (g) | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| TOTAL: | 2.3866E+05 | 1.0000E+00 | 1.0000E+00 |
| PHASE: AQUEOUS | mol | MOLALITY | ACTIVITY |
| H2O | 7.3724E+04 | 5.5508E+01 | 7.4661E-01 |
| H<+> | 5.0774E-05 | 3.8229E-08 | 1.7570E-07 |
| OH<-> | 1.1845E-04 | 8.9182E-08 | 4.4051E-08 |
| O2 (aq) | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (CO2)<0> | 2.0059E-04 | 1.5103E-07 | 4.7041E-07 |
| (CO3)<2-> | 1.5935E-05 | 1.1998E-08 | 2.4799E-10 |
| (HCO3)<-> | 3.6225E-03 | 2.7275E-06 | 9.1008E-07 |
| (HSO4)<-> | 3.6198E-48 | 2.7254E-51 | 1.3676E-51 |
| (SO4)<2-> | 5.8526E-42 | 4.4066E-45 | 7.9965E-47 |
| Cl<-> | 8.2977E+03 | 6.2476E+00 | 6.2600E+00 |
| Na<+> | 7.9598E+03 | 5.9931E+00 | 6.0385E+00 |
| H2S<0> | 1.1775E-09 | 8.8659E-13 | 8.8659E-13 |
| HS<-> | 2.4708E-09 | 1.8603E-12 | 5.1806E-13 |
| H2SiO4<2-> | 4.0800E-15 | 3.0720E-18 | 3.9219E-20 |
| H3SiO4<-> | 1.8273E-10 | 1.3758E-13 | 8.6432E-14 |
| SiO2<0> | 2.2250E-07 | 1.6752E-10 | 1.6752E-10 |
| Al(OH)4<-> | 2.1028E-06 | 1.5832E-09 | 3.2675E-10 |
| Al<3+> | 6.8966E-11 | 5.1926E-14 | 1.2832E-14 |
| Fe<2+> | 1.6280E+02 | 1.2258E-01 | 1.4062E-01 |
| Fe<3+> | 2.0505E-34 | 1.5438E-37 | 1.1326E-37 |
| (AsO4)<3-> | 1.2178E-17 | 9.1691E-21 | 2.0561E-24 |
| (H2AsO4)<-> | 1.4648E-17 | 1.1029E-20 | 3.1400E-20 |
| (HAsO4)<2-> | 3.2564E-16 | 2.4518E-19 | 9.9823E-21 |
| H2AsO3<-> | 1.7014E-01 | 1.2810E-04 | 3.5675E-05 |
| HAsO2<0> | 1.6013E+01 | 1.2057E-02 | 1.2057E-02 |
| B(OH)3<0> | 3.2260E+02 | 2.4289E-01 | 2.4289E-01 |
| Cr<3+> | 4.6059E-03 | 3.4679E-06 | 6.6318E-13 |
| CrO4<2-> | 5.4004E-54 | 4.0661E-57 | 2.4853E-58 |
| Cu<+> | 9.5137E-09 | 7.1631E-12 | 2.0079E-12 |
| Cu<2+> | 2.4092E-18 | 1.8139E-21 | 1.1186E-22 |
| Cu(CO3)2<2-> | 1.4150E-46 | 1.0654E-49 | 3.4040E-52 |
| Cu(HCO3)<-> | 9.4428E-12 | 7.1097E-15 | 1.9800E-15 |
| Cu(OH)<+> | 2.5998E-20 | 1.9575E-23 | 5.4871E-24 |
| Cu(OH)2(aq) | 1.7916E-22 | 1.3490E-25 | 1.3490E-25 |
| Cu(OH)3<-> | 1.1197E-25 | 8.4305E-29 | 2.3478E-29 |
| Cu(OH)4<2-> | 3.0908E-30 | 2.3271E-33 | 7.4355E-36 |
| Cu2(OH)<3+> | 1.5125E-34 | 1.1388E-37 | 2.1778E-44 |
| Cu2(OH)2<2+> | 3.5395E-36 | 2.6650E-39 | 8.8855E-42 |
| Cu3(OH)4<2+> | 1.6179E-55 | 1.2181E-58 | 4.0614E-61 |
| CuCO3<0> | 2.0587E-22 | 1.5500E-25 | 1.5500E-25 |
| Mn<2+> | 1.4043E-07 | 1.0574E-10 | 8.9315E-11 |
| MoO4<2-> | 3.9493E+01 | 2.9736E-02 | 6.7572E-03 |
| (NH4)<+> | 8.6564E+01 | 6.5176E-02 | 1.8270E-02 |
| NO3<-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| Ni<2+> | 5.5501E-05 | 4.1788E-08 | 8.0687E-08 |
| HNi(P2O7)<-> | 1.8255E-12 | 1.3745E-15 | 3.8277E-16 |

| | | | |
|-------------------------------|------------|------------|-------------|
| Ni (HPO4) <0> | 4.5863E-08 | 3.4531E-11 | 3.4531E-11 |
| Ni (P2O7) <2-> | 1.4029E-09 | 1.0563E-12 | 3.3749E-15 |
| (H2P2O7) <2-> | 1.1214E-08 | 8.4433E-12 | 2.6978E-14 |
| (H2PO4) <-> | 5.2089E-03 | 3.9219E-06 | 1.0922E-06 |
| (H3P2O7) <-> | 4.0199E-15 | 3.0267E-18 | 8.4289E-19 |
| (HP2O7) <3-> | 2.6596E-04 | 2.0025E-07 | 3.4367E-14 |
| (HPO4) <2-> | 1.5855E-01 | 1.1938E-04 | 3.8143E-07 |
| (P2O7) <4-> | 2.0970E+01 | 1.5789E-02 | 7.7886E-17 |
| (PO4) <3-> | 7.5053E-03 | 5.6509E-06 | 9.6981E-13 |
| H3PO4 <0> | 3.5180E-08 | 2.6488E-11 | 2.6488E-11 |
| H4P2O7 <0> | 1.9670E-21 | 1.4810E-24 | 1.4810E-24 |
| Sn(OH)6 <2-> | 1.0786E-11 | 8.1212E-15 | 2.5949E-17 |
| Sn <2+> | 5.6310E-10 | 4.2397E-13 | 1.4136E-15 |
| Sn <4+> | 1.3524E-18 | 1.0183E-21 | 6.1567E-36 |
| SnO3 <2-> | 1.4673E-13 | 1.1047E-16 | 3.5299E-19 |
| Ti (OH)4 <0> | 1.7739E-07 | 1.3356E-10 | 1.3356E-10 |
| (UO2) (H2PO4) (H3PO4) <+> | 3.1092E-32 | 2.3410E-35 | 6.5621E-36 |
| (UO2) (H2PO4) <+> | 3.4642E-22 | 2.6083E-25 | 7.3114E-26 |
| (UO2) (H2PO4) 2 <0> | 4.8474E-27 | 3.6497E-30 | 3.6497E-30 |
| (UO2) (H3PO4) <2+> | 2.2338E-27 | 1.6819E-30 | 5.6076E-33 |
| (UO2) (HASO4) <0> | 1.0158E-31 | 7.6483E-35 | 7.6483E-35 |
| (UO2) (HPO4) <0> | 3.2389E-19 | 2.4387E-22 | 2.4387E-22 |
| U +III <3+> | 9.3115E-18 | 7.0109E-21 | 1.3407E-27 |
| U +IV (CO3)4 <4-> | 8.9365E-22 | 6.7285E-25 | 4.0813E-29 |
| U +IV (CO3)5 <6-> | 2.5294E-13 | 1.9044E-16 | 7.6288E-40 |
| U +IV (OH) <3+> | 1.9942E-15 | 1.5015E-18 | 1.0588E-19 |
| U +IV (OH) 2 (CO3) 2 <2-> | 1.7895E-12 | 1.3474E-15 | 2.4288E-17 |
| U +IV (OH) 2 <2+> | 5.0319E-10 | 3.7886E-13 | 1.2749E-13 |
| U +IV (OH) 3 <+> | 1.7174E-07 | 1.2931E-10 | 1.4000E-10 |
| U +IV (OH) 4 <0> | 4.2000E-06 | 3.1623E-09 | 3.1623E-09 |
| U +IV (SO4) <2+> | 1.0400E-59 | 7.8302E-63 | 2.6107E-65 |
| U +IV (SO4) 2 <0> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| U +IV <4+> | 1.2211E-20 | 9.1943E-24 | 8.3978E-26 |
| U +IV Cl <3+> | 1.9706E-13 | 1.4837E-16 | 2.8374E-23 |
| (U +V O2) (CO3) 3 <5-> | 8.2058E-07 | 6.1783E-10 | 5.3790E-36 |
| (U +V O2) <+> | 1.9113E-10 | 1.4391E-13 | 4.0339E-14 |
| (U +VI O2) (CO3) <0> | 1.0487E-19 | 7.8958E-23 | 7.8958E-23 |
| (U +VI O2) (CO3) 2 <2-> | 1.8332E-20 | 1.3802E-23 | 9.1003E-26 |
| (U +VI O2) (CO3) 3 <4-> | 9.9220E-22 | 7.4705E-25 | 3.8081E-30 |
| (U +VI O2) (H2AsO4) <+> | 1.0091E-34 | 7.5980E-38 | 2.1298E-38 |
| (U +VI O2) (H2AsO4) 2 <0> | 6.6711E-53 | 5.0228E-56 | 5.0228E-56 |
| (U +VI O2) (OH) <+> | 5.0179E-19 | 3.7781E-22 | 9.0447E-22 |
| (U +VI O2) (OH) 2 <0> | 6.6115E-19 | 4.9780E-22 | 4.9780E-22 |
| (U +VI O2) (OH) 3 <-> | 1.1292E-18 | 8.5021E-22 | 1.7287E-23 |
| (U +VI O2) (OH) 4 <2-> | 2.4723E-24 | 1.8614E-27 | 1.6157E-28 |
| (U +VI O2) (PO4) <-> | 2.8899E-18 | 2.1758E-21 | 6.0594E-22 |
| (U +VI O2) (SO4) <0> | 5.6439E-63 | 4.2494E-66 | 4.2494E-66 |
| (U +VI O2) (SO4) 2 <2-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2) (SO4) 3 <4-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |

| | | | |
|--------------------------|------------|------------|-------------|
| (U +VI O2)<2+> | 3.4146E-20 | 2.5710E-23 | 3.6791E-23 |
| (U +VI O2)11(CO3)6(OH)12 | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)2(CO3)(OH)3<-> | 1.8980E-32 | 1.4291E-35 | 3.9798E-36 |
| (U +VI O2)2(OH)<3+> | 8.2000E-32 | 6.1740E-35 | 1.1807E-41 |
| (U +VI O2)2(OH)2<2+> | 1.1301E-35 | 8.5088E-39 | 6.2057E-38 |
| (U +VI O2)3(CO3)6<6-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)3(OH)4<2+> | 4.9354E-49 | 3.7159E-52 | 2.2901E-53 |
| (U +VI O2)3(OH)5<+> | 1.6841E-48 | 1.2680E-51 | 2.2413E-50 |
| (U +VI O2)3(OH)7<-> | 6.2645E-49 | 4.7167E-52 | 9.5900E-54 |
| (U +VI O2)3O(OH)2(HCO3)< | 2.2449E-53 | 1.6903E-56 | 4.7380E-57 |
| (U +VI O2)4(OH)7<+> | 1.4373E-63 | 1.0822E-66 | 7.0398E-66 |
| (U +VI O2)Cl<+> | 1.6599E-18 | 1.2498E-21 | 3.5034E-22 |
| (U +VI O2)Cl2<0> | 1.6087E-19 | 1.2112E-22 | 1.2112E-22 |
| (U +VI O2)SiO(OH)3<+> | 1.4175E-24 | 1.0673E-27 | 2.9917E-28 |
| V<2+> | 8.9749E-02 | 6.7574E-05 | 2.2530E-07 |
| V<3+> | 2.9662E+01 | 2.2333E-02 | 4.2709E-09 |
| VO<2+> | 5.0757E-05 | 3.8216E-08 | 1.2742E-10 |
| VO2<+> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| Zr(OH)<3+> | 1.0247E-17 | 7.7150E-21 | 1.4754E-27 |
| Zr(OH)2<2+> | 1.1744E-14 | 8.8426E-18 | 2.9482E-20 |
| Zr(OH)4<0> | 5.0597E-07 | 3.8096E-10 | 3.8096E-10 |
| Zr<4+> | 3.5483E-17 | 2.6716E-20 | 1.6153E-34 |
| TOTAL: | 9.0660E+04 | | 1.0000E+00 |
| | mol | | ACTIVITY |
| Fe3O4_Magnetite | 4.7783E+04 | | 1.0000E+00 |
| NaCl_Halite | 3.0964E+04 | | 1.0000E+00 |
| U +IV (OH)4(am) | 1.9453E+04 | | 1.0000E+00 |
| ZrO2(monoclinic) | 1.5274E+04 | | 1.0000E+00 |
| Mn(OH)2 | 2.3728E+03 | | 1.0000E+00 |
| Eskolaite_Cr2O3 | 2.0938E+03 | | 1.0000E+00 |
| Ni(cr) | 2.0501E+03 | | 1.0000E+00 |
| U +IV (SiO4)(c)_Coffinit | 1.0690E+03 | | 1.0000E+00 |
| FeCO3_Siderite | 1.0212E+03 | | 1.0000E+00 |
| Rutile-TiO2 | 1.8953E+02 | | 1.0000E+00 |
| Cassiterite_SnO2 | 1.8768E+02 | | 1.0000E+00 |
| Na2[B4O5(OH)4]:8H2O_Bora | 1.6898E+02 | | 1.0000E+00 |
| FeAl2O4_Hercynite | 1.2706E+02 | | 1.0000E+00 |
| Cu(OH)2(s) | 8.3477E+01 | | 1.0000E+00 |
| Ni3S2(cr) | 1.0891E+01 | | 1.0000E+00 |
| SnO2(cr)_Cassiterite | 0.0000E+00 | | 1.0000E+00 |
| CuO(s) | 0.0000E+00 | | 1.2150E-01 |
| Gibbsite_Al(OH)3 | 0.0000E+00 | | 1.1579E-01 |
| Anatase_TiO2 | 0.0000E+00 | | 8.1922E-02 |
| Fe(OH)2(cr) | 0.0000E+00 | | 6.2891E-02 |
| V2O4 | 0.0000E+00 | | 2.6935E-02 |
| Fe(OH)2_precipitated | 0.0000E+00 | | 1.3833E-02 |
| FeS | 0.0000E+00 | | 5.9032E-03 |
| AlOOH_Boehmite | 0.0000E+00 | | 3.3891E-03 |
| FeCl2:4H2O | 0.0000E+00 | | 1.9615E-03 |

| | | |
|--|------------|------------|
| Dawsonite_NaAlCO ₃ (OH) ₂ | 0.0000E+00 | 1.7333E-03 |
| MnHPO ₄ | 0.0000E+00 | 3.0154E-04 |
| Fe ₂ (SiO ₄)_Fayalite | 0.0000E+00 | 6.8521E-05 |
| Ni(OH) ₂ (cr) | 0.0000E+00 | 4.8766E-05 |
| SnS_Herzenbergite | 0.0000E+00 | 1.7479E-05 |
| Na(HCO ₃)_Nahcolite | 0.0000E+00 | 1.3566E-05 |
| FeSiO ₃ (cr) | 0.0000E+00 | 7.5026E-06 |
| Chamosite-7A_Fe ₂ Al ₂ SiO ₅ (| 0.0000E+00 | 6.2668E-06 |
| Fe ₃ Si ₂ O ₅ (OH) ₄ _Greenalite | 0.0000E+00 | 4.0721E-06 |
| Ni(CO ₃)(cr) | 0.0000E+00 | 2.0101E-06 |
| Tridymite_SiO ₂ | 0.0000E+00 | 1.1269E-06 |
| Chalcedony-SiO ₂ | 0.0000E+00 | 8.9573E-07 |
| NiO(cr) | 0.0000E+00 | 6.6034E-07 |
| Coesite_SiO ₂ | 0.0000E+00 | 2.5905E-07 |
| Fe(OH) ₃ (cr) | 0.0000E+00 | 1.0454E-07 |
| Al ₂ O ₃ _Corundum | 0.0000E+00 | 9.8501E-08 |
| SiO ₂ (am)-GWB | 0.0000E+00 | 8.6632E-08 |
| Na ₂ (CO ₃):7H ₂ O | 0.0000E+00 | 3.1998E-09 |
| Na ₂ (CO ₃):10H ₂ O_Natron | 0.0000E+00 | 3.1417E-09 |
| Na ₂ (CO ₃):H ₂ O_Thermonatri | 0.0000E+00 | 2.0797E-09 |
| NiCl ₂ :6H ₂ O(cr) | 0.0000E+00 | 5.6122E-10 |
| Rhodochrosite_MnCO ₃ | 0.0000E+00 | 2.7220E-10 |
| Ni(CO ₃):5.5H ₂ O(cr) | 0.0000E+00 | 1.5963E-10 |
| Daphnite-7A_Fe ₅ AlAlSi ₃ O ₁₁ | 0.0000E+00 | 1.4961E-10 |
| NiCO ₃ | 0.0000E+00 | 1.3365E-10 |
| NiCl ₂ :4H ₂ O(cr) | 0.0000E+00 | 1.1914E-10 |
| FeS ₂ | 0.0000E+00 | 2.8436E-11 |
| NiCl ₂ :2H ₂ O(cr) | 0.0000E+00 | 1.6023E-11 |
| Ni ₂ SiO ₄ | 0.0000E+00 | 3.0749E-12 |
| MnCl ₂ :4H ₂ O | 0.0000E+00 | 1.6488E-12 |
| Na ₃ (CO ₃)(HCO ₃):2H ₂ O_Tron | 0.0000E+00 | 2.8223E-13 |
| Na.96Al.96Si ₂ .04O ₆ :H ₂ O_A | 0.0000E+00 | 1.1744E-13 |
| Kaolinite_Al ₂ Si ₂ O ₅ (OH) ₄ | 0.0000E+00 | 3.9935E-14 |
| NaOH | 0.0000E+00 | 3.2792E-14 |
| U +VI O ₃ :2H ₂ O(c)_Schoepi | 0.0000E+00 | 8.3476E-15 |
| NiCl ₂ (cr) | 0.0000E+00 | 7.2216E-15 |
| NiS ₂ (cr) | 0.0000E+00 | 4.9978E-16 |
| Delafossite_CuFeO ₂ | 0.0000E+00 | 3.6794E-16 |
| Minnesotaites_Fe ₃ Si ₄ O ₁₀ (O | 0.0000E+00 | 2.0525E-17 |
| Ni ₂ (SiO ₄)(oliv) | 0.0000E+00 | 1.4534E-17 |
| Ni ₉ S ₈ (cr) | 0.0000E+00 | 7.1021E-18 |
| (U +VI O ₂)(CO ₃)(c)_Ruthe | 0.0000E+00 | 5.2167E-18 |
| Goethite_FeOOH | 0.0000E+00 | 3.5984E-18 |
| Na.96Al.96Si ₂ .04O ₆ _Analc | 0.0000E+00 | 5.7635E-20 |
| alpha-NiS | 0.0000E+00 | 9.3808E-21 |
| beta-NiS | 0.0000E+00 | 2.5249E-21 |
| Na ₄ (U +VI O ₂)(CO ₃) ₃ (c) | 0.0000E+00 | 1.0994E-21 |
| Beidellite-Na_Na.33Al ₂ .3 | 0.0000E+00 | 7.0154E-25 |
| Beidellite-H_H.33Al ₂ .33S | 0.0000E+00 | 2.3683E-26 |

| | | |
|--|------------|-------------|
| Na ₂ (U +VI 2O ₇)(c) | 0.0000E+00 | 1.9208E-26 |
| Pyrophyllite_Al ₂ Si ₄ O ₁₀ (O | 0.0000E+00 | 1.0064E-27 |
| Hematite_Fe ₂ O ₃ | 0.0000E+00 | 1.5388E-34 |
| (U +VI O ₂) ₂ (SiO ₄):2H ₂ O(c | 0.0000E+00 | 5.2262E-35 |
| Cronstedtite-7A_Fe ₂ Fe ₂ Si | 0.0000E+00 | 8.4569E-37 |
| Trevorite_NiFe ₂ O ₄ | 0.0000E+00 | 6.4695E-38 |
| (U +VI O ₂) ₃ (PO ₄) ₂ :4H ₂ O(c | 0.0000E+00 | 3.7703E-43 |
| Na ₂ (SO ₄)_Thenardite | 0.0000E+00 | 5.2951E-45 |
| Na ₂ (SO ₄):10H ₂ O_Mirabilit | 0.0000E+00 | 2.5399E-45 |
| Fe(SO ₄):7H ₂ O_Melanterite | 0.0000E+00 | 3.1967E-46 |
| alpha-NiSO ₄ :6H ₂ O | 0.0000E+00 | 2.8446E-52 |
| NiSO ₄ :7H ₂ O(cr) | 0.0000E+00 | 2.1851E-52 |
| beta-NiSO ₄ :6H ₂ O | 0.0000E+00 | 1.9343E-52 |
| NiSO ₄ :6H ₂ O(alpha) | 0.0000E+00 | 1.3779E-52 |
| Morenosite_NiSO ₄ :7H ₂ O | 0.0000E+00 | 1.0751E-52 |
| Nontronite-Na_Na.33Fe ₂ Al | 0.0000E+00 | 5.8718E-55 |
| U +IV (OH) ₂ (SO ₄)(c) | 0.0000E+00 | 1.9709E-55 |
| MnSO ₄ :H ₂ O | 0.0000E+00 | 1.4420E-55 |
| Nontronite-H_H.33Fe ₂ Al.3 | 0.0000E+00 | 1.2514E-55 |
| MnSO ₄ :4H ₂ O | 0.0000E+00 | 1.0478E-55 |
| MnSO ₄ :5H ₂ O | 0.0000E+00 | 9.4343E-56 |
| Na(U +VI O ₂)(OH) ₃ (c)_Cla | 0.0000E+00 | 3.4106E-56 |
| UO ₂ (OH) ₂ :2H ₂ O | 0.0000E+00 | 9.3981E-57 |
| U +VI O ₃ :2H ₂ O(c)_Metasch | 0.0000E+00 | 5.3658E-57 |
| Na ₂ CrO ₄ :6H ₂ O | 0.0000E+00 | 4.4123E-58 |
| Na ₂ CrO ₄ :4H ₂ O | 0.0000E+00 | 3.3190E-58 |
| NiSO ₄ (cr) | 0.0000E+00 | 1.1848E-58 |
| Na ₂ (U +VI O ₂) ₂ (Si ₂ O ₅) ₃ :4 | 0.0000E+00 | 1.8431E-64 |
| Na ₃ (HSO ₄)(SO ₄) | 0.0000E+00 | <1.0000E-75 |
| Na ₆ (CO ₃)(SO ₄) ₂ _Burkeite | 0.0000E+00 | <1.0000E-75 |
| FeNa ₂ (SO ₄) ₂ :4H ₂ O | 0.0000E+00 | <1.0000E-75 |
| Fe ₂ (SO ₄) ₃ | 0.0000E+00 | <1.0000E-75 |
| NaFe ₃ (SO ₄) ₂ (OH) ₆ _Jarosit | 0.0000E+00 | <1.0000E-75 |
| Cu ₂ CO ₃ (OH) ₂ (s) | 0.0000E+00 | <1.0000E-75 |
| Cu ₃ (CO ₃) ₂ (OH) ₂ (s) | 0.0000E+00 | <1.0000E-75 |
| MnSO ₄ :Na ₂ SO ₄ :2H ₂ O | 0.0000E+00 | <1.0000E-75 |
| beta-Ni(OH) ₂ | 0.0000E+00 | <1.0000E-75 |
| Na ₂ Ni(SO ₄) ₂ :4H ₂ O(cr) | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-Na_Na3.46 | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-NH ₄ _(NH ₄) | 0.0000E+00 | <1.0000E-75 |
| BSK3+3SF | 0.0000E+00 | <1.0000E-75 |
| HAW-Kokille | 0.0000E+00 | <1.0000E-75 |

```
*****
Cp_EQUIL      H_EQUIL      S_EQUIL      G_EQUIL      V_EQUIL
  J.K-1        J          J.K-1        J            m3
*****
4.77220E+05  -1.33764E+11  -7.28220E+06  -1.31593E+11  1.52377E+02
```

Properties for AQUEOUS:

pH = 6.7552
 Eh/V = -0.4470
 Total solute molality = 12.7516
 Ionic strength = 6.6848
 Osmotic coefficient = 1.2720
 Debye-Hueckel slope = 0.3905

Mole fraction of system components:

| | GAS | AQUEOUS |
|----|------------|------------|
| H | 9.9970E-01 | 6.1794E-01 |
| O | 3.0440E-04 | 3.1163E-01 |
| NA | 0.0000E+00 | 3.3060E-02 |
| CL | 0.0000E+00 | 3.4464E-02 |
| S | 1.1475E-13 | 1.5153E-14 |
| C | 1.8085E-07 | 1.5955E-08 |
| FE | 0.0000E+00 | 6.7618E-04 |
| SI | 0.0000E+00 | 9.2489E-13 |
| AL | 0.0000E+00 | 8.7340E-12 |
| U | 0.0000E+00 | 2.1569E-11 |
| P | 0.0000E+00 | 1.7491E-04 |
| AS | 0.0000E+00 | 6.7216E-05 |
| N | 0.0000E+00 | 3.5954E-04 |
| NI | 0.0000E+00 | 2.3072E-10 |
| B | 0.0000E+00 | 1.3399E-03 |
| ZR | 0.0000E+00 | 2.1015E-12 |
| MN | 0.0000E+00 | 5.8328E-13 |
| MO | 0.0000E+00 | 1.6403E-04 |
| CU | 0.0000E+00 | 3.9554E-14 |
| CR | 0.0000E+00 | 1.9130E-08 |
| SN | 0.0000E+00 | 2.3842E-15 |
| Ti | 0.0000E+00 | 7.3677E-13 |
| V | 0.0000E+00 | 1.2357E-04 |

System density/g.cm-3 including gaseous phase = 0.159492

System density/g.cm-3 excluding gaseous phase = 5.33978

| | Cp/J.K-1 | H/J | S/J.K-1 | G/J | V/m3 |
|----------------|-------------|--------------|--------------|--------------|-------------|
| AQUEOUS | 4.60394E+05 | -2.10526E+10 | 2.77702E+04 | -2.10608E+10 | 1.49674E+00 |
| NaCl_Halite | 0.00000E+00 | -1.18934E+10 | -8.59042E-06 | -1.18934E+10 | 8.36497E-01 |
| FeAl2O4_Hercy | 0.00000E+00 | -2.38834E+08 | 7.29037E-08 | -2.38834E+08 | 1.27062E-07 |
| Fe3O4_Magnetit | 0.00000E+00 | -4.85024E+10 | 5.55780E-05 | -4.85024E+10 | 2.12750E+00 |
| Na2[B4O5(OH)4] | 0.00000E+00 | -9.32415E+08 | -9.11721E-07 | -9.32415E+08 | 1.68978E-07 |
| Cu(OH)2(s) | 0.00000E+00 | -3.82380E+07 | -5.95742E-09 | -3.82380E+07 | 8.34765E-08 |
| FeCO3_Siderit | 0.00000E+00 | -6.92437E+08 | 1.72311E-07 | -6.92437E+08 | 1.02124E-06 |
| Cassiterite_S | 0.00000E+00 | -9.75238E+07 | -8.78174E-09 | -9.75238E+07 | 1.87685E-07 |
| Eskolaite_Cr2 | 0.00000E+00 | -2.20069E+09 | -1.88218E-06 | -2.20069E+09 | 2.09379E-06 |
| Rutile-TiO2 | 0.00000E+00 | 7.94539E+07 | 3.74870E-08 | 7.94539E+07 | 1.89526E-07 |
| Mn(OH)2 | 0.00000E+00 | -1.62285E+09 | -4.96806E-07 | -1.62285E+09 | 2.37283E-06 |
| Ni(cr) | 0.00000E+00 | 7.99531E+00 | 4.56241E-15 | 7.99531E+00 | 2.05008E-06 |

| | | | | | |
|---------------|-------------|--------------|--------------|--------------|-------------|
| Ni3S2(cr) | 0.00000E+00 | -2.29985E+06 | -2.27639E-09 | -2.29985E+06 | 1.08909E-08 |
| U +IV (OH)4(a | 0.00000E+00 | -2.85945E+10 | -1.27428E-05 | -2.85945E+10 | 1.94529E-05 |
| U +IV (SiO4)(| 0.00000E+00 | -2.01500E+09 | 7.44687E-07 | -2.01500E+09 | 1.06896E-06 |
| ZrO2(monoclin | 0.00000E+00 | -1.59274E+10 | -6.87257E-06 | -1.59274E+10 | 1.52744E-05 |

| | Dens/g.cm-3 | Thermal exp | Bulk mod/bar | Cv/J.K-1 | Grueneisen |
|--|-------------|-------------|--------------|-------------|-------------|
| AQUEOUS | 1.23546E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| NaCl_Halite | 2.16333E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| FeAl2O4_Hercy | 1.73806E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| Fe3O4_Magneti | 5.20018E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Na2[B4O5(OH)4 | 3.81372E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Cu(OH)2(s) | 9.75607E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| FeCO3_Siderit | 1.15854E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Cassiterite_S | 1.50709E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Eskolaite_Cr2 | 1.51990E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Rutile-TiO2 | 7.98788E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Mn(OH)2 | 8.89527E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Ni(cr) | 5.86934E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Ni3S2(cr) | 2.40212E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| U +IV (OH)4(a | 3.06058E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| U +IV (SiO4)(| 3.30112E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| ZrO2(monoclin | 1.23223E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| System Volume : 1.52377208410995337D+02 m3 | | | | | |

System composition in [mol]:

| |
|---|
| 500 1 IA -9.62561369952013816E-11 << EA |
| 500 2 IA 7.12210450028764899E+05 << H |
| 500 3 IA 3.97337025260537572E+05 << O |
| 500 4 IA 3.92619034774110987E+04 << NA |
| 500 8 IA 3.92619034774098473E+04 << CL |
| 500 9 IA 2.17817402151881048E+01 << S |
| 500 10 IA 1.02132931469438972E+03 << C |
| 500 11 IA 1.44660489927476912E+05 << FE |
| 500 12 IA 1.06895515479519031E+03 << SI |
| 500 13 IA 2.54123094443164774E+02 << AL |
| 500 16 IA 2.05218938456351898E+04 << U |
| 500 23 IA 4.21126758834627282E+01 << P |
| 500 27 IA 1.61833169606625802E+01 << AS |
| 500 29 IA 8.65642870911205478E+01 << N |
| 500 31 IA 2.08275138260861968E+03 << NI |
| 500 32 IA 9.98510776061260344E+02 << B |
| 500 33 IA 1.52744485003946957E+04 << ZR |
| 500 37 IA 2.37282560746371837E+03 << MN |
| 500 39 IA 3.94934333959687578E+01 << MO |
| 500 41 IA 8.34765366820889909E+01 << CU |
| 500 44 IA 4.18758829989179867E+03 << CR |
| 500 45 IA 1.87684609552659538E+02 << SN |
| 500 46 IA 1.89525769319153994E+02 << Ti |
| 500 47 IA 2.97517740938127453E+01 << V |

Aq. phase composition in [mol]:

```

500 | 1|IA | -9.62561369952013816E-11<< EA
500 | 2|IA | 1.48778377768739301E+05<< H
500 | 3|IA | 7.50296950019459036E+04<< O
500 | 4|IA | 7.95978555196814705E+03<< NA
500 | 8|IA | 8.29774230112769510E+03<< CL
500 | 9|IA | 3.64830097396324719E-09<< S
500 | 10|IA | 3.84151106640733073E-03<< C
500 | 11|IA | 1.62800565483434553E+02<< FE
500 | 12|IA | 2.22680766817209320E-07<< SI
500 | 13|IA | 2.10285089800944589E-06<< AL
500 | 16|IA | 5.19301023383186877E-06<< U
500 | 23|IA | 4.21126758834627282E+01<< P
500 | 27|IA | 1.61833169606625802E+01<< AS
500 | 29|IA | 8.65642870911205478E+01<< N
500 | 31|IA | 5.55483564376417895E-05<< NI
500 | 32|IA | 3.22597277739663241E+02<< B
500 | 33|IA | 5.05972565919156119E-07<< ZR
500 | 37|IA | 1.40434439986077670E-07<< MN
500 | 39|IA | 3.94934333959687578E+01<< MO
500 | 41|IA | 9.52319272117555123E-09<< CU
500 | 44|IA | 4.60587602073415578E-03<< CR
500 | 45|IA | 5.74031613244723557E-10<< SN
500 | 46|IA | 1.77388878195528909E-07<< Ti
500 | 47|IA | 2.97517740938127453E+01<< V

```

```

Gas Phase : *****
Total Volume : 1.47916427764369075D+02 m3
Total Mass : 4.83444571231442678D+05 g
Pressure : 3.99999999999999953D+06 Pa

```

```

Solution: *****
Total Volume : 1.49674366984964458D+00 m3
Total Mass : 1.84917421025139256D+06 g
Density : 1.23546486115231179D+06 g / m3

```

System component molalities:

| | | |
|-------|--------------------------|------------|
| 1 EA | -7.21865029397567103E-14 | mol/kg H2O |
| 2 H | 1.12018691274988186E+02 | mol/kg H2O |
| 3 O | 5.64915975488305833E+01 | mol/kg H2O |
| 4 NA | 5.99310715530855553E+00 | mol/kg H2O |
| 8 CL | 6.24756263006341328E+00 | mol/kg H2O |
| 9 S | 2.74689041922392306E-12 | mol/kg H2O |
| 10 C | 2.89236277899348400E-06 | mol/kg H2O |
| 11 FE | 1.22576321625373755E-01 | mol/kg H2O |
| 12 SI | 1.67661514025567016E-10 | mol/kg H2O |
| 13 AL | 1.58328521304085840E-09 | mol/kg H2O |
| 16 U | 3.90993784779453050E-09 | mol/kg H2O |
| 23 P | 3.17076104021376953E-02 | mol/kg H2O |

| | | |
|-------|-------------------------|------------|
| 27 AS | 1.21847946832677866E-02 | mol/kg H2O |
| 29 N | 6.51762594573546267E-02 | mol/kg H2O |
| 31 NI | 4.18236459083682181E-08 | mol/kg H2O |
| 32 B | 2.42890972486889706E-01 | mol/kg H2O |
| 33 ZR | 3.80958480024646102E-10 | mol/kg H2O |
| 37 MN | 1.05736346995454842E-10 | mol/kg H2O |
| 39 MO | 2.97355219845788880E-02 | mol/kg H2O |
| 41 CU | 7.17023267348546679E-12 | mol/kg H2O |
| 44 CR | 3.46787088120743245E-06 | mol/kg H2O |
| 45 SN | 4.32201715266013658E-13 | mol/kg H2O |
| 46 Ti | 1.33560200616575169E-10 | mol/kg H2O |
| 47 V | 2.24008007553249747E-02 | mol/kg H2O |

Solid Phases : *****

Total Volume : 2.96403697677661837D+00 m3

Total Mass : 2.19704013611463743D+04 kg

| | mol | activity |
|------------------------------|-------------------------|----------|
| mole fraction | | |
| 53 NaCl_Halite | 3.09641611762821522D+04 | |
| 79 FeAl2O4_Hercynite | 1.27061546170156944D+02 | |
| 85 Fe3O4_Magnetite | 4.77831295639078235D+04 | |
| 107 Na2[B4O5(OH)4]:8H2O_Bora | 1.68978374580399276D+02 | |
| 140 Cu(OH)2(s) | 8.34765366725657998D+01 | |
| 144 FeCO3_Siderite | 1.02123912409987997D+03 | |
| 166 Cassiterite_SnO2 | 1.87684609552085504D+02 | |
| 173 Eskolaite_Cr2O3 | 2.09379184700788892D+03 | |
| 183 Rutile-TiO2 | 1.89525769141765124D+02 | |
| 192 Mn(OH)2 | 2.37282560732328375D+03 | |
| 215 Ni(cr) | 2.05007871682513587D+03 | |
| 217 Ni3S2(cr) | 1.08908700783757801D+01 | |
| 283 U +IV (OH)4(am) | 1.94529386858696707D+04 | |
| 284 U +IV (SiO4)(c)_Coffinit | 1.06895515457250963D+03 | |
| 400 ZrO2(monoclinic) | 1.52744484998887237D+04 | |

Other information: *****

Total volume of solid phases = 2.96403697677661837D+00 m3

Charge balance after calculation = -9.62561369952013816D-11

Relative error of charge balance = -5.68813690485893110D-15

kw = 13.984 (Should be at current T: 13.9960)

Eh = -0.447

pH = 6.755

pHCl= 2.979

phi = 0.0010

T = 298.1500 K

CPU-Time: 11.220 s

End of calculation

* Nathan: *

* Kaum war der Vater tot, so kommt ein jeder mit *

* seinem Ring, und jeder will der Fuerst des Hauses *

* sein. Man untersucht, man zankt, man klagt. *

* Umsonst; der rechte Ring war nicht erweislich; - *

* *

* Fast so unerweislich, als uns ist - *

* der rechte Glaube. *

* *

* Aus: "Nathan, der Weise" von G. E. Lessing *

Anlage 4: Output-Datei der geochemischen Modellierung mit IP21-Lösung

This is NATHAN 2.7.1

This program contains ChemApp
Copyright GTT-Technologies, Kaiserstrasse 100,
D-52134 Herzogenrath, Germany
<http://www.gtt-technologies.de>

No warranty is given for the correctness of
calculations done with NATHAN. Propositions
and error messages are welcome.

Helge C. Moog
Gesellschaft fuer Anlagen- und Reaktorsicherheit (GRS)
Theodor-Heuss-Str. 4, 38122 Braunschweig,
Germany
<http://www.grs.de/geochemie>
helge.moog@grs.de

Used input file:
IP21+3BSK3+3SF.ni

Version of prepare.f: 236
Active Chemapp-Version: 554
Number of phases = 406
Number of system components = 48
Name of paramaterfile used:
tdb-hmw-118#1-1-1.dat

> Berechnung von n*3.0684 m3 IP21-Loesung

> mit BSK3+3SF

>

Step 1:
Charge balance before calculation = (no charged constituents set for aq. phase)

Active CHEMAPP-stream:

SYSTEM UNITS:

Pressure : Pa
Volume : m3

Temperature : K
Energy : J
Amount : mol

TARGET LIMITS:

Pressure/bar : 2.00000E+01 6.00000E+01
Volume/dm3 : 1.00000E-07 1.00000E+50
Temperature/K: 298.15 6000.00

T = 298.15 K
P = 4.00000E+06 Pa

STREAM CONSTITUENT

BSK3+3SF
AMOUNT/mol = 3.000000000000000E+00
H
AMOUNT/mol = 1.02045272143284E+06
O
AMOUNT/mol = 5.21747746144681E+05
NA
AMOUNT/mol = 4.42624462838430E+03
K
AMOUNT/mol = 5.21997303213620E+03
MG
AMOUNT/mol = 3.86432443059239E+04
CA
AMOUNT/mol = 5.82853342308084E+00
CL
AMOUNT/mol = 8.11836706250860E+04
S
AMOUNT/mol = 2.88034635706537E+03

ELIMINATED PHASES:

K2HAsO4: 3H2O
K3AsO4: 7H2O
KH2AsO4
Na2HAsO4: 7H2O
Na3AsO4: 12H2O
NaH2AsO4: H2O

T = 298.15 K
P = 4.00000E+06 Pa
V = 1.49299E+02 m3

| STREAM CONSTITUENTS | AMOUNT/mol |
|---------------------|------------|
| BSK3+3SF | 3.0000E+00 |
| H | 1.0205E+06 |
| O | 5.2175E+05 |
| NA | 4.4262E+03 |

| | |
|----|------------|
| K | 5.2200E+03 |
| MG | 3.8643E+04 |
| CA | 5.8285E+00 |
| CL | 8.1184E+04 |
| S | 2.8803E+03 |

| | EQUIL AMOUNT | MOLE FRACTION | FUGACITY |
|----------------|--------------|---------------|-------------|
| PHASE: GAS | mol | | Pa |
| H2 | 2.2831E+05 | 9.9975E-01 | 3.9990E+06 |
| H2O(g) | 5.7235E+01 | 2.5063E-04 | 1.0025E+03 |
| CO2(g) | 7.4869E-05 | 3.2785E-10 | 1.3114E-03 |
| H2S(g) | 2.7226E-06 | 1.1922E-11 | 4.7688E-05 |
| O2(g) | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| TOTAL: | 2.2837E+05 | 1.0000E+00 | 1.0000E+00 |
| PHASE: AQUEOUS | mol | MOLALITY | ACTIVITY |
| H2O | 6.2018E+04 | 5.5508E+01 | 3.0763E-01 |
| H<+> | 7.6401E-07 | 6.8382E-10 | 1.6939E-08 |
| OH<-> | 6.8004E-03 | 6.0866E-06 | 1.8827E-07 |
| O2(aq) | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (CO2)<0> | 4.7750E-08 | 4.2738E-11 | 4.2626E-10 |
| (CO3)<2-> | 4.0121E-05 | 3.5910E-08 | 9.9618E-12 |
| (HCO3)<-> | 4.2912E-04 | 3.8407E-07 | 3.5245E-09 |
| (HSO4)<-> | 3.1040E-49 | 2.7782E-52 | 2.1205E-50 |
| (SO4)<2-> | 9.5258E-41 | 8.5259E-44 | 1.2860E-44 |
| Ca(CO3)<0> | 4.3796E-07 | 3.9199E-10 | 3.9199E-10 |
| Ca<2+> | 5.8285E+00 | 5.2167E-03 | 2.8756E-02 |
| Cl<-> | 1.3526E+04 | 1.2106E+01 | 2.9324E+02 |
| K<+> | 1.4866E+01 | 1.3306E-02 | 2.6232E-03 |
| Mg(CO3)<0> | 3.5900E-03 | 3.2131E-06 | 3.2131E-06 |
| Mg(OH)<+> | 3.8141E+01 | 3.4137E-02 | 1.1098E-02 |
| Mg<2+> | 7.3565E+03 | 6.5843E+00 | 3.9592E+02 |
| Na<+> | 4.4249E+01 | 3.9605E-02 | 1.2891E-01 |
| H2S<0> | 5.1444E-08 | 4.6044E-11 | 4.6044E-11 |
| HS<-> | 3.0241E-06 | 2.7067E-09 | 2.7907E-10 |
| H2SiO4<2-> | 1.4718E-15 | 1.3173E-18 | 1.2162E-19 |
| H3SiO4<-> | 3.7718E-13 | 3.3759E-16 | 2.5841E-14 |
| SiO2<0> | 3.1777E-08 | 2.8442E-11 | 2.8442E-11 |
| Al(OH)4<-> | 3.1616E-10 | 2.8297E-13 | 2.1506E-11 |
| Al<3+> | 1.0640E-17 | 9.5230E-21 | 2.5313E-18 |
| Fe<2+> | 3.4557E+00 | 3.0929E-03 | 4.2633E-03 |
| Fe<3+> | 3.7642E-34 | 3.3691E-37 | 3.3100E-40 |
| (AsO4)<3-> | 6.2548E-12 | 5.5982E-15 | 4.1671E-22 |
| (H2AsO4)<-> | 1.4456E-20 | 1.2939E-23 | 5.9152E-20 |
| (HAsO4)<2-> | 6.7023E-13 | 5.9988E-16 | 1.9505E-19 |
| H2AsO3<-> | 1.7681E+00 | 1.5825E-03 | 1.6316E-04 |
| HAsO2<0> | 1.4415E+01 | 1.2902E-02 | 1.2902E-02 |
| B(OH)3<0> | 9.9851E+02 | 8.9370E-01 | 8.9370E-01 |
| Cr<3+> | 6.0826E-06 | 5.4441E-09 | 2.2469E-15 |
| CrO4<2-> | 6.3852E-55 | 5.7150E-58 | 2.9124E-57 |

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|------------------------|------------|------------|-------------|
| Cu<+> | 7.8061E-09 | 6.9867E-12 | 1.1404E-12 |
| Cu<2+> | 5.0088E-19 | 4.4830E-22 | 6.1241E-24 |
| Cu(CO3)2<2-> | 7.5870E-50 | 6.7906E-53 | 3.0071E-56 |
| Cu(HCO3)<-> | 4.8959E-13 | 4.3820E-16 | 4.5180E-17 |
| Cu(OH)<+> | 8.7880E-21 | 7.8655E-24 | 1.2839E-24 |
| Cu(OH)2(aq) | 1.5072E-22 | 1.3490E-25 | 1.3490E-25 |
| Cu(OH)3<-> | 1.0873E-24 | 9.7319E-28 | 1.0034E-28 |
| Cu(OH)4<2-> | 3.4266E-28 | 3.0669E-31 | 1.3582E-34 |
| Cu2(OH)<3+> | 7.5521E-37 | 6.7594E-40 | 2.7897E-46 |
| Cu2(OH)2<2+> | 1.9683E-37 | 1.7617E-40 | 4.8645E-43 |
| Cu3(OH)4<2+> | 8.9970E-57 | 8.0526E-60 | 2.2235E-62 |
| CuCO3<0> | 3.8086E-25 | 3.4088E-28 | 3.4088E-28 |
| Mn<2+> | 1.6700E-09 | 1.4947E-12 | 4.8897E-12 |
| MoO4<2-> | 3.9493E+01 | 3.5348E-02 | 7.7375E-08 |
| (NH4)<+> | 8.6564E+01 | 7.7478E-02 | 1.2646E-02 |
| NO3<-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| Ni<2+> | 1.6698E-09 | 1.4945E-12 | 5.3871E-11 |
| HNi(P2O7)<-> | 2.0193E-18 | 1.8074E-21 | 1.8635E-22 |
| Ni(HPO4)<0> | 1.4380E-12 | 1.2870E-15 | 1.2870E-15 |
| Ni(P2O7)<2-> | 4.2998E-14 | 3.8484E-17 | 1.7042E-20 |
| (H2P2O7)<2-> | 4.7849E-12 | 4.2826E-15 | 1.8965E-18 |
| (H2PO4)<-> | 6.3698E-05 | 5.7012E-08 | 5.8782E-09 |
| (H3P2O7)<-> | 6.1905E-20 | 5.5407E-23 | 5.7127E-24 |
| (HP2O7)<3-> | 4.1454E-06 | 3.7103E-09 | 2.5059E-17 |
| (HPO4)<2-> | 5.3722E-02 | 4.8083E-05 | 2.1293E-08 |
| (P2O7)<4-> | 2.0983E+01 | 1.8780E-02 | 5.8908E-19 |
| (PO4)<3-> | 9.2895E-02 | 8.3144E-05 | 5.6156E-13 |
| H3PO4<0> | 1.5356E-11 | 1.3744E-14 | 1.3744E-14 |
| H4P2O7<0> | 1.0812E-27 | 9.6769E-31 | 9.6769E-31 |
| Sn(OH)6<2-> | 2.0302E-10 | 1.8171E-13 | 8.0469E-17 |
| Sn<2+> | 3.1325E-11 | 2.8037E-14 | 7.7416E-17 |
| Sn<4+> | 7.4970E-23 | 6.7100E-26 | 3.1329E-39 |
| SnO3<2-> | 3.9479E-11 | 3.5335E-14 | 1.5648E-17 |
| Ti(OH)4<0> | 2.5335E-08 | 2.2675E-11 | 2.2675E-11 |
| (UO2)(H2PO4)(H3PO4)<+> | 6.8648E-39 | 6.1442E-42 | 1.0029E-42 |
| (UO2)(H2PO4)<+> | 1.4741E-25 | 1.3193E-28 | 2.1535E-29 |
| (UO2)(H2PO4)2<0> | 6.4643E-33 | 5.7857E-36 | 5.7857E-36 |
| (UO2)(H3PO4)<2+> | 6.4433E-32 | 5.7669E-35 | 1.5924E-37 |
| (UO2)(HAsO4)<0> | 9.1379E-32 | 8.1787E-35 | 8.1787E-35 |
| (UO2)(HPO4)<0> | 8.3242E-22 | 7.4504E-25 | 7.4504E-25 |
| U +III <3+> | 1.1286E-19 | 1.0101E-22 | 4.1689E-29 |
| U +IV (CO3)4<4-> | 9.9033E-18 | 8.8638E-21 | 3.1850E-37 |
| U +IV (CO3)5<6-> | 2.0427E+02 | 1.8282E-01 | 2.3915E-49 |
| U +IV (OH)<3+> | 7.9130E-20 | 7.0824E-23 | 1.3563E-21 |
| U +IV (OH)2(CO3)2<2-> | 5.3884E-15 | 4.8228E-18 | 2.1456E-21 |
| U +IV (OH)2<2+> | 3.8759E-12 | 3.4690E-15 | 6.9798E-15 |
| U +IV (OH)3<+> | 2.6287E-08 | 2.3528E-11 | 3.2758E-11 |
| U +IV (OH)4<0> | 3.5331E-06 | 3.1623E-09 | 3.1623E-09 |
| U +IV (SO4)<2+> | 5.0919E-60 | 4.5574E-63 | 1.2584E-65 |

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|--------------------------|------------|------------|-------------|
| U +IV (SO4)2<0> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| U +IV <4+> | 2.0873E-28 | 1.8682E-31 | 2.5170E-28 |
| U +IV Cl<3+> | 1.0784E-14 | 9.6522E-18 | 3.9837E-24 |
| (U +V O2)(CO3)3<5-> | 5.8070E-08 | 5.1974E-11 | 1.9795E-40 |
| (U +V O2)<+> | 1.5677E-10 | 1.4031E-13 | 2.2903E-14 |
| (U +VI O2)(CO3)<0> | 1.9394E-22 | 1.7358E-25 | 1.7358E-25 |
| (U +VI O2)(CO3)2<2-> | 2.0381E-23 | 1.8241E-26 | 8.0363E-30 |
| (U +VI O2)(CO3)3<4-> | 4.3970E-16 | 3.9354E-19 | 1.3509E-35 |
| (U +VI O2)(H2AsO4)<+> | 1.5030E-35 | 1.3452E-38 | 2.1958E-39 |
| (U +VI O2)(H2AsO4)2<0> | 1.0899E-53 | 9.7549E-57 | 9.7549E-57 |
| (U +VI O2)(OH)<+> | 3.2696E-20 | 2.9264E-23 | 2.1155E-22 |
| (U +VI O2)(OH)2<0> | 5.5598E-19 | 4.9762E-22 | 4.9762E-22 |
| (U +VI O2)(OH)3<-> | 3.6921E-20 | 3.3045E-23 | 7.3854E-23 |
| (U +VI O2)(OH)4<2-> | 1.7573E-21 | 1.5728E-24 | 2.9502E-27 |
| (U +VI O2)(PO4)<-> | 2.0808E-19 | 1.8624E-22 | 1.9202E-23 |
| (U +VI O2)(SO4)<0> | 4.1787E-62 | 3.7400E-65 | 3.7400E-65 |
| (U +VI O2)(SO4)2<2-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)(SO4)3<4-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)<2+> | 7.3758E-22 | 6.6015E-25 | 2.0135E-24 |
| (U +VI O2)11(CO3)6(OH)12 | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)2(CO3)(OH)3<-> | 4.0505E-34 | 3.6253E-37 | 3.7379E-38 |
| (U +VI O2)2(OH)<3+> | 4.0914E-34 | 3.6619E-37 | 1.5113E-43 |
| (U +VI O2)2(OH)2<2+> | 3.2523E-39 | 2.9109E-42 | 3.3950E-39 |
| (U +VI O2)3(CO3)6<6-> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| (U +VI O2)3(OH)4<2+> | 3.9710E-50 | 3.5542E-53 | 1.2524E-54 |
| (U +VI O2)3(OH)5<+> | 9.8914E-48 | 8.8531E-51 | 5.2387E-51 |
| (U +VI O2)3(OH)7<-> | 2.0468E-50 | 1.8319E-53 | 4.0942E-53 |
| (U +VI O2)3O(OH)2(HCO3)< | 1.6670E-56 | 1.4920E-59 | 2.4354E-60 |
| (U +VI O2)4(OH)7<+> | 3.6641E-65 | 3.2795E-68 | 1.6448E-66 |
| (U +VI O2)Cl<+> | 6.1476E-18 | 5.5023E-21 | 8.9813E-22 |
| (U +VI O2)Cl2<0> | 1.6251E-17 | 1.4546E-20 | 1.4546E-20 |
| (U +VI O2)SiO(OH)3<+> | 3.3506E-26 | 2.9989E-29 | 4.8950E-30 |
| V<2+> | 3.0421E-02 | 2.7228E-05 | 7.5183E-08 |
| V<3+> | 3.7190E-01 | 3.3286E-04 | 1.3738E-10 |
| VO<2+> | 7.0865E-06 | 6.3426E-09 | 1.7513E-11 |
| VO2<+> | NOT CALCD. | NOT CALCD. | <1.0000E-75 |
| Zr(OH)<3+> | 8.6861E-21 | 7.7744E-24 | 3.2086E-30 |
| Zr(OH)2<2+> | 1.1088E-16 | 9.9242E-20 | 2.7403E-22 |
| Zr(OH)4<0> | 7.2263E-08 | 6.4678E-11 | 6.4678E-11 |
| Zr<4+> | 1.9669E-21 | 1.7604E-24 | 8.2194E-38 |
| TOTAL: | 8.4374E+04 | | 1.0000E+00 |
| | mol | | ACTIVITY |
| Fe3O4_Magnetite | 4.7714E+04 | | 1.0000E+00 |
| MgCl2:6H2O_Bischoffite | 2.3830E+04 | | 1.0000E+00 |
| U +IV (OH)4(am) | 1.9249E+04 | | 1.0000E+00 |
| ZrO2(monoclinic) | 1.5274E+04 | | 1.0000E+00 |
| KMgCl3:6H2O_Carnallite | 5.2051E+03 | | 1.0000E+00 |
| NaCl_Halite | 4.3820E+03 | | 1.0000E+00 |
| Mn(OH)2 | 2.3728E+03 | | 1.0000E+00 |

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| Eskolaite_Cr2O3 | 2.0938E+03 | 1.0000E+00 |
| Mg(OH)2_Brucite | 1.7050E+03 | 1.0000E+00 |
| FeS | 1.5136E+03 | 1.0000E+00 |
| U +IV (SiO4)(c)_Coffinit | 1.0690E+03 | 1.0000E+00 |
| Ni3S2(cr) | 6.9425E+02 | 1.0000E+00 |
| Rutile-TiO2 | 1.8953E+02 | 1.0000E+00 |
| Cassiterite_SnO2 | 1.8768E+02 | 1.0000E+00 |
| Mg4Al2O7·10H2O_Hydrotalc | 1.2706E+02 | 1.0000E+00 |
| Cu(OH)2(s) | 8.3477E+01 | 1.0000E+00 |
| V2O4 | 1.4675E+01 | 1.0000E+00 |
| SnO2(cr)_Cassiterite | 0.0000E+00 | 1.0000E+00 |
| CuO(s) | 0.0000E+00 | 2.9486E-01 |
| Mg(CO3)_Magnesite | 0.0000E+00 | 2.4728E-01 |
| CaMoO4_Powellite | 0.0000E+00 | 1.9725E-01 |
| Na2[B4O5(OH)4]·8H2O_Bora | 0.0000E+00 | 1.0673E-01 |
| KCl_Sylvite | 0.0000E+00 | 9.5247E-02 |
| Anatase_TiO2 | 0.0000E+00 | 8.1922E-02 |
| Ni(cr) | 0.0000E+00 | 7.1857E-02 |
| Fe(OH)2(cr) | 0.0000E+00 | 3.4828E-02 |
| Fe(OH)2_precipitated | 0.0000E+00 | 7.6604E-03 |
| SnS_Herzenbergite | 0.0000E+00 | 5.3487E-03 |
| FeAl2O4_Hercynite | 0.0000E+00 | 4.5563E-03 |
| FeCl2·4H2O | 0.0000E+00 | 3.7614E-03 |
| Gibbsite_Al(OH)3 | 0.0000E+00 | 1.7831E-03 |
| Dolomite-ord_CaMg(CO3)2 | 0.0000E+00 | 1.5329E-03 |
| FeCO3_Siderite | 0.0000E+00 | 1.2179E-03 |
| CaCl2·6H2O_Antarcticite | 0.0000E+00 | 1.5008E-04 |
| AlOOH_Boehmite | 0.0000E+00 | 1.2666E-04 |
| CaMg(CO3)2_Dolomite | 0.0000E+00 | 1.1456E-04 |
| Ca(CO3)_Calcite | 0.0000E+00 | 6.6533E-05 |
| CaCl2·4H2O | 0.0000E+00 | 4.8945E-05 |
| Dolomite-dis_CaMg(CO3)2 | 0.0000E+00 | 4.3762E-05 |
| Ca(CO3)_Aragonite | 0.0000E+00 | 4.3465E-05 |
| Mg4Al4Si2O10(OH)8_Amesit | 0.0000E+00 | 3.9905E-05 |
| Fe2(SiO4)_Fayalite | 0.0000E+00 | 2.1014E-05 |
| Ripidolite-14A_Mg3Fe2Al2 | 0.0000E+00 | 1.5768E-05 |
| Mg(CO3)·3H2O_Nesquehonit | 0.0000E+00 | 1.5688E-05 |
| Mg2CaCl6·12H2O_Tachyhydr | 0.0000E+00 | 9.3309E-06 |
| MgMoO4·5H2O(s) | 0.0000E+00 | 6.6583E-06 |
| Mg2Al2SiO5(OH)4_Amesit-7 | 0.0000E+00 | 2.6728E-06 |
| FeSiO3(cr) | 0.0000E+00 | 1.7120E-06 |
| MnHPO4 | 0.0000E+00 | 9.2156E-07 |
| Ni(OH)2(cr) | 0.0000E+00 | 5.9472E-07 |
| FeS2 | 0.0000E+00 | 2.5008E-07 |
| Tridymite_SiO2 | 0.0000E+00 | 1.9132E-07 |
| Chalcedony-SiO2 | 0.0000E+00 | 1.5207E-07 |
| Fe3Si2O5(OH)4_Greenalite | 0.0000E+00 | 4.8380E-08 |
| Coesite_SiO2 | 0.0000E+00 | 4.3980E-08 |
| Fe(OH)3(cr) | 0.0000E+00 | 2.3850E-08 |

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| NiO(cr) | 0.0000E+00 | 1.9544E-08 |
| SiO2(am)-GWB | 0.0000E+00 | 1.4708E-08 |
| Ripidolite-7A_Mg3Fe2Al2S | 0.0000E+00 | 6.6152E-09 |
| Dawsonite_NaAlCO3(OH)2 | 0.0000E+00 | 5.3557E-09 |
| Saponite-Mg_Mg3.165Al.33 | 0.0000E+00 | 1.4602E-09 |
| Na(HCO3)_Nahcolite | 0.0000E+00 | 1.1216E-09 |
| Chamosite-7A_Fe2Al2SiO5(| 0.0000E+00 | 1.1062E-09 |
| Al2O3_Corundum | 0.0000E+00 | 3.3392E-10 |
| Saponite-Ca_Ca.165Mg3Al. | 0.0000E+00 | 2.7708E-10 |
| Saponite-Na_Na.33Mg3Al.3 | 0.0000E+00 | 2.2311E-10 |
| Ca(OH)2_Portlandite | 0.0000E+00 | 1.4564E-10 |
| Saponite-K_K.33Mg3Al.33S | 0.0000E+00 | 1.3465E-10 |
| Ni(CO3)(cr) | 0.0000E+00 | 5.3909E-11 |
| Ca2Cl2(OH)2:H2O | 0.0000E+00 | 2.3102E-11 |
| Saponite-H_H.33Mg3Al.33S | 0.0000E+00 | 1.2387E-11 |
| 2MgCl2:MnCl2:12H2O | 0.0000E+00 | 7.5150E-12 |
| MgCl2:MnCl2:8H2O | 0.0000E+00 | 7.2905E-12 |
| MnCl2:4H2O | 0.0000E+00 | 5.7093E-12 |
| NiCl2:4H2O(cr) | 0.0000E+00 | 5.0310E-12 |
| K(HCO3)_Kalicinite | 0.0000E+00 | 4.7405E-12 |
| CSH(0.8)_Ca0.8SiO2.8_H2O | 0.0000E+00 | 4.6294E-12 |
| MnCl2:KCl:2H2O | 0.0000E+00 | 4.2975E-12 |
| NiCl2:6H2O(cr) | 0.0000E+00 | 4.0236E-12 |
| NiCl2:2H2O(cr) | 0.0000E+00 | 3.9854E-12 |
| Talc_Mg3Si4O10(OH)2 | 0.0000E+00 | 8.6072E-13 |
| Rhodochrosite_MnCO3 | 0.0000E+00 | 5.9861E-13 |
| NiS2(cr) | 0.0000E+00 | 9.6792E-14 |
| Ni9S8(cr) | 0.0000E+00 | 1.9141E-14 |
| Na2(CO3):H2O_Thermonatri | 0.0000E+00 | 1.5687E-14 |
| CSH(1.1)_Ca(1.1)SiO(3.1) | 0.0000E+00 | 1.1830E-14 |
| NiCl2(cr) | 0.0000E+00 | 1.0580E-14 |
| Ni2SiO4 | 0.0000E+00 | 6.3643E-15 |
| CaCl2:MnCl2:8H2O | 0.0000E+00 | 4.5321E-15 |
| NiCO3 | 0.0000E+00 | 3.5844E-15 |
| U +VI O3:2H2O(c)_Schoepi | 0.0000E+00 | 3.4383E-15 |
| NaOH | 0.0000E+00 | 2.9918E-15 |
| Delafossite_CuFeO2 | 0.0000E+00 | 1.2002E-15 |
| Daphnite-7A_Fe5AlAlSi3O1 | 0.0000E+00 | 3.1375E-16 |
| Na2(CO3):7H2O | 0.0000E+00 | 1.1812E-16 |
| Ni(CO3):5.5H2O(cr) | 0.0000E+00 | 3.2639E-17 |
| Na.96Al.96Si2.04O6:H2O_A | 0.0000E+00 | 1.3025E-17 |
| Na2(CO3):10H2O_Natron | 0.0000E+00 | 8.1124E-18 |
| Na2Ca(CO3)2:2H2O_Pirsson | 0.0000E+00 | 6.7689E-18 |
| KNa(CO3):6H2O | 0.0000E+00 | 4.4492E-18 |
| Goethite_FeOOH | 0.0000E+00 | 1.9924E-18 |
| Kaolinite_Al2Si2O5(OH)4 | 0.0000E+00 | 6.6250E-19 |
| CaNa2(CO3)2:5H2O_Gayluss | 0.0000E+00 | 3.0069E-19 |
| alpha-NiS | 0.0000E+00 | 3.4995E-20 |
| Minnesotaite_Fe3Si4O10(O | 0.0000E+00 | 1.7059E-20 |

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| (U +VI O2)(CO3)(c)_Ruthe | 0.0000E+00 | 1.1468E-20 |
| K2(CO3):1.5H2O | 0.0000E+00 | 1.0189E-20 |
| beta-NiS | 0.0000E+00 | 9.4189E-21 |
| Ni2(SiO4)(oliv) | 0.0000E+00 | 2.1616E-21 |
| CSH(1.8)_Ca(1.8)SiO(3.8) | 0.0000E+00 | 4.3254E-22 |
| Maximum-Microcline_KAlSi | 0.0000E+00 | 2.4624E-22 |
| Na3(CO3)(HCO3):2H2O_Tron | 0.0000E+00 | 7.2520E-23 |
| MgCl2:2MnCl2:12H2O | 0.0000E+00 | 5.5251E-23 |
| Na.96Al.96Si2.04O6_Analc | 0.0000E+00 | 1.5514E-23 |
| Mesolite_Na.676Ca.657Al1 | 0.0000E+00 | 1.0399E-23 |
| KMgAlSi4O10(OH)2_Celadon | 0.0000E+00 | 8.1396E-24 |
| Ca4Al2Cl2O6:10H2O_Friede | 0.0000E+00 | 5.8935E-24 |
| Smectite-high-Fe-Mg_Ca.0 | 0.0000E+00 | 2.8519E-24 |
| Ferroaluminoceladonite_K | 0.0000E+00 | 1.6692E-25 |
| Scolecite_CaAl2Si3O10:3H | 0.0000E+00 | 1.0214E-25 |
| Smectite-low-Fe-Mg_Ca.02 | 0.0000E+00 | 9.9348E-27 |
| C2AH8_Ca2Al2O5:8H2O_Dica | 0.0000E+00 | 2.5434E-27 |
| Na2(U +VI O2)(c) | 0.0000E+00 | 2.2841E-27 |
| Sepiolite_Mg4Si6O15(OH)2 | 0.0000E+00 | 1.2278E-27 |
| Illite_K0.6Mg0.25Al1.8Al | 0.0000E+00 | 8.3585E-28 |
| C2ASH8_Ca2Al2SiO7:8H2O_G | 0.0000E+00 | 5.0047E-28 |
| K2Na(HCO3)(CO3):2H2O_Tro | 0.0000E+00 | 1.9192E-28 |
| Montmorillonite-Mg_Mg0.4 | 0.0000E+00 | 7.6125E-29 |
| Montmorillonite-Ca_Ca0.1 | 0.0000E+00 | 1.4461E-29 |
| Montmorillonite-Na_Na0.3 | 0.0000E+00 | 1.1644E-29 |
| C3AS(0.5)_Ca3Al2Si(0.5)O | 0.0000E+00 | 1.0195E-29 |
| Montmorillonite-K_K0.33M | 0.0000E+00 | 7.0274E-30 |
| Beidellite-Mg_Mg.165Al2. | 0.0000E+00 | 1.9640E-30 |
| Montmorillonite-H_H0.33M | 0.0000E+00 | 6.4649E-31 |
| Beidellite-Ca_Ca.165Al2. | 0.0000E+00 | 3.7309E-31 |
| Beidellite-Na_Na.33Al2.3 | 0.0000E+00 | 3.0041E-31 |
| Beidellite-K_K.33Al2.33S | 0.0000E+00 | 1.8130E-31 |
| C3AH6_Ca3Al2O6:6H2O_Hydr | 0.0000E+00 | 5.5419E-32 |
| C3ASH4_Ca3Al2SiO8:4H2O_S | 0.0000E+00 | 3.0680E-32 |
| Beidellite-H_H.33Al2.33S | 0.0000E+00 | 1.6679E-32 |
| Pyrophyllite_Al2Si4O10(O | 0.0000E+00 | 1.1679E-33 |
| Na4(U +VI O2)(CO3)3(c) | 0.0000E+00 | 8.0997E-34 |
| Ca4Cl2(OH)6:13H2O | 0.0000E+00 | 7.9725E-34 |
| Hematite_Fe2O3 | 0.0000E+00 | 1.1449E-34 |
| (U +VI O2)2(SiO4):2H2O(c | 0.0000E+00 | 8.8665E-36 |
| Ca(U +VI O2)2(SiO3OH)2:5 | 0.0000E+00 | 4.7012E-38 |
| Cronstedtite-7A_Fe2Fe2Si | 0.0000E+00 | 3.2762E-38 |
| Trevorite_NiFe2O4 | 0.0000E+00 | 1.4247E-39 |
| Ferroccladonite_KFeFeSi4 | 0.0000E+00 | 9.0744E-40 |
| Mordenite_Ca.2895Na.361A | 0.0000E+00 | 2.5975E-40 |
| Mg2Cl(OH)3:4H2O_Oxychlor | 0.0000E+00 | 2.4116E-41 |
| Ca(SO4)_Anhydrite | 0.0000E+00 | 7.8715E-42 |
| Mg(SO4):H2O_Kieserite | 0.0000E+00 | 1.9329E-42 |
| Ca(SO4):2H2O_Gypsum | 0.0000E+00 | 1.2462E-42 |

| | | |
|--------------------------|------------|-------------|
| Bassanite_CaSO4:0.5H2O | 0.0000E+00 | 9.4850E-43 |
| K3AlCl6 | 0.0000E+00 | 8.9019E-43 |
| CaSO4:0.5H2O(beta) | 0.0000E+00 | 6.4408E-43 |
| Mg(SO4):6H2O_Hexahydrite | 0.0000E+00 | 1.7725E-43 |
| KMgCl(SO4):3H2O_Kainite | 0.0000E+00 | 1.6627E-43 |
| Mg(SO4):7H2O_Epsomite | 0.0000E+00 | 9.6661E-44 |
| KAlCl4 | 0.0000E+00 | 1.8307E-45 |
| Na2(SO4)_Thenardite | 0.0000E+00 | 3.8808E-46 |
| K2Si4O9 | 0.0000E+00 | 1.5652E-47 |
| K2(SO4)_Arcanite | 0.0000E+00 | 5.0173E-48 |
| Fe(SO4):7H2O_Melanterite | 0.0000E+00 | 3.1428E-48 |
| (U +VI O2)3(PO4)2:4H2O(c | 0.0000E+00 | 5.9725E-49 |
| Na2(SO4):10H2O_Mirabilit | 0.0000E+00 | 2.6258E-50 |
| Stilbite_Ca1.019Na.136K. | 0.0000E+00 | 4.0477E-52 |
| K(HSO4)_Mercurite | 0.0000E+00 | 1.3624E-53 |
| K2CrO4:MgCrO4 | 0.0000E+00 | 4.0939E-54 |
| U +IV (OH)2(SO4)(c) | 0.0000E+00 | 1.7353E-54 |
| MgCrO4 | 0.0000E+00 | 5.9388E-55 |
| MnSO4:H2O | 0.0000E+00 | 5.2311E-55 |
| alpha-NiSO4:6H2O | 0.0000E+00 | 1.4947E-55 |
| beta-NiSO4:6H2O | 0.0000E+00 | 1.0164E-55 |
| NiSO4:6H2O(alpha) | 0.0000E+00 | 7.2403E-56 |
| NiSO4:7H2O(cr) | 0.0000E+00 | 4.7308E-56 |
| MnSO4:4H2O | 0.0000E+00 | 2.6591E-56 |
| Morenosite_NiSO4:7H2O | 0.0000E+00 | 2.3277E-56 |
| MnSO4:5H2O | 0.0000E+00 | 9.8652E-57 |
| MgCrO4:5H2O | 0.0000E+00 | 5.0550E-57 |
| Laumontite_K0.2Na0.2Ca1. | 0.0000E+00 | 3.5953E-57 |
| Na(U +VI O2)(OH)3(c)_Cla | 0.0000E+00 | 3.1106E-57 |
| UO2(OH)2:2H2O | 0.0000E+00 | 1.5950E-57 |
| U +VI O3:2H2O(c)_Metasch | 0.0000E+00 | 9.1067E-58 |
| Nontronite-Mg_Mg.165Fe2A | 0.0000E+00 | 3.6078E-58 |
| Nontronite-Ca_Ca.165Fe2A | 0.0000E+00 | 6.8536E-59 |
| Nontronite-Na_Na.33Fe2Al | 0.0000E+00 | 5.5186E-59 |
| Nontronite-K_K.33Fe2Al.3 | 0.0000E+00 | 3.3305E-59 |
| Nontronite-H_H.33Fe2Al.3 | 0.0000E+00 | 1.9343E-59 |
| Chabazite_K0.6Na0.2Ca1.5 | 0.0000E+00 | 1.6459E-59 |
| NiSO4(cr) | 0.0000E+00 | 1.2721E-59 |
| Na2CrO4:4H2O | 0.0000E+00 | 5.1092E-62 |
| K2CrO4 | 0.0000E+00 | 1.8194E-62 |
| Na2CrO4:6H2O | 0.0000E+00 | 1.1531E-62 |
| Na2CrO4:MgCrO4:2H2O | 0.0000E+00 | 1.1073E-68 |
| Na2(U +VI O2)2(Si2O5)3:4 | 0.0000E+00 | 1.5128E-71 |
| K2Ca(SO4)2:H2O_Syngenite | 0.0000E+00 | <1.0000E-75 |
| K2Mg(SO4)2:4H2O_Leonite | 0.0000E+00 | <1.0000E-75 |
| K2Mg(SO4)2:6H2O_Picromer | 0.0000E+00 | <1.0000E-75 |
| K2MgCa2(SO4)4:2H2O_Polyh | 0.0000E+00 | <1.0000E-75 |
| K3(HSO4)(SO4) | 0.0000E+00 | <1.0000E-75 |
| K8(HCO3)4(CO3)2:3H2O | 0.0000E+00 | <1.0000E-75 |

| | | |
|---------------------------|------------|-------------|
| K8(HSO4)6(SO4)_Misenite | 0.0000E+00 | <1.0000E-75 |
| Na2Ca(SO4)2_Glauberite | 0.0000E+00 | <1.0000E-75 |
| Na2Mg(SO4)2:4H2O_Bloedite | 0.0000E+00 | <1.0000E-75 |
| Na3(HSO4)(SO4) | 0.0000E+00 | <1.0000E-75 |
| Na4Ca(SO4)3:2H2O | 0.0000E+00 | <1.0000E-75 |
| Na6(CO3)(SO4)2_Burkeite | 0.0000E+00 | <1.0000E-75 |
| NaK3(SO4)2_Glaserite | 0.0000E+00 | <1.0000E-75 |
| FeK2(SO4)2:6H2O | 0.0000E+00 | <1.0000E-75 |
| FeNa2(SO4)2:4H2O | 0.0000E+00 | <1.0000E-75 |
| Fe2(SO4)3 | 0.0000E+00 | <1.0000E-75 |
| KFe3(SO4)2(OH)6_Jarosite | 0.0000E+00 | <1.0000E-75 |
| NaFe3(SO4)2(OH)6_Jarosit | 0.0000E+00 | <1.0000E-75 |
| Cu2CO3(OH)2(s) | 0.0000E+00 | <1.0000E-75 |
| Cu3(CO3)2(OH)2(s) | 0.0000E+00 | <1.0000E-75 |
| MnSO4:K2SO4:4H2O | 0.0000E+00 | <1.0000E-75 |
| MnSO4:Na2SO4:2H2O | 0.0000E+00 | <1.0000E-75 |
| beta-Ni(OH)2 | 0.0000E+00 | <1.0000E-75 |
| K2Ni(SO4)2:6H2O(cr) | 0.0000E+00 | <1.0000E-75 |
| Na2Ni(SO4)2:4H2O(cr) | 0.0000E+00 | <1.0000E-75 |
| Ca(U +VI 6O19):11H2O(c)_ | 0.0000E+00 | <1.0000E-75 |
| K2(U +VI 6O19):11H2O(c)_ | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-Ca_Ca1.73 | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-K_K3.467A | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-Na_Na3.46 | 0.0000E+00 | <1.0000E-75 |
| Clinoptilolite-NH4_(NH4) | 0.0000E+00 | <1.0000E-75 |
| Erionite_K1.5Na0.9Ca0.9A | 0.0000E+00 | <1.0000E-75 |
| KAl(SO4)2 | 0.0000E+00 | <1.0000E-75 |
| KAl(SO4)2:12H2O | 0.0000E+00 | <1.0000E-75 |
| KAl(SO4)2:3H2O | 0.0000E+00 | <1.0000E-75 |
| KAl3(OH)6(SO4)2_Alunite | 0.0000E+00 | <1.0000E-75 |
| Phillipsite_K0.7Na0.7Ca1 | 0.0000E+00 | <1.0000E-75 |
| Stellerite_Ca2.0Al4.0Si1 | 0.0000E+00 | <1.0000E-75 |
| C9S6H16_Ca9Si6O21:16H2O_ | 0.0000E+00 | <1.0000E-75 |
| Ca6Al2(SO4)3(OH)12:26H2O | 0.0000E+00 | <1.0000E-75 |
| BSK3+3SF | 0.0000E+00 | <1.0000E-75 |
| HAW-Kokille | 0.0000E+00 | <1.0000E-75 |

| Cp_EQUIL | H_EQUIL | S_EQUIL | G_EQUIL | V_EQUIL |
|----------|---------|---------|---------|---------|
| J.K-1 | J | J.K-1 | J | m3 |

| | | | | |
|-------------|--------------|--------------|--------------|-------------|
| 2.24676E+06 | -1.87874E+11 | -6.86749E+06 | -1.85827E+11 | 1.49299E+02 |
|-------------|--------------|--------------|--------------|-------------|

Properties for AQUEOUS:

pH = 7.7711

Eh/V = -0.5071

Total solute molality = 20.0088

Ionic strength = 22.8350

Osmotic coefficient = 3.2704

Debye-Hueckel slope = 0.3905

Mole fraction of system components:

| | GAS | AQUEOUS |
|----|------------|------------|
| H | 9.9987E-01 | 5.8113E-01 |
| O | 1.2530E-04 | 3.1217E-01 |
| NA | 0.0000E+00 | 2.0179E-04 |
| K | 0.0000E+00 | 6.7793E-05 |
| MG | 0.0000E+00 | 3.3721E-02 |
| CA | 0.0000E+00 | 2.6579E-05 |
| CL | 0.0000E+00 | 6.1681E-02 |
| S | 5.9603E-12 | 1.4025E-11 |
| C | 1.6390E-10 | 4.6575E-03 |
| FE | 0.0000E+00 | 1.5759E-05 |
| SI | 0.0000E+00 | 1.4491E-13 |
| AL | 0.0000E+00 | 1.4418E-15 |
| U | 0.0000E+00 | 9.3149E-04 |
| P | 0.0000E+00 | 1.9204E-04 |
| AS | 0.0000E+00 | 7.3799E-05 |
| N | 0.0000E+00 | 3.9475E-04 |
| NI | 0.0000E+00 | 7.6212E-15 |
| B | 0.0000E+00 | 4.5534E-03 |
| ZR | 0.0000E+00 | 3.2953E-13 |
| MN | 0.0000E+00 | 7.6153E-15 |
| MO | 0.0000E+00 | 1.8010E-04 |
| CU | 0.0000E+00 | 3.5600E-14 |
| CR | 0.0000E+00 | 2.7738E-11 |
| SN | 0.0000E+00 | 1.2487E-15 |
| Ti | 0.0000E+00 | 1.1553E-13 |
| V | 0.0000E+00 | 1.8347E-06 |

System density/g.cm-3 including gaseous phase = 0.195482

System density/g.cm-3 excluding gaseous phase = 3.69814

| | Cp/J.K-1 | H/J | S/J.K-1 | G/J | V/m3 |
|---------------|-------------|--------------|--------------|--------------|-------------|
| AQUEOUS | 2.23393E+06 | -2.15185E+10 | 1.32399E+05 | -2.15580E+10 | 1.49629E+00 |
| KMgCl3:6H2O_C | 0.00000E+00 | -1.31636E+10 | 1.98773E-06 | -1.31636E+10 | 8.98297E-01 |
| Mg(OH)2_Brucl | 0.00000E+00 | -1.41840E+09 | -1.61167E-09 | -1.41840E+09 | 4.19946E-02 |
| MgCl2:6H2O_Bi | 0.00000E+00 | -5.03941E+10 | 6.55493E-06 | -5.03941E+10 | 3.08768E+00 |
| NaCl_Halite | 0.00000E+00 | -1.68313E+09 | -1.21570E-06 | -1.68313E+09 | 1.18380E-01 |
| FeS | 0.00000E+00 | -1.54337E+08 | -3.63949E-08 | -1.54337E+08 | 1.51363E-06 |
| Fe3O4_Magneti | 0.00000E+00 | -4.84327E+10 | 5.54982E-05 | -4.84327E+10 | 2.12444E+00 |
| Cu(OH)2(s) | 0.00000E+00 | -3.82380E+07 | -5.95742E-09 | -3.82380E+07 | 8.34765E-08 |
| Cassiterite_S | 0.00000E+00 | -9.75238E+07 | -8.78174E-09 | -9.75238E+07 | 1.87685E-07 |
| Eskolaite_Cr2 | 0.00000E+00 | -2.20070E+09 | -1.88218E-06 | -2.20070E+09 | 2.09379E-06 |
| Rutile-TiO2 | 0.00000E+00 | 7.94539E+07 | 3.74870E-08 | 7.94539E+07 | 1.89526E-07 |
| V2O4 | 0.00000E+00 | -1.93435E+07 | -1.60075E-08 | -1.93435E+07 | 1.46747E-08 |
| Mn(OH)2 | 0.00000E+00 | -1.62285E+09 | -4.96806E-07 | -1.62285E+09 | 2.37283E-06 |
| Ni3S2(cr) | 0.00000E+00 | -1.46606E+08 | -1.45111E-07 | -1.46606E+08 | 6.94250E-07 |
| U +IV (OH)4(a | 0.00000E+00 | -2.82943E+10 | -1.26090E-05 | -2.82943E+10 | 1.92487E-05 |

```
U|+IV|(SiO4)( 0.00000E+00 -2.01500E+09 7.44687E-07 -2.01500E+09 1.06896E-06
Mg4Al2O7:10H2 0.00000E+00 -8.13865E+08 -5.47679E-07 -8.13865E+08 1.27062E-07
ZrO2(monoclin 0.00000E+00 -1.59274E+10 -6.87257E-06 -1.59274E+10 1.52744E-05
```

| | Dens/g.cm-3 | Thermal exp | Bulk mod/bar | Cv/J.K-1 | Grueneisen |
|----------------|-------------|-------------|--------------|-------------|-------------|
| AQUEOUS | 1.31271E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| KMgCl3:6H2O_C | 1.61000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Mg(OH)2_Brucei | 2.36783E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| MgCl2:6H2O_Bi | 1.56905E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| NaCl_Halite | 2.16333E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| FeS | 8.79110E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Fe3O4_Magnetit | 5.20018E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Cu(OH)2(s) | 9.75607E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Cassiterite_S | 1.50709E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Eskolaite_Cr2 | 1.51990E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Rutile-TiO2 | 7.98788E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| V2O4 | 1.65881E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| Mn(OH)2 | 8.89527E+04 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| Ni3S2(cr) | 2.40212E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| U +IV (OH)4(a | 3.06058E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| U +IV (SiO4)(| 3.30112E+05 | 0.00000E+00 | 1.41931E+08 | 0.00000E+00 | 0.00000E+00 |
| Mg4Al2O7:10H2 | 4.43332E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |
| ZrO2(monoclin | 1.23223E+05 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 | 0.00000E+00 |

System Volume : 1.49298686869624248D+02 m3

System composition in [mol]:

```
500 | 1|IA | 1.94165253718742746E-09<< EA
500 | 2|IA | 1.02045272143285570E+06<< H
500 | 3|IA | 5.62979546390846954E+05<< O
500 | 4|IA | 4.42624462838427826E+03<< NA
500 | 5|IA | 5.21997303213619853E+03<< K
500 | 6|IA | 3.86432443059244906E+04<< MG
500 | 7|IA | 5.82853342308118982E+00<< CA
500 | 8|IA | 8.11836706250888528E+04<< CL
500 | 9|IA | 2.90212809728055436E+03<< S
500 | 10|IA | 1.02132931469436824E+03<< C
500 | 11|IA | 1.44660489927478979E+05<< FE
500 | 12|IA | 1.06895515479518963E+03<< SI
500 | 13|IA | 2.54123094442875271E+02<< AL
500 | 16|IA | 2.05218938456352407E+04<< U
500 | 23|IA | 4.21126758835316437E+01<< P
500 | 27|IA | 1.61833169606622072E+01<< AS
500 | 29|IA | 8.65642870911768370E+01<< N
500 | 31|IA | 2.08275138260861968E+03<< NI
500 | 32|IA | 9.98510776061514093E+02<< B
500 | 33|IA | 1.52744485003946138E+04<< ZR
500 | 37|IA | 2.37282560746382160E+03<< MN
500 | 39|IA | 3.94934333958794284E+01<< MO
500 | 41|IA | 8.34765366820889909E+01<< CU
```


500 | 44 | IA | 4.18758829989179867E+03 << CR
 500 | 45 | IA | 1.87684609552691967E+02 << SN
 500 | 46 | IA | 1.89525769319154023E+02 << Ti
 500 | 47 | IA | 2.97517740938139355E+01 << V

Aq. phase composition in [mol]:

500 | 1 | IA | 1.94165253718742746E-09 << EA
 500 | 2 | IA | 1.27434903943133788E+05 << H
 500 | 3 | IA | 6.84557245287656406E+04 << O
 500 | 4 | IA | 4.42494242725455891E+01 << NA
 500 | 5 | IA | 1.48663170557705033E+01 << K
 500 | 6 | IA | 7.39462934255757500E+03 << MG
 500 | 7 | IA | 5.82853342308118982E+00 << CA
 500 | 8 | IA | 1.35258647871565263E+04 << CL
 500 | 9 | IA | 3.07555423013874318E-06 << S
 500 | 10 | IA | 1.02132923982514910E+03 << C
 500 | 11 | IA | 3.45565861587647838E+00 << FE
 500 | 12 | IA | 3.17776108520413768E-08 << SI
 500 | 13 | IA | 3.16161025989136205E-10 << AL
 500 | 16 | IA | 2.04265039607543997E+02 << U
 500 | 23 | IA | 4.21126758835316437E+01 << P
 500 | 27 | IA | 1.61833169606622072E+01 << AS
 500 | 29 | IA | 8.65642870911768370E+01 << N
 500 | 31 | IA | 1.67124217305270618E-09 << NI
 500 | 32 | IA | 9.98510776061514093E+02 << B
 500 | 33 | IA | 7.22631447005471318E-08 << ZR
 500 | 37 | IA | 1.66995770552764042E-09 << MN
 500 | 39 | IA | 3.94934333958794284E+01 << MO
 500 | 41 | IA | 7.80662549768159283E-09 << CU
 500 | 44 | IA | 6.08256423901209180E-06 << CR
 500 | 45 | IA | 2.73826872733568198E-10 << SN
 500 | 46 | IA | 2.53347302121996252E-08 << Ti
 500 | 47 | IA | 4.02325709988830604E-01 << V

Gas Phase : *****

Total Volume : 1.41531560489057028D+02 m3

Total Mass : 4.61277240217084240D+05 g

Pressure : 3.99999999999999953D+06 Pa

Solution: *****

Total Volume : 1.49628813455203247D+00 m3

Total Mass : 1.96419917478533578D+06 g

Density : 1.31271452966068545D+06 g / m3

System component molalities:

| | | |
|------|-------------------------|------------|
| 1 EA | 1.73610344726753064E-12 | mol/kg H2O |
| 2 H | 1.14058132952689959E+02 | mol/kg H2O |
| 3 O | 6.12699651985366813E+01 | mol/kg H2O |
| 4 NA | 3.96045868172047594E-02 | mol/kg H2O |
| 5 K | 1.33058080227420692E-02 | mol/kg H2O |

| | | | | |
|----|----|-------------------------|--------|-----|
| 6 | MG | 6.61841921319812609E+00 | mol/kg | H2O |
| 7 | CA | 5.21671551136136408E-03 | mol/kg | H2O |
| 8 | CL | 1.21060622832346816E+01 | mol/kg | H2O |
| 9 | S | 2.75271501315612322E-09 | mol/kg | H2O |
| 10 | C | 9.14120877561372169E-01 | mol/kg | H2O |
| 11 | FE | 3.09292005292859627E-03 | mol/kg | H2O |
| 12 | SI | 2.84419327149048852E-11 | mol/kg | H2O |
| 13 | AL | 2.82973778932806456E-13 | mol/kg | H2O |
| 16 | U | 1.82823452007624315E-01 | mol/kg | H2O |
| 23 | P | 3.76921317181735521E-02 | mol/kg | H2O |
| 27 | AS | 1.44845631801034241E-02 | mol/kg | H2O |
| 29 | N | 7.74776820203523925E-02 | mol/kg | H2O |
| 31 | NI | 1.49581281165519017E-12 | mol/kg | H2O |
| 32 | B | 8.93697655247882583E-01 | mol/kg | H2O |
| 33 | ZR | 6.46777225924889030E-11 | mol/kg | H2O |
| 37 | MN | 1.49466317397183836E-12 | mol/kg | H2O |
| 39 | MO | 3.53478296576856116E-02 | mol/kg | H2O |
| 41 | CU | 6.98716836106189790E-12 | mol/kg | H2O |
| 44 | CR | 5.44408085383030444E-09 | mol/kg | H2O |
| 45 | SN | 2.45083418199158353E-13 | mol/kg | H2O |
| 46 | TI | 2.26753576724400237E-11 | mol/kg | H2O |
| 47 | V | 3.60093803975938907E-04 | mol/kg | H2O |

Solid Phases : *****

Total Volume : 6.27083824601520767D+00 m3

Total Mass : 2.67597014210345114D+04 kg

| | mol | activity |
|-----|--------------------------|-------------------------|
| 29 | KMgCl3:6H2O_Carnallite | 5.20510671508042833D+03 |
| 33 | Mg(OH)2_Brucite | 1.70501681511172137D+03 |
| 39 | MgCl2:6H2O_Bischoffite | 2.38302452442896501D+04 |
| 53 | NaCl_Halite | 4.38199520411173307D+03 |
| 83 | FeS | 1.51362716974443674D+03 |
| 85 | Fe3O4_Magnetite | 4.77144690330395533D+04 |
| 140 | Cu(OH)2(s) | 8.34765366742823716D+01 |
| 166 | Cassiterite_SnO2 | 1.87684609552418152D+02 |
| 173 | Eskolaite_Cr2O3 | 2.09379414690461726D+03 |
| 183 | Rutile-TiO2 | 1.89525769293819280D+02 |
| 187 | V2O4 | 1.46747241919125528D+01 |
| 192 | Mn(OH)2 | 2.37282560746215177D+03 |
| 217 | Ni3S2(cr) | 6.94250460868982827D+02 |
| 283 | U +IV (OH)4(am) | 1.92486736512642856D+04 |
| 284 | U +IV (SiO4)(c)_Coffinit | 1.06895515476341211D+03 |
| 373 | Mg4Al2O7:10H2O_Hydrotalc | 1.27061547221279554D+02 |
| 400 | ZrO2(monoclinic) | 1.52744485003223508D+04 |

Other information: *****

Total volume of solid phases = 6.27083824601520767D+00 m3

Charge balance after calculation = 1.94165253718742746D-09

Relative error of charge balance = 6.50838978115876322D-14
kw = 13.984 (Should be at current T: 13.9960)
Eh = -0.507
pH = 7.771
pHCl= 2.652
phi = 0.0029
T = 298.1500 K

CPU-Time: 9.510 s

End of calculation

* Nathan: *
* Kaum war der Vater tot, so kommt ein jeder mit *
* seinem Ring, und jeder will der Fuerst des Hauses *
* sein. Man untersucht, man zankt, man klagt. *
* Umsonst; der rechte Ring war nicht erweislich; - *
* * *
* Fast so unerweislich, als uns ist - *
* der rechte Glaube. *
* * *
* Aus: "Nathan, der Weise" von G. E. Lessing *
